

Supporting Information for

Microbelt-void-Microbelt Structured SnO_2/C as Advanced Electrode with Outstanding Rate Capability and High Reversibility

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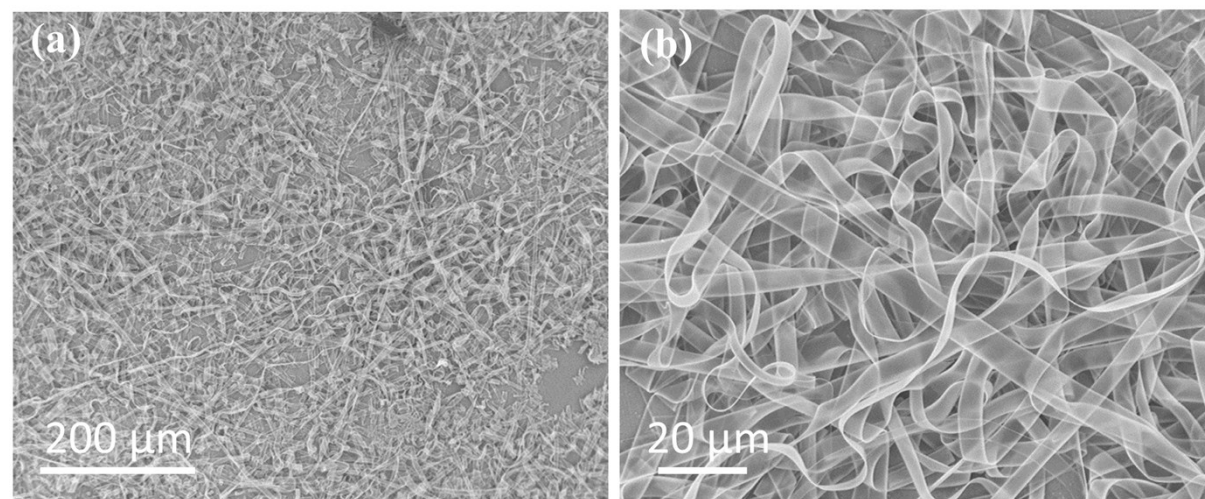


Figure S1. (a, b) SEM images of electrospinning products without annealing treatment.

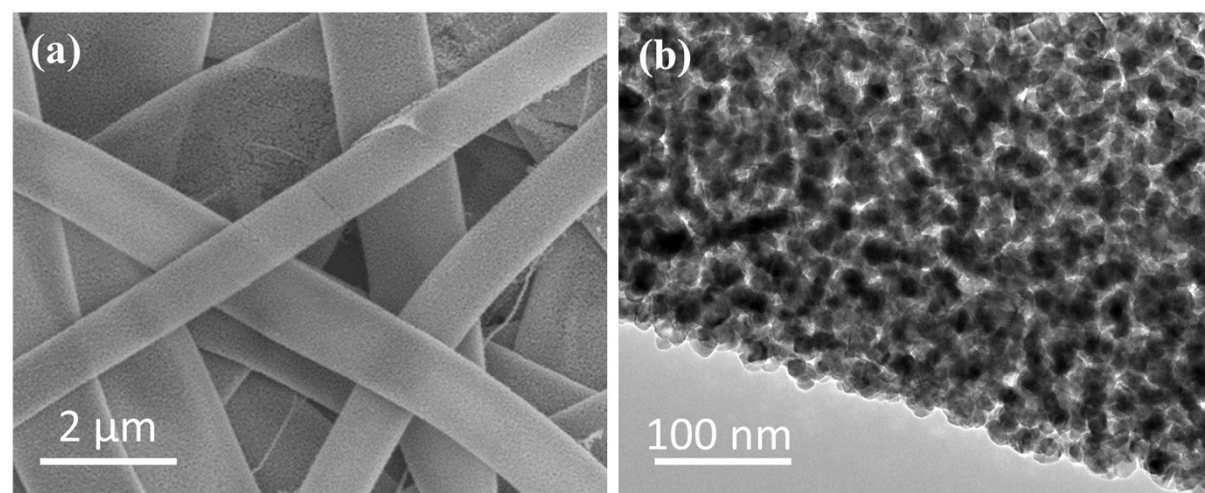


Figure S2. Morphological characterization of SnO_2 microbelts. (a-b) SEM images, (c-d) TEM of SnO_2 microbelts.

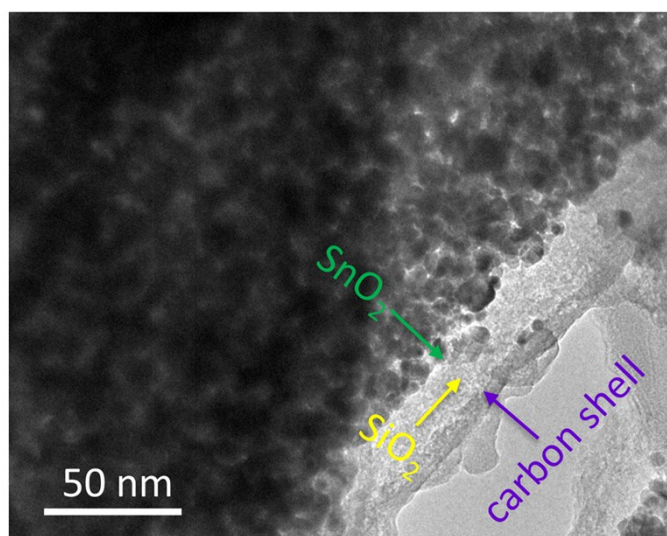


Figure S3. TEM images of the SnO₂ microbelt coated with SiO₂ and carbonized polydopamine.

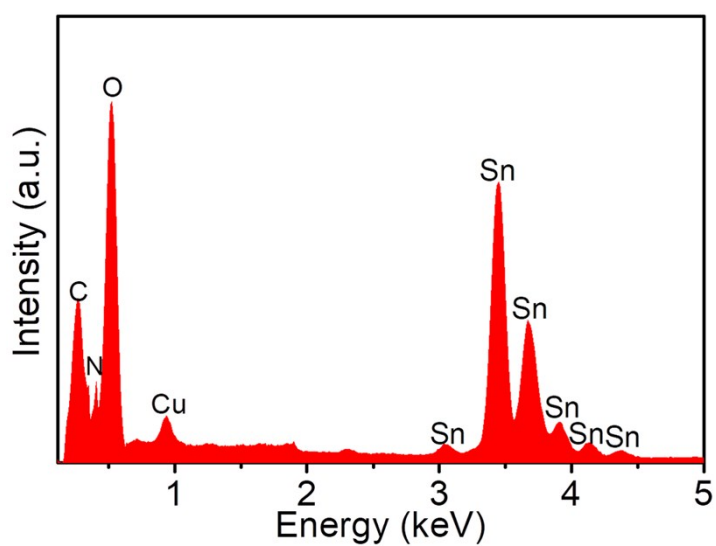


Figure S4. EDX spectrum of microbelt-void-microbelt structured SnO₂@C, showing the mass ratio of Sn, O, C and N are 54.47, 27.72, 15.29 and 2.51wt%, respectively.

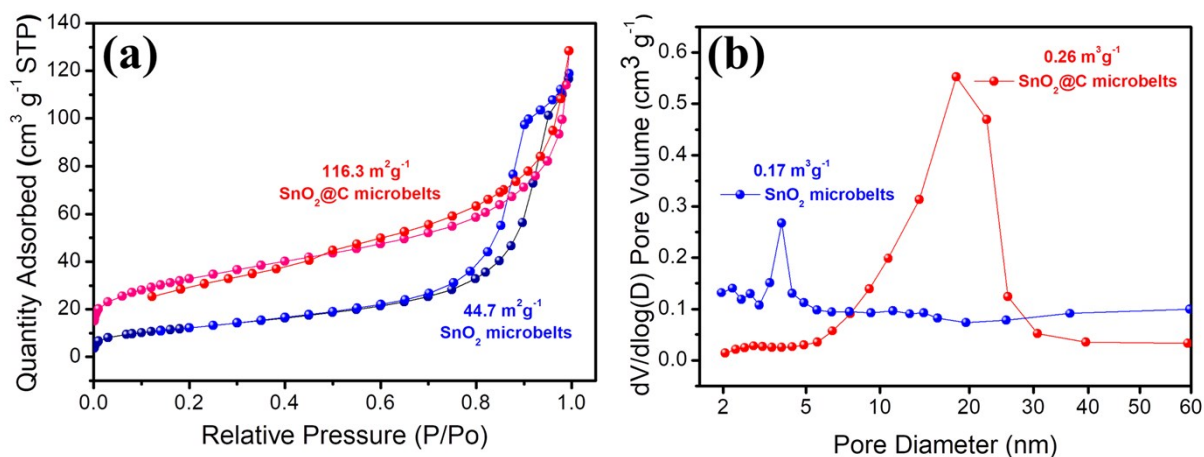


Figure S5. (a) Nitrogen adsorption-desorption isotherms of the SnO₂ microbelt and microbelt-void-microbelt structured SnO₂@C sample. (b) The corresponding pore size distribution.

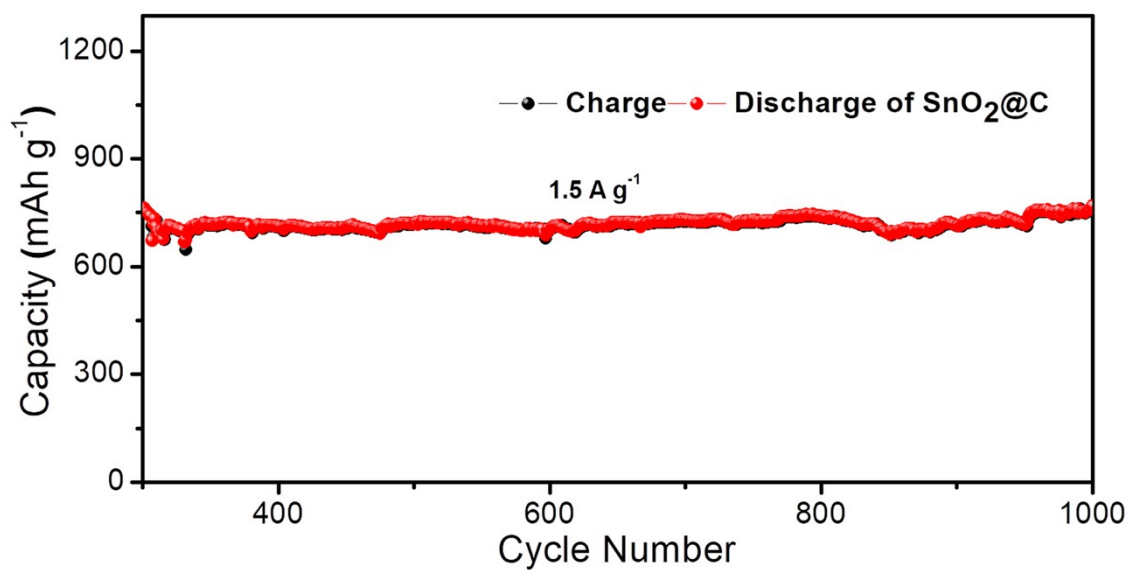


Figure S6. Cyclic performance of microbelt-void-microbelt structured SnO₂@C at 1.5 A g⁻¹ after 300 cycles at 0.3 A g⁻¹.

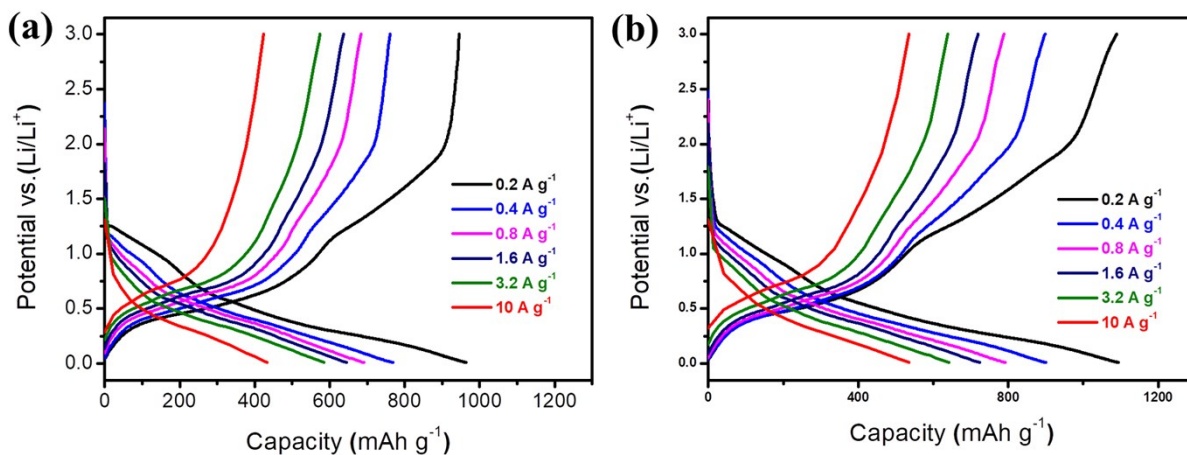


Figure S7. Charge and discharge profiles of (a) SnO_2 and (b) microbelt-void-microbelt structured $\text{SnO}_2@C$ electrode at various current densities.

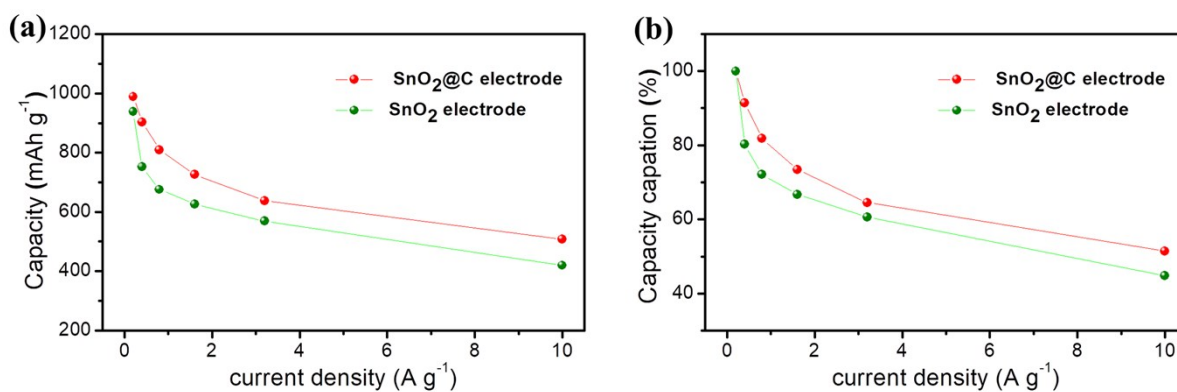


Figure S8. (a) Capacity and (b) capacity retention rate vs. current density plot of the two microbelt electrodes.

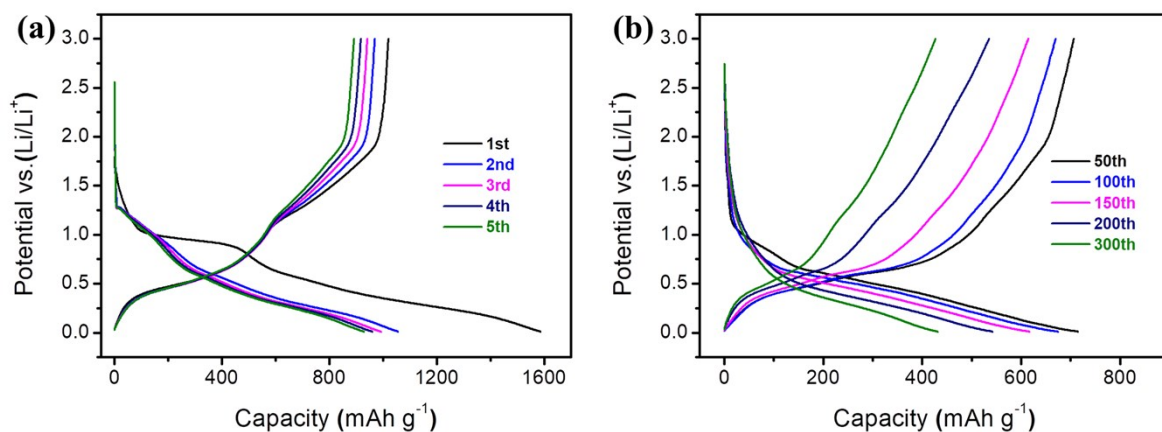


Figure S9. (a) Galvanostatic discharge-charge profiles of the SnO₂ microbelt electrode in the initial 5 cycles. (b) Galvanostatic discharge-charge profiles of the SnO₂ microbelt electrode in the 50th, 100th, 200th, 250th, and 300th cycles.

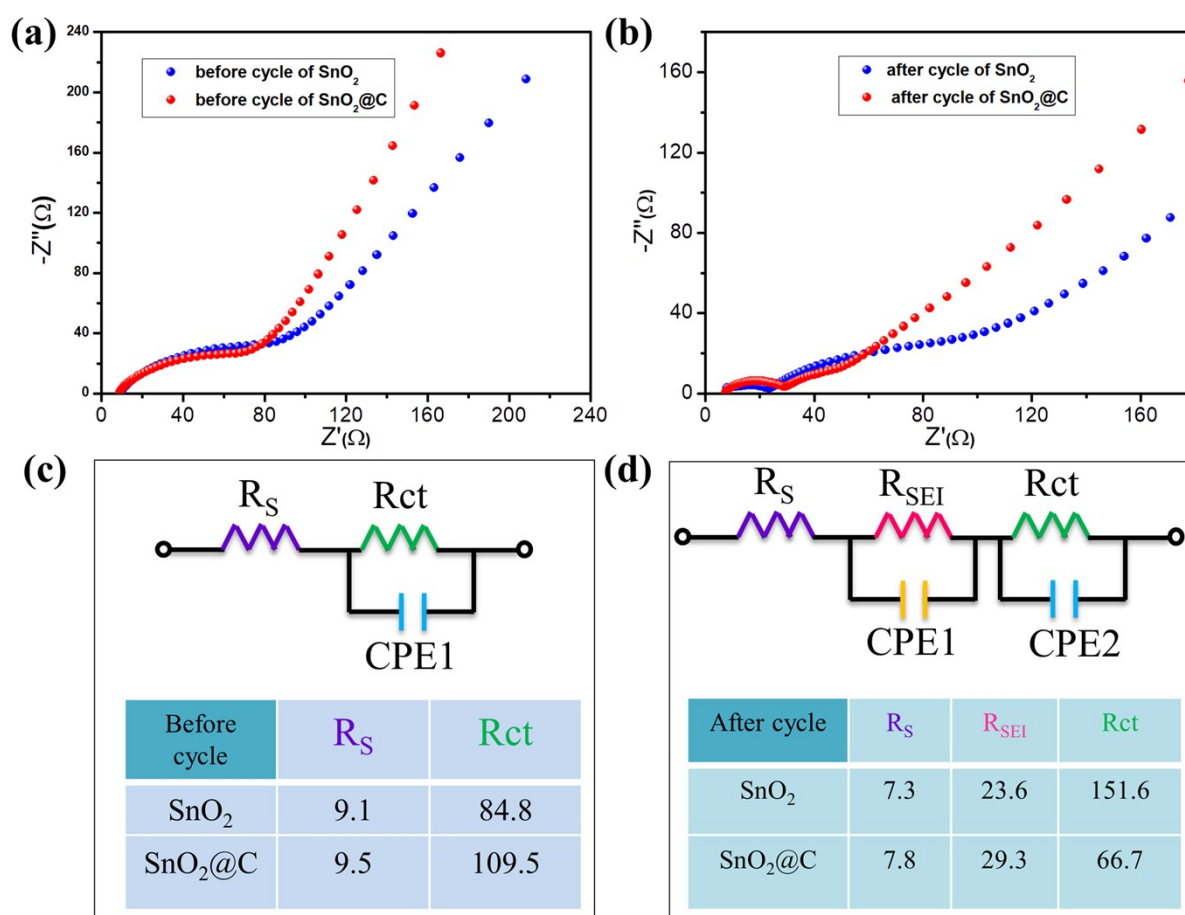


Figure S10. Electrochemical impedance spectroscopy of SnO₂ and SnO₂@C electrode: (a) Before cycle, (b) after cycle, (c,d) the corresponding equivalent circuit and calculated resistances.