

Supporting Information

Ion-matching porous carbons with ultra-high surface area and superior energy storage performance for supercapacitors

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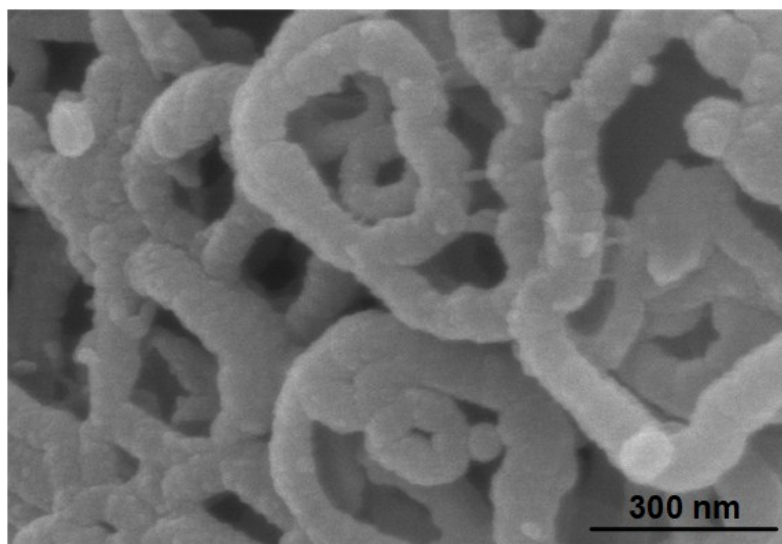


Figure S1. SEM image of PPy rings.

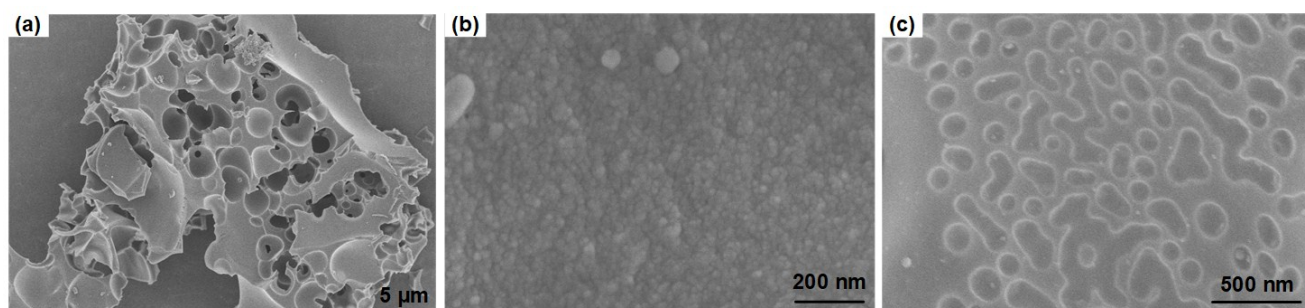


Figure S2. SEM images of PPC-700: (a) at low-magnification; (b) accidented surface; (c) wormlike pits.

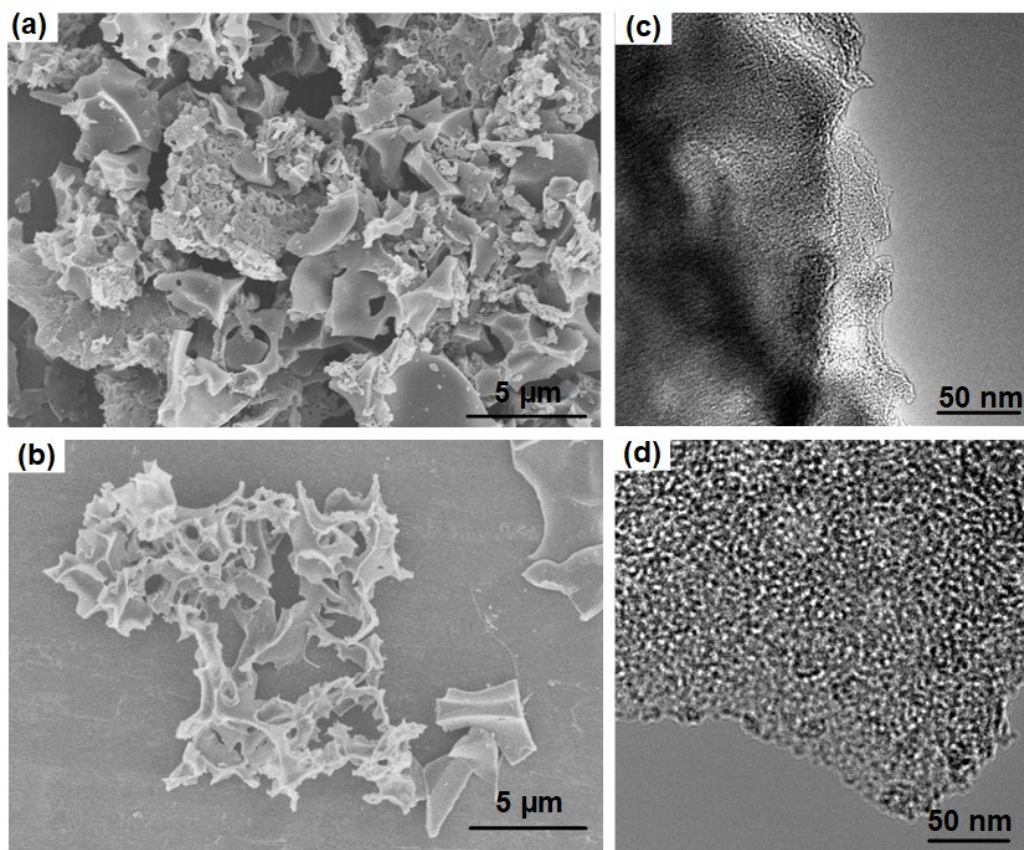


Figure S3. The morphology of the PPC-600 and PPC-800: (a) SEM image of PPC-600. (b) SEM image of PPC-800. (c) TEM image of PPC-600. (d) TEM image of PPC-800.

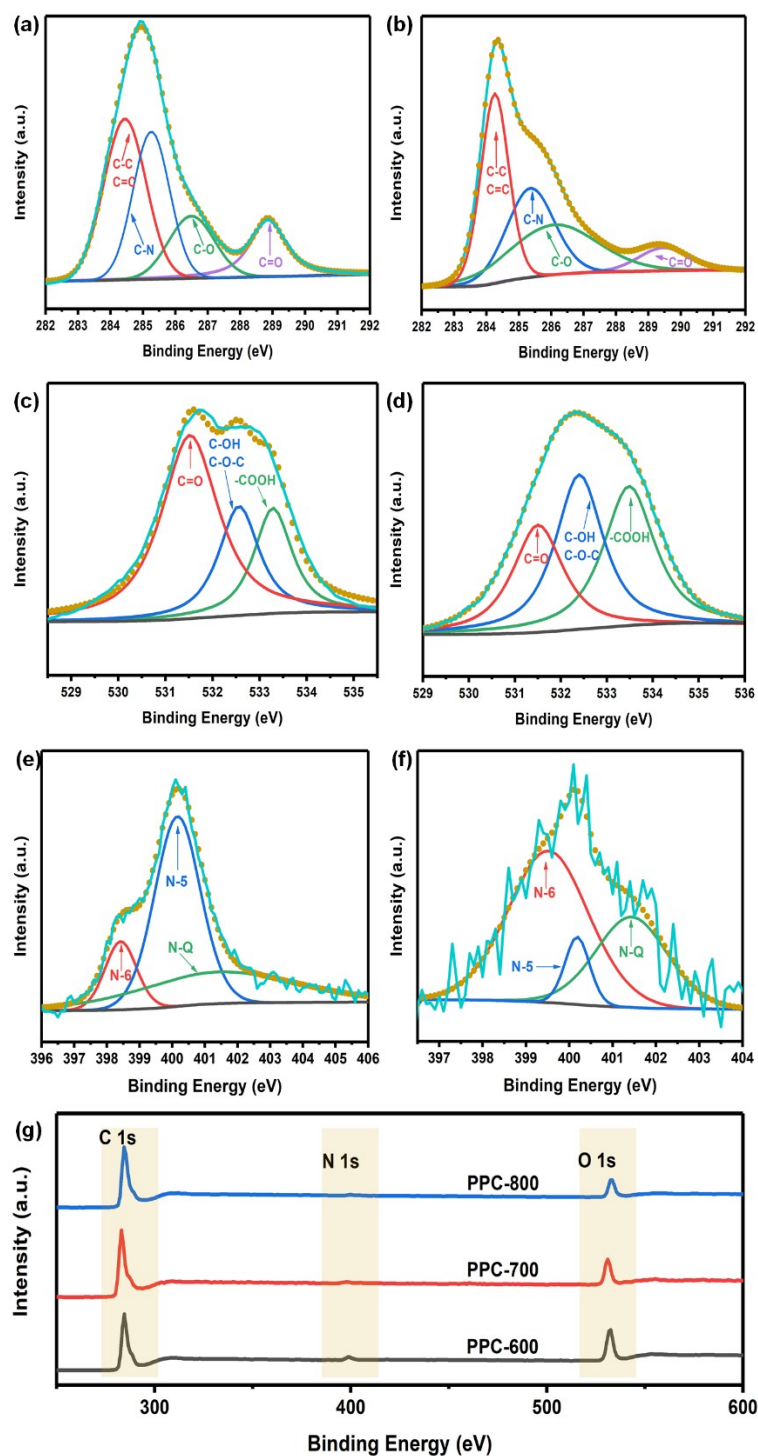


Figure S4. High-resolution XPS spectra of PPC-600 (a) C 1s, (c) O 1s, and (e) N 1s; PPC-800 (b) C 1s, (d) O 1s, and (f) N 1s; (g) XPS survey spectra of PPC-600, PPC-700 and PPC-800.

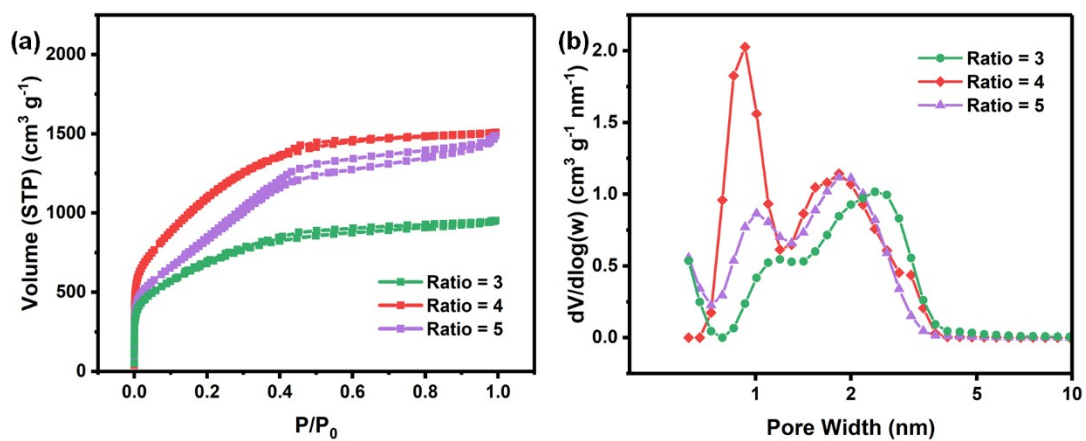


Figure S5. (a) Nitrogen adsorption-desorption isotherms and (b) the corresponding pore size distribution of PPC samples prepared by different KOH amount at 700 °C.

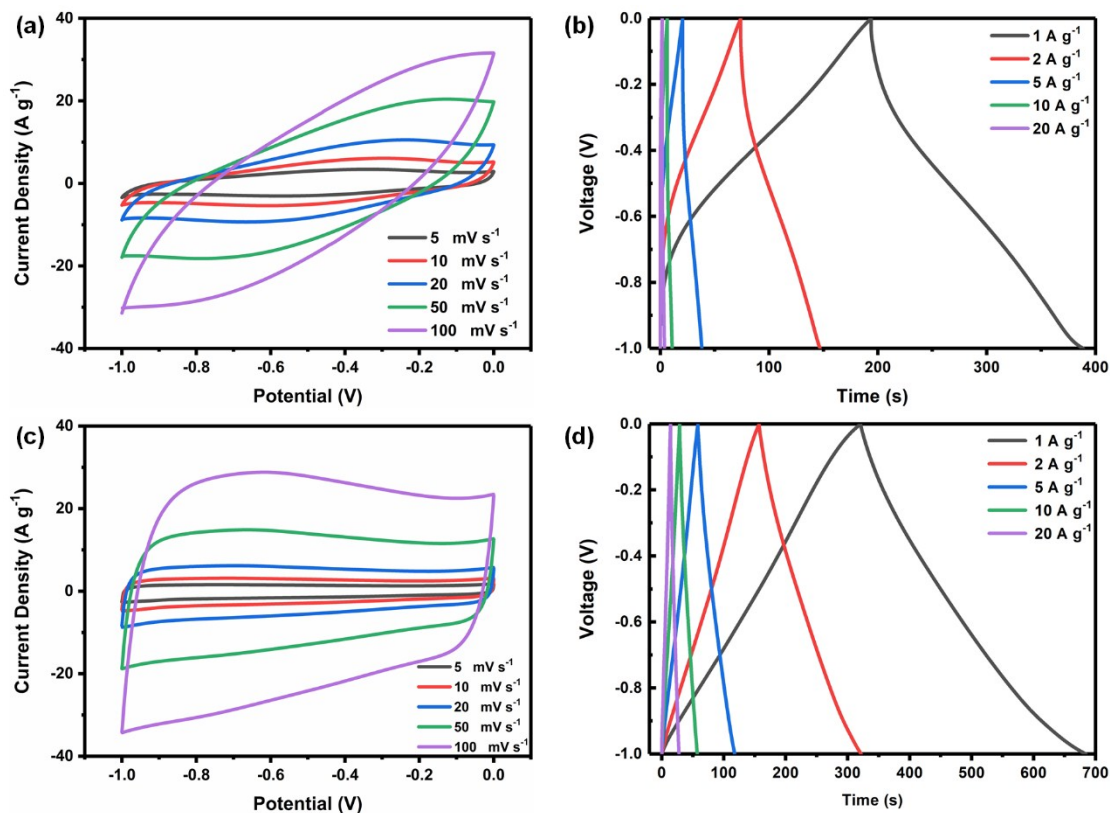


Figure S6. Electrochemical performance of PPC-600 and PPC-800 in 3-electrode test using 6 M KOH electrolyte with a potential window of 1.0 V: (a) Cyclic voltammetry of PPC-600 at various scan rates of 5, 10, 20, 50 and 100 mV s⁻¹; (b) Galvanostatic charge-discharge profiles of PPC-600 at current densities of 1, 2, 5, 10 and 20 A g⁻¹; (c) Cyclic voltammetry of PPC-800 at various scan rates of 5, 10, 20, 50 and 100 mV s⁻¹; (d) Galvanostatic charge-discharge profiles of PPC-800 at current densities of 1, 2, 5, 10 and 20 A g⁻¹.

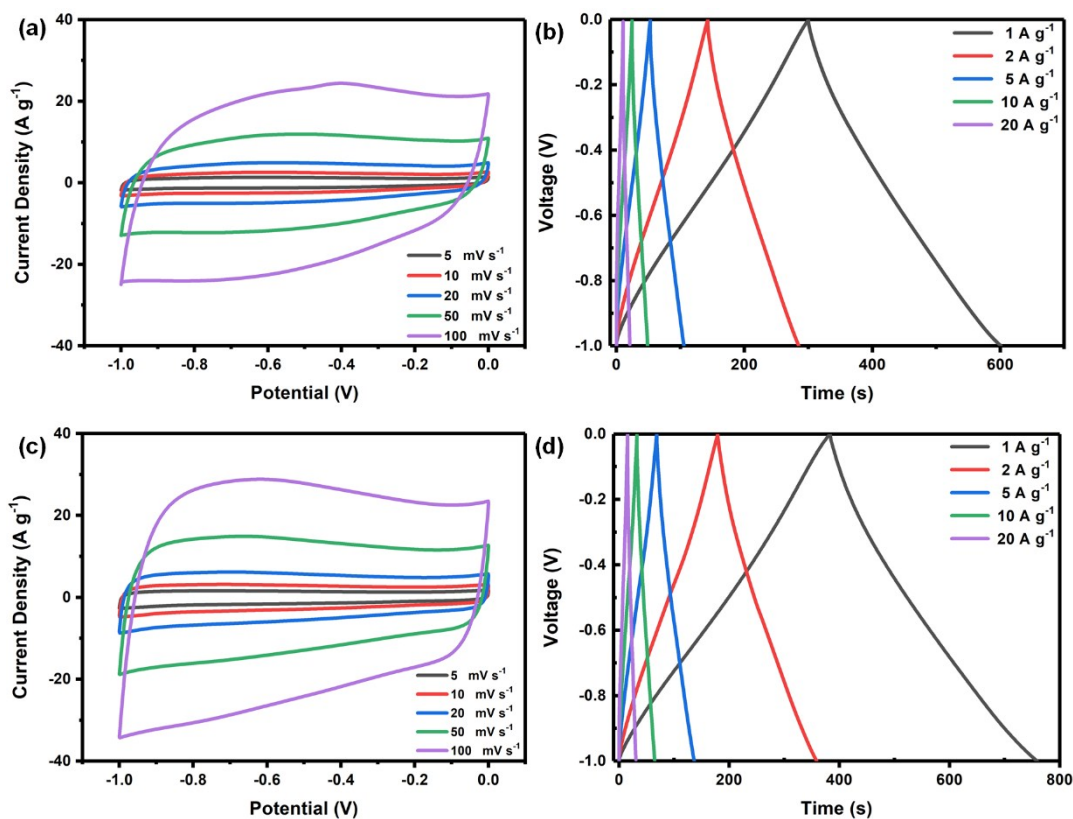


Figure S7. Electrochemical performance of PPC samples prepared by different KOH amount at 700 °C in 3-electrode test using 6 M KOH electrolyte with a potential window of 1.0 V: (a) Cyclic voltammety at various scan rates of 5, 10, 20, 50 and 100 mV s⁻¹ when mass ratio of KOH : PPy equals 3; (b) Galvanostatic charge-discharge profiles at current densities of 1, 2, 5, 10 and 20 A g⁻¹ when mass ratio of KOH : PPy equals 3; (c) Cyclic voltammety at various scan rates of 5, 10, 20, 50 and 100 mV s⁻¹ when mass ratio of KOH : PPy equals 5; (d) Galvanostatic charge-discharge profiles at current densities of 1, 2, 5, 10 and 20 A g⁻¹ when mass ratio of KOH : PPy equals 5.

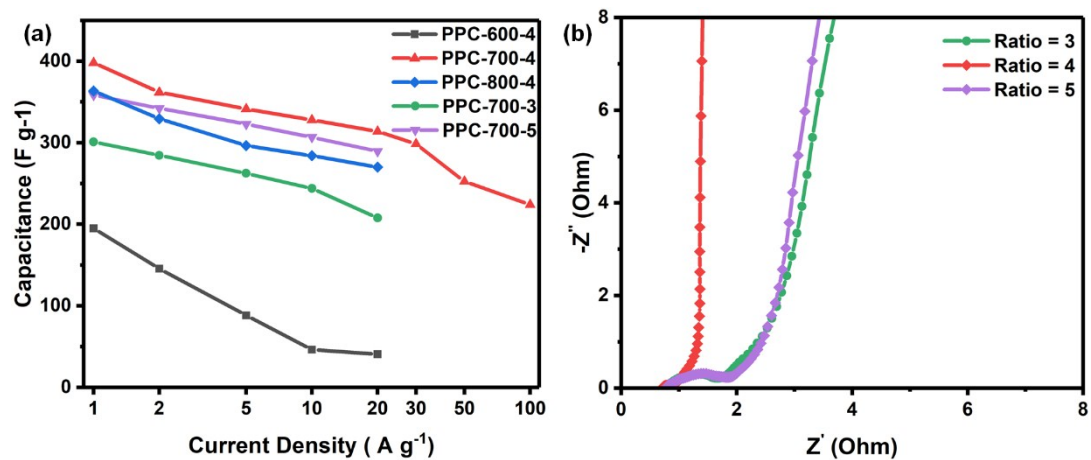


Figure S8. (a) capacitive performance at different current densities (PPC-X-Y: X means temperature, Y means KOH amount.) (b) Nyquist plots of PPC samples prepared by different KOH amount at 700 °C in a frequency range of 0.01 to 10⁵Hz with 5mV amplitude.

Table S1. Textural properties of PPC samples prepared by different KOH amount at 700 °C

ratio ^a	SSA ^b (m ² g ⁻¹)	Pore Volume (cm ³ g ⁻¹)			
		V _{Total}	V ^c _{Small} (%)	V ^d _{Close} (%)	V ^e _{Large} (%)
3	3156.38	2.301	0.241(11%)	0.514(22%)	1.546(67%)
4	4091.54	2.334	0.062(2%)	0.946(41%)	1.326(57%)
5	3699.61	2.060	0.263(13%)	0.611(30%)	1.186(57%)

a) KOH : PPy

b) Specific Surface Area calculated by Brunauer-Emmett-Teller (BET) method in the relative pressure range of 0.05-0.20

c) Volume of pores smaller than 0.5 nm in width

d) Volume of pores between 0.5 and 1 nm in width

e) Volume of pores larger than 1 nm in width

Table S2. Summary of capacitive performance of carbon-based supercapacitors

Carbon source	SSA (m ² g ⁻¹)	Capacitance (F g ⁻¹)	Current density (A g ⁻¹)	Electrolyte	Cycles (Current density/Scanning rate)	Capacitance retention	Ref.
PPy	4091.54	397.9	1A g ⁻¹	6 M KOH	30K (10A g ⁻¹)	95%	This work
PPy	2870	318.2	0.5A g ⁻¹	6 M KOH	10K (5A g ⁻¹)	95.8%	[1]
PANi	4073	225	0.5A g ⁻¹	0.5 M H ₂ SO ₄	10K (5A g ⁻¹)	96%	[2]
chitosan	2905	374.7	1A g ⁻¹	3 M KOH	20K (200mV/s)	98%	[3]
popcorn	3301	311	10A g ⁻¹	6 M KOH	10K (5A g ⁻¹)	95%	[4]
ZIF-8/PAN	417.9	307.2	1A g ⁻¹	2 M H ₂ SO ₄ (two-electrode)	10K (5A g ⁻¹)	98.2%	[5]
non-fat powdered milk	4051	247	25 A g ⁻¹	6 M KOH (two-electrode)	75K (25A g ⁻¹)	80%	[6]
commercial silicone resin	2896	322	0.5 A g ⁻¹	6 M KOH	10K (5A g ⁻¹)	93.3%	[7]
petroleum asphalt	3581	194	20 A g ⁻¹	6 M KOH	10K (2A g ⁻¹)	95.1%	[8]
phenol formaldehyde resin	1224	388	1A g ⁻¹	1 M H ₂ SO ₄	8K (5A g ⁻¹)	98%	[9]
polyphosphazene	1798	105	0.3A g ⁻¹	6 M KOH (two-electrode)	10K (5A g ⁻¹)	89.5%	[10]
PMMA/bacterial cellulose	2076	266	0.5A g ⁻¹	1 M H ₂ SO ₄	10K (5A g ⁻¹)	95%	[11]
bradyrhizobium japonicum	1275	358	1A g ⁻¹	6 M KOH (two-electrode)	8K (1A g ⁻¹)	91%	[12]
carbon aerogel	450	151	0.5A g ⁻¹	6 M KOH	0.5K (0.5A g ⁻¹)	97.8%	[13]

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