

Supplementary Figures and Tables

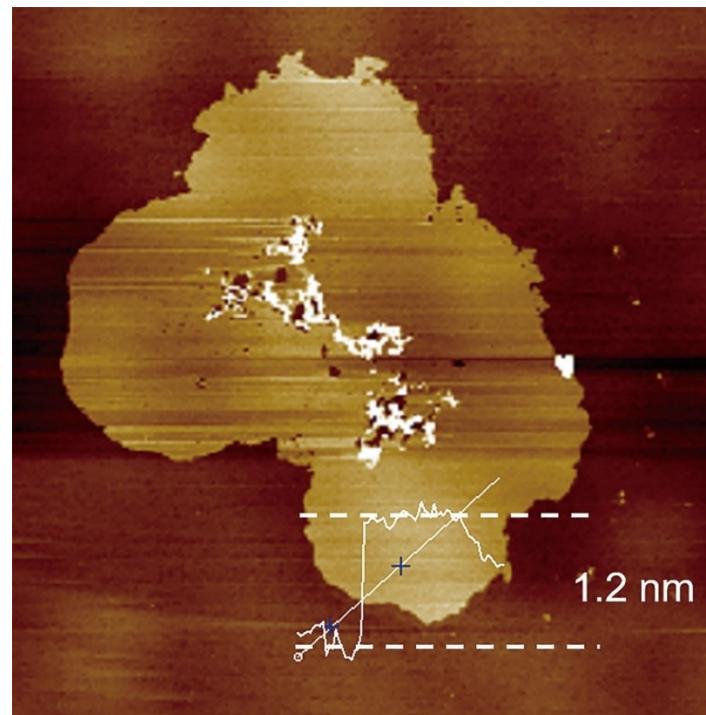


Figure S1 The AFM image of MoS₂ single layer.

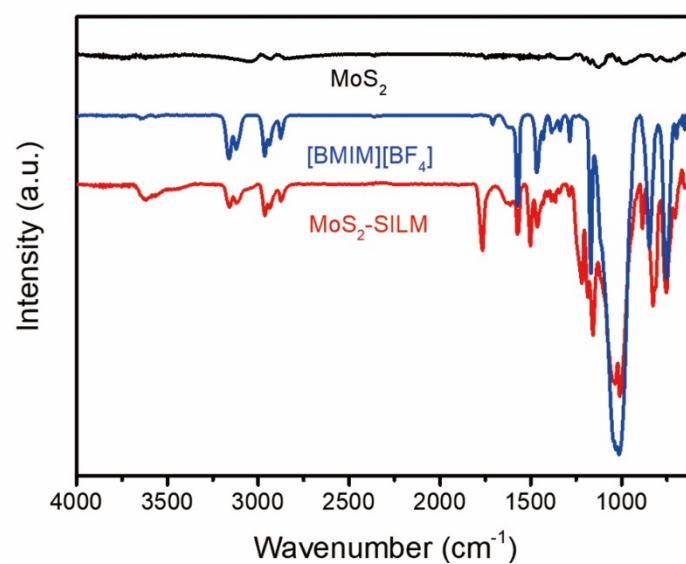


Figure S2 The FTIR spectra of MoS₂, [BMIM][BF₄] and MoS₂-SILM. Detail data are shown in Table S1.

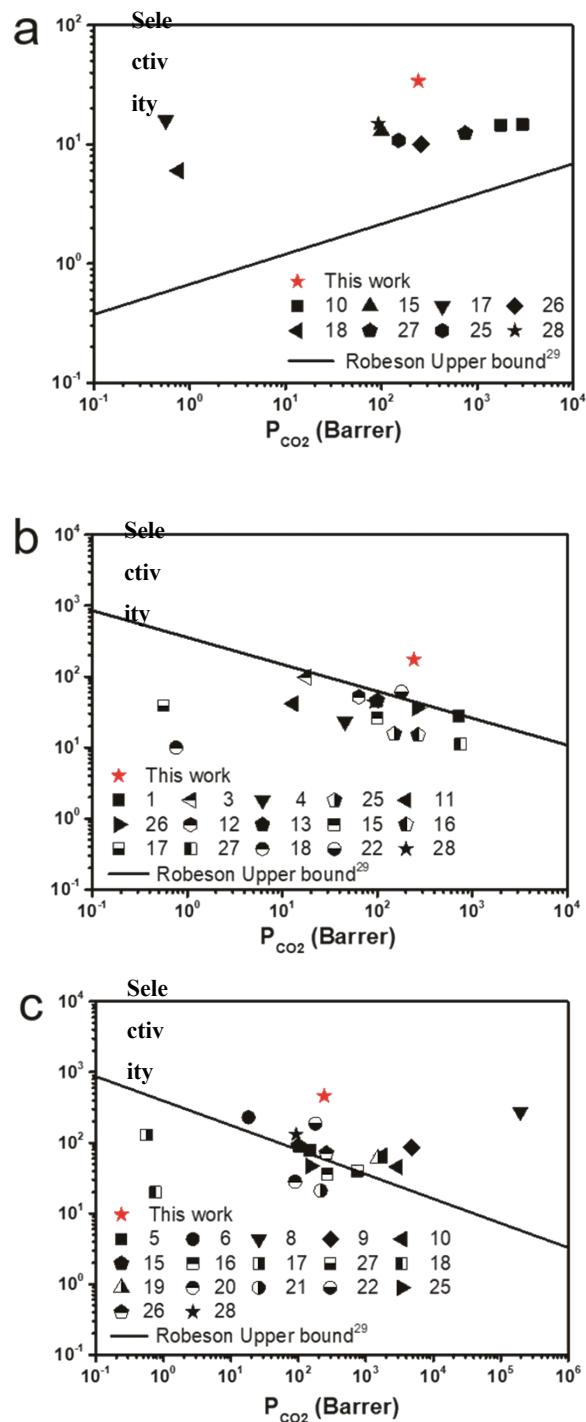


Figure S3 The separation performance (selectivity vs. permeability) of MoS₂-SILM under EEF compared with other membranes. (a) CO₂/H₂, (b) CO₂/CH₄ and (c) CO₂/N₂. The detail data is shown in Table S2 and from the Supplementary references.

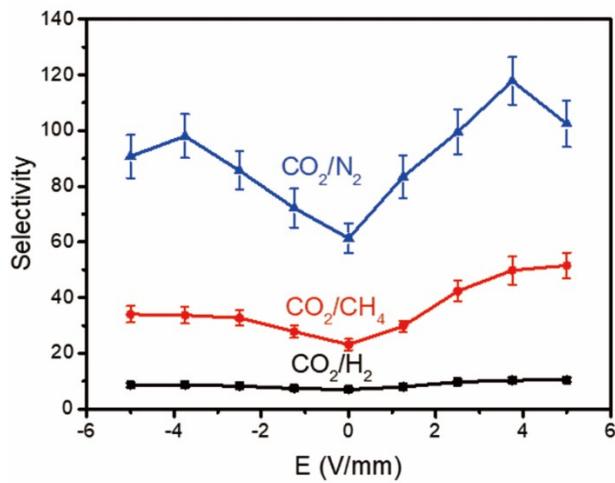


Figure S4 The mixed gas selectivity of MoS₂-SILM under EEF.

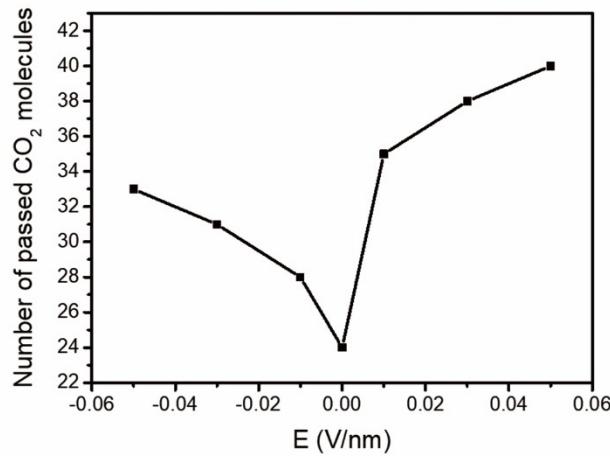


Figure S5 Molecular simulated CO₂ permeance under EEF.

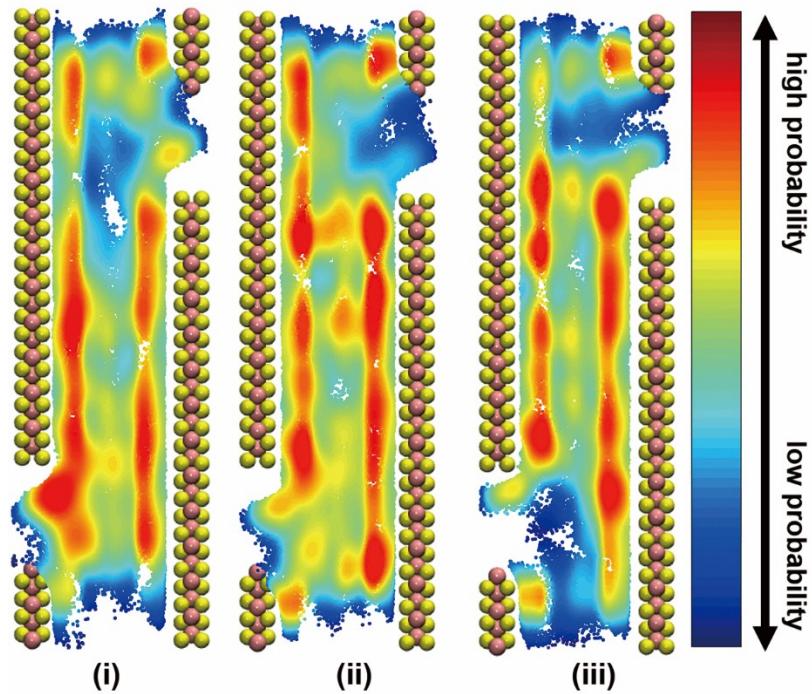


Figure S6 The density map of $[\text{BMIM}]^+$ inside the nanochannel under EEF of (i) $E=-0.03 \text{ V/nm}$, (ii) $E=0 \text{ V/nm}$, (iii) $E=0.03 \text{ V/nm}$. Blue indicates the region that low probability to find $[\text{BMIM}]^+$, while red region indicates the high probability of finding $[\text{BMIM}]^+$.

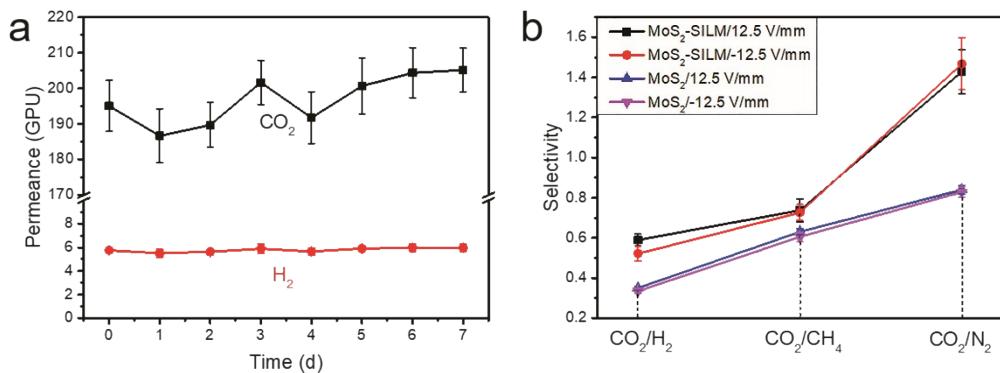


Figure S7 The durability of $\text{MoS}_2\text{-SILM}$. (a) is the permeance of CO_2 and H_2 tested under 3.75 V/mm for a week; (b) is the gas selectivity of $\text{MoS}_2\text{-SILM}$ and MoS_2 under 12.5 V/mm and -12.5 V/mm after 1 day.

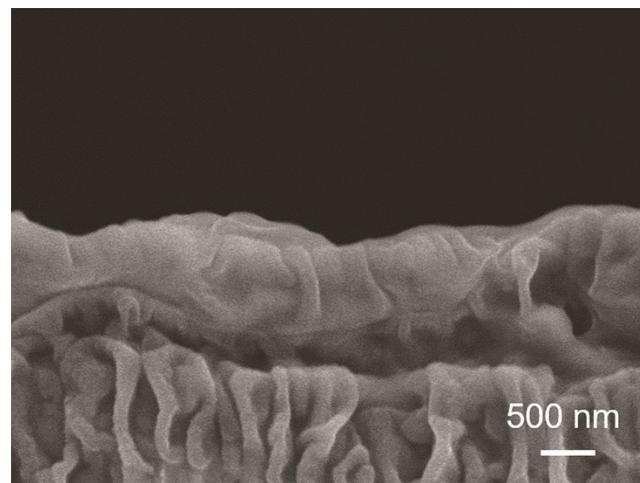


Figure S8 Cross-section SEM image of MoS₂-SILM after testing under 12.5 and -12.5 V/mm.

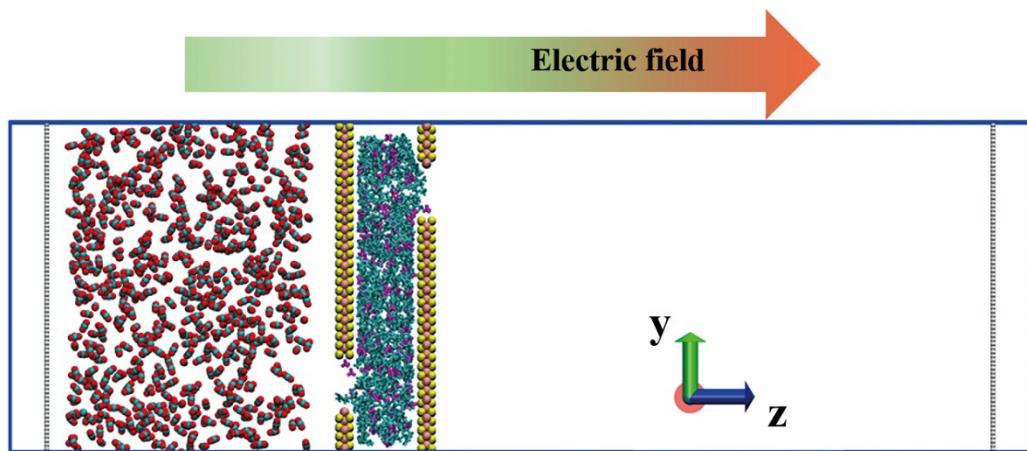


Fig. S9 A bichamber system for simulations of CO₂ permeance through the MoS₂-SILM. White, cyan, and red balls denote helium, carbon, and oxygen, respectively. Green-cyan and magenta stick represent [BMIM]⁺ and [BF₄]⁻ ions.

Table S1 The data of FTIR peaks in Fig. S2.

[BMIM][BF ₄] (cm ⁻¹)	MoS ₂ -SILM (cm ⁻¹)	Shifts (cm ⁻¹)	Assignment
3160.81	3158.88	-1.93	vC4-H, vC5-H
3122.24	3118.38	-3.86	vC2-H
2939.03	2937.10	-1.93	vCH ₂
2877.32	2875.39	-1.93	vCH ₂
1708.65	1764.57	55.92	
1621.86	1614.15	-7.71	
1465.66	1463.73	-1.93	vCH ₃
1384.66	1386.59	1.93	wCH ₂
1373.09	1363.45	-9.64	vC2N1C5, wCH ₂
1338.38	1340.31	1.93	vN-Bu, N-Me
1284.38	1288.24	3.86	vBF ₄
1253.52			tCH ₂ , rC2-H
	1218.81		vBF ₄
1168.67	1157.10	-11.57	vN-Bu, N-Me
1031.74	1035.60	3.86	vBF
1014.39	1010.53	-3.86	vBF ₄
908.32	919.89	11.57	
846.61	827.32	-19.29	rC4-H, rC5-H
808.04	813.82	5.78	vC-C-C
696.19	705.83	9.64	vN-Bu, N-Me

Table S2 The data of gas separation performance of different membranes in Figure 3.

Number	Membrane	P _{CO₂} (Barrett)	P _{CO₂} (GPU)	CO ₂ /H ₂	CO ₂ /CH ₄	CO ₂ /N ₂	Ref
1	ZIF-90/6FDADAM	720	12		28		1
2	PSF	-	19.86		23.12		2
3	DDR	17.9	35.8		98		3
4	MIL-53-NH ₂ /organosilica	45.6	430		23.2		4
5	MMP-3/mPSf	150	3000			78	5
6	PAMAM dendrimer	18.3	61			230	6
7	DNMDAm-DGBAmE-TMC/PDMS/PS	-	1601			138	7
8	PVAm-PIP/PS	195000	6500			277	8
9	DAmBS-DGBAmE-TMC/PDMS/PS	4781	5831			86	9
10	SIPN:PEGMEA:PEGDA:PE GDME	2980	11.9-14.9	14.7		45.7	10
10	SIPN:PEGDA:PEGDME	1767	7.1-8.8	14.5		65.9	10
11	Matrimid® ZIF-8	12.96	180		41.5		11
12	6FDA-DAT/Ni2(dodbc)	63.9	0.91-1.6		52		12
13	6FDA-ODA UiO-66	100	1.26-2.52		46.1		13
14	Pebax/[emim][BF ₄]/GO	-	981			44	14
15	Pebax/GO	100	20	13	26	91	15
16	Pebax/[emim][BF ₄]	269	306		15	36	16
17	PAN/[emim][Ac]GO/PTMSP	0.56	37	16	39	130	17
18	GO	0.76	110	6	10	20	18
19	0.1water-[BMIM][BF ₄]-PES	1518	13.8			60	19
20	[BMIM][BF ₄]/AgO	89.7	14.1		-	28.2	20
21	[BMIM][BF ₄]/CuO	214.84	52.4		-	21	21
22	Pebax/0.5%Ag/50%[BMIM][BF ₄]	180	3.2		61	187.5	22
23	[BMIM][BF ₄]/LiBF ₄	-	13.36		8.25	8.4	23
24	[BMIM][BF ₄]/cyanuric chloride	-	19.2		10.7	11	24
25	Pebax®/PEG	151	2.52	10.8	15.5	47	25
26	IL-Pebax 165/ZIF-8	261	20	10	36	71	26
27	PEO-PBT/PEG-DBE (PAN-PDMS)	750	15	12.4	11.2	40	27
28	MoS ₂ -[BMIM][BF ₄] (1940 nm)	92.9	47.88	14.95	43.52	131.42	28
29	MoS ₂ -[BMIM][BF ₄] (1120 nm) under positive EEF	243	195	34	174	462	This work

Table S3 Lennard-Jones parameters and atom charge of MoS₂ used in this work.^{30, 31}

	CO ₂		MoS ₂	
	C	O	Mo	S
ϵ (Kcal/mol)	0.0559	0.1600	0.0135	0.4612
σ (Å)	2.7570	2.5650	4.2000	3.1300
q (e)	0.6512	-0.3256	0.7600	-0.3800

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