## **Supporting Information**

## Structural Impact of Zn-insertion into Monoclinic V<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>: Implications for Zn-ion Batteries

Min Je Park,<sup>a</sup> Hooman Yaghoobnejad Asl,<sup>a</sup> Soosairaj Therese<sup>b</sup> and Arumugam Manthiram\*<sup>a</sup>

<sup>a</sup> Materials Science and Engineering Program & Texas Materials Institute, The University of Texas at Austin, Austin, Texas 78712, USA

<sup>b</sup> Department of Chemistry and Chemical Technology, Bronx Community College, Bronx, New York 10453, USA

\* To whom correspondence should be addressed.

Tel. : (+1) 512-471-1791; Fax. : (+1) 512-475-8482; E-mail : manth@austin.utexas.edu



Fig. S1 Potential measured against equivalent volume of  $KMnO_4$  for the redox titration of Zninserted  $V_2(PO_4)_3$  obtained via microwave irradiation at 140 °C for 10 min.



**Fig. S2** XRD patterns of  $V_2(PO_4)_3$  before microwave irradiation, after microwave irradiation at 50 °C for 10 min and 1 h, and after microwave irradiation at 80 °C for 10 min, 1 h, and 3 h.



**Fig. S3** (a) Linear sweep voltammetry (LSV) of  $\text{Li}_3\text{V}_2(\text{PO}_4)_3$  electrode for electrochemical preparation of  $\text{V}_2(\text{PO}_4)_3$  and (b) discharge profile of the electrode after the LSV step.



Fig. S4 Cyclic voltammetry (CV) of V<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> electrode at 25 °C (black) and 80 °C (red).



Fig. S5 CV of (a)  $LiV_2(PO_4)_3$  electrode and (b)  $Li_3V_2(PO_4)_3$  electrode in non-aqueous Zn-ion set-

up.



**Fig. S6** XRD patterns of Zn-inserted  $V_2(PO_4)_3$  via microwave-irradiation (black) and  $V_2(PO_4)_3$  electrode after electrochemical Zn-insertion (red) (\*: Peaks due to Toray paper).



Fig. S7 Observed PDF curves of  $Li_3V_2(PO_4)_3$  (top),  $V_2(PO_4)_3$  (middle), and  $Zn_{0.36}V_2(PO_4)_3$ 

(bottom) in the range of 1 to 20 Å.

**Table S1** Refined cell parameters and statistics, atomic coordinates, and isotropic thermal parameters obtained from Rietveld refinement of the powder XRD of  $V_2(PO4)_3$  prepared *via* chemical oxidation

Refined Cell Parameters and Statistics									
Space Group : P 2 <sub>1</sub> /c $a = 8.3058(7)$ Å; $b = 8.58721(9)$ Å; $c = 14.02173(7)$ Å; $\alpha = \gamma = 90.0000$ °; $\beta = 125.663(9)$ ° V = 812.52(5) Å <sup>3</sup> $R_{wp} = 11.008$ %; $R_F^2 = 4.681$ %									
Atomic Coordinates									
Label	х	у	Z	U <sub>iso</sub> (Å <sup>2</sup> )	Fractions	Site			
V1	0.3736(3)	0.46588(4)	0.11718(5)	0.012018(4)	1	4e			
V2	0.13433(6)	0.4708(3)	0.38658(7)	0.01(2)	1	4e			
P1	0.25922(3)	0.11785(5)	0.14878(3)	0.016616(5)	1	4e			
P2	0.037(4)	0.6(1)	0.143(5)	0.016(6)	1	4e			
P3	0.54312(7)	0.251(6)	0.00590(9)	0.016616(5)	1	4e			
01	0.11384(6)	0.08264(7)	0.17768(9)	0.0063040(5)	1	4e			
O2	0.43921(1)	0.00892(9)	0.21929(9)	0.0063040(5)	1	4e			
O3	0.15709(3)	0.07947(3)	0.01980(7)	0.0063040(5)	1	4e			
O4	0.30249(2)	0.2940(7)	0.1605(3)	0.0063040(5)	1	4e			
05	0.173(9)	0.600(1)	0.106(6)	0.0063040(5)	1	4e			
06	0.104(8)	0.490(2)	0.241(8)	0.0063040(5)	1	4e			
07	0.183(3)	0.088(9)	0.455(4)	0.0063040(5)	1	4e			
08	0.055(1)	0.771(7)	0.193(7)	0.006(3)	1	4e			
09	0.55540(4)	0.32259(8)	0.11224(2)	0.0063040(5)	1	4e			
O10	0.37451(5)	0.35090(2)	0.4383(1)	0.0063040(5)	1	4e			
011	0.54495(6)	0.12182(9)	0.42613(5)	0.0063040(5)	1	4e			
O12	0.73979(4)	0.1657(3)	0.05686(4)	0.0063040(5)	1	4e			

**Table S2** Elemental ratio determined via ICP measurement of  $LiV_2(PO_4)_3$  electrode after firstreduction and subsequent oxidation in a Zn-ion cell

Elemental ratio (per 2 V atoms)						
	After 1 <sup>st</sup> reduction	After subsequent oxidation				
Li : V	1.14	0.38				
Zn : V	0.79	0.58				

Table S3 Refined cell parameters and statistics, atomic coordinates, and isotropic thermal

parameters obtained from Rietveld refinement of the powder XRD of  $Zn_{0.36}V_2(PO4)_3$  prepared

via microwave-assisted chemical insertion

Refined Cell Parameters and Statistics											
Space Group : P 2 <sub>1</sub> /c $a = 8.508(7)$ Å; $b = 8.682(4)$ Å; $c = 14.590(8)$ Å; $\alpha = \gamma = 90.0000$ °; $\beta = 125.18(2)$ ° V = 881.00(3) Å <sup>3</sup> R <sub>wp</sub> = 16.222 %; R <sub>F</sub> <sup>2</sup> = 15.800 %											
Atomic Coordinates											
Label	х	У	Z	U <sub>iso</sub> (Å <sup>2</sup> )	Fractions	Site					
V1	0.1612(9)	0.4781(6)	0.3971(8)	0.0066(8)	1	4e					
V2	0.3717(5)	0.4631(9)	0.1174(2)	0.0066(8)	1	4e					
P1	0.0317(2)	0.5985(2)	0.1392(5)	0.014(9)	1	4e					
P2	0.2575(4)	0.1186(1)	0.1477(2)	0.014(9)	1	4e					
P3	0.532(4)	0.2397(5)	0.4976(5)	0.014(9)	1	4e					
01	0.211(8)	0.6096(3)	0.1444(1)	0.016(9)	1	4e					
O2	0.0761(8)	0.5002(2)	0.2332(9)	0.016(9)	1	4e					
O3	0.1249(7)	0.0387(7)	0.4723(8)	0.016(9)	1	4e					
O4	0.0111(6)	0.257(8)	0.3317(4)	0.016(9)	1	4e					
05	0.0921(8)	0.1107(6)	0.1646(3)	0.016(9)	1	4e					
O6	0.5989(5)	0.4866(8)	0.2839(9)	0.016(9)	1	4e					
07	0.1889(2)	0.1066(9)	0.0314(3)	0.016(9)	1	4e					
08	0.3690(9)	0.265(5)	0.202(7)	0.016(9)	1	4e					
O9	0.3515(7)	0.3263(7)	0.4155(9)	0.016(9)	1	4e					
O10	0.5134(7)	0.3634(5)	0.0709(9)	0.016(9)	1	4e					
O11	0.699(9)	0.1700(3)	0.0823(8)	0.016(9)	1	4e					
O12	0.5683(2)	0.1674(9)	0.4153(2)	0.016(9)	1	4e					
Zn1	0.0174(4)	0.787(6)	0.3026(9)	0.00(5)	0.36(5)	4e					