

Supporting Information

Hollow-Shell Structured Porous CoSe₂ Microspheres Encapsulated by MXene Nanosheets for Advanced Lithium Storage

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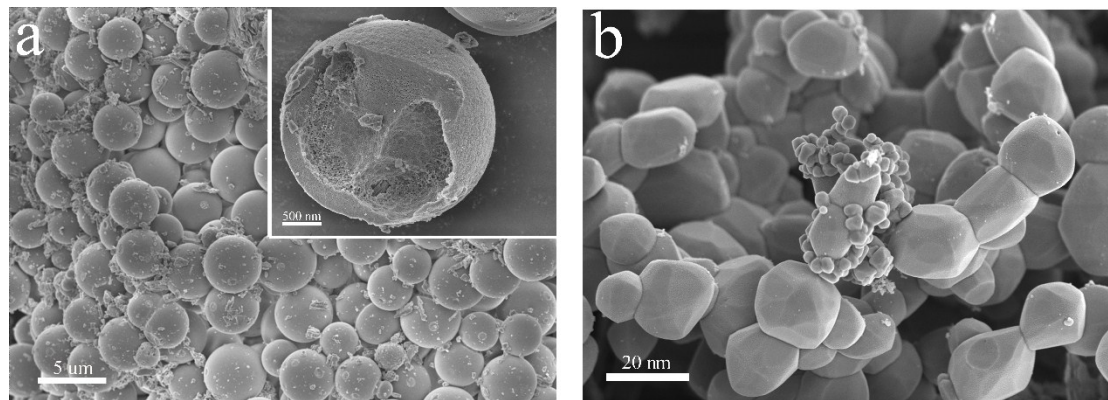


Figure S1 FESEM images of (a) Co-MOF and (b) pure CoSe₂.

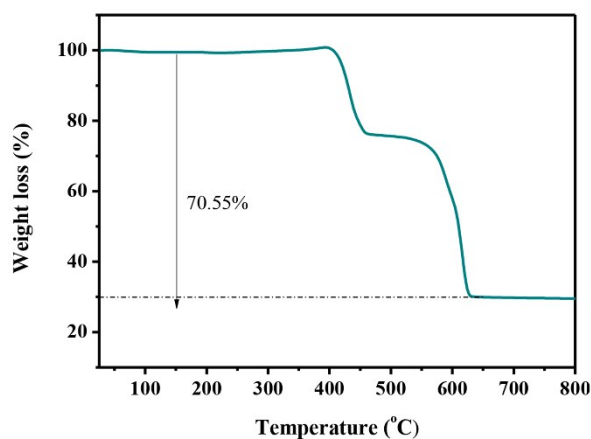


Figure S2 TGA of CoSe₂ hollow sphere.

The mass loading of CoSe₂ in MOF-derived CoSe₂ hollow sphere was determined by TGA in air atmosphere at a heating rate of 10 °C min⁻¹ from 25 to 800 °C, as displayed in [Figure S3](#). During the thermal treatment process, the significant mass loss between 400 and 630 °C could be ascribed to the combustion of carbon, and the oxidation of CoSe₂ with conversion of Co₃O₄ and Se. When the temperature was further increased, no mass loss could be observed probably owing to the complete transformation of CoSe₂ to Co₃O₄. Based on the reaction equation: $3\text{CoSe}_2 (\text{s}) + 2\text{O}_2 (\text{g}) = \text{Co}_3\text{O}_4 (\text{s}) + 6\text{Se} (\text{g})$ and 70.55 wt.% of the original weight lost, the accurate content of CoSe₂ in CoSe₂ hollow sphere was calculated to be 79.55 wt.%.

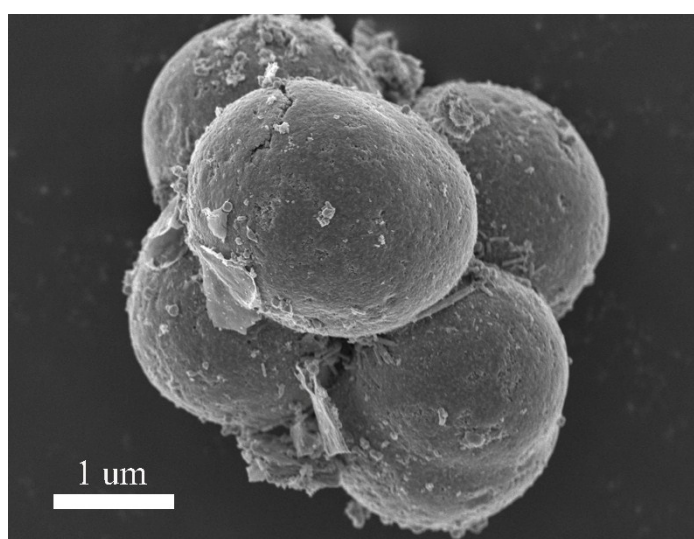


Figure S3 FESEM images of CoSe₂@MXene.

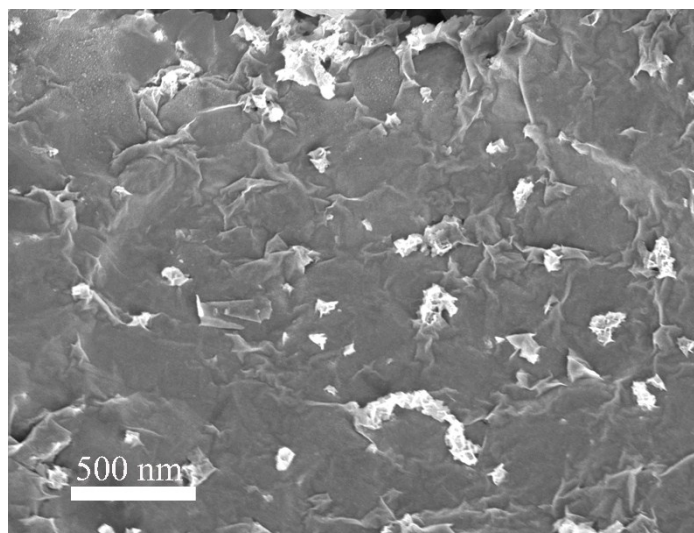


Figure S4 FESEM images of MXene flakes.

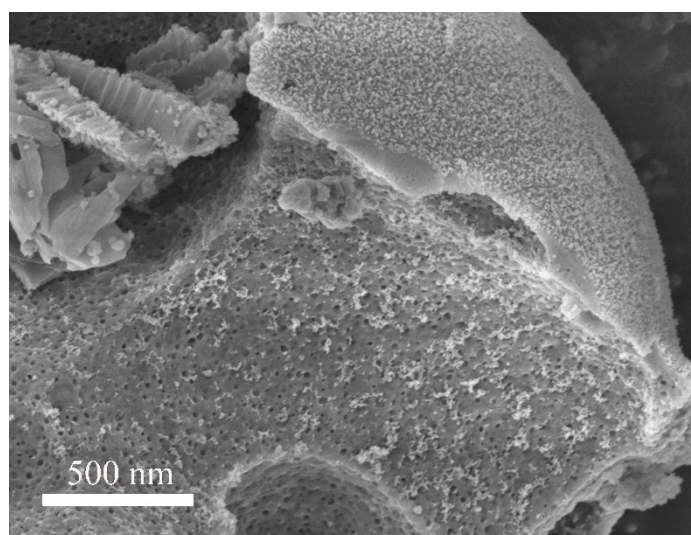


Figure S5 FESEM images of Co-MOF.

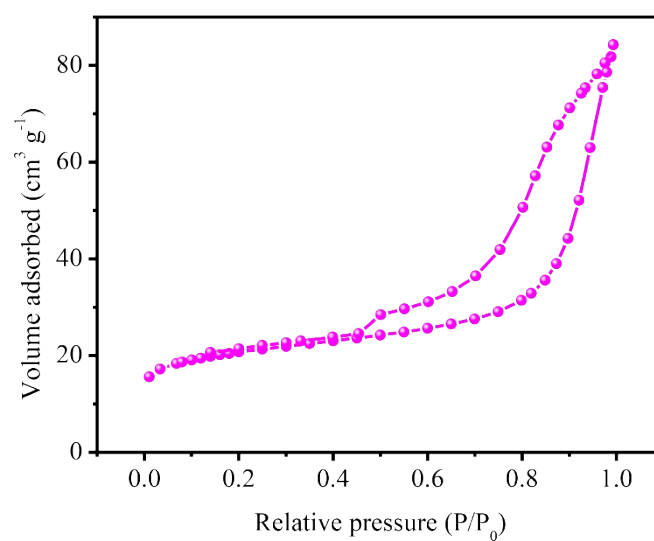


Figure S6 N₂ adsorption-desorption isotherms of CoSe₂ spheres.

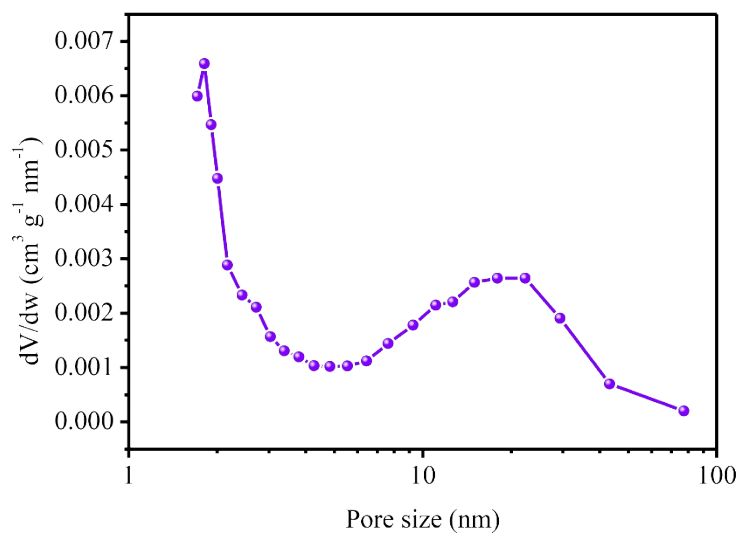


Figure S7 Pore size distribution of CoSe₂ spheres.

The CoSe₂ spheres possess a specific surface area of 69.19 m²/g, and a pore volume of 0.116 cm³/g at $P/P_0 = 0.999$.

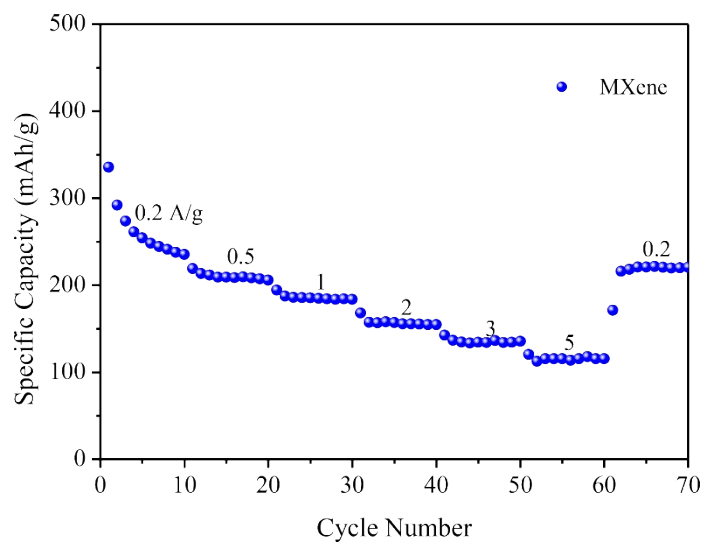


Figure S8 Rate performance of MXene.

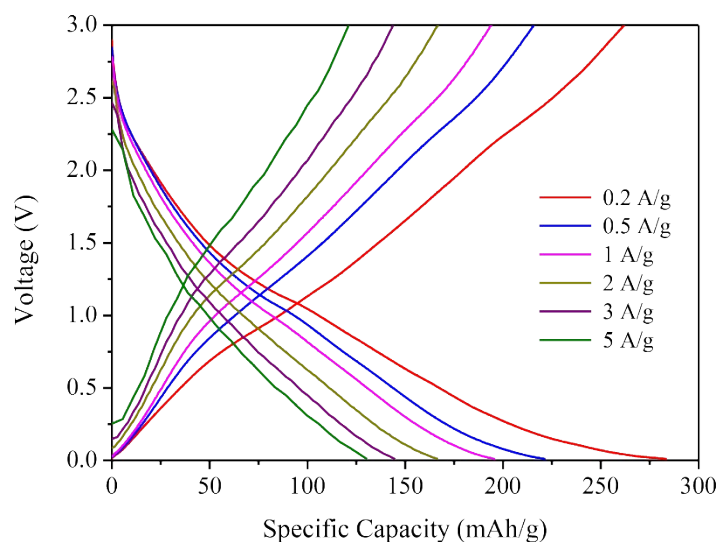


Figure S9 Galvanostatic charge/discharge curves of MXene at various rates.

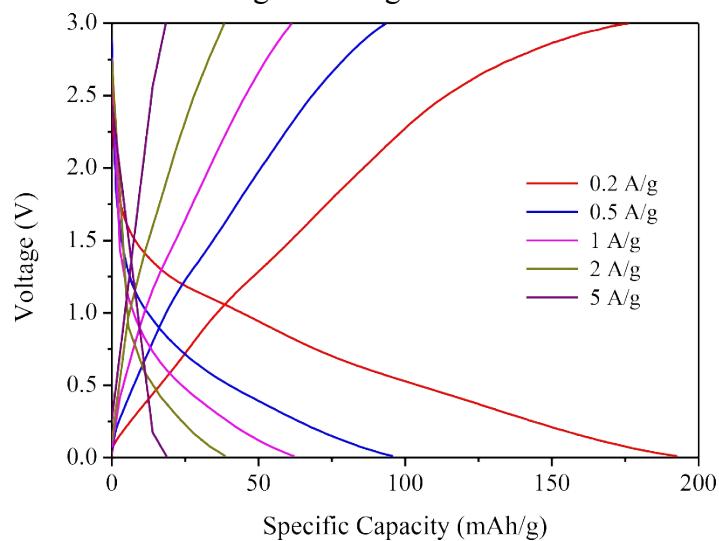


Figure S10 Galvanostatic charge/discharge curves of pure CoSe₂ at various rates.

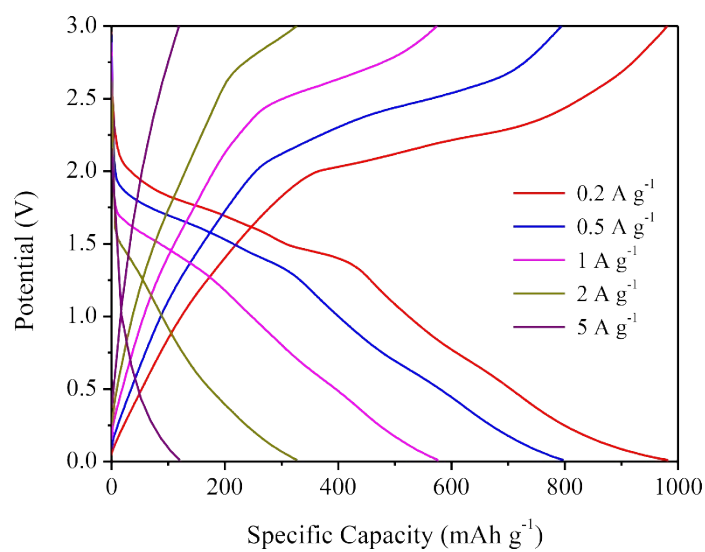


Figure S11 Galvanostatic charge/discharge curves of CoSe₂ spheres at various rates.

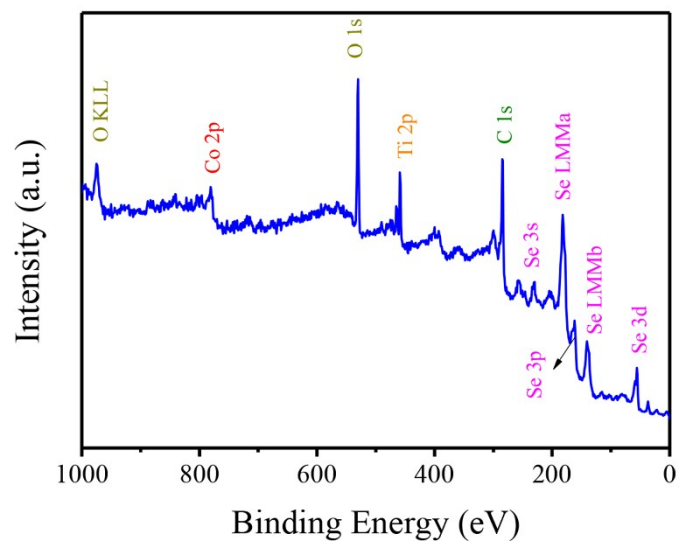


Figure S12 XPS survey spectrum of CoSe₂@MXene.

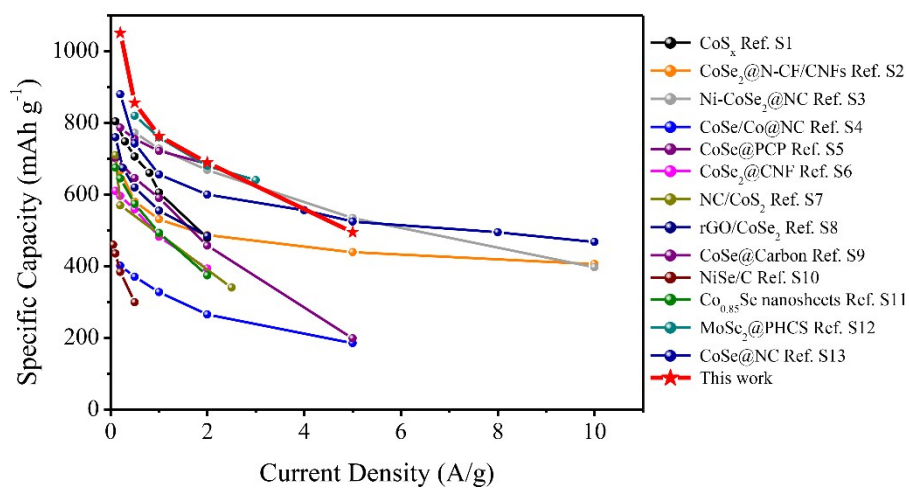


Figure S13 Comparison of rate capability of the produced CoSe₂@MXene with other typical anode materials for LIBs.

Table S1 Summary and comparison of cycling and rate performances of recently reported TMDs for LIBs.

Materials	Cycling Performance			Rate Performance		Reference
	Capacity (mAh g ⁻¹)	Cycles	Current (A g ⁻¹)	Capacity (mAh g ⁻¹)	Current (A g ⁻¹)	
CoSe₂@MXene	910	100	0.2	1051, 856, 763, 669, and 465	0.2, 0.5, 1, 2, and 5	This work
CoS _x	1012.1	100	0.5	804, 747, 705, 660, 605, and 478	0.1, 0.3, 0.5, 0.8, 1, and 2	1
CoSe ₂ @N-CF/CNTs	428	500	1	666, 580, 531, 487, 439, and 406	0.2, 0.5, 1, 2, 5, and 10	2
Ni-CoSe ₂ @NC	645	300	1.5	772, 728, 669, 534, and 397	0.5, 1, 2, 5, and 10	3
CoSe/Co@NC	630	100	0.2	401, 370, 328, 265, and 185	0.2, 0.5, 1, 2, and 5	4
CoSe@PCP	675	100	0.2	401, 370, 328, 265, and 185	0.2, 0.5, 1, 2, and 5	5
CoSe ₂ @CNF	1405	300	0.2	610, 596, 559, 482, and 393	0.1, 0.2, 0.5, 1, and 2	6
NC/CoS ₂	560	50	0.1	710, 570, 490, and 340	0.1, 0.2, 1, and 2.5	7
rGO/CoSe ₂	1577	200	0.2	760, 674, 620, 555, and 484	0.1, 0.25, 0.5, 1, and 2	8
CoSe@Carbon	860	100	0.2	787, 755, 722, and 686	0.2, 0.5, 1, and 2	9
NiSe/C	428	50	0.1	460, 435, 384, and 299	0.05, 0.1, 0.2, and 0.5	10
Co _{0.85} Se nanosheets	516	50	0.2	675, 645, 574, 493, and 374	0.1, 0.2, 0.5, 1, and 2	11
MoSe ₂ @PHCs	681	100	0.5	820, 760, 680, and 640	0.5, 1, 2, and 3	12
CoSe@NC	796	100	1	880, 742, 656, 600, 556, 525, 495, and 374	0.2, 0.5, 1, 2, 4, 5, 8, and 2	13

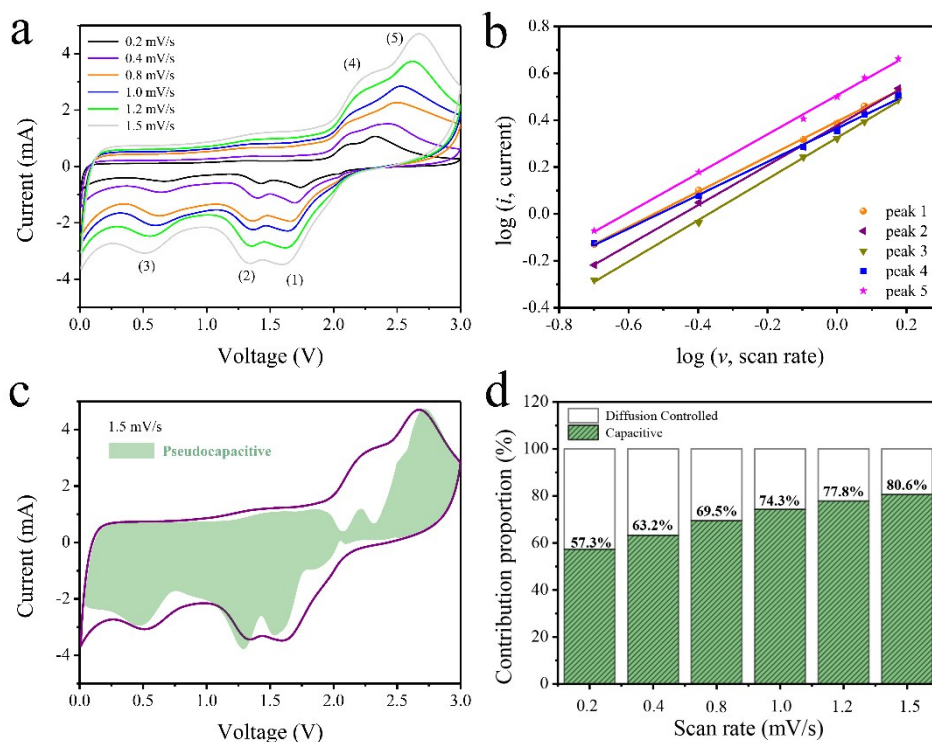


Figure S14 (a) CV profiles of CoSe₂@MXene at different scan rates. (b) The plots of $\log(i)$ vs $\log(v)$ (peak current: i , scan rate: v), calculated from CV curves. (c) The shaded region shows the CV profile with the pseudocapacitive contribution at a scan rate of 1.5 mV s⁻¹. (d) Contribution ratio of pseudocapacitive at different scan rates.

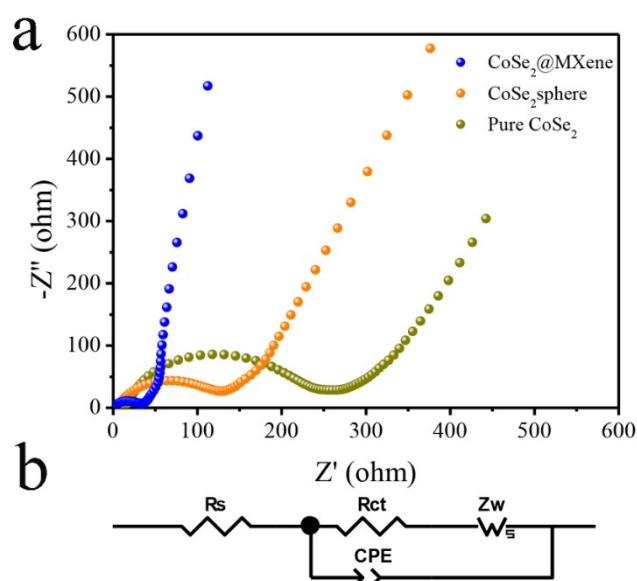


Figure S15 (a) Nyquist plots of pure CoSe₂, CoSe₂ sphere and CoSe₂@MXene

electrodes, (b) equivalent circuit model.

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