

*Supporting Information*

**An amorphous Zn-P/graphite composite with chemical bonding for ultra-reversible lithium storage**

Wenwu Li,<sup>\*,a,c</sup> Jiale Yu,<sup>a</sup> Jiajun Wen,<sup>a</sup> Jun, Liao,<sup>a</sup> Ziyao Ye,<sup>a</sup> Bote Zhao,<sup>c</sup> Xinwei Li,<sup>\*,d</sup> Haiyan Zhang, Meilin Liu<sup>c</sup> and Zaiping Guo<sup>\*,b</sup>

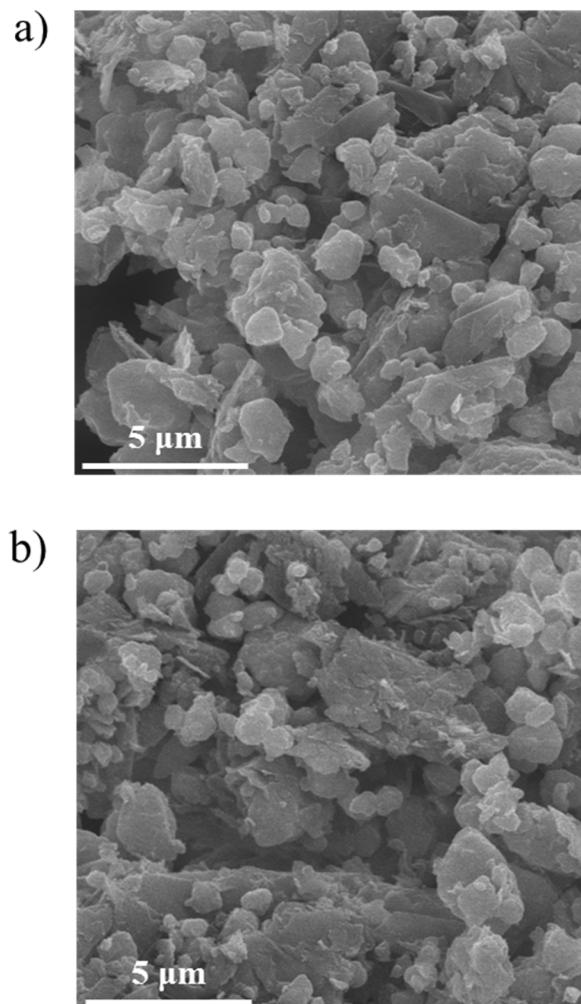
<sup>a</sup>School of Materials and Energy, Guangdong University of Technology, Guangzhou 510006, PR China,

<sup>b</sup>Institute for Superconducting and Electronic Materials, School of Mechanical, Materials and Mechatronics Engineering, University of Wollongong, North Wollongong, NSW 2500, Australia.

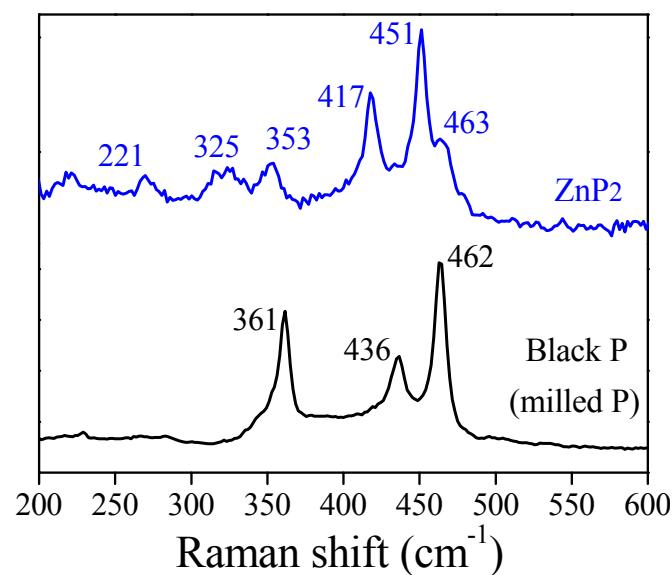
<sup>c</sup>School of Materials Science & Engineering, Georgia Institute of Technology, Atlanta, GA 30332, USA

<sup>d</sup>Department of Mechanical and Energy Engineering, Southern University of Science and Technology, Shenzhen 518071, PR China.

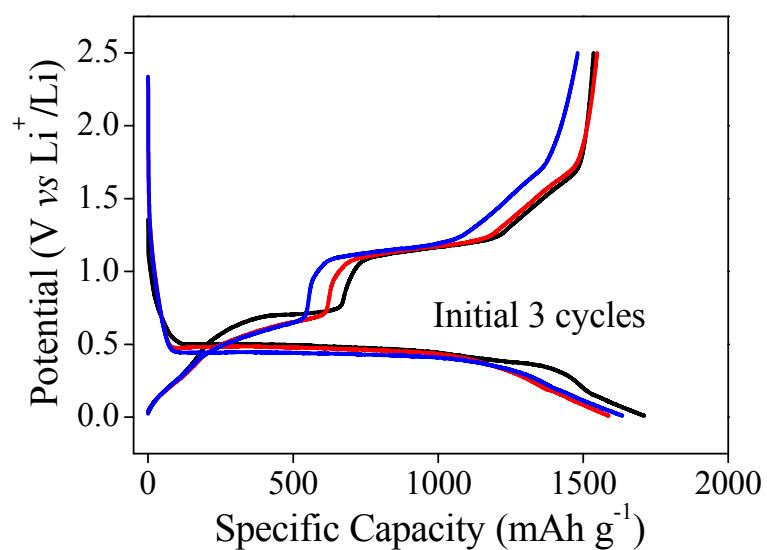
\*Corresponding author E-mail: wenwuligatech@gmail.com; lixw@sustc.edu.cn;  
[zguo@uow.edu.au](mailto:zguo@uow.edu.au)



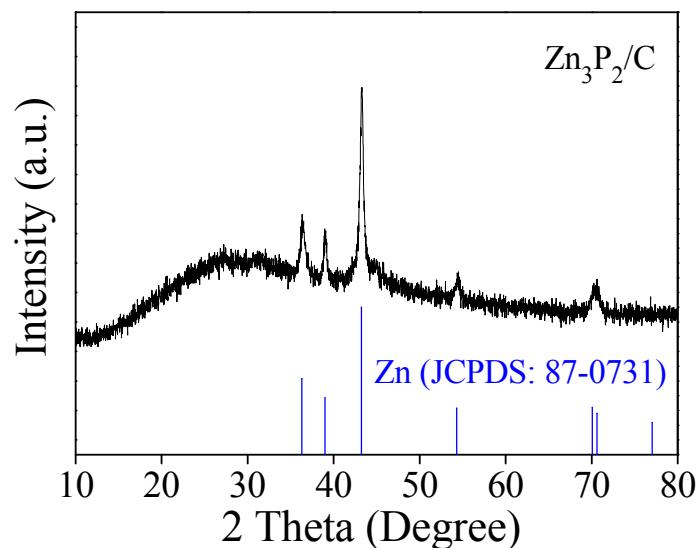
**Figure S1.** a), b) FESEM images of the  $\text{ZnP}_2$  and  $\text{Zn}_3\text{P}_2$  powders, respectively.



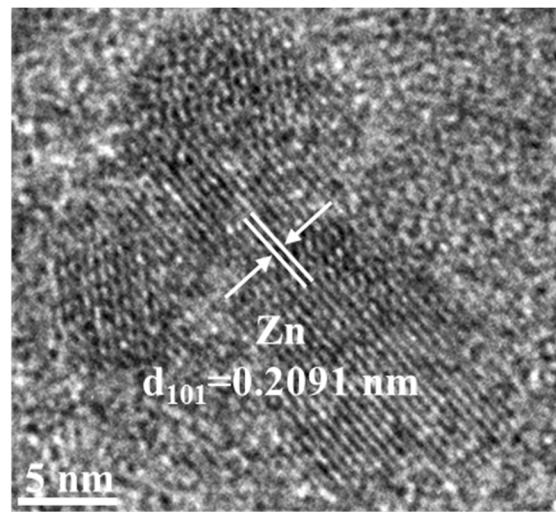
**Figure S2.** The Raman fingerprint peaks of the alfa- $\text{ZnP}_2$  differing from the milled P counterparts.



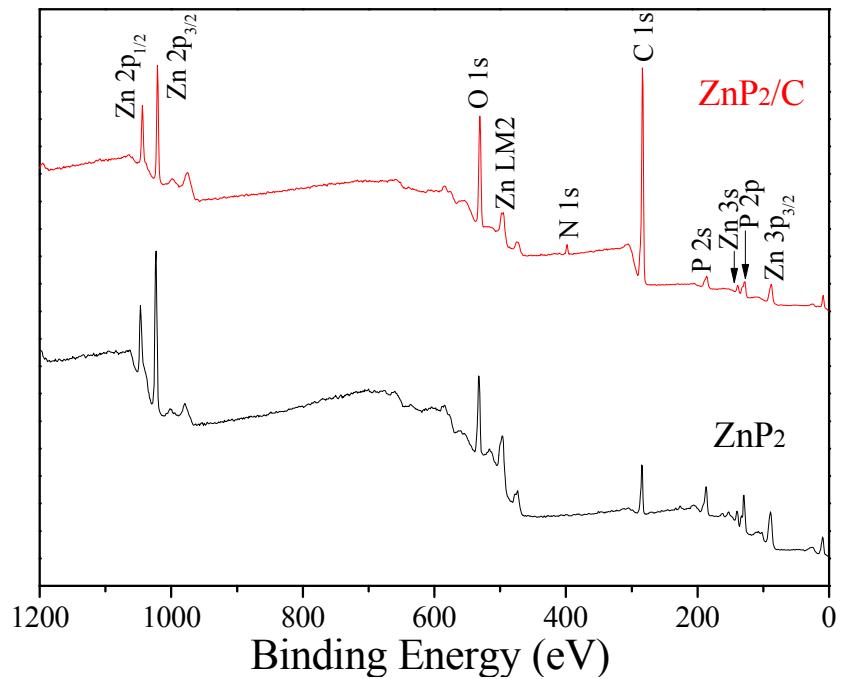
**Figure S3.** The initial three discharge/charge profiles of ZnP<sub>2</sub>.



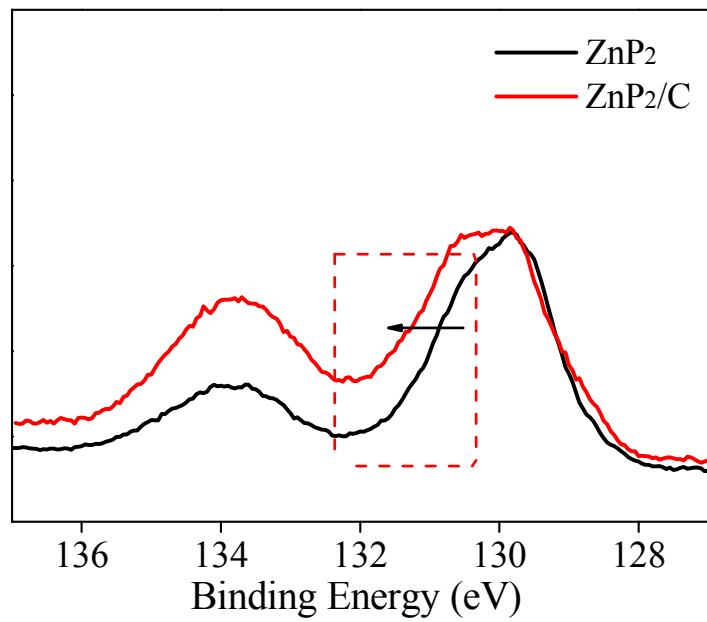
**Figure S4.** The XRD pattern of Zn<sub>3</sub>P<sub>2</sub>/C (10 h) milled which degraded into a mixture containing crystalline Zn.



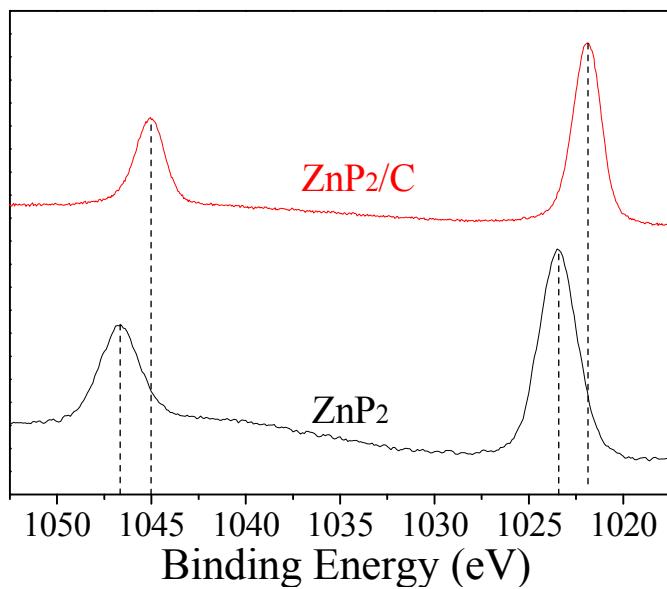
**Figure S5.** The crystalline Zn within the 3Zn-2P/C composite.



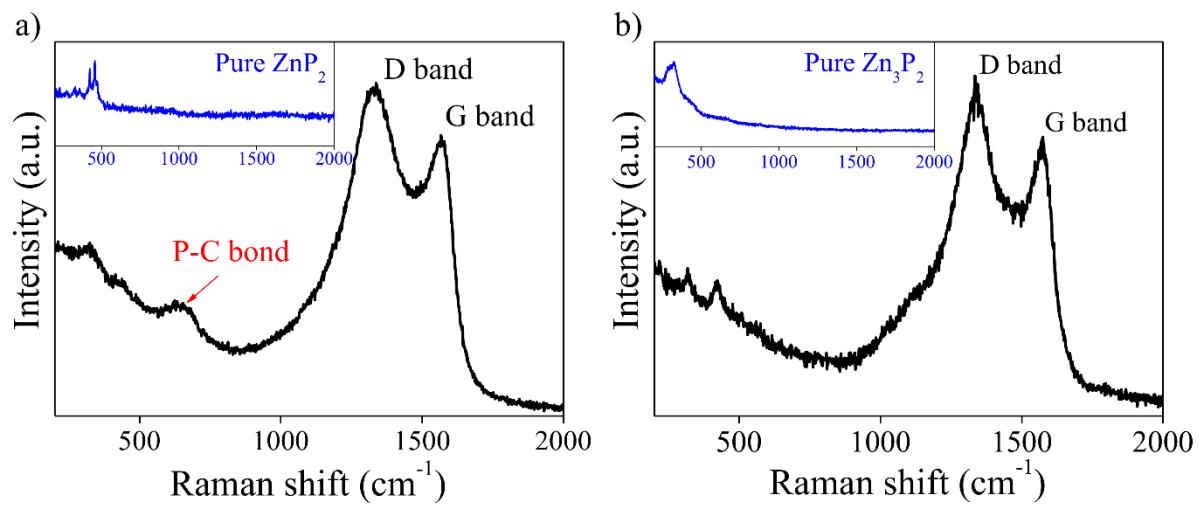
**Figure S6.** The XPS spectra of survey spectrum for ZnP<sub>2</sub> and ZnP<sub>2</sub>/C.



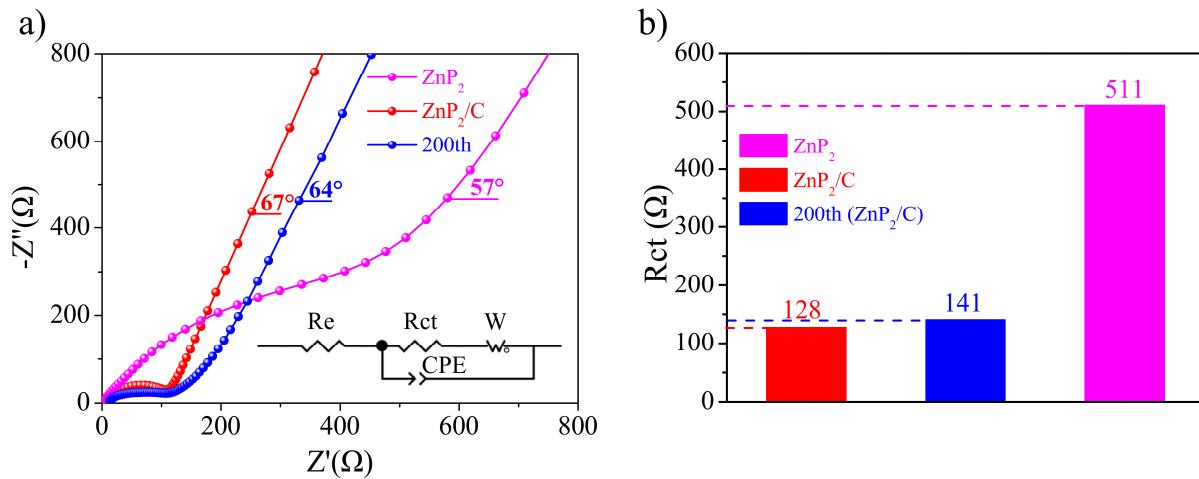
**Figure S7.** The XPS spectra of P 2p for  $\text{ZnP}_2$  and  $\text{ZnP}_2/\text{C}$ .



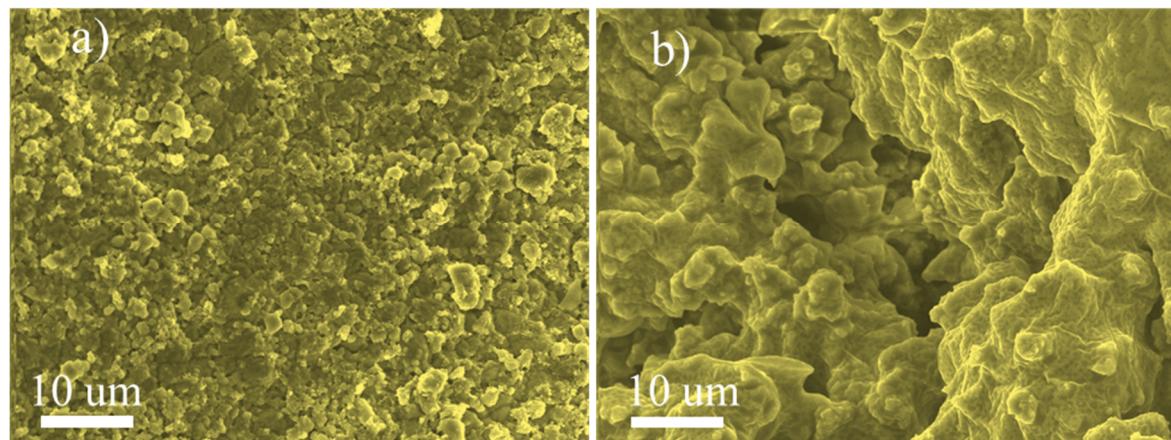
**Figure S8.** The XPS spectra of Zn 2p for  $\text{ZnP}_2$  and  $\text{ZnP}_2/\text{C}$ .



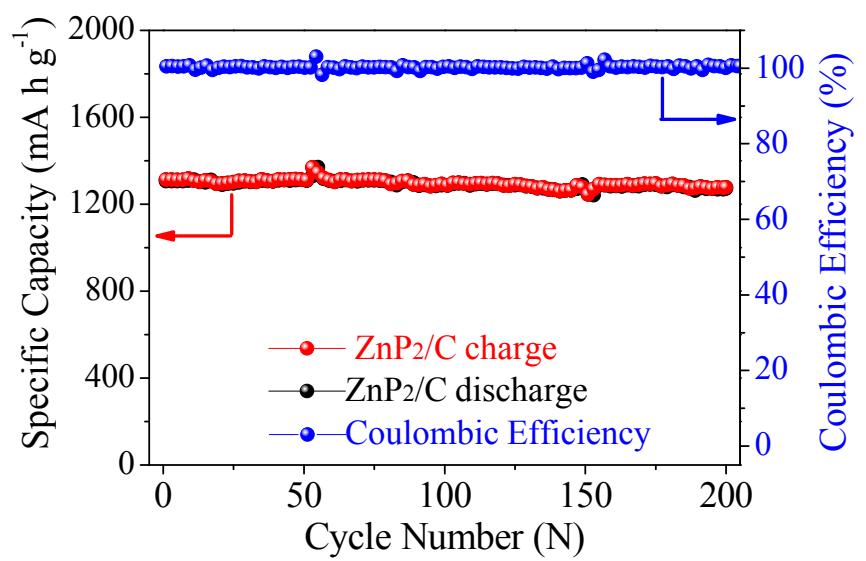
**Figure S9.** The Raman Spectra: a)  $\text{ZnP}_2$  and its carbon composite; b)  $\text{Zn}_3\text{P}_2$  and its carbon composite



**Figure S10.** The electrochemical impedance spectra of the pristine  $\text{ZnP}_2$  electrode (pink), the pristine  $\text{ZnP}_2/\text{C}$  electrode (red), the  $\text{ZnP}_2/\text{C}$  electrode (blue) after 200 cycles.



**Figure S11.** The electrode morphology after 50 cycles: a) the  $\text{ZnP}_2/\text{C}$  electrode; b) the  $\text{ZnP}_2$  electrode.



**Figure S12.** The cycle stability of the Li-ion full cell  $\text{LiCoO}_2//\text{ZnP}_2/\text{C}$  at a current density of  $200 \text{ mA h g}^{-1}$ .