

Electronic Supplementary Information (ESI) for

**Ultrafast Li-Ion Migration in Holey-Graphene-Based Composites
Constructed by A Generalized *Ex-Situ* Method towards High
Capacity Energy Storage**

Chengling Zhu,^{a1} Zeyu Hui,^{a1} Hui Pan,^a Shenmin Zhu,^{*a} Qing Zhang,^b Jianfeng Mao,^{*b} Zaiping Guo,^{*bc} Yao Li,^a Muhammad Imtiaz^{ad} and Zhixin Chen^c

^aState Key Laboratory of Metal Matrix Composites, Shanghai Jiao Tong University, Shanghai 200240, China. E-mail: smzhu@sjtu.edu.cn

^bInstitute for Superconducting & Electronic Materials, Australian Institute of Innovative Materials, University of Wollongong, Wollongong 2522, Australia. E-mail: jmao@uow.edu.au; zguo@uow.edu.au

^cEngineering Materials Institute, School of Mechanical, Materials & Mechatronics Engineering, University of Wollongong, Wollongong, 2522, Australia

^dDepartment of Physics, Islamia College Peshawar, Peshawar 25120, Pakistan

¹These authors contributed equally to this work.

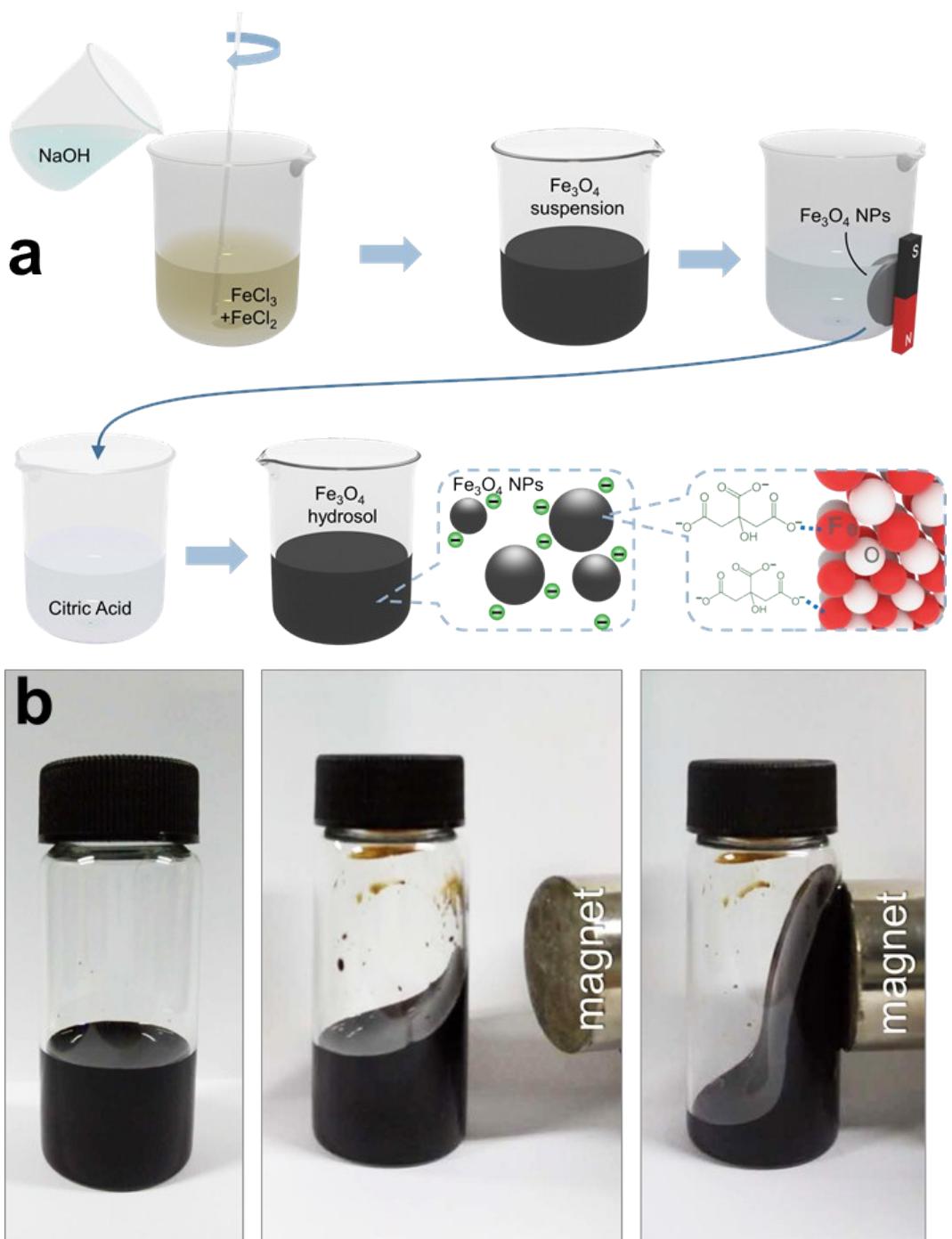


Fig. S1 (a) The schematic synthesis route of Fe_3O_4 hydrosol. (b) A simple demonstration to show the superparamagnetism of the Fe_3O_4 hydrosol, which is magnetofluid in fact.

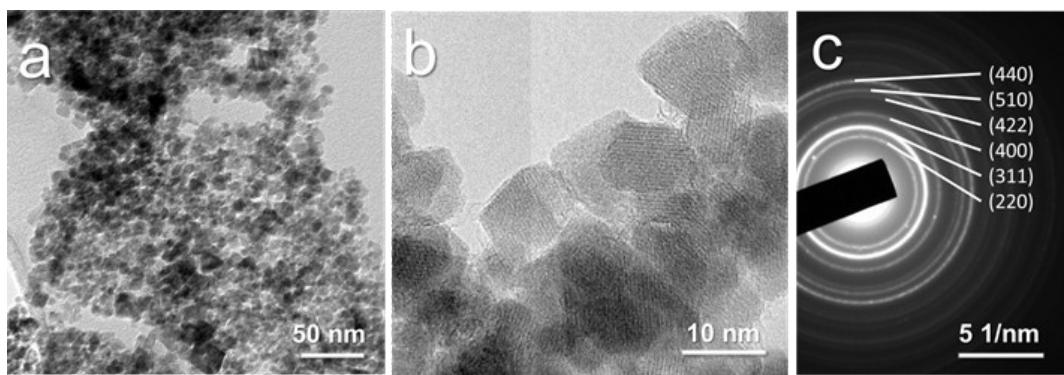


Fig. S2 TEM images and SAED pattern of the Fe_3O_4 nanoparticles.

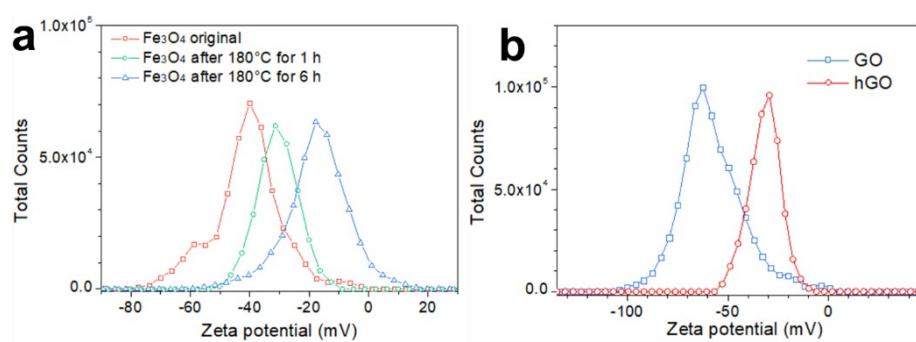


Fig. S3 (a) The zeta potential of Fe_3O_4 in the hydrosol, original and after hydrothermal process. This hydrothermal experiment was conducted on the Fe_3O_4 hydrosol solely to test its stability. (b) The zeta potential of GO and hGO.

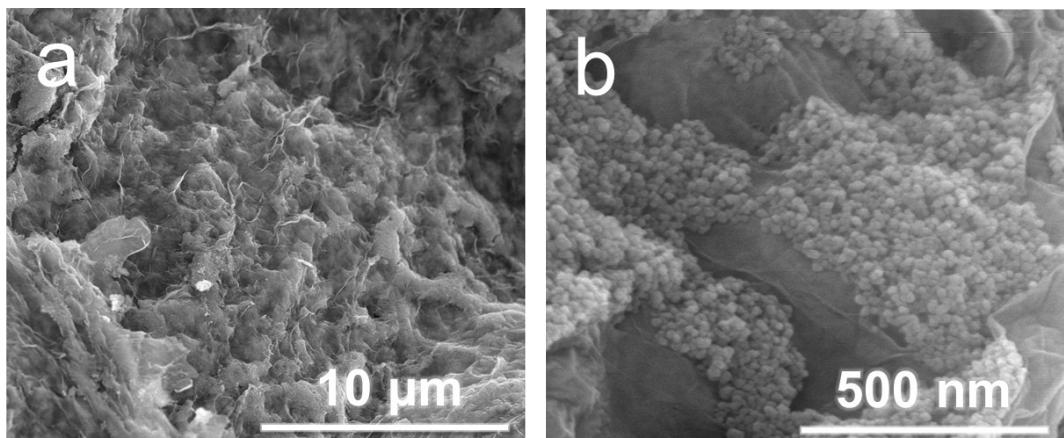


Fig. S4 SEM images of $\text{Fe}_3\text{O}_4/\text{G}$ in (a) micron-scale and (b) submicron scale.

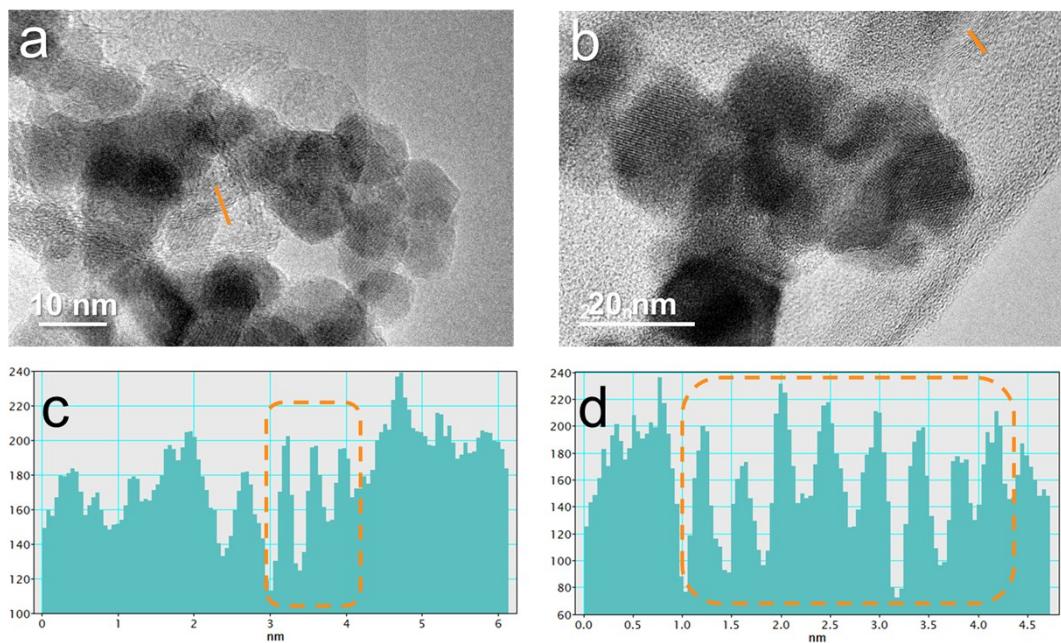


Fig. S5 TEM images of (a) Fe₃O₄@3DhG and (b) Fe₃O₄/G that rGO layers can be identified. (c, d) The line profiles of (a) and (b), respectively, with the orange boxes indicating rGO fringes.

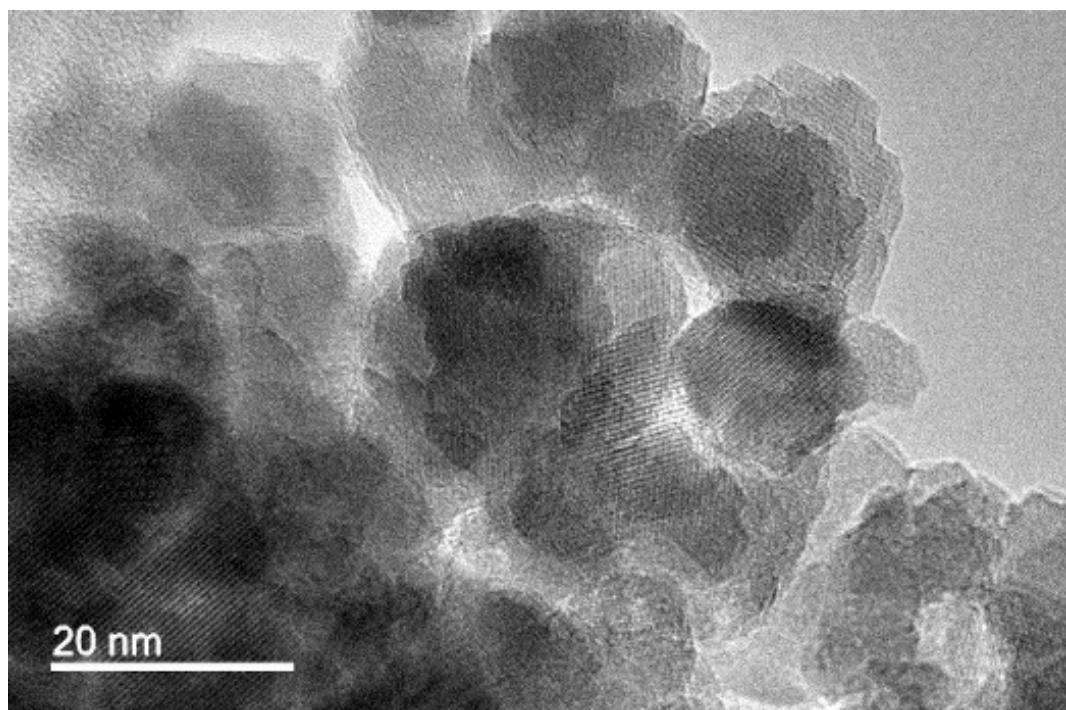


Fig. S6 HRTEM image of Fe₃O₄/G.

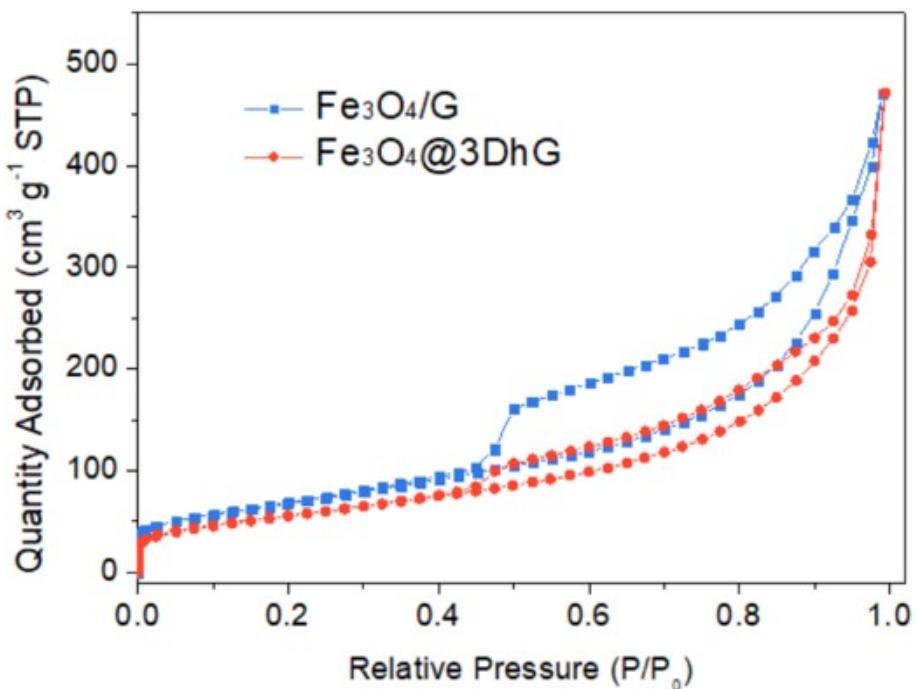


Fig. S7 The nitrogen adsorption isotherms of $\text{Fe}_3\text{O}_4@3\text{DhG}$ and $\text{Fe}_3\text{O}_4/\text{G}$.

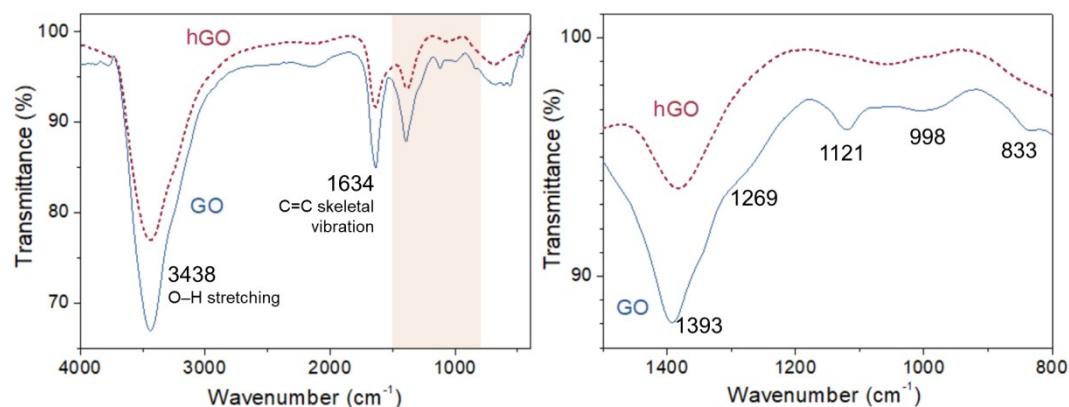


Fig. S8 (a) FT-IR spectra of hGO (red) and GO (blue). (b) The magnified part of (a) between $800-1500 \text{ cm}^{-1}$.

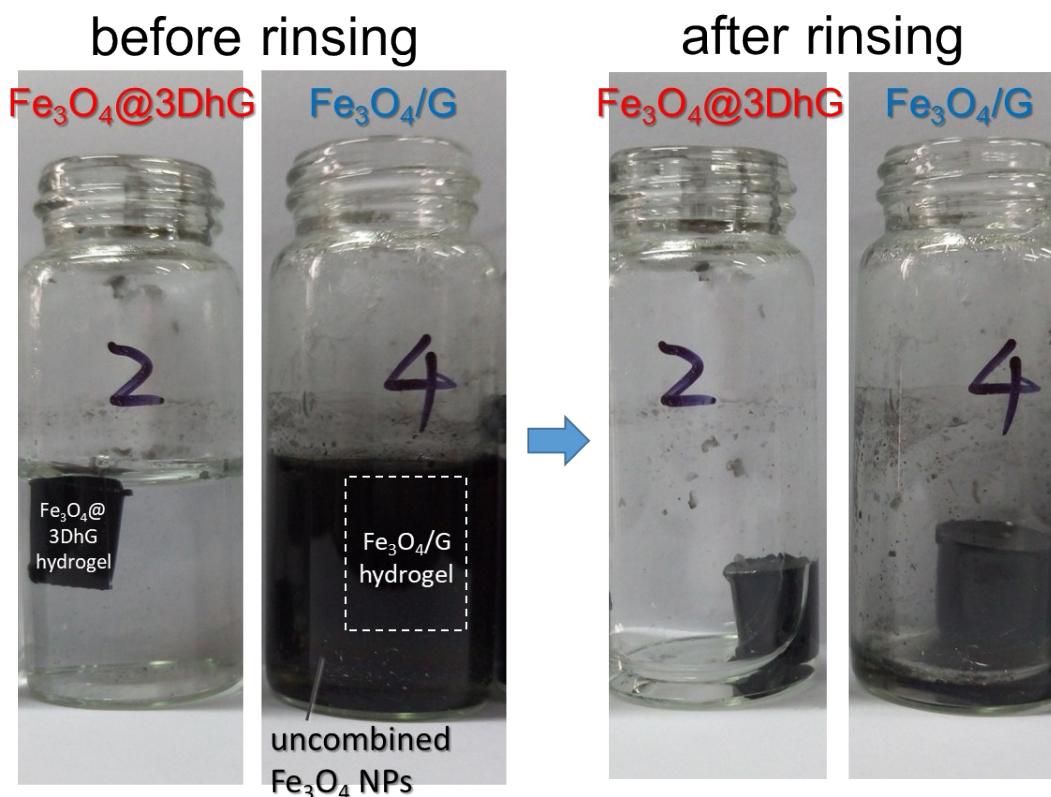


Fig. S9 Photos of $\text{Fe}_3\text{O}_4@\text{3DhG}$ and $\text{Fe}_3\text{O}_4/\text{G}$ after hydrothermal process. Many uncombined Fe_3O_4 NPs were found in the container of $\text{Fe}_3\text{O}_4/\text{G}$, and were removed after rinsing.

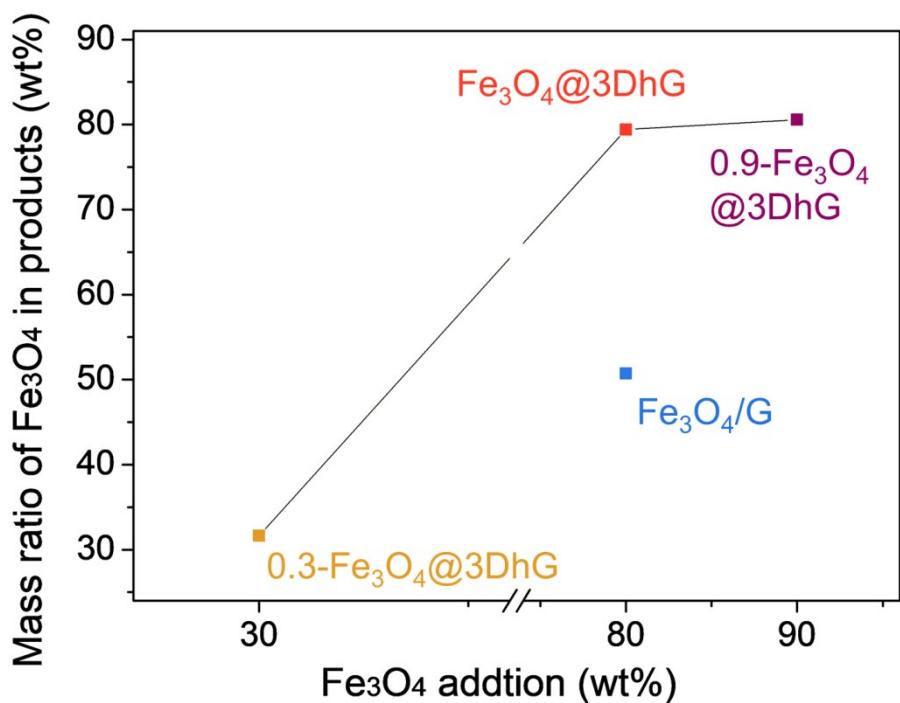


Fig. S10 The relation between the Fe_3O_4 addition and the mass ratio of Fe_3O_4 in products.

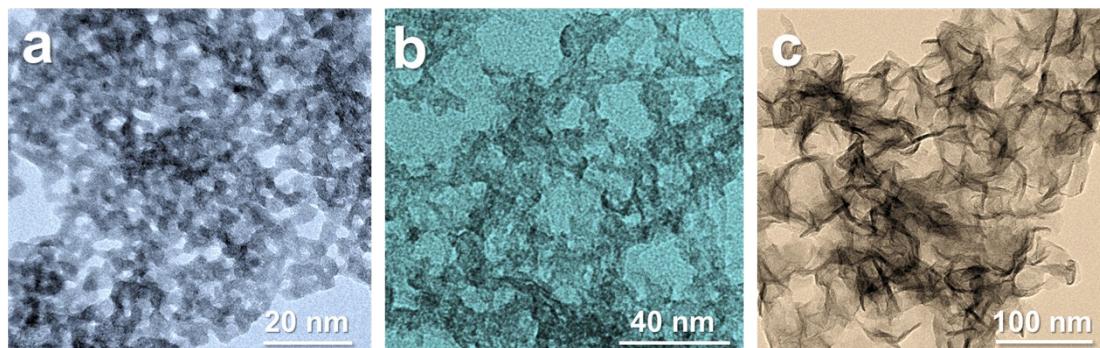


Fig. S11 TEM images of the (a) SnO_2 NPs, (b) $\delta\text{-MnO}_2$ NPs and (c) MoS_2 NSs.

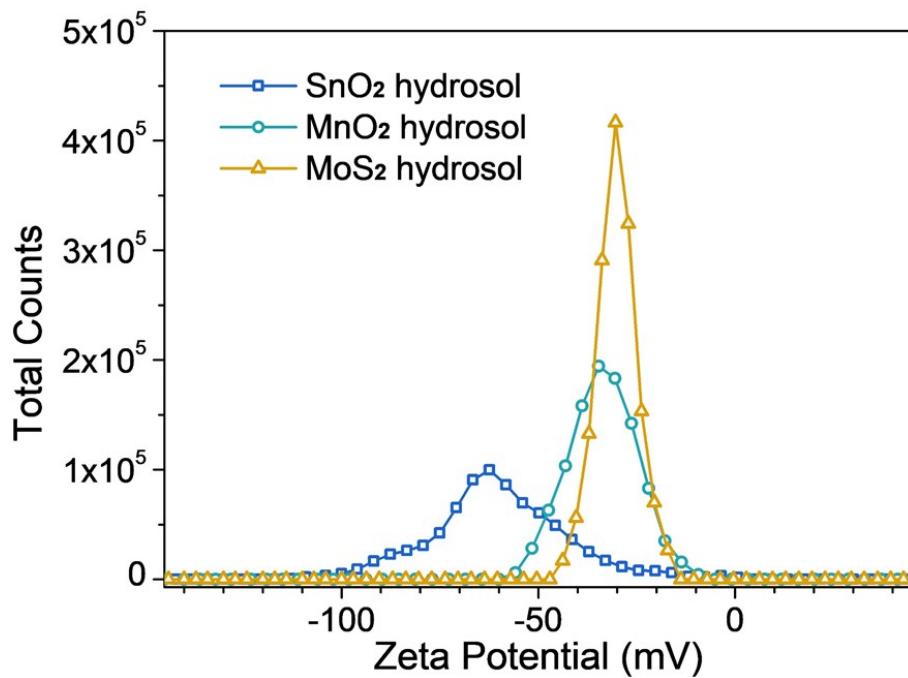


Fig. S12 The zeta potential of SnO₂ NPs, δ-MnO₂ NPs and MoS₂ NSs in corresponding hydrosols.

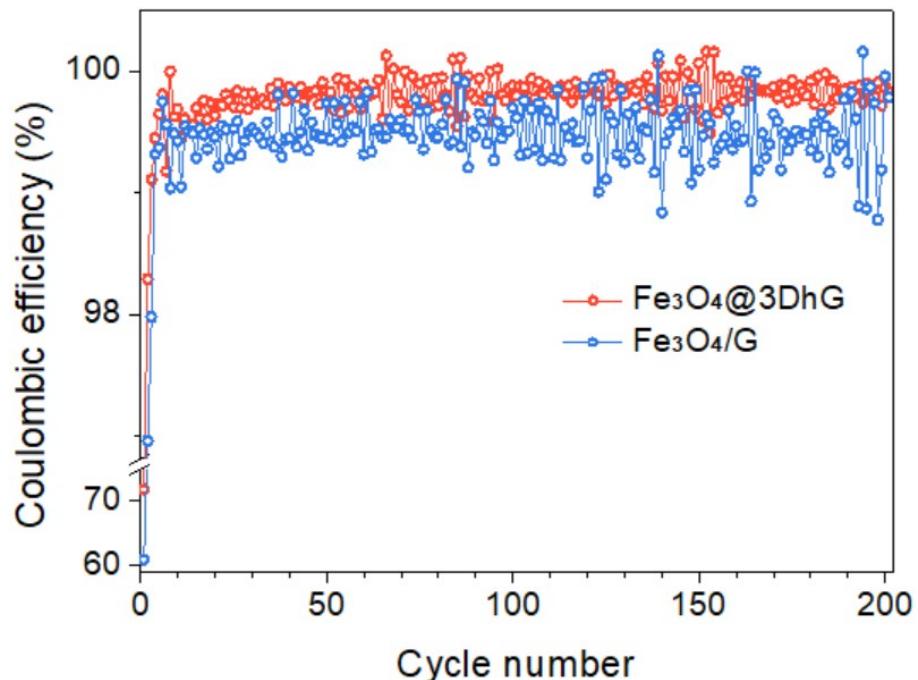


Fig. S13 The coulombic efficiency of Fe₃O₄@3DhG and Fe₃O₄/G half cells at 0.2 A g⁻¹.

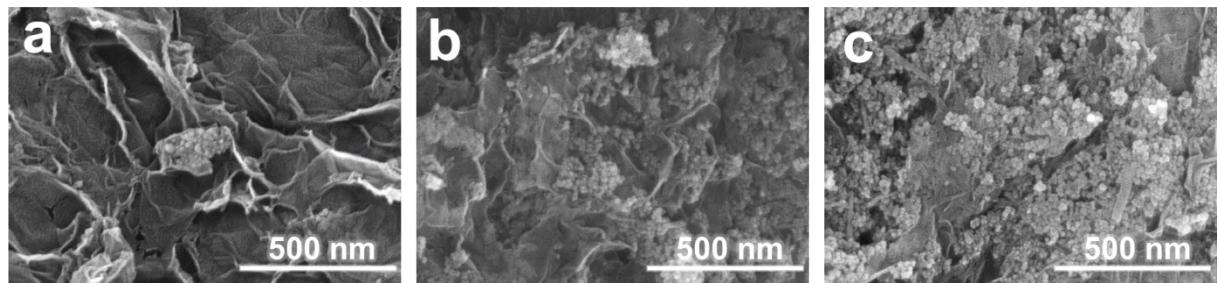


Fig. S14 SEM images of (a) 0.3-Fe₃O₄@3DhG, (b) Fe₃O₄@3DhG, and (c) 0.9-Fe₃O₄@3DhG.

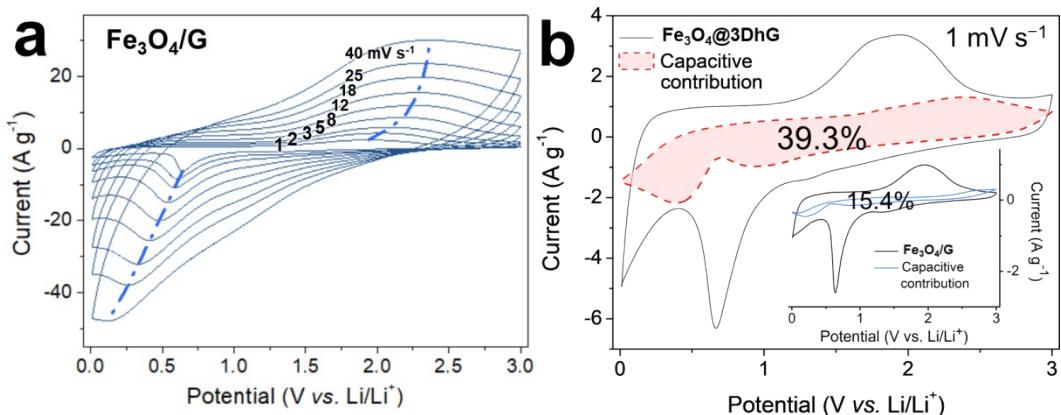


Fig. S15 (a) CV curves of Fe₃O₄/G at scan rates of 1–40 mV s⁻¹. (b) The calculated contribution of surface capacitive effect of Fe₃O₄@3DhG and (inset) Fe₃O₄/G at 1 mV s⁻¹.

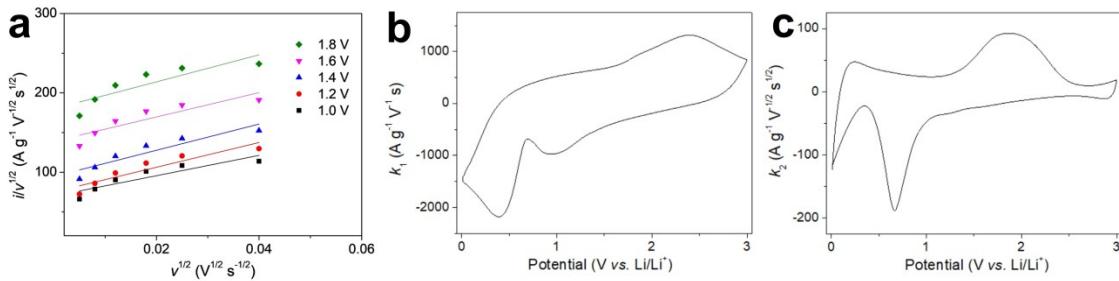


Fig. S16 (a) The example of linear fitting of $i/v^{1/2}$ vs. $v^{1/2}$ based on the data in Fig. 3a. The solved (b) k_1 - and (c) k_2 -value vs. potential.

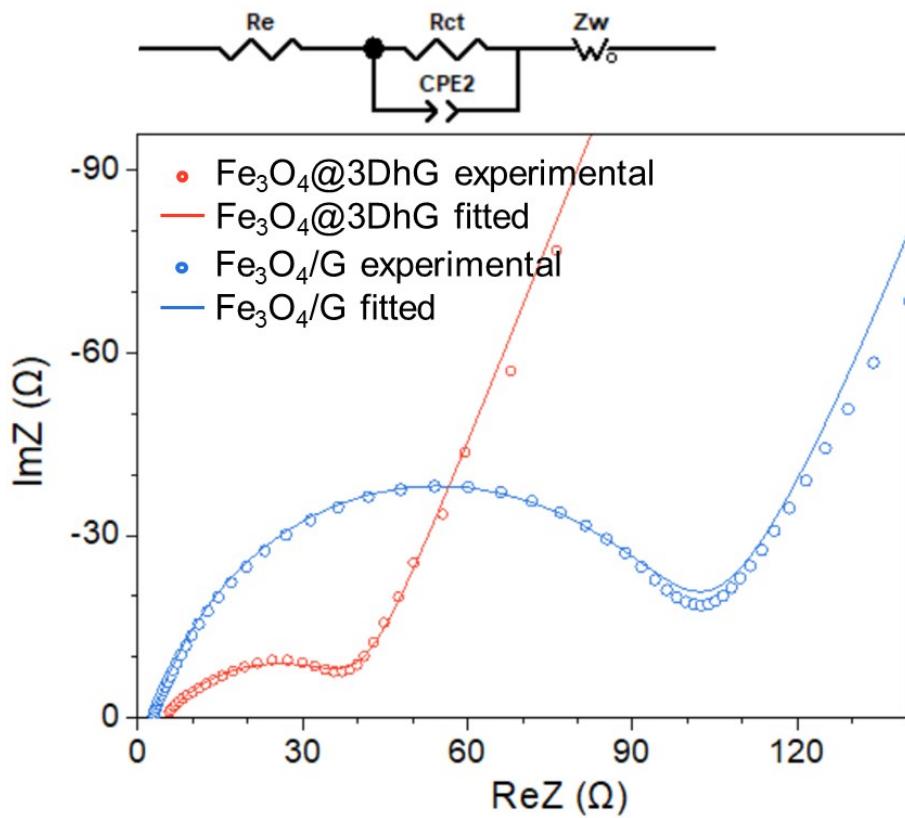


Fig. S17 The experimental and fitted Nyquist plots of $\text{Fe}_3\text{O}_4@\text{3DhG}$ and $\text{Fe}_3\text{O}_4/\text{G}$ before cycling test.

Table S1 The composition and performance of the reported **graphene-based** composites synthesized by *ex-situ*-combined method for LIB anodes.

Active Materials	Loading Content (wt%)	Particle Size (nm)	Capacity (mAh g ⁻¹)	Cycle Number	Current Density (A g ⁻¹)	Reference
SnO ₂	48	5	924	200	0.1	1
			505	1000	1	
Si	32.8	<100	1261	70	0.05	2
MnO	84	100–200	1202	100	0.2	3
			812	1000	2	
Co ₃ O ₄	76.2	270	600	500	1	4
Co ₃ S ₄	70.8	30	710	200	0.5	5
ZnO	58	3.5	650	100	0.0978	6
Fe ₂ O ₃	80	400	758	300	0.2	7
Fe ₃ O ₄ +TiO ₂	60	6	703	200	0.5	8
Fe ₃ O ₄	70.5	100	1048	50	0.2	9
			1080	450	1	
Fe ₃ O ₄	83.7	200	1059	150	0.093	10
Fe ₃ O ₄	63	57.6	550	60	0.375	11
Fe ₃ O ₄	79.4	10	1516	200	0.2	This Work
			554	2000	5	

Table S2 The composition and performance of the reported iron oxide-graphene composites for LIB anodes in recent years.

Composition	Synthesis Route	Loading Content (wt%)	Particle Size (nm)	Capacity (mAh g ⁻¹)	Cycle Number	Current Density (A g ⁻¹)	Reference
Fe ₃ O ₄ @carbon microboxes	<i>ex situ</i>	/	650	857	100	0.1	12
Fe ₃ O ₄ nanorods/rGO	<i>in situ</i>	75	10–80	1149	100	0.1	13
				1053	250	0.5	
Fe ₃ O ₄ -Fe@Bamboo-like CNT	<i>in situ</i>	34.4	70	764.1	100	0.3	14
Fe ₃ O ₄ /carbon/CNT microspheres	<i>in situ</i>	36.7	20	1317	130	0.1	15
				525	300	5	
Fe ₃ O ₄ /N-doped carbon	<i>in situ</i>	55	5	1963	170	0.1	16
				1741	600	1	
graphene@Fe ₃ O ₄ /amorphous carbon	<i>in situ</i>	37.4	2–3	853	500	0.2	17
Fe ₃ O ₄ /N-doped graphene	<i>in situ</i>	73.5	15	~1,200	150	0.1	18
				>900	1000	3	
Fe ₃ O ₄ /graphene foam	<i>in situ</i>	79.4	20–180	1220	500	1	19
hollow Fe ₃ O ₄ /carbon	<i>in situ</i>	67	100–12 ₀	1300	100	0.1	20
Fe ₃ O ₄ /C nanofibers	<i>ex situ</i>	27.7	30	780	150	0.5	21
				611	300	1	
Fe ₃ O ₄ nanorods/N-doped graphene	<i>in situ</i>	85	20	929	50	0.1	22
Fe ₃ O ₄ /graphene	<i>in situ</i>	74.3	15	1070	160	0.2	23
				445	600	2	
Graphene-Wrapped Fe ₃ O ₄ - Graphene Nanoribbons	<i>in situ</i>	60	10	708	300	0.4	24
Fe ₃ O ₄ @Polypyrrole Nanocages	<i>in situ</i>	88.7	650	652	500	2	25
				950	100	0.2	
Fe ₃ O ₄ /Graphene Nanoribbons	<i>in situ</i>	58	6	1700	100	0.1	26
				1058	100	1	
Fe ₃ O ₄ /Few layer Graphene	<i>ex situ</i>	73.4	10	1413	100	1	27
				768	100	5	
γ-Fe ₂ O ₃ /RGO	<i>in situ</i>	64	50	830	500	0.5	28
yolk-shell octahedron	<i>in situ</i>	/	400	1176	200	0.1	29
				744	500	1	
Fe ₂ O ₃ /C hollow microfibers	<i>in situ</i>	/	100	997.8	200	1	30
hollow Fe ₂ O ₃ /C Nanofibers	<i>in situ</i>	62	17	812	300	1	31
Porous Fe ₂ O ₃ Nanoframeworks/Graphene	<i>in situ</i>	66.7	20–40	1129	130	0.2	32
				523.5	1200	5	
nitridated α-Fe ₂ O ₃ nanorods/carbon cloth	<i>in situ</i>	/	100–30 ₀	332	100	5	33

Porous α -Fe ₂ O ₃ nanorods/CNT/graphene	<i>in situ</i>	/	50	~1200	300	0.2	34
Fe ₂ O ₃ /few layered graphene	<i>ex situ</i>	~75	28	729	300	0.2	7
Monolithic Fe ₂ O ₃ /graphene hybrid	<i>in situ</i>	42	30-50	810	100	0.1	35
α -Fe ₂ O ₃ /GO/CNTs	<i>in situ</i>	83	50	875	100	0.2	36
α -Fe ₂ O ₃ /Graphene	<i>in situ</i>	48.3	180	1111.6	350	0.2	37
				658.5	200	1	
α -Fe ₂ O ₃ Hollow Nanobarrels/rGO	<i>in situ</i>	52	200	916	100	0.5	38
				478	1000	5	
Hollow Fe ₂ O ₃ /N-Doped Graphene	<i>in situ</i>	86	30	1483	100	0.1	39
				729	300	1	
Fe ₂ O ₃ /Fe ₃ C–Graphene Thin Films	<i>in situ</i>	24	40	984	100	0.17	40
				518	1000	6.6	
FeOOH/rGO	<i>in situ</i>	81.4	200	1135	200	1	41
				783	200	5	
Fe ₃ O ₄ @3DhG	<i>ex situ</i>	79.4	10	1516	200	0.2	This Work
				554	2000	5	

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