

Deep eutectic solvent route synthesis of zinc and copper vanadate *n*-type semiconductors – mapping oxygen vacancies and their effect on photovoltage

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Supporting Information

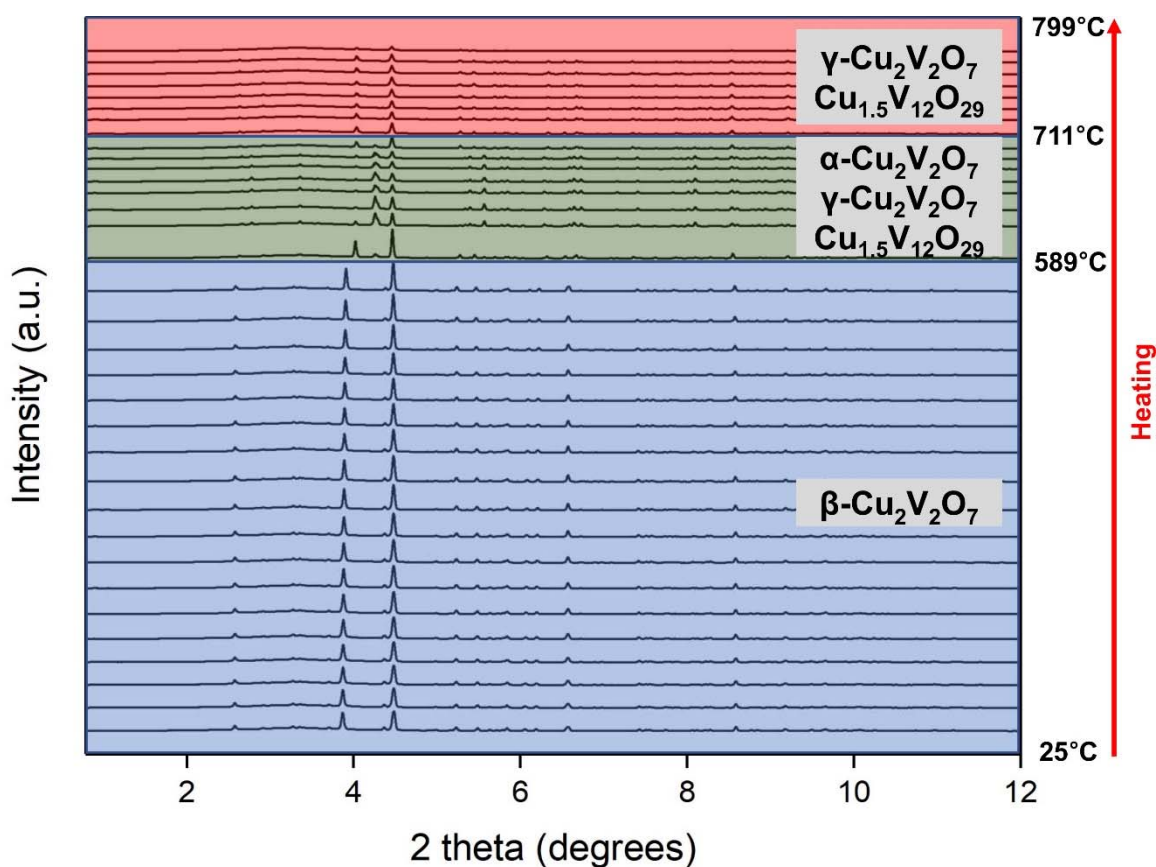


Figure S1. In-situ high-temperature powder X-ray diffraction (HT PXRD) patterns of $\beta\text{-Cu}_2\text{V}_2\text{O}_7$.

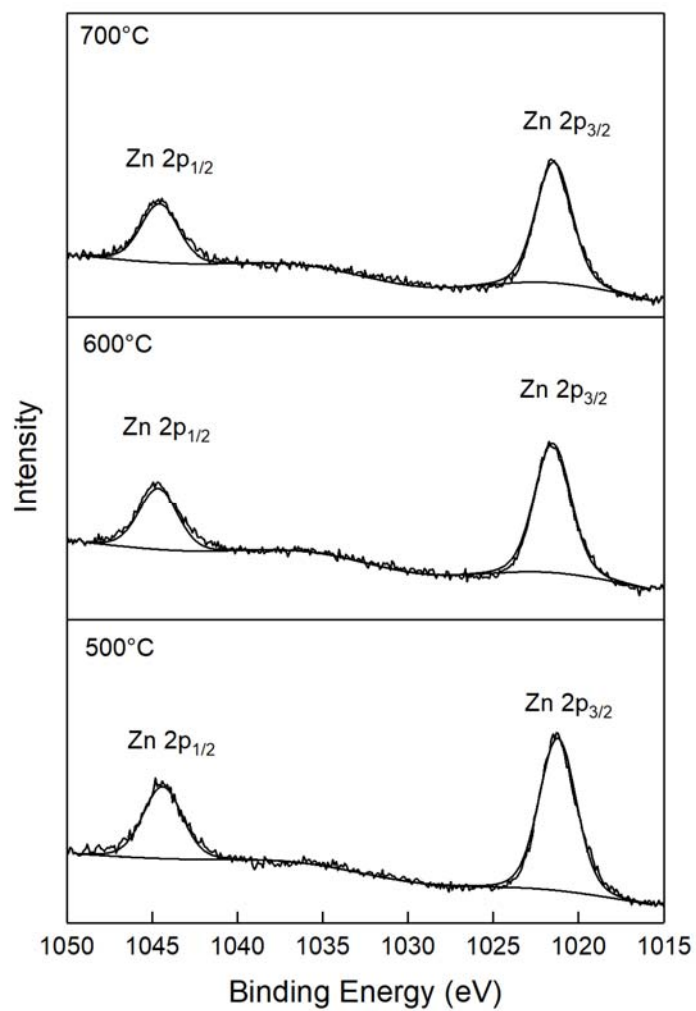


Figure S2. XPS data of the Zn 2p region for α -Zn₂V₂O₇.

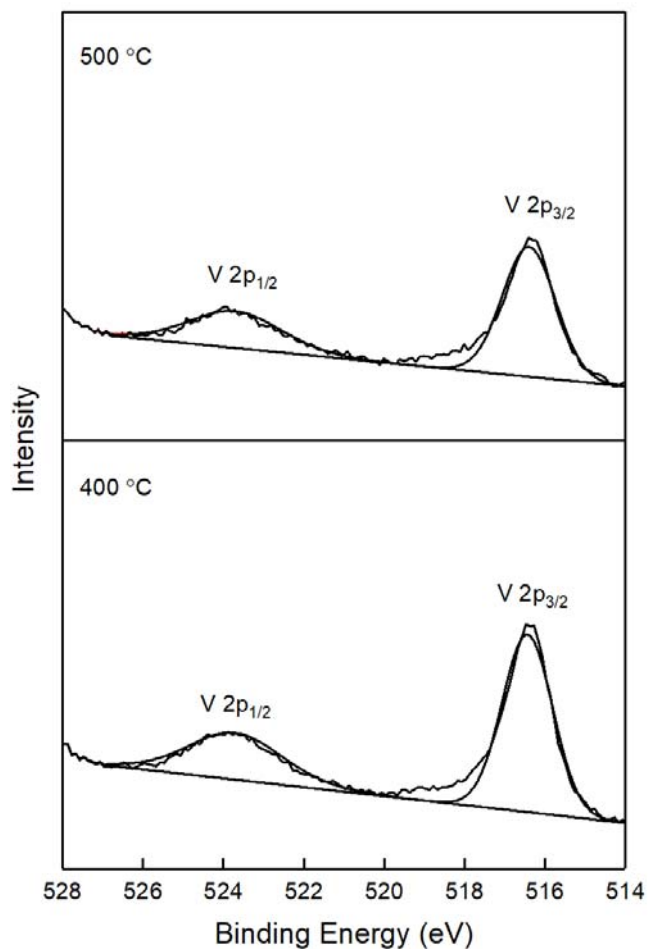


Figure S3. XPS data of the V 2p region for β - $\text{Cu}_2\text{V}_2\text{O}_7$.

Table S1. Estimated particle size of α - $\text{Zn}_2\text{V}_2\text{O}_7$ and β - $\text{Cu}_2\text{V}_2\text{O}_7$ calculated by Debye-Scherrer equation and atomic % ratio of V:Zn(Cu) by EDS.

$\text{M}_2\text{V}_2\text{O}_7$	Calcination Temperature (°C)	Particle size (nm)	V:Zn(Cu) ratio ^[a]
α - $\text{Zn}_2\text{V}_2\text{O}_7$	500	50.5 ± 3.5	1.95(8) : 2
	600	57.4 ± 4.4	1.93(4) : 2
	700	56.6 ± 6.0	1.94(7) : 2
β - $\text{Cu}_2\text{V}_2\text{O}_7$	400	42.0 ± 7.9	1.92(6) : 2
	500	61.0 ± 3.5	1.95(3) : 2

^[a] The average atomic % ratio from different spots was taken.