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Supporting Information

Sequential Growth of Hierarchical N-Doped Carbon-MoS₂

Nanocomposites with Variable Nanostructures

Qinglong Liu^{a,b}, Haodong Shi^{c,d}, Tianyu Yang^b, Yan Yang^{e*}, Zhong-Shuai Wu^c, Jianqiang Yu^{a*}, S.

Ravi P. Silva^f and Jian Liu^{b,f*}

^a School of Material Science and Engineering, School of Chemistry and Chemical Engineering, Qingdao University, Qingdao, 266071, Shandong, China.

^b State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, and Dalian National Laboratory for Clean Energy, 457 Zhongshan Road, Dalian 116023, China

^c Dalian National Laboratory for Clean Energy, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 457 Zhongshan Road, Dalian 116023, China

^d University of Chinese Academy of Sciences, 19 A Yuquan Rd, Shijingshan District, Beijing 100049, China

^e Collaborative Innovation Center for Vessel Pollution Monitoring and Control, Dalian Maritime University, Dalian, 116026, China

^f Department of Chemical and Process Engineering, University of Surrey, Guildford, Surrey, GU27XH, UK

*E-mail: jianqyu@qdu.edu.cn; yyang@dlmu.edu.cn; jian.liu@surrey.ac.uk

Experimental Section

All these nanomaterials follow the same chemicals addition sequence, as follows: CTAB, F127 and cysteine were dissolved in the solution of water and ethanol. After 30 minutes stirring at 30 °C, sodium molybdate dihydrate (Na₂MoO₄ 2H₂O) was added, and subsequently 3-aminophenol and formaldehyde solution were added in every 10 minutes. The only difference is the addition amount of Na₂MoO₄ 2H₂O and cysteine.

Synthesis of NC-MoS2 yolk-shell nanocomposites.

0.13 g of CTAB, 0.10 g F127 and 0.60 g cysteine were dissolved in the solution containing 20 mL of water and 8 mL of ethanol. After 30 minutes stirring at 30 °C, 0.30 g of sodium molybdate dihydrate (Na₂MoO₄ 2H₂O) was added, and subsequently 0.20 g of 3-aminophenol and 0.28 mL of formaldehyde solution were added in every 10 minutes. The suspension was kept at 30 °C for 24 h. Then the resulting mixture was transferred into a Teflon-lined autoclave and kept at 220 °C under a static condition for 24 h. The solid product was recovered by centrifugation and washed with water and ethanol for three times each and then vacuum-dried at 80 °C for 6 h. To obtain NC-MoS₂ samples, the sample is pyrolysed under nitrogen flow in the tube furnace using a heating rate of 1°C min⁻¹ up to 350 °C, dwelling for 2 h, and resuming heating rate at 1 °C min⁻¹ up to 800 °C and then dwelling for 6 h.

Synthesis of NC-MoS₂ core-shell nanocomposites (Figure 1C).

0.13 g of CTAB, 0.10 g F127 and 0.60 g cysteine were dissolved in the solution containing 20 mL of water and 8 mL of ethanol. After 30 minutes stirring at 30 °C, 0.10 g of sodium molybdate dihydrate (Na₂MoO₄ $2H_2O$) was added, and subsequently 0.20

g of 3-aminophenol and 0.28 mL of formaldehyde solution were added in every 10 minutes. The suspension was kept at 30 °C for 24 h. Then the resulting mixture was transferred into a Teflon-lined autoclave and kept at 220 °C under a static condition for 24 h. The solid product was recovered by centrifugation and washed with water and ethanol for three times each and then vacuum-dried at 80 °C for 6 h. To obtain NC-MoS₂ samples, the sample is pyrolysed under nitrogen flow in the tube furnace using a heating rate of 1°C min⁻¹ up to 350 °C, dwelling for 2 h, and resuming heating rate at 1 °C min⁻¹ up to 800 °C and then dwelling for 6 h.

For NC-MoS₂ core-shell nanocomposites (Figure S1), the synthesis process is similar, the only difference is the Na₂MoO₄ 2H₂O amount, 0.05 g instead of 0.10 g.

Synthesis of NC-MoS₂ hollow nanocomposites.

0.13 g of CTAB, 0.10 g F127 and 0.30 g cysteine were dissolved in the solution containing 20 mL of water and 8 mL of ethanol. After 30 minutes stirring at 30 °C, 0.05 g of sodium molybdate dihydrate (Na₂MoO₄ 2H₂O) was added, and subsequently 0.20 g of 3-aminophenol and 0.28 mL of formaldehyde solution were added in every 10 minutes. The suspension was kept at 30 °C for 24 h. Then the resulting mixture was transferred into a Teflon-lined autoclave and kept at 220 °C under a static condition for 24 h. The solid product was recovered by centrifugation and washed with water and ethanol for three times each and then vacuum-dried at 80 °C for 6 h. To obtain NC-MoS₂ samples, the sample is pyrolysed under nitrogen flow in the tube furnace using a heating rate of 1°C min⁻¹ up to 350 °C, dwelling for 2 h, and resuming heating rate at 1 °C min⁻¹ up to 800 °C and then dwelling for 6 h.

Synthesis of NC-MoS₂ nanorods with different length.

For short NC-MoS₂ nanorods (Figure 2A): 0.13 g of CTAB, 0.10 g F127 and 0.10 g cysteine were dissolved in the solution containing 20 mL of water and 8 mL of ethanol. After 30 minutes stirring at 30 °C, 0.05 g of sodium molybdate dihydrate (Na₂MoO₄ 2H₂O) was added, and subsequently 0.20 g of 3-aminophenol and 0.28 mL of formaldehyde solution were added in every 10 minutes. The suspension was kept at 30 °C for 24 h. Then the resulting mixture was transferred into a Teflon-lined autoclave and kept at 220 °C under a static condition for 24 h. The solid product was recovered by centrifugation and washed with water and ethanol for three times each and then vacuum-dried at 80 °C for 6 h. To obtain NC-MoS₂ samples, the sample is pyrolysed under nitrogen flow in the tube furnace using a heating rate of 1°C min⁻¹ up to 350 °C, dwelling for 2 h, and resuming heating rate at 1 °C min⁻¹ up to 800 °C and then dwelling for 6 h.

For long NC-MoS₂ nanorods (Figure 2A), the synthesis process is similar, the only difference is the cysteine amount, 0.20g instead of 0.10 g.



Scheme S1. The possible detail formation mechanism of NC-MoS $_2$ nanocomposites

with different nanostructures through a synergistic self-assembling process.

Cysteine	$Na_2MoO_4 \bullet 2H_2O$	Mo /APF	Cysteine/APF	
(g)	(g)	mole ratio	mole ratio	
0.60	0.40	1.06	2.70	
0.00	0.40	1.00		
0.60	0.20	0.90	2.70	
0.60	0.50	0.80		
0.60	0.10	0.26	2.70	
0.60	0.10	0.26		
0.60	0.05	0.12	2 70	
0.60	0.05	0.13	2.70	
0.20	0.05	0.12	1.35	
0.30	0.05	0.13		
0.20	0.05	0.12	0.00	
0.20	0.05	0.13	0.90	
0.10	0.05	0.12	0.45	
0.10	0.05	0.13		
	Cysteine (g) 0.60 0.60 0.60 0.60 0.30 0.20 0.10	Cysteine (g) Na2MoO4 • 2H2O (g) 0.60 0.40 0.60 0.30 0.60 0.10 0.60 0.05 0.30 0.05 0.20 0.05 0.10 0.05	Cysteine (g)Na2MoO4 • 2H2O (g)Mo /APF mole ratio0.600.401.060.600.300.800.600.100.260.600.050.130.300.050.130.200.050.130.100.050.13	

Table S1. Synthesis parameters of $NC-MoS_2$ nanocomposites with different nanostructures.

Sample	Sbet (m ² g ⁻¹) ^a	V _t (cm ³ g ⁻¹) ^b	Mo content ^c wt%	N content ^d wt%	C content ^d wt%	MoS ₂ /C ratio by wt
Dual-Shell spheres	6.4	0.029	35.25	3.38	61.37	0.96
Yolk-Shell spheres	43.7	0.028	27.56	6.68	65.76	0.70
Core-Shell spheres	10.4	0.058	12.37	7.87	79.76	0.26
Hollow spheres	19.9	0.090	9.40	10.05	80.55	0.19
Long nanorods	14.8	0.025	9.74	11.82	78.44	0.21

Table S2. Physicochemical parameters of NC-MoS₂ nanocomposites with different nanostructures.

^a S_{BET} is the BET specific surface area.

^b V_t is the total pore volume determined at the relative pressure of 0.98.

^c The Mo content was obtained by ICP.

^d The N content was obtained by elemental analysis.



Figure S1. Structural and elemental distribution information of NC-MoS₂ nanocomposites yolk-shell spheres. In this frame, the upper row images are: SEM image (left), TEM image (middle) and dark field TEM image (right), while bottom row images are elemental mapping of C, N, S and Mo of area in dark field TEM image (bottom row).



Figure S2. The HRTEM images of the NC-MoS₂ nanocomposites: short nanorod (A) and (B) long nanorod.



Figure S3. CV curves from 3.0 to 0 V at a scanning rate of 0.1 mV \cdot S⁻¹.



Figure S4. The Raman spectra of $NC-MoS_2$ nanocomposites with dual-shell, core-shell and hollow nanostructures.