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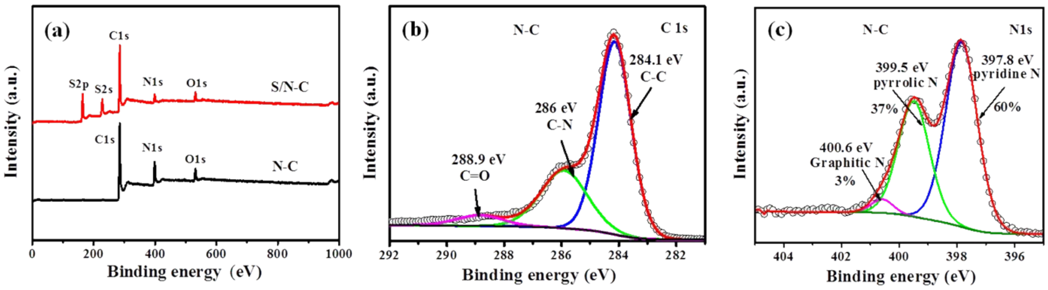

Figure S1. (a) XPS spectra of N-C and S/N-C samples; (b, c) high-resolution XPS spectra of C 1s and N 1s for the N-C sample.

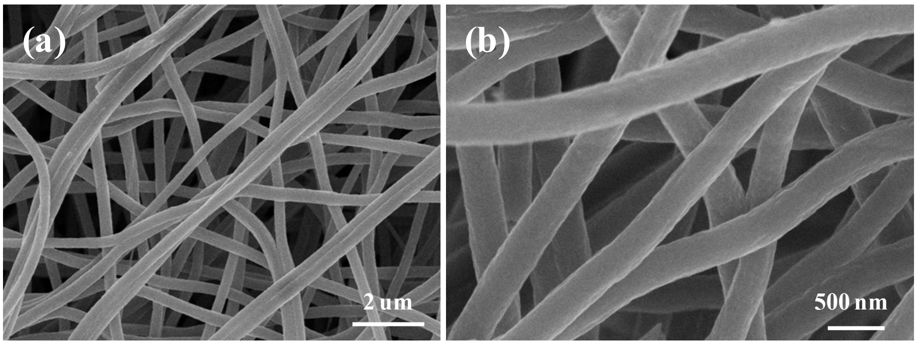

Figure S2. SEM images of N-C nanofibers: (a) low magnification and (b) high magnification.

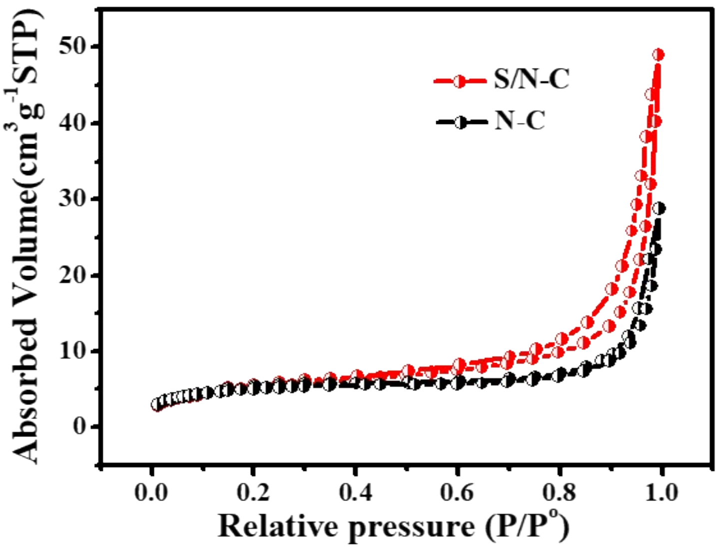

Figure S3. Nitrogen adsorption/desorption isotherms of N-C and S/N-C nanofibers.

Table S1 Element contents (%) in the N-C and S/N-C nanofibers.

	S/N-C	***N-C***
C	**49.45**	**70.7**
N	**12.59**	**17.36**
O	**10.01**	**11.94**
S	**27.95**	**0**

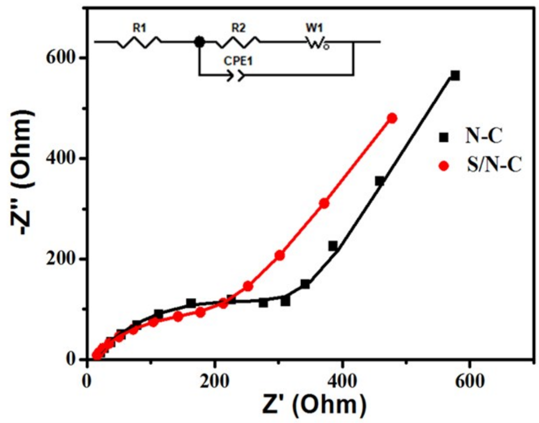

Figure S4. Nyquist plots of EIS data for N-C and S/N-C nanofiber electrodes. The inset is the equivalent circuit diagram.

Table S2 Various Carbon-based anodes for sodium ion batteries

Materials	Doped amount (S, N, %)	Capacity ($mAh g^{-1}$)	Cycles	Tested current ($mA g^{-1}$)	References
S/N–doped mesoporous carbon nanofibers	**27.95 w% for S 12.59 w% for N**	**335.8 (426.3)**	**800 (400)**	**5000 (100)**	**This work**
S-doped N-rich carbon nanosheets	9.19 w% for S 20.01 w% for N	211	1 000	1 000	Ref. 1
N, S-doped carbon nanofibers	2.8 w% for S 10.9 w% for N	164.3	6 000	10 000	Ref. 2
Nitrogen and sulfur co-doped carbon nanosheets	4.96 w% for N 3.01 w% for S	248	500	100	Ref. 3
N/S codoped ordered mesoporous carbon	20.32 at% for N 0.82 at% for S	220	3 000	5 000	Ref. 4
pTTPN@600	7.52 w% for N 1.63 w% for S	74	2 000	10 000	Ref. 5
Sulfurized polyacrylonitrile derived carbon	~8 w% for N 3.39 w% for S	126.5	10 000	10 000	Ref. 6
Sulfur-doped disordered carbon	26.91 w% for S	271	1 000	1 000	Ref. 7
Covalent sulfur–carbon complex	40.1 w% for S	590	200	100	Ref. 8
Sulfur-doped carbon spheres	25.5 w% for S	238.2	600	1 000	Ref. 9
N-doped porous carbon	2.14 at% for N	198.6	500	200	Ref. 10
Nitrogen-doped 3D porous carbon monolith	15.02 w% for N	175	3 000	500	Ref. 11
Ultrathin carbon nanocups		212	1 000	1500	Ref. 12
$FeS@Fe_3C@GC$		575.7	100	100	Ref. 13
TiO_2–Sn@carbon nanofibers		413	400	100	Ref. 14
Mo_2C embedded in S-doped carbon		102.7	500	5 000	Ref. 15

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