

Supporting Information

An efficient multi-doping strategy to enhance Li-ion conductivity in garnet-type solid electrolyte $\text{Li}_7\text{La}_3\text{Zr}_2\text{O}_{12}$

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Table S1. XRD refined atomic occupation factors, fractional coordinates, and isotropic thermal factors of the undoped LLZO, mono-doped LLZTO, dual-doped LLBZTO, and ternary-doped LGLBZTO.

<i>Garnet</i>	<i>GOF</i>	<i>R</i> _{wp}	<i>atom</i>	<i>site</i>	<i>Occupancy</i>	<i>X</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} (Å)
LLZO	1.77	10.36%	Li	24d	0.32*	0.375	0	0.25	0.021(1)
			Li	96h	0.36*	0.094(2)	0.685(2)	0.585(3)	0.021(1)
			Al	24d	0.06	3/8	0	1/4	0.01
			La	24c	1	0.125	0	0.25	0.0108(6)
			Zr	16a	1	0	0	0	0.00075(7)
			O	96h	1	0.9724(3)	0.0551(6)	0.1487(4)	0.0107(4)
LLZTO	2.34	8.86%	Li	24d	0.52*	0.375	0	0.25	0.14(4)
			Li	96h	0.33*	0.697(5)	0.599(5)	0.079(5)	0.09(4)
			La	24c	1	0.125	0	0.25	0.0014(5)
			Zr	16a	0.875**	0	0	0	0.0035(3)
			Ta	16a	0.125**	0	0	0	0.0015(3)
			O	96h	1	0.9724(7)	0.0651(7)	0.1426(7)	0.0107(4)
LLBZTO	2.20	9.92%	Li	24d	0.65*	0.375	0	0.25	0.20(7)
			Li	96h	0.33*	0.651(4)	0.599(4)	0.047(4)	0.20(5)
			La	24c	0.9833**	0.125	0	0.25	0.0014(5)
			Ba	24c	0.0167**	0.125	0	0.25	0.0014(5)
			Zr	16a	0.875**	0	0	0	0.0035(3)
			Ta	16a	0.125**	0	0	0	0.0015(3)
LGLBZTO	2.37	10.82%	Li	24d	0.77*	0.375	0	0.25	0.20(5)
			Li	96h	0.33*	0.7342(4)	0.5961(4)	0.055(4)	0.18(5)
			Ga	24d	0.0598(5)	0.375	0	0.25	0.20(7)
			La	24c	0.9833**	0.125	0	0.25	0.0014(5)
			Ba	24c	0.0167**	0.125	0	0.25	0.0014(5)
			Zr	16a	0.875**	0	0	0	0.0035(3)
			Ta	16a	0.125**	0	0	0	0.0015(3)
			O	96h	1	0.9731(9)	0.05726(9)	0.13875(9)	0.006(3)

* From neutron data **fixed

Table S2. The d(0.1), d(0.5), and d(0.9) values of undoped LLZO mono-doped LLZTO, dual-doped LLBZTO, and ternary-doped LGLBZTO.

Garnet	d(0.1)	d(0.5)	d(0.9)	*Span
<i>LLZO</i>	2.217 μm	2.734 μm	5.007 μm	1.02 μm
<i>LLZTO</i>	0.506 μm	3.493 μm	6.102 μm	1.60 μm
<i>LLBZTO</i>	0.538 μm	3.668 μm	7.926 μm	2.01 μm
<i>LGLBZTO</i>	0.553 μm	3.68 μm	8.138 μm	2.06 μm

$$*Span = \frac{D_{v0.9} - D_{v0.1}}{D_{v0.5}}$$

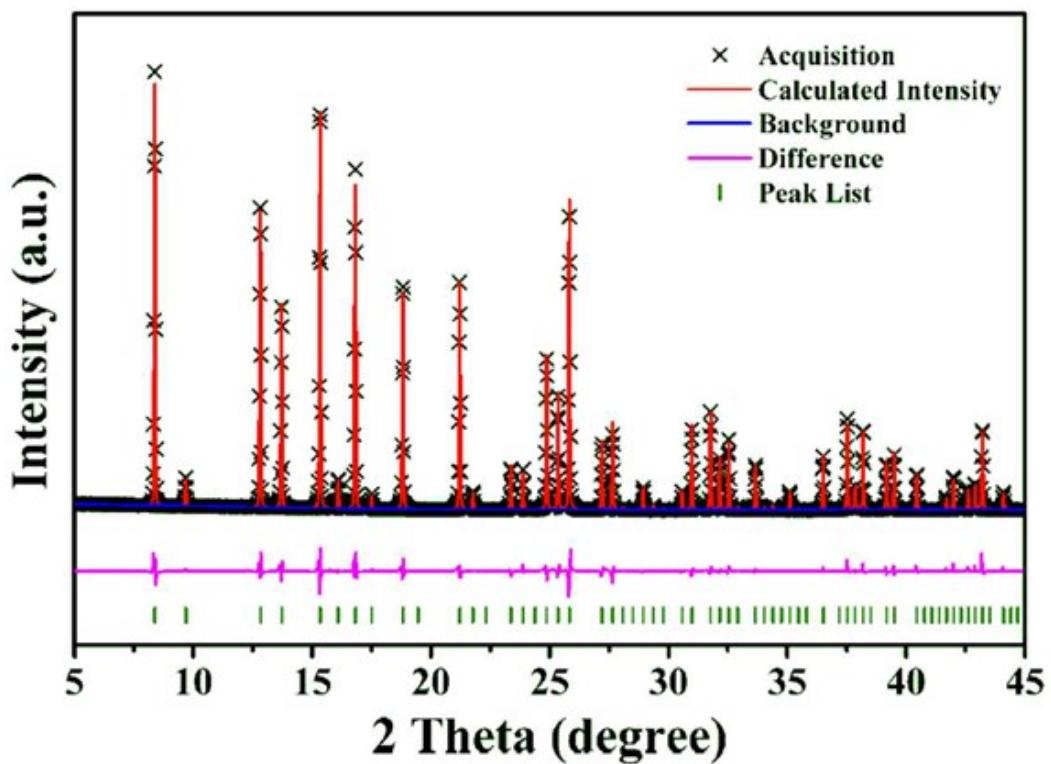


Fig. S1 Rietveld analysis of Synchrotron X-ray powder diffraction pattern of the undoped LLZO; observed (crosses), calculated, and difference profiles; vertical bars correspond to the calculated Bragg reflections for cubic garnet.

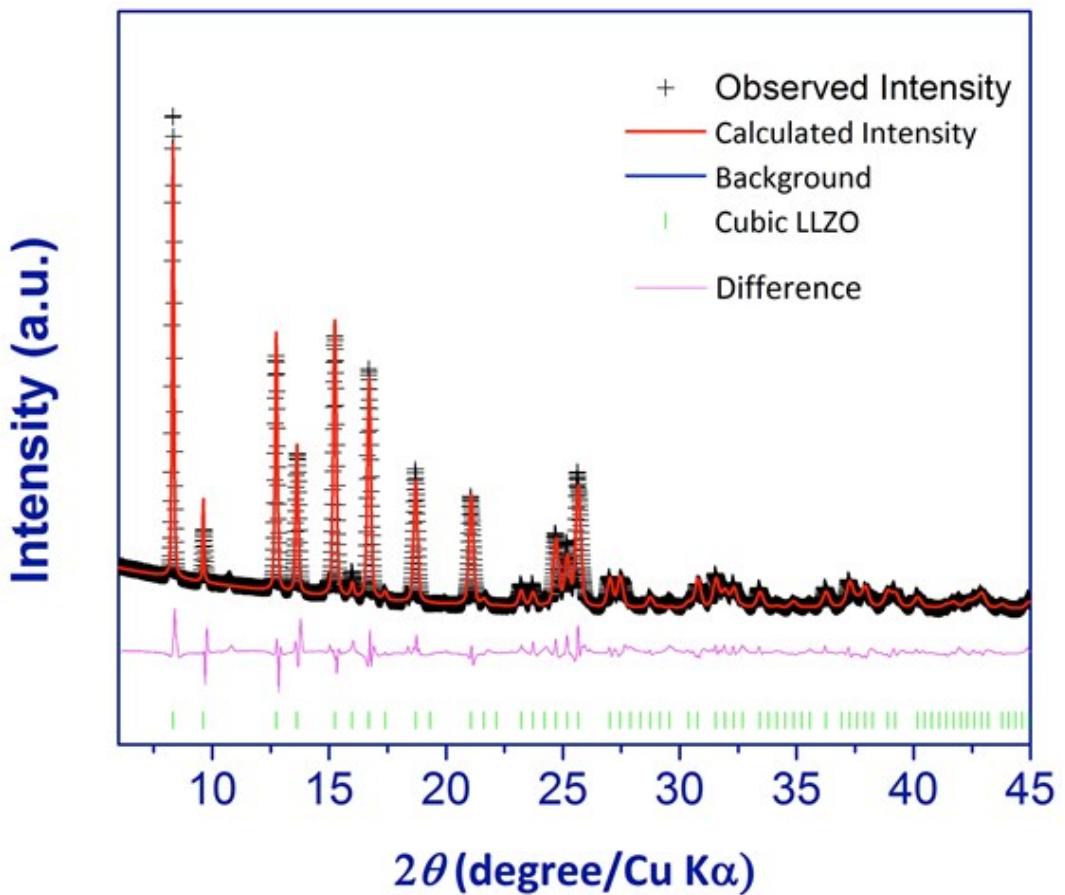


Fig. S2 Rietveld analysis of Synchrotron X-ray powder diffraction pattern for the mono-doped LLZTO; observed (crosses), calculated, and difference profiles; vertical bars correspond to the calculated Bragg reflections for cubic garnet.

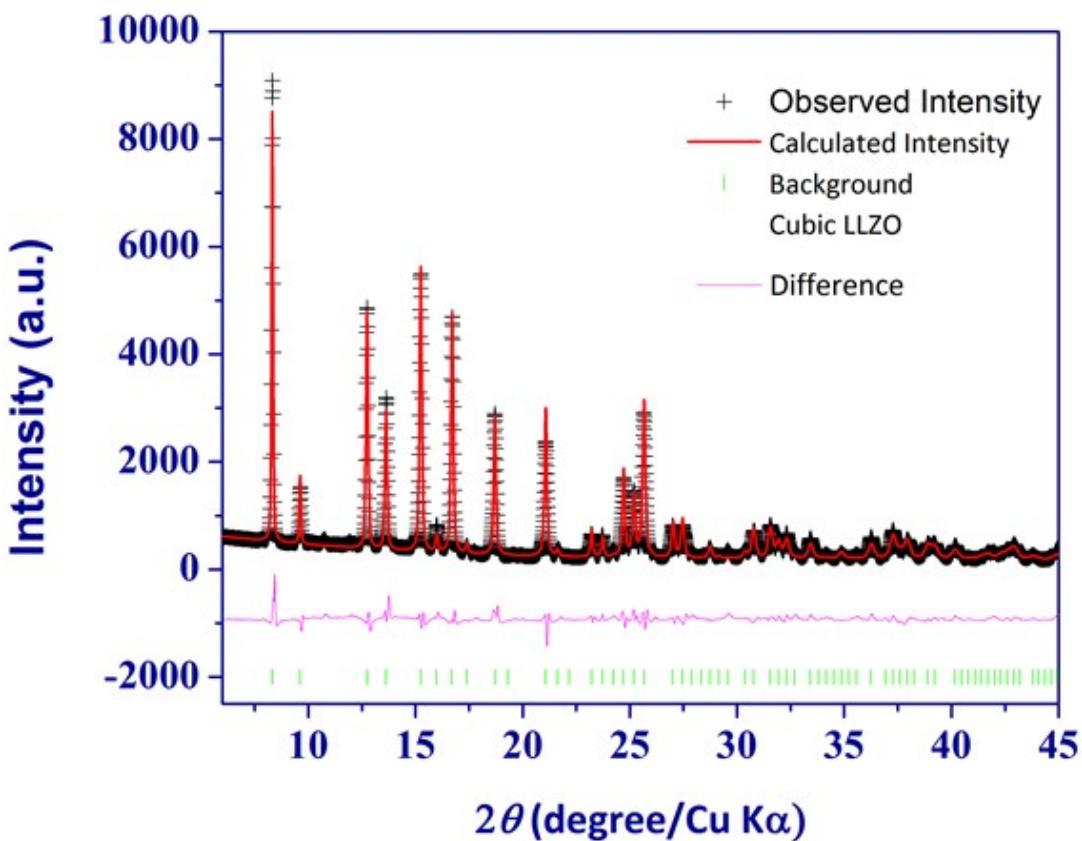


Fig. S3 Rietveld analysis of Synchrotron X-ray powder diffraction pattern for the dual-doped LLBZTO; observed (crosses), calculated, and difference profiles; vertical bars correspond to the calculated Bragg reflections for cubic garnet.

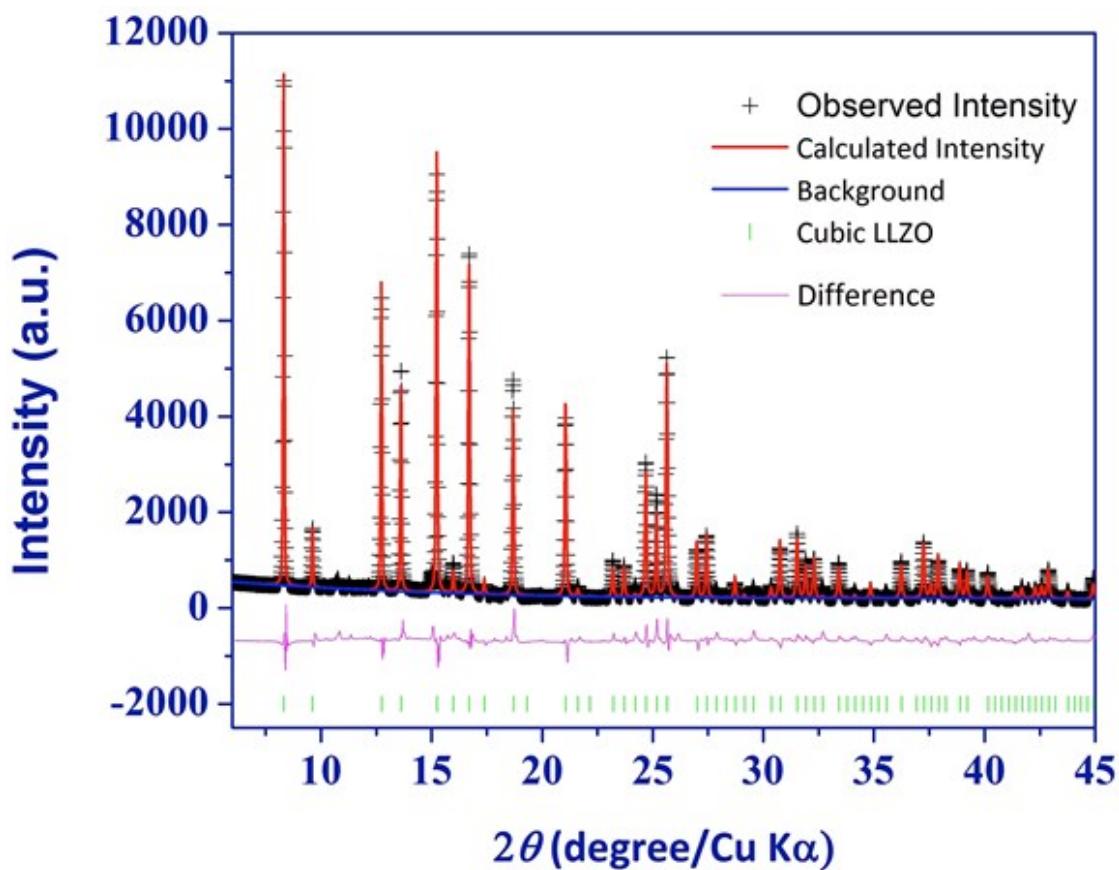


Fig. S4 Rietveld analysis of Synchrotron X-ray powder diffraction pattern for the ternary-doped LGLBZTO; observed (crosses), calculated, and difference profiles; vertical bars correspond to the calculated Bragg reflections for cubic garnet.

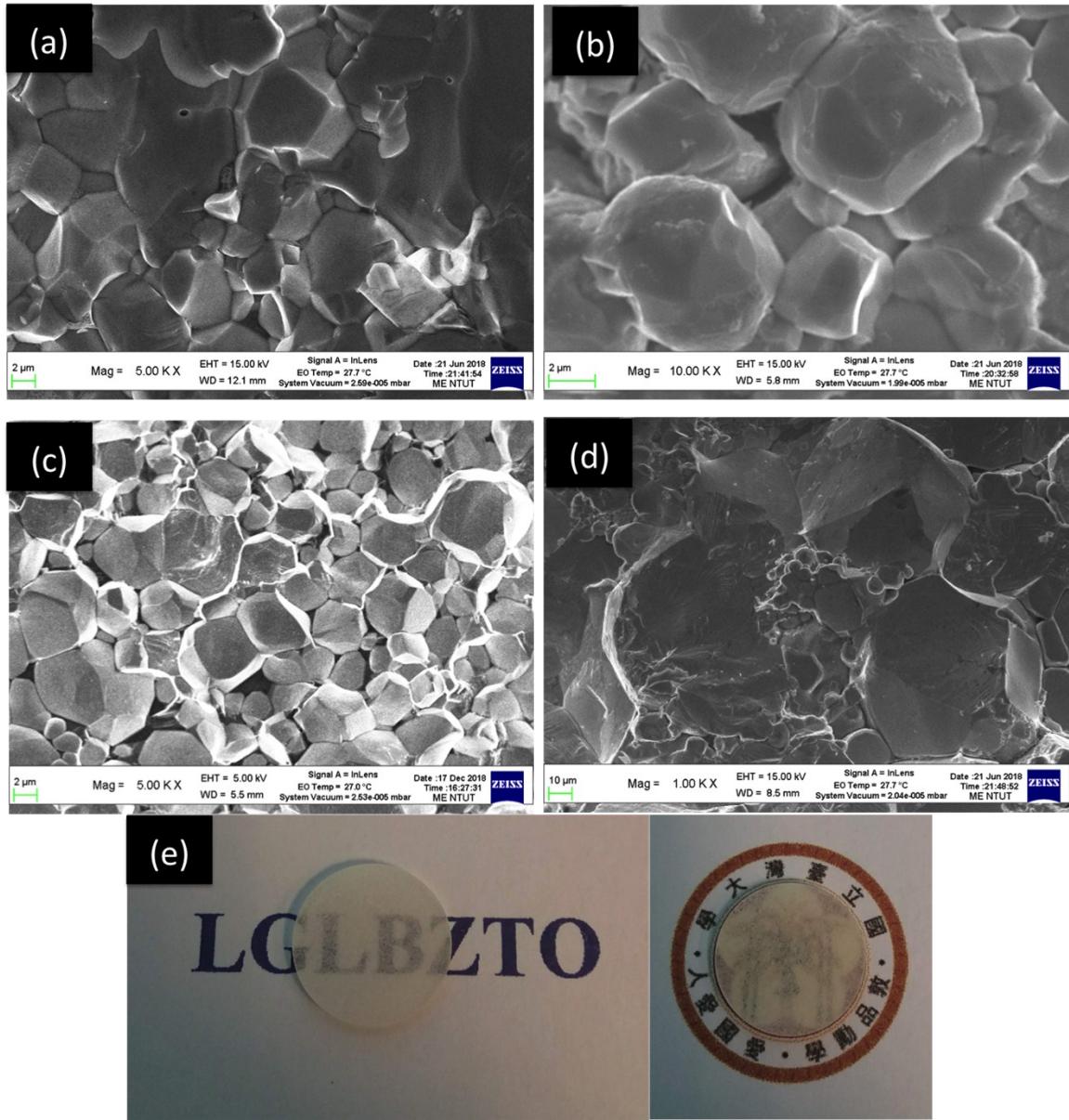


Fig. S5 FE-SEM images of the (a) undoped LLZO, (b) mono-doped LLZTO, (c) dual-doped LLBZTO, and (d) ternary-doped LGLBZTO (e) Optical photographs of LGLBZTO with 0.5mm thickness.

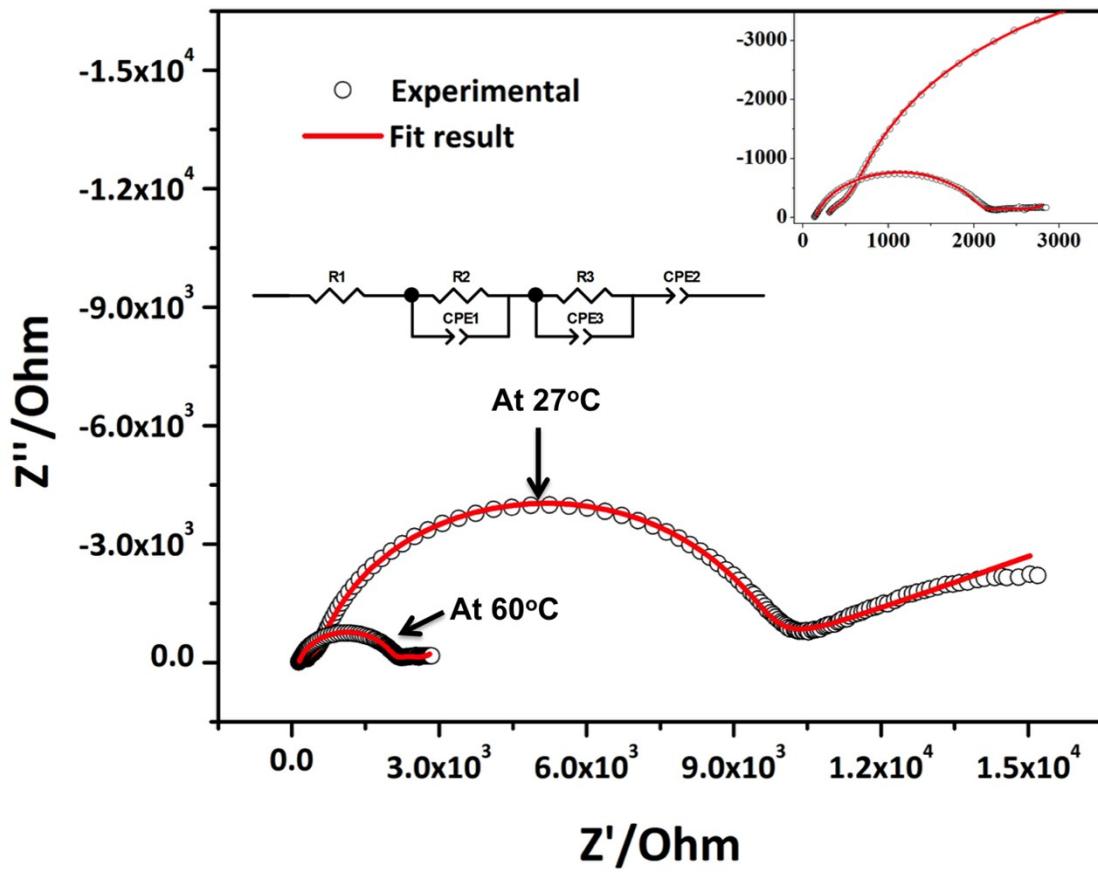


Fig. 6 Impedance spectra of the symmetrical cell $\text{Li}|\text{LGLBZTO}|\text{Li}$ measured at 27 and 60 $^\circ\text{C}$.

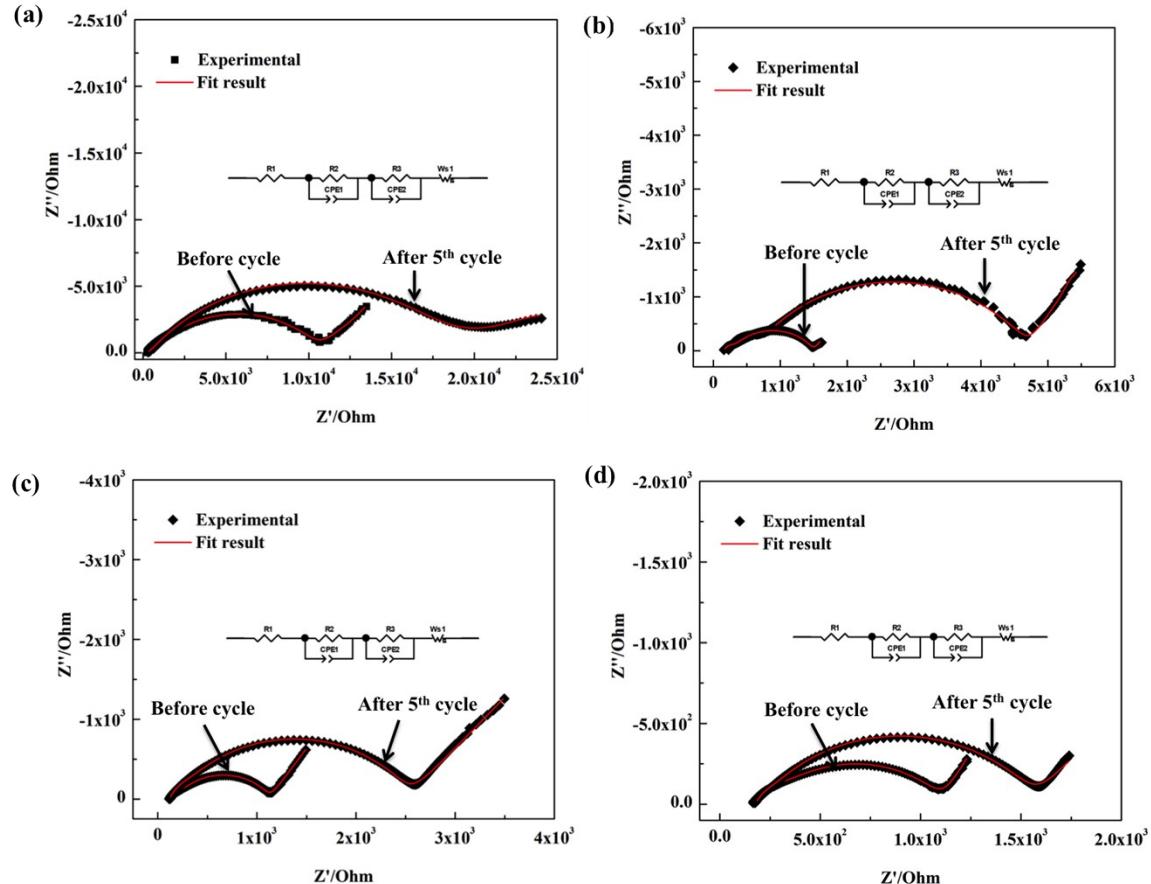


Fig. 7 Impedance spectra of the symmetrical cell with the configuration of (a) Li|LLZO|LFP (b) Li|LLZTO|LFP (c) Li|LLBZTO|LFP and (d) Li|LGLBZTO|LFP recorded before and after 5th cycle at 60 °C.