

1 [Supporting Information]

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3 **One-step fabrication of silver nanosphere-wetted carbon nanotube**
4 **electrodes via electric-field-driven combustion wave for high-performance**
5 **flexible supercapacitors**

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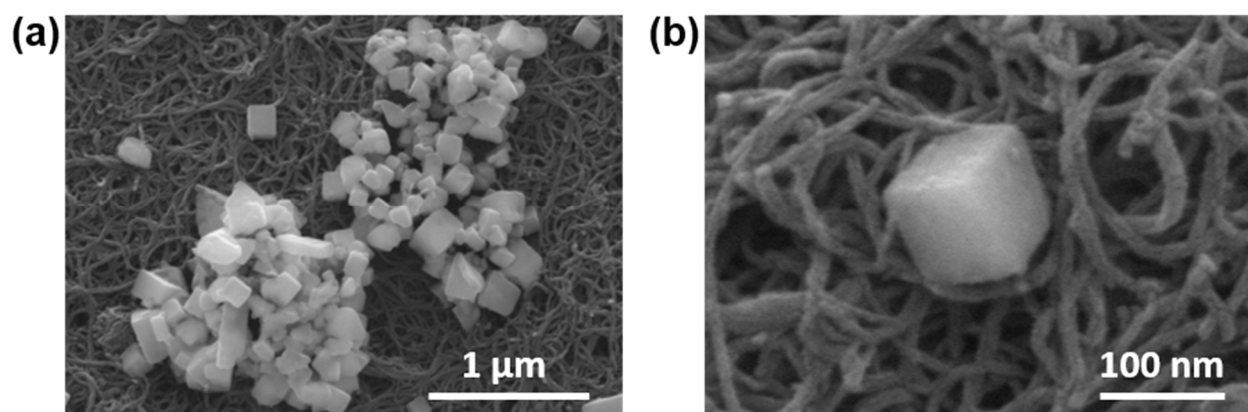
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19 **Keywords:** combustion; supercapacitor; silver/silver oxides; carbon nanotubes; wetted hybrids;
20 chemical synthesis

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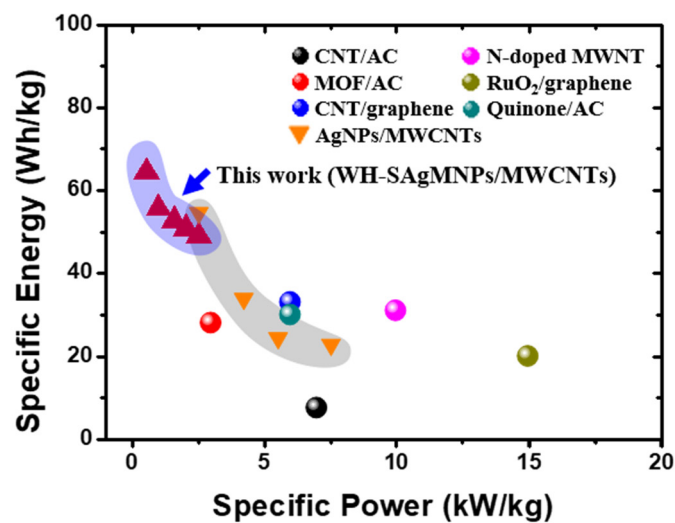
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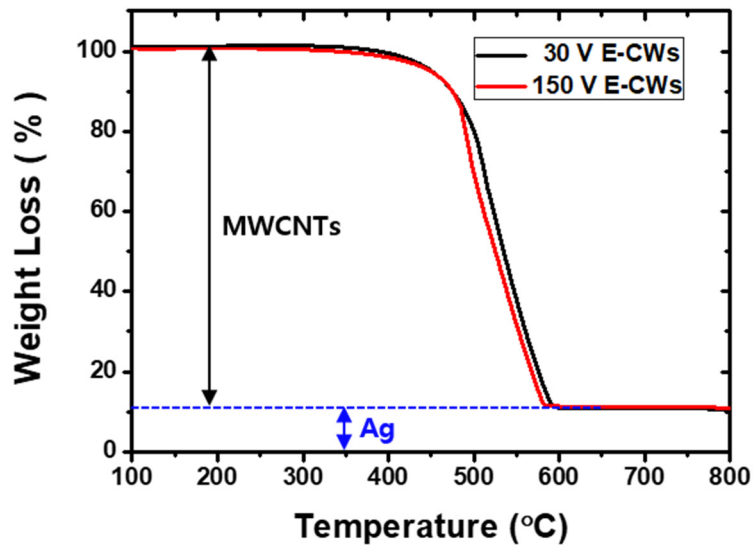
Fig. S1. Hybrid structures of Ag₂O/Ag and MWCNTs before the application of E-CWs.

Scanning electron microscopy (SEM) images of (a) as-prepared Ag₂O/MWCNTs and (b) interfacial contacts between Ag₂O and MWCNTs before the application of E-CWs



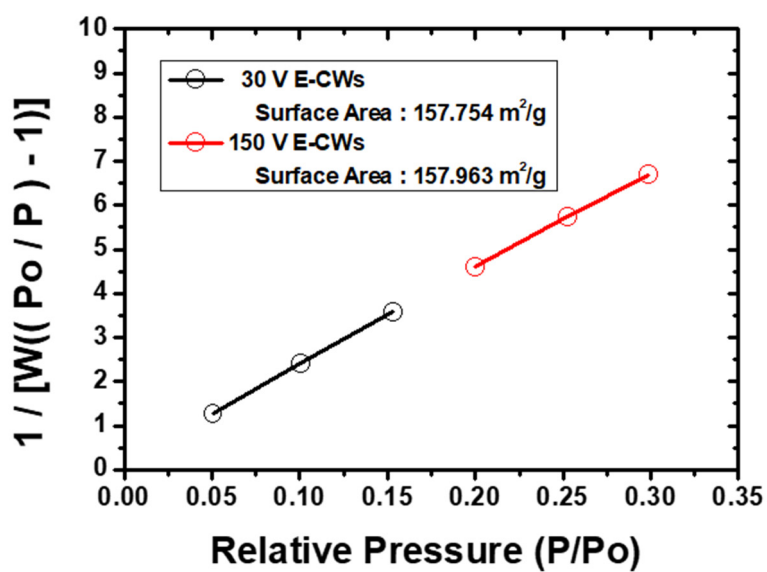
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Fig. S2. Ragone plots of the specific energy and power of flexible all-solid-state symmetric supercapacitors based on WH-SAgMNPs/MWCNTs electrodes with different hybrid electrodes of metal oxides and carbon-based materials.^{1,2} The specific energy and power density reported in this work were obtained from the full-cell of WH-SAgMNPs/MWCNTs electrode/gel-type electrolyte/ WH-SAgMNPs/MWCNTs electrode fabricated by 150 V E-CWs.



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Fig. S3. Thermogravimetric analysis (TGA) of hybrids of AgMNPs/MWCNTs, fabricated by 30 V and 150 V E-CWs.



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2 **Fig. S4. BET (Brunauer–Emmett–Teller) specific surface areas of hybrids of**
 3 **AgMNPs/MWCNTs, fabricated by 30 V and 150 V E-CWs.**

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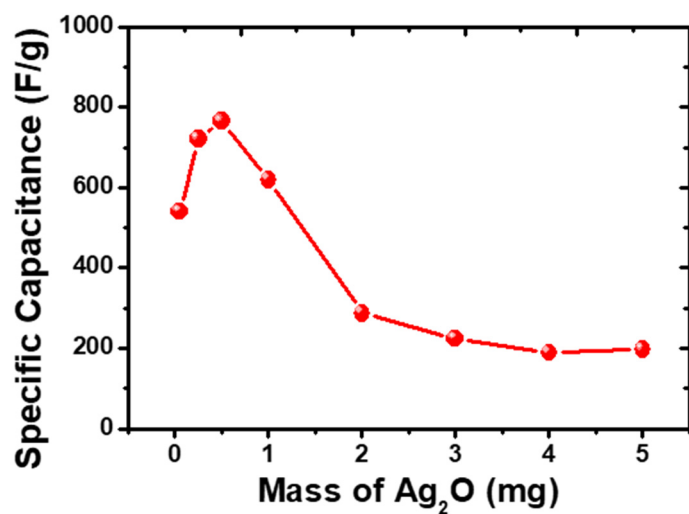
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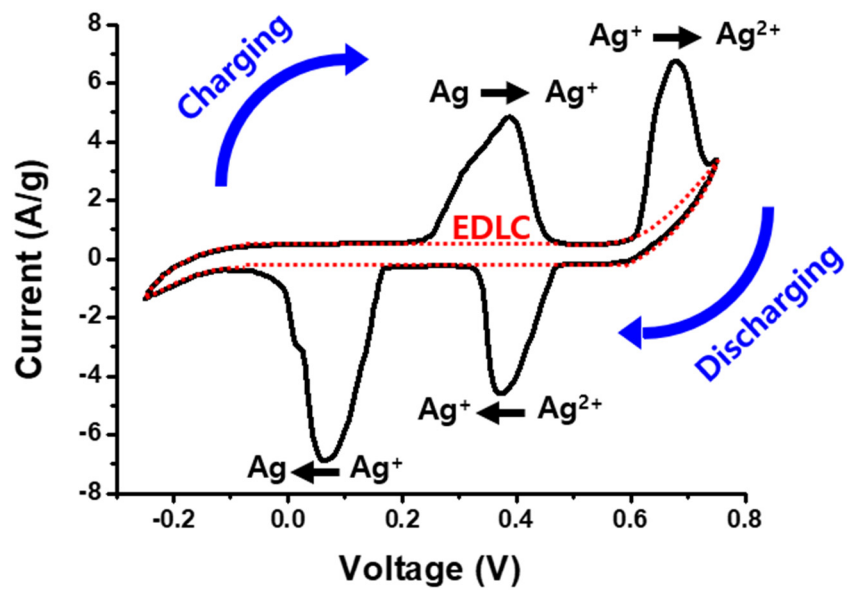
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Fig. S5. Maximum specific capacitances of WH-SAgMNPs/MWCNTs electrodes in different mass of Ag₂O (0.05, 0.25, 0.5, 1, 2, 3, 4 and 5 mg) at the scan rate of 10 mV/s. The mass of the MWCNTs was 20 mg.



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Fig. S6. Classification of dominant charge-discharge mechanisms in CV curves of WH-SAgMNPs/MWCNTs electrodes.

1 **Ragone plots for flexible all-solid-state symmetric supercapacitors based on the WH-** 2 **SAgMNPs/MWCNTs electrodes**

3 The Ragone plots for flexible all-solid-state symmetric supercapacitors based on the WH-
4 SAgMNPs/MWCNTs electrodes show the overall comparison of the energy storage
5 performances of different hybrid electrodes of metal/metal oxides and carbon-based materials,
6 in terms of specific energy and power density (**Fig. S2**).^{1, 2} The specific energy and power
7 density reported in this work were obtained from the full-cell of the WH-SAgMNPs/MWCNTs
8 electrode/gel-type electrolyte/WH-SAgMNPs/MWCNTs electrode fabricated by 150 V E-CWs.
9 The specific energy (~ 64.5 Wh/kg) and areal energy density (~ 8.39 $\mu\text{Wh}/\text{cm}^2$) were much
10 higher than other full-cells using hybrid electrodes composed of other metal/metal
11 oxides/carbon-based materials, while the specific power exhibited moderate performance
12 compared to the others, owing to the relatively slow charge-discharge rates of the metallic
13 electrodes. Because the electrochemical reactions of the WH-SAgMNPs/MWCNTs electrode
14 mainly use the redox reaction of Ag in the metallic supercapacitors, one charge-discharge cycle
15 intrinsically requires more time than metal oxide-based supercapacitors (**Fig. S6**). However,
16 the obtained specific energy from the H-Ag/MWCNTs fabricated by E-CWs significantly
17 surpasses other supercapacitors. These characteristics imply that the metallic supercapacitor
18 takes a long time for charging at certain voltages and discharging in the constant currents. This
19 outstanding specific energy and comparable specific power indicates that the developed WH-
20 SAgMNPs/MWCNTs as the metallic electrodes for the supercapacitors can fill the gap between
21 conventional supercapacitors and batteries.

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References

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