1 Optical motion control of liquid crystalline droplets by host-guest

2	molecular interaction
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9	Video information
10	Movie S1 A movie of the motion of the BHAB-doped 5CB droplet during the on-off
11	irradiation of a UV light. The UV light was irradiated from the right side with an intensity of 58
12	mW/cm ² . The diameter of the droplet was 110 μ m, and the concentration of SDS was 0.3 wt%.
13	The file name is S1_BHAB_doped_5CB_motion.wmv
14	
15	Movie S2 A movie of the motion of the BMAB-doped 5CB droplet during the on-off
16	irradiation of a UV light. The UV light was irradiated from the right side with an intensity of 58
17	mW/cm ² . The diameter of the droplet was 110 μ m, and the concentration of SDS was 0.3 wt%.
18	S2_The file name is BMAB_doped_5CB_motion.wmv
19	
20	Movie S3 A movie of the convective flow inside the LC droplet when the UV light was
21	irradiated from the right side. The UV light intensity was 32 mW/cm ² . The diameter of the droplet
22	was 400 μ m, and the concentration of SDS was 1 wt%. The file name is
23	S3_BHAB_doped_5CB_convective_flow.wmv
24	
25	Movie S4 A movie of the convective flow inside the LC droplet when the UV light was
26	irradiated from the right side. The UV light intensity was 32 mW/cm ² . The diameter of the droplet
27	was 430 μ m, and the concentration of SDS was 1 wt%. The file name is
28	S4_BMAB_doped_5CB_convective_flow.wmv
29	
30	Movie S5 A movie of the behavior of an isotropic droplet doped with BHAB (2 wt%)
31	when the UV light was irradiated from the right side. The UV light intensity was 104 $\rm mW/cm^2.$
32	The diameter of the droplet was 90 $\mu m,$ and the concentration of SDS was 0.3 wt%. At first the
33	nematic droplet was gradually heated and the droplet changed into the isotropic phase, which was
34	recognized by the pattern change. The file name is S5_BHAB_doped_isotropic_5CB.wmv.

Movie S6 A movie of the behavior of toluene droplet doped with BHAB (2 wt%) when
the UV light was irradiated from the right side. The UV light intensity was 58 mW/cm². The
diameter of the droplet was 100 µm and the concentration of SDS was 0.3 wt%. The file name is
S6 BHAB doped toluene.wmv



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43Fig.S1. (a) The schematic diagram of the microfluidic setup consisted of a glass capillary for the 44 formation of LC particles, and the geometry of micro-capillaries and the formation mechanism of a single emulsion is shown. A square glass capillary (inner diameter: 0.90×0.90 mm²) was attached on 45a glass slide, and a tapered cylindrical capillary was inserted into it. The tapered capillary was prepared 46 47by tapering a tip of cylindrical glass capillary (inner diameter: 0.70×0.70 mm², outer diameter: 0.8748 \times 0.87 mm²) using a micropipette puller (P-1000, Sutter Instrument). The tip diameter was 50 μ m. 49Syringe needles were connected to the syringe pump via a micro- tube, and the two types of liquids 50(inner fluid and outer fluid) were introduced. The inner fluid was sheared by the outer fluid in the inlet of the tapered cylindrical capillary, and droplets were formed. The droplets were collected from the 5152tapered capillary on the right side. The hydrophilic treatment was applied for the tapered capillary by 53plasma cleaner (PDC-32G, Harrick Plasma). The microfluidic device was operated at room 54temperature (25 °C). (b) A microscopic image during the formation of photo-responsive LC particles 55is shown.

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58 Fig.S2. A schematic drawing of the observation setup is shown. A rubber spacer (thickness: 0.2 mm)

59 with an open space was sandwiched with two coverslips (thickness: 0.12~0.17 mm), and an SDS

60 solution including LC droplets was pipetted into the open space. The cell was placed on an inverted

61 optical microscope (IX71, OLYMPUS), and the behavior of the LC droplets under light illumination

62 was observed. A UV-LED was illuminated from the top side at an oblique angle of 40 degrees.



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66 Fig.S3. UV/Vis absorption spectra for BHAB, BMAB, 5CB and SDS are shown. BHAB, BMAB and

- 5CB were solved in cyclohexane with the concentrations 7.9×10^{-5} , 7.5×10^{-5} and 3.3×10^{-5} mol/L,
- respectively. The solven for SDS was water and its concentration was 3.5×10^{-2} mol/L.

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