

## Supporting Information

### Hierarchical Self-assembly of NiFe-LDH Nanosheets on $\text{CoFe}_2\text{O}_4@\text{Co}_3\text{S}_4$

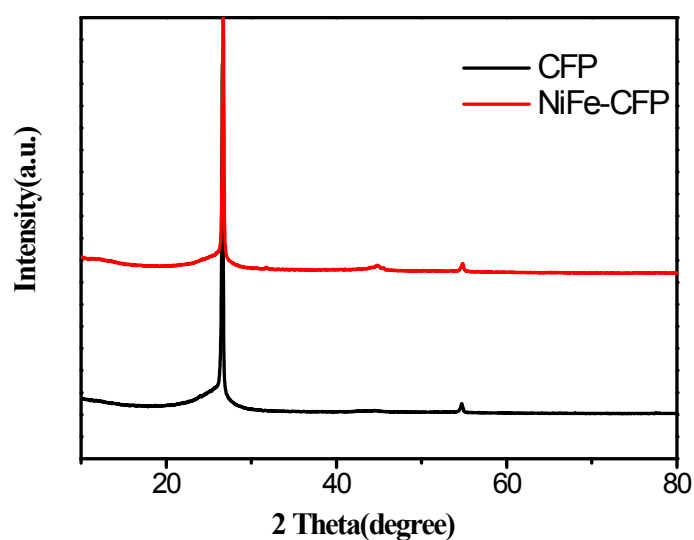
#### Nanowires for Enhanced Overall Water Splitting

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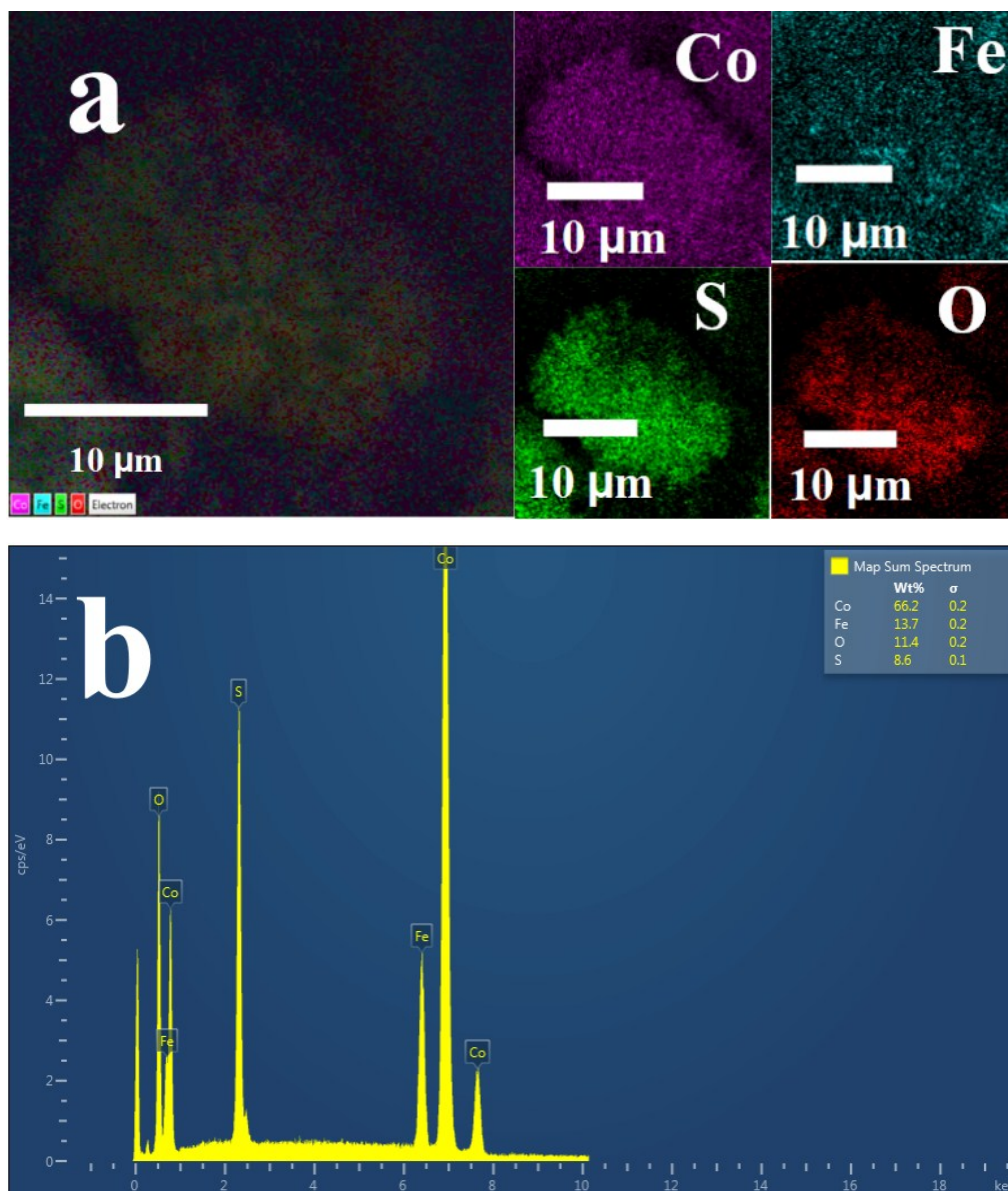
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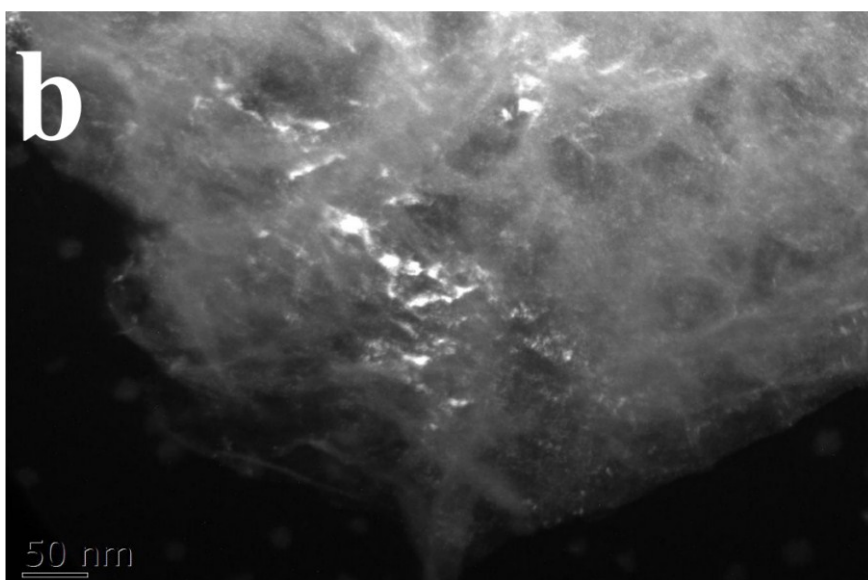
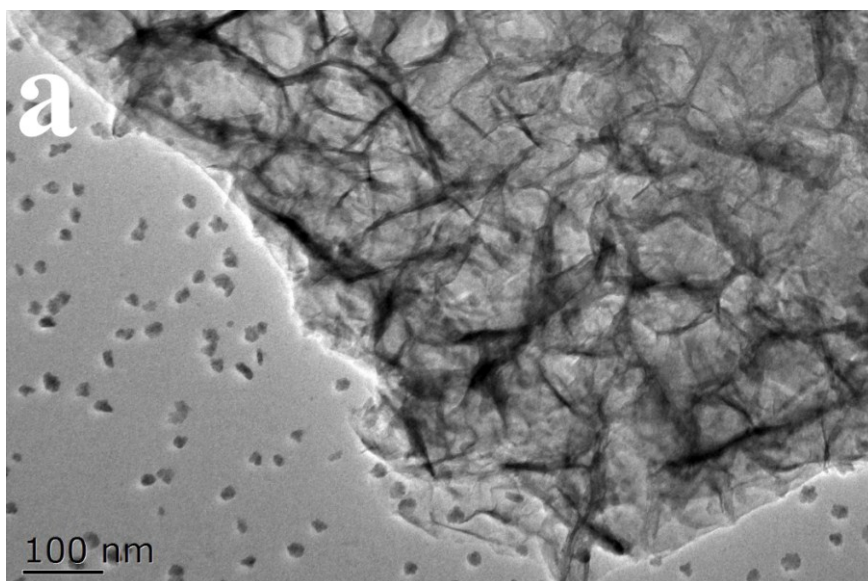
\*Corresponding author: E-mail: [gyq@nuc.edu.cn](mailto:gyq@nuc.edu.cn)



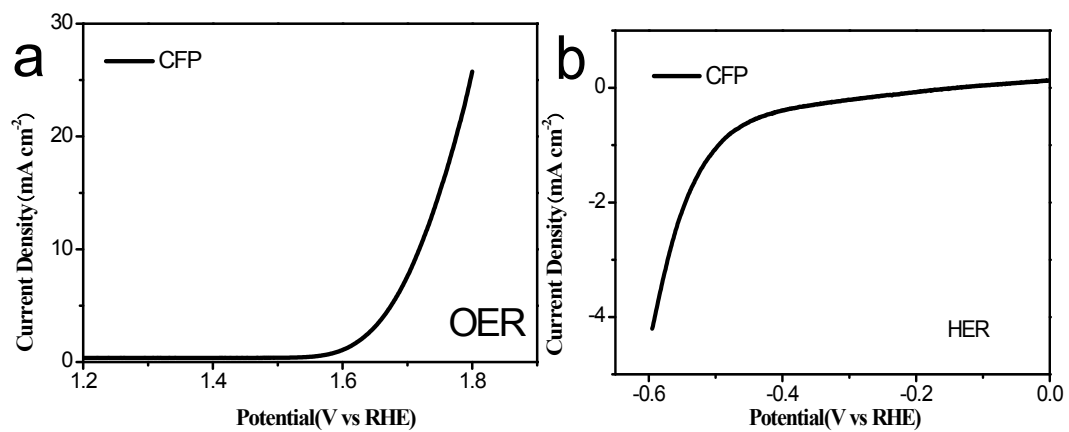
**Fig. S1** XRD patterns of the bare carbon fiber paper and NiFe-CFP.



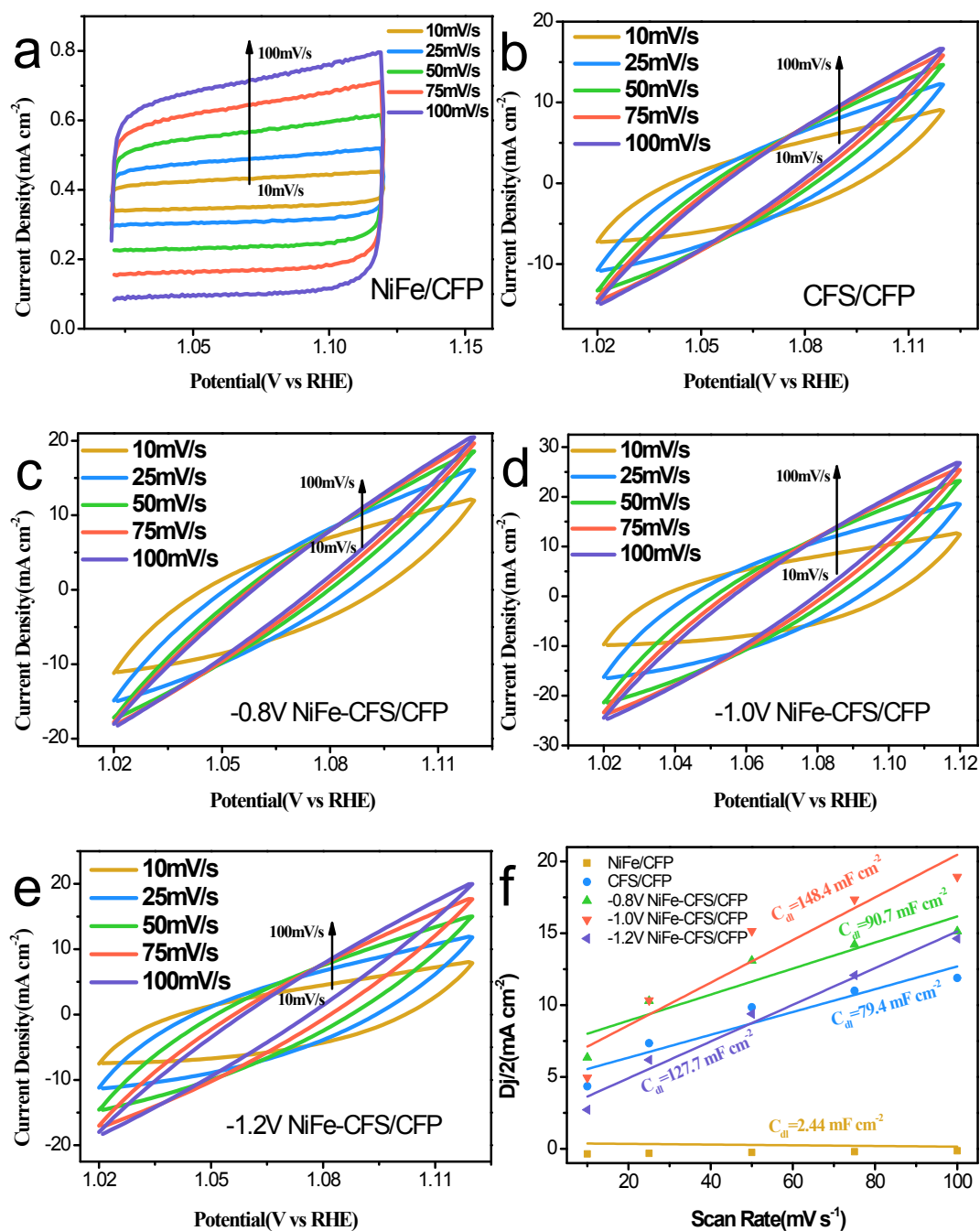
**Fig. S2** (a) SEM elemental mapping analysis of  $\text{CoFe}_2\text{O}_4@\text{Co}_3\text{S}_4/\text{CFP}$  (b) The corresponding EDX spectrum.



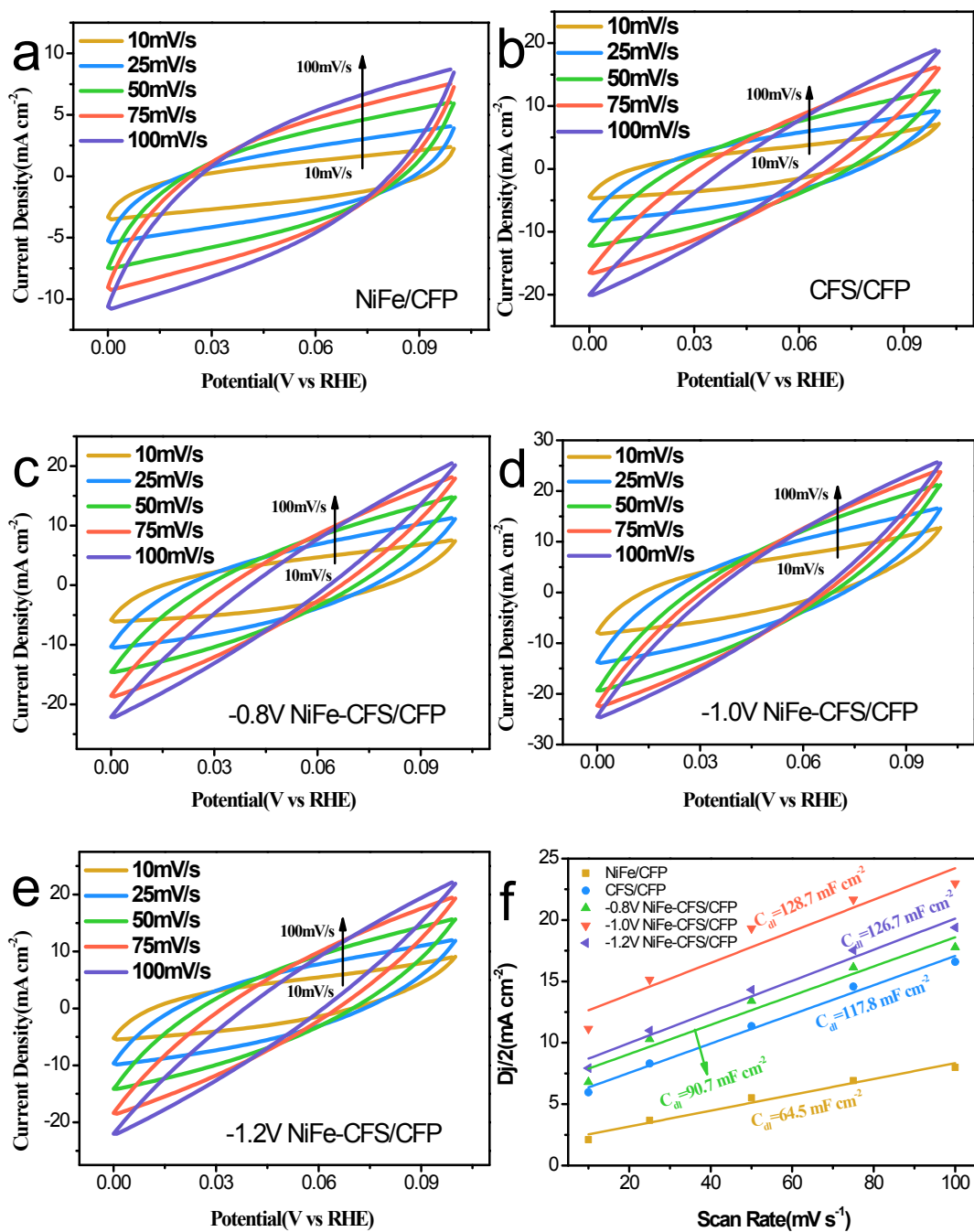
**Fig. S3** (a) TEM and (a) high-angle annular dark field scanning transmission electron microscopy (HAADF-STEM) images of NiFe-LDH.



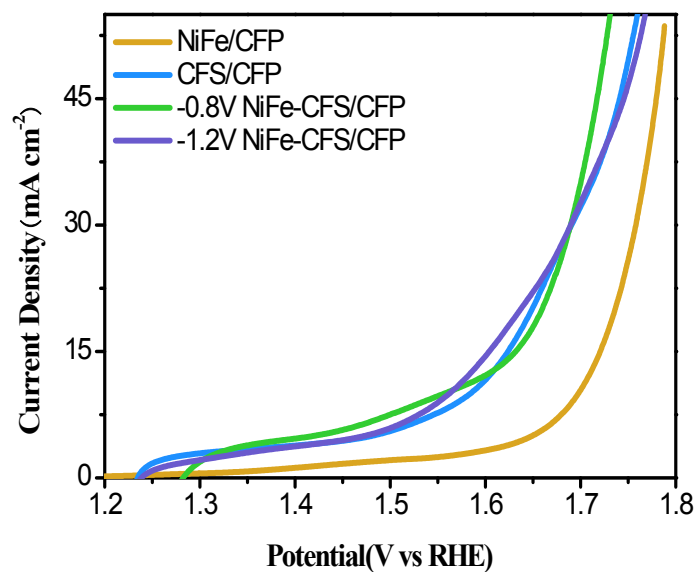
**Fig. S4** LSV curves of bare CFP in 1 M KOH to show the background activity for catalyzing OER and HER.



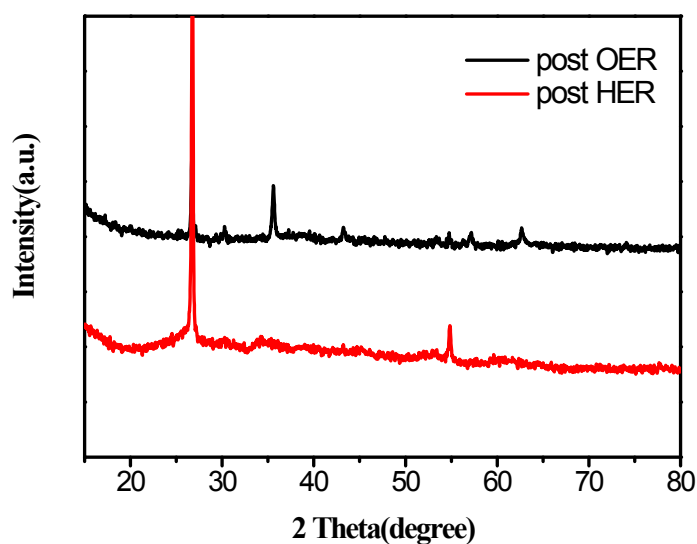
**Fig. S5** Cyclic voltammograms of (a) NiFe/CFP, (b) CFS/CFP, (c) -0.8V NiFe-CFS/CFP, (d) -1.0V NiFe-CFS/CFP, and (e) -1.2V NiFe-CFS/CFP composite electrodes between the potential regions of 0.02 and 0.12 V (vs Ag/AgCl) with scan rates of 10, 25, 50, 75, and 100 mV/s toward OER 1 M KOH solution. (f) Current density differences plotted against scan rates under a non-Faradaic range.



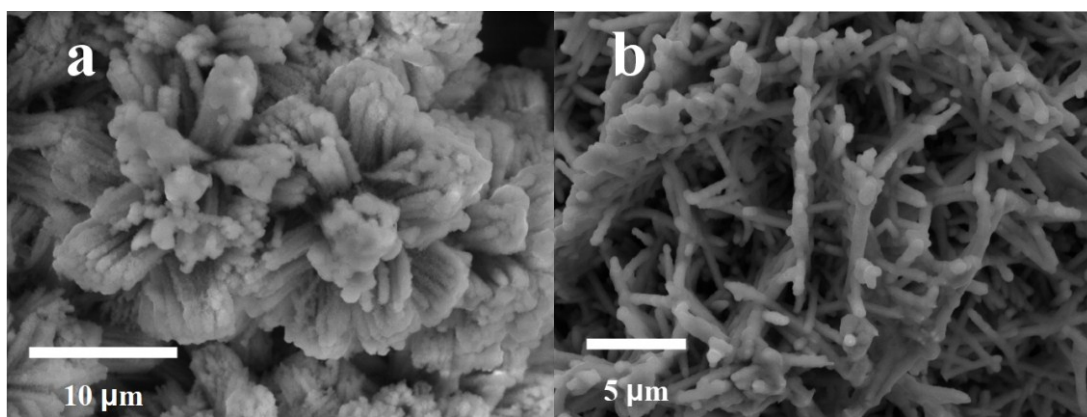
**Fig. S6** Cyclic voltammograms of (a) NiFe/CFP, (b) CFS/CFP, (c) -0.8V NiFe-CFS/CFP, (d) -1.0V NiFe-CFS/CFP, and (e) -1.2V NiFe-CFS/CFP composite electrodes between the potential regions of 0 and 0.1 V (vs Ag/AgCl) with scan rates of 10, 25, 50, 75, and 100 mV/s toward HER 1 M KOH solution. (f) Current density differences plotted against scan rates under a non-Faradaic range.



**Fig. S7** (a) The polarization curves of NiFe/CFP, CFS/CFP, -0.8V NiFe-CFS/CFP, and -1.2V NiFe-CFS/CFP catalysts in a two-electrode configuration.

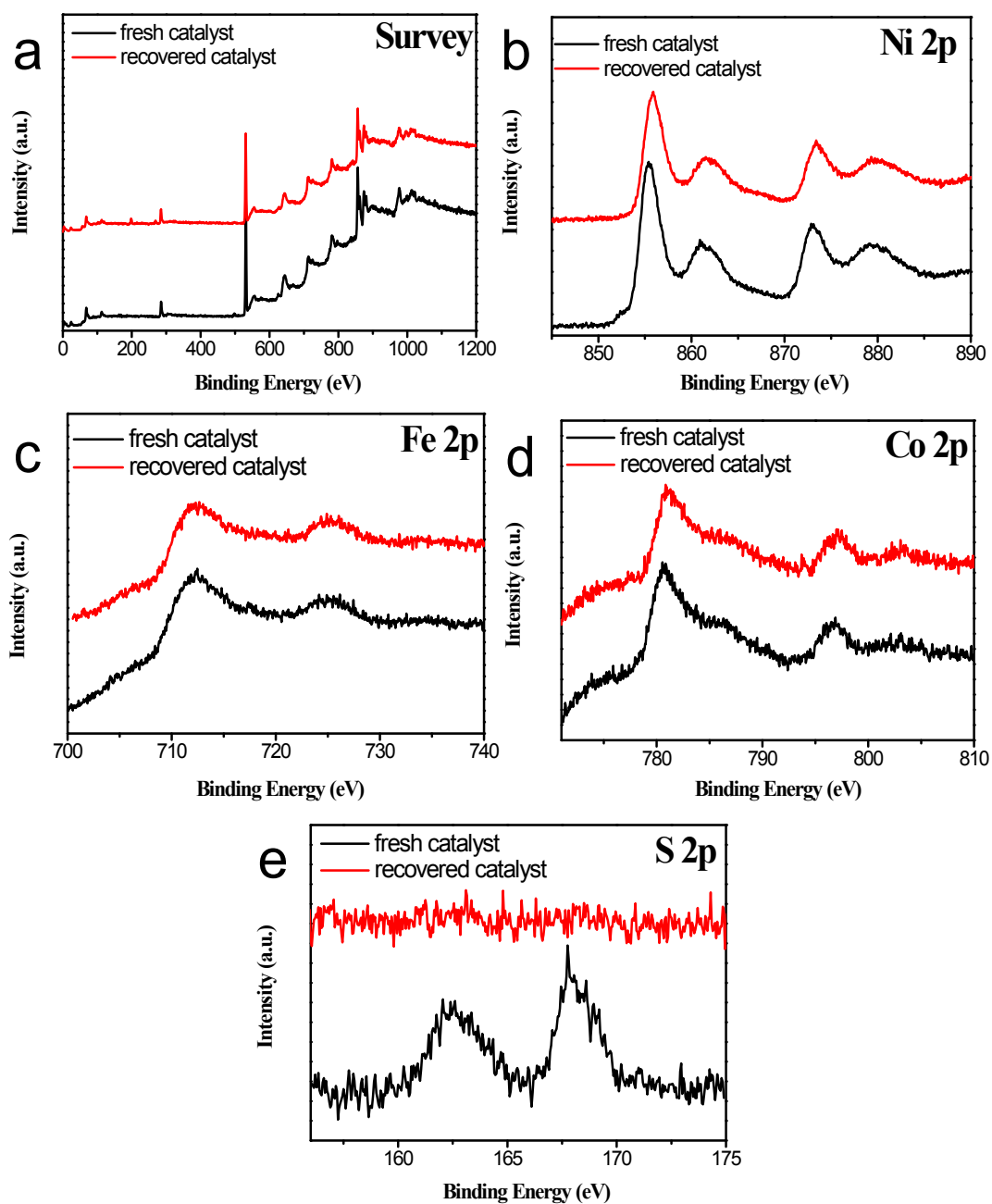


**Fig. S8** XRD patterns of post-OER and post-HER of -1.0V NiFe-CFS/CFP electrocatalysts.



**Fig. S9** Low magnification SEM images of -1.0V NiFe-CFS/CFP after long-term (a) OER and (b) HER electrolysis.





**Fig. S10** XPS high-resolution spectra of (a) survey (b) Ni 2p (c) Fe 2p (d) Co 2p and (e) S 2p of fresh and recovered -1.0V NiFe-CoFe<sub>2</sub>O<sub>4</sub>@Co<sub>3</sub>S<sub>4</sub>/CFP electrocatalysts.

**Table S1** Comparison of OER properties for electrocatalysts of catalytic materials

Electrode	Electrolyte	$\eta$ (mV)	Reference
-1.0V NiFe-CFS/CFP	1 M KOH	233.3 mV@ 50 mAcm <sup>-2</sup>	This work
NiFe/CFP	1 M KOH	292.9 mV@ 50 mAcm <sup>-2</sup>	This work
CFS/CFP	1 M KOH	285.2 mV@ 50 mAcm <sup>-2</sup>	This work
Ru <sub>2</sub> O/CFP	1 M KOH	304.8 mV@ 50 mAcm <sup>-2</sup>	This work
Ir <sub>2</sub> O/CFP	1 M KOH	445.3 mV@50 mA cm <sup>-2</sup>	This work
3D porous CoFe <sub>2</sub> O <sub>4</sub> /C	1 M KOH	240 mV@10 mA cm <sup>-2</sup>	<i>Adv. Mater.</i> <b>2017</b> , 29, 1604437.
Co <sub>1</sub> Mn <sub>1</sub> CH*	1 M KOH	322 mV@50 mA cm <sup>-2</sup>	<i>J. Am. Chem. Soc.</i> <b>2017</b> , 139, 8320-8328.
CoFeO <sub>x</sub> /NF	1 M KOH	270 mV@10 mA cm <sup>-2</sup>	<i>J. Am. Chem. Soc.</i> <b>2016</b> , 138, 8946-8957.
NiFe/NiCo <sub>2</sub> O <sub>4</sub> /NF	1 M KOH	320 mV@600 mA cm <sup>-2</sup>	<i>Adv. Funct. Mater.</i> <b>2016</b> , 26, 3515-3523.
Ni <sub>3</sub> Fe(OH) <sub>9</sub> /NF	1 M KOH	370 mV@100 mA cm <sup>-2</sup>	<i>Nat. Commun.</i> <b>2015</b> , 6, 6616.
Carbon-Confined NiCo@NiCoO <sub>2</sub>	1 M KOH	330 mV@ 10 mAcm <sup>-2</sup>	<i>Adv. Mater.</i> <b>2018</b> , 1705442
NiFe/Cu <sub>2</sub> O NWs/CF	1 M KOH	215 mV@ 10 mAcm <sup>-2</sup>	<i>ChemSusChem</i> <b>2017</b> , 10, 1475-1481.
Ni <sub>0.6</sub> Co <sub>1.4</sub> P	1 M KOH	300 mV@ 10 mAcm <sup>-2</sup>	<i>Adv. Funct. Mater.</i> <b>2018</b> , 28, 1706008
Co-NC@Mo <sub>2</sub> C	1 M KOH	347 mV@ 10 mAcm <sup>-2</sup>	<i>Nano Energy</i> <b>2019</b> , 57, 746
NiCo <sub>2</sub> O <sub>4</sub> /NF*	1 M KOH	420 mV@ 50 mAcm <sup>-2</sup>	<i>Adv. Funct. Mater.</i> <b>2016</b> , 26, 4661.
NiFe-LDH/CNT	1 M KOH	247 mV@ 10 mAcm <sup>-2</sup>	<i>J Am Chem Soc.</i> <b>2013</b> , 135, 8452-8455.

**Table S2** Comparison of HER properties for electrocatalysts of catalytic materials

Electrode	Electrolyte	$\eta$ (mV)	Reference
-1.0V NiFe-CFS/CFP	1 M KOH	98 mV@ 10 mAcm <sup>-2</sup>	This work
Pt-C/CFP	1 M KOH	35.8 mV@ 10 mAcm <sup>-2</sup>	This work
Ni <sub>0.75</sub> Fe <sub>0.125</sub> V <sub>0.125</sub> -LDHs	1 M KOH	125 mV@ 10 mAcm <sup>-2</sup>	<i>Small</i> , <b>2018</b> , 14, 1703257.
Ni <sub>5</sub> P <sub>4</sub> /NiP <sub>2</sub> /NiFe LDH	1 M KOH	124 mV@ 10 mAcm <sup>-2</sup>	<i>J. Mater. Chem. A</i> , <b>2018</b> , 6, 13619-13623.
Co <sub>3</sub> O <sub>4</sub> @MoS <sub>2</sub>	1 M KOH	90 mV@ 10 mAcm <sup>-2</sup>	<i>J. Mater. Chem. A</i> <b>2017</b> , 28, 2067-2072.
MoS <sub>2</sub> nanowall	1 M NaOH	150 mV@ 10 mAcm <sup>-2</sup>	<i>Nano Res.</i> <b>2017</b> , 10, 1178-1188.
MoS <sub>2</sub> /Ni <sub>3</sub> S <sub>2</sub> /NF	1 M KOH	110 mV@ 10 mAcm <sup>-2</sup>	<i>Angew. Chem.</i> <b>2016</b> , 55, 6702-6707
NCNT/Ni-NiFe <sub>2</sub> O <sub>4</sub> /Ni foam	1 M KOH	140 mV@ 10 mAcm <sup>-2</sup>	<i>Catal. Sci. Technol.</i> <b>2019</b> , 9, 1595-1601
NiCo <sub>2</sub> O <sub>4</sub> /NiFe LDH	1 M KOH	192 mV@ 10 mAcm <sup>-2</sup>	<i>ACS Appl. Mater. Interfaces</i> <b>2017</b> , 9, 1488.
Cu@NC NT/CF	1 M KOH	123 mV@ 10 mAcm <sup>-2</sup>	<i>ACS Appl. Mater. Interfaces</i> <b>2017</b> , 9, 36857-36864
NiMoN	1 M KOH	109 mV@ 10 mAcm <sup>-2</sup>	<i>Adv. Energy Mater.</i> <b>2016</b> , 6, 1600221-1600227.
Fe <sub>3</sub> C embedded Fe/N doped Carbon fiber	1 M KOH	86 mV@ 10 mAcm <sup>-2</sup>	<i>J. Mater. Chem. A</i> <b>2017</b> , 5, 7507-7515.
Co <sub>1</sub> Mn <sub>1</sub> CH/NF	1 M KOH	180 mV@ 10 mAcm <sup>-2</sup>	<i>J. Am. Chem. Soc.</i> <b>2017</b> , 139, 8320-8328.

**Table S3** Comparison of overall water splitting performance for the recently reported catalysts and ours.

Electrode	Electrolyte	$\eta$ (mV)	Reference
-1.0V NiFe-FCS/CFP	1 M KOH	1.53 V@ 10 mAcm <sup>-2</sup>	This work
NF@Ni/NiP	1 M KOH	1.61V@ 10 mAcm <sup>-2</sup>	<i>Adv. Funct. Mater.</i> <b>2016</b> , 26, 3314-3323
Cu@NiFe LDH	1 M KOH	1.54V@ 10 mAcm <sup>-2</sup>	<i>Energy Environ. Sci.</i> <b>2017</b> , 10, 1820-1827
NC@CuCo <sub>2</sub> N <sub>x</sub> /CF	1 M KOH	1.62V@ 10 mAcm <sup>-2</sup>	<i>Adv. Funct. Mater.</i> <b>2017</b> , 27, 1704169.
NiFe LDH/NF	1 M KOH	1.70 V@ 10 mAcm <sup>-2</sup>	<i>Science</i> , <b>2014</b> , 345, 1593-1596.
Ni <sub>0.69</sub> Co <sub>0.31</sub> P	1 M KOH	1.59V@ 10 mAcm <sup>-2</sup>	<i>Nanoscale</i> <b>2016</b> , 8, 19129.
Co <sub>1</sub> Mn <sub>1</sub> CH	1 M KOH	1.68 V@ 10 mAcm <sup>-2</sup>	<i>J. Am. Chem. Soc.</i> <b>2017</b> 139, 8320-8328.
NiS/Ni <sub>2</sub> P/CC	1 M KOH	1.62 V@ 10 mAcm <sup>-2</sup>	<i>J. Mater. Chem. A</i> <b>2018</b> , 6, 8233.
Co <sub>3</sub> S <sub>4</sub> @MoS <sub>2</sub>	1 M KOH	1.58 V@ 10 mAcm <sup>-2</sup>	<i>Nano Energy</i> <b>2018</b> , 47, 494.
Ni-Co-P hollow nanobricks	1 M KOH	1.62 V@ 10 mAcm <sup>-2</sup>	<i>Energy &amp; Environmental Science</i> <b>2018</b> , 11, 872.
NiCo <sub>2</sub> S <sub>4</sub> @NiFe LDH	1 M KOH	1.60 V@ 10 mAcm <sup>-2</sup>	<i>ACS Appl. Mater. Interfaces</i> <b>2017</b> , 9, 15364-15372.