

Supporting Information

Boosting Inverted Perovskite Solar Cells Performance by Using Functionalized 9,9-bis(4-diphenylaminophenyl)fluorenes with Triphenylamine as Dopant-free Hole Transporting Materials

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Experimental details

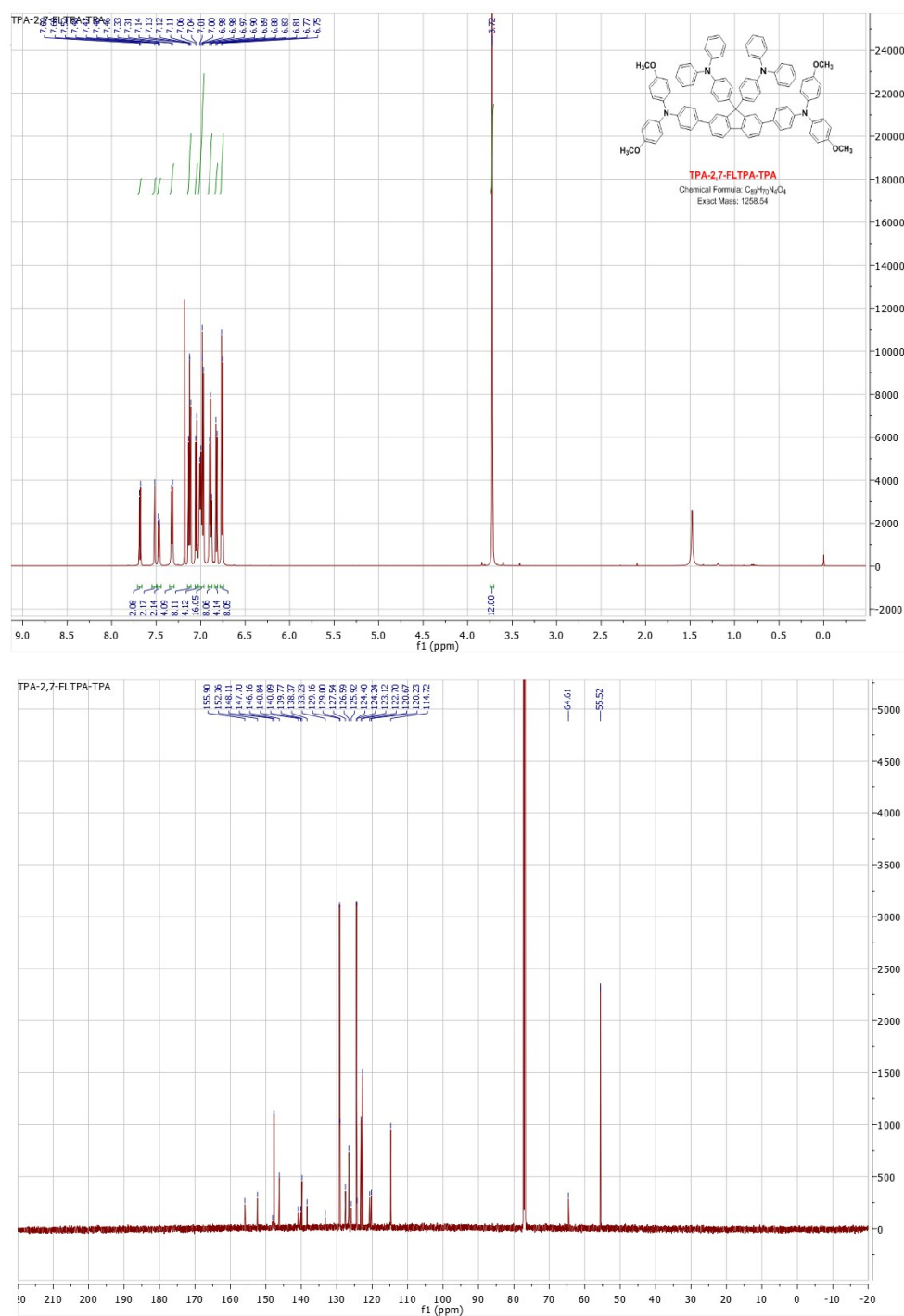


Figure S1. ^1H (600 MHz, CDCl_3) spectrum of TPA-2,7-FLTPA-TPA.

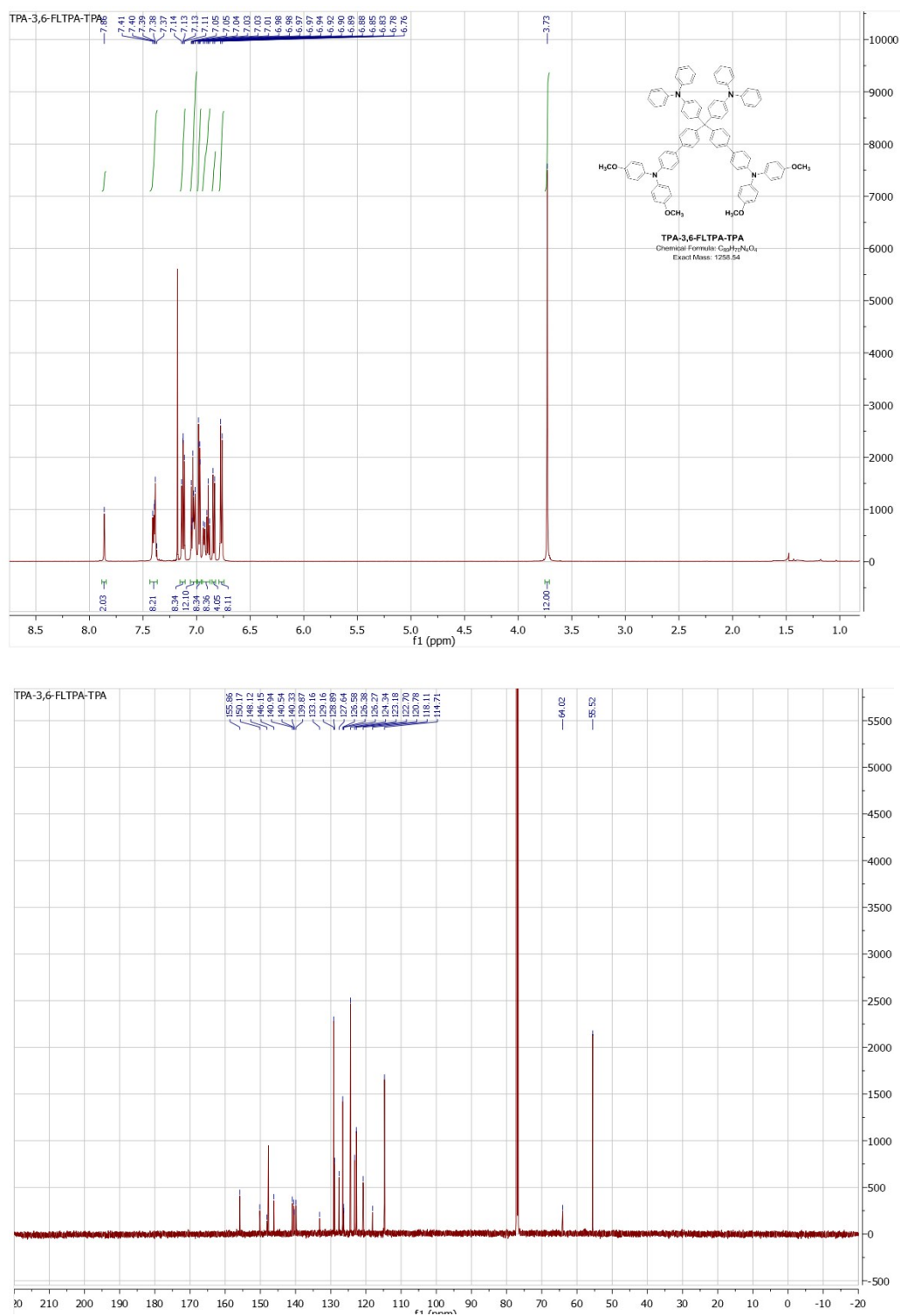


Figure S2. 1H (600 MHz, $CDCl_3$) spectrum of TPA-3,6-FLTPA-TPA.

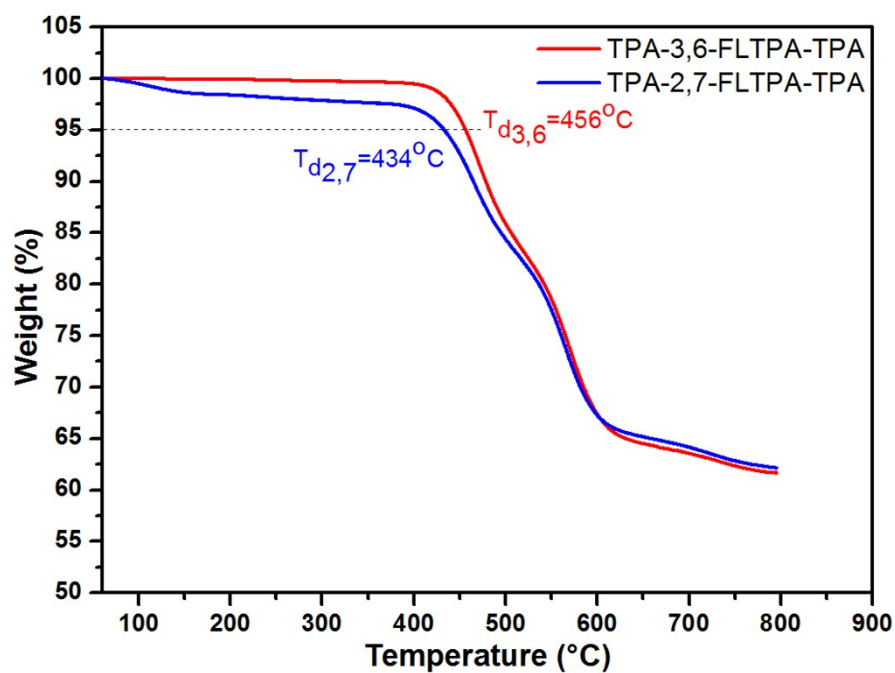


Figure S3. Thermogravimetric analysis (TGA) curve of **TPA-2,7-FLTPA-TPA** and **TPA-3,6-FLTPA-TPA**.

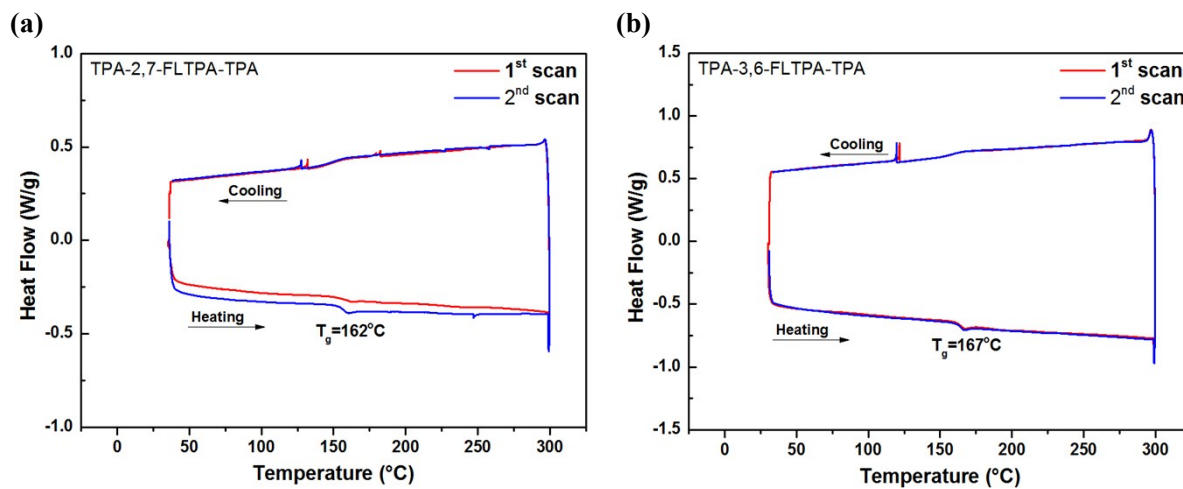


Figure S4. Differential scanning calorimetry (DSC) of (a) **TPA-2,7-FLTPA-TPA** and (b) **TPA-3,6-FLTPA-TPA** with scan rate of 10 °C/min under N₂ atmosphere.

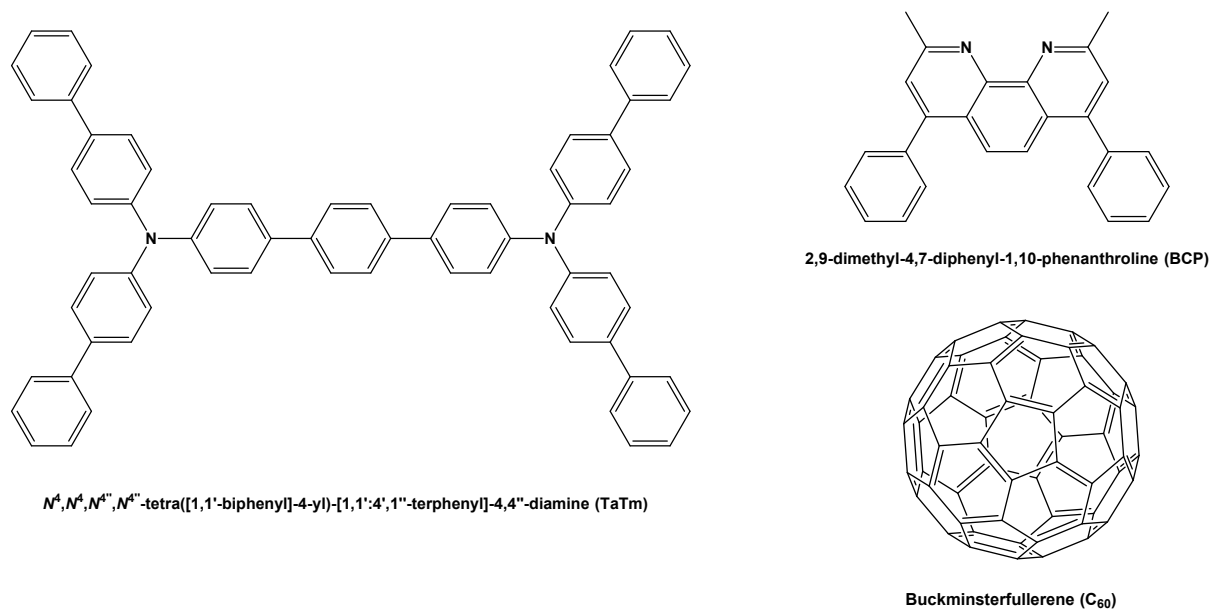


Figure S5. The structures of molecules used in PSC devices.

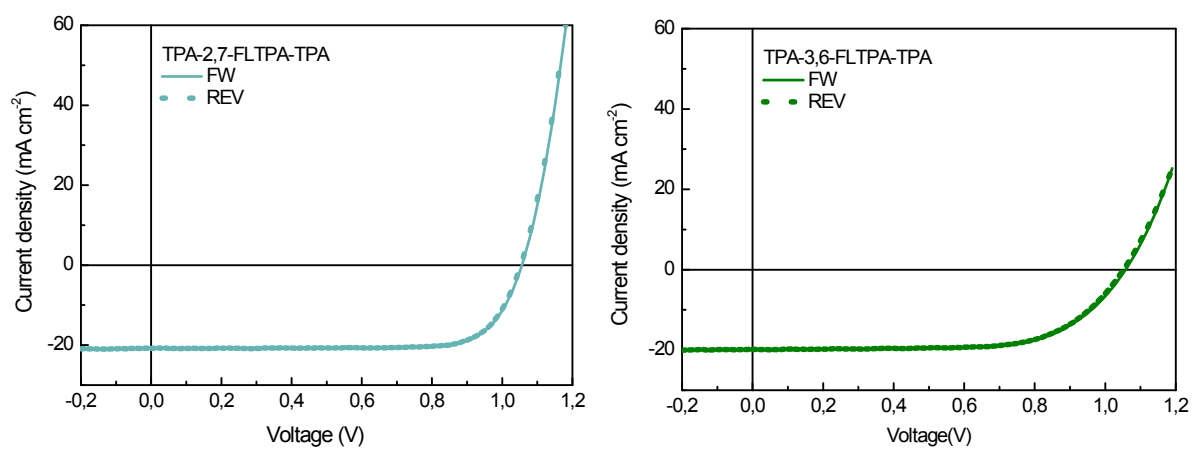


Figure S6. Forward and reverse current density versus voltage curves for the perovskite solar cells. Scan speed was 0.01 V/s.

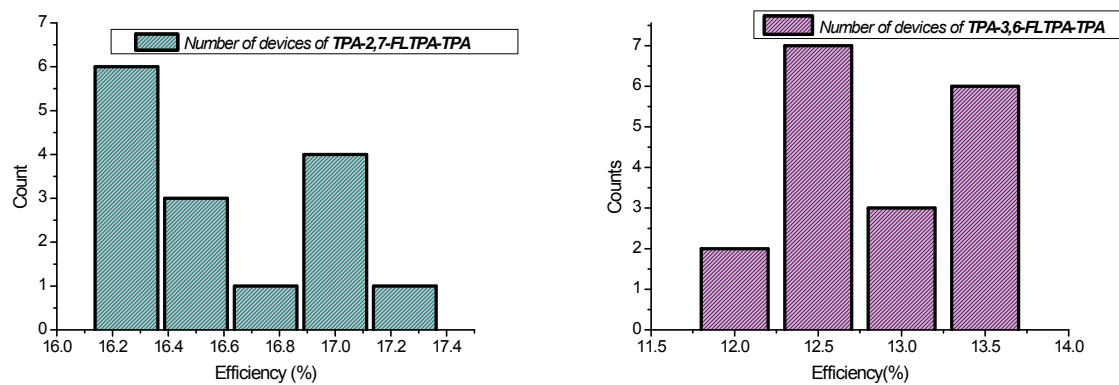


Figure S7. Statistics of the PCE for the two different HTMs based devices (8 cells of each HTM).

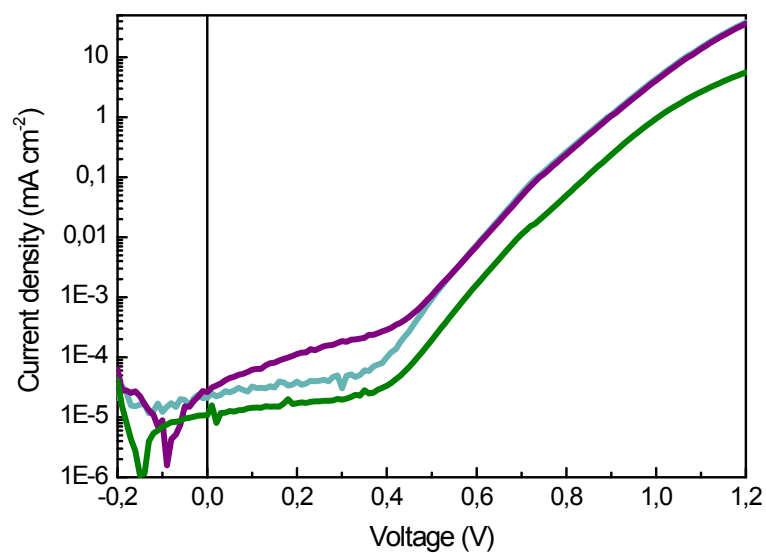


Figure S8. Current density versus voltage plotted on a semilog scale.

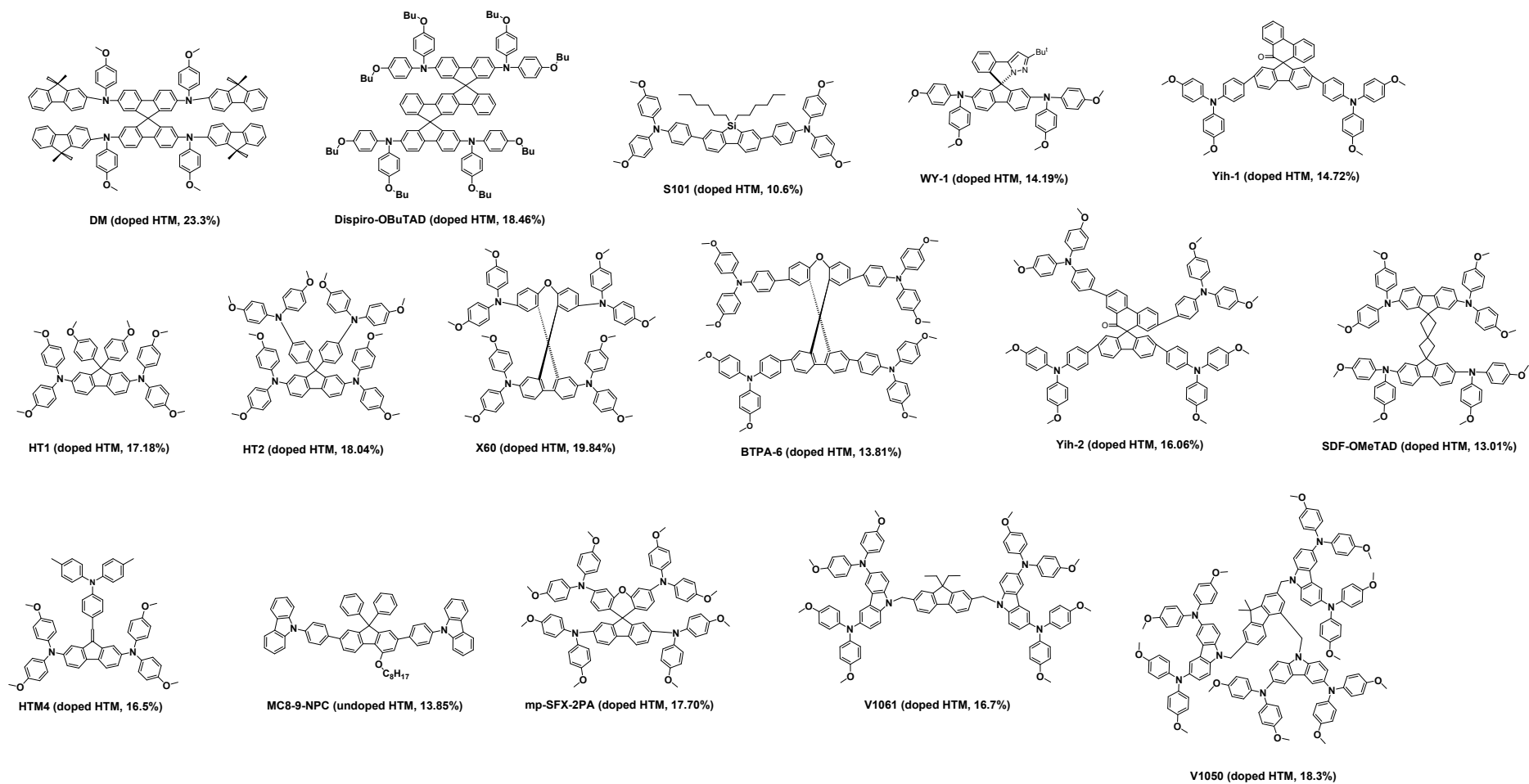


Figure S9. Chemical structures of reported small molecular fluorene-based hole transporting materials.

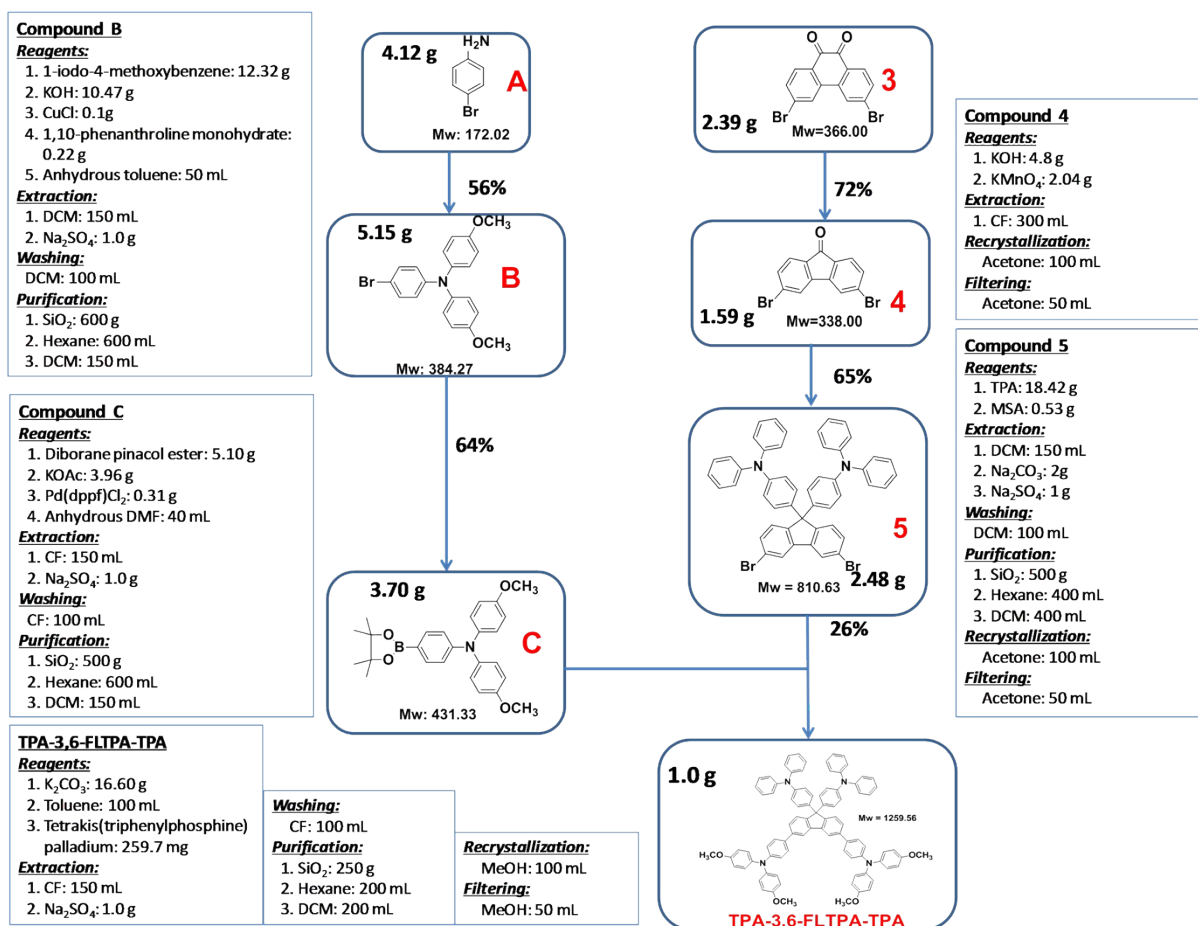


Figure S10. Flowchart describing the synthesis of 1g of TPA-3,6-FLTPA-TPA.

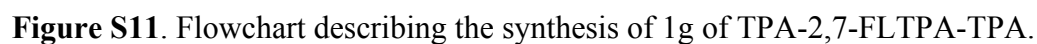


Table S1. Density functional theory (DFT) calculations of **TPA-2,7-FLTPA-TPA** and **TPA-3,6-FLTPA-TPA**

Compound	TPA-2,7-FLTPA-TPA	TPA-3,6-FLTPA-TPA
HOMO, eV	-4.85	-4.92
LUMO, eV	-1.56	-1.29
gap, eV	3.29	3.63
VIS peak, nm	431	384
osc strength	1.6251	0.5552
composition	H->L	H->L 81%
VIS peak, nm		380
osc strength		0.6395
composition		H-1>L 88%
Dihedrals angle		
FLTPA core to TPA end-capping unit	31	31
FL penta core to inside TPA unit	21	20

Table S2. A summary of fluorene-based hole transporting materials for perovskite solar cells.

HTM	Role of Fluorene	HOMO [eV]	T _d [°C]	T _g [°C]	Hole mobility [10 ⁻⁴ cm ² V ⁻¹ s ⁻¹]	Device structure	Active area [cm ²]	Additives	V _{oc} [V]	J _{sc} [mAcm ⁻²]	FF [%]	PCE [%] ^{b)}	References
TPA-2,7-FLTPA-TPA	Central core	-5.45	434	162	-	ITO/MoO ₃ /HTM/MAPbI ₃ /C ₆₀ /B CP/Ag	0.01	None	1.052	20.82	78	17.1 (15.9) ^{c)}	This study
TPA-3,6-FLTPA-TPA	Central core	-5.57	456	167	-	ITO/MoO ₃ /HTM/MAPbI ₃ /C ₆₀ /B CP/Ag	0.01	None	1.051	19.85	67	13.9 (15.9)	This study
DP	End-capper	-5.27	460	161	-	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.95} (MAPbBr ₃) _{0.05} /HTM/Au	0.0939	Li-TFSI + <i>t</i> BP	1.144	24.91	81.29	23.2 (23.05)	1
Dispiro-OBuTAD	Central core	-5.16	420	98	0.102	FTO/TiO ₂ /MAPbI ₃ /Dispiro/Au	-	Li-TFSI + <i>t</i> BP	1.08	22.79	75	18.46 (17.82)	2
S101	Central core	-5.32	124	45	0.72	c-TiO ₂ /mp-TiO ₂ /MAPbI ₃ /HTM/Au	0.2	Li-TFSI + (CF ₃ SO ₂) ₂ N + <i>t</i> BP + FK102	0.92	18.9	65	10.6 (12.0)	3
V1050	Central core	-5.11	400	166	1.7	FTO/SnO ₂ /FA _{0.83} Cs _{0.17} Pb(I _{0.8} Br _{0.2}) ₃ /HTM/Au	0.0913	Li-TFSI + <i>t</i> BP + Co[t-BuPyPz] ₃ [TFSI] ₃	1.05	22.0	79.5	18.3 (18.9)	4
V1061	Central core	-5.10	420	146	2.0	FTO/SnO ₂ /FA _{0.83} Cs _{0.17} Pb(I _{0.8} Br _{0.2}) ₃ /HTM/Au	0.0913	Li-TFSI + <i>t</i> BP + Co[t-BuPyPz] ₃ [TFSI] ₃	0.96	21.6	79.7	16.7 (18.9)	4
Yih-1	Central core	-5.34	435	114	0.336	c-TiO ₂ /MAPbI ₃ /HTM/Au	-	Li-TFSI + <i>t</i> BP	1.03	20.91	68	14.72 (16.08)	5
Yih-2	Central core	-5.35	427	159	0.494	c-TiO ₂ /MAPbI ₃ /HTM/Au	-	Li-TFSI + <i>t</i> BP	1.02	22.18	71	16.06 (16.08)	5
X60	Central core	-5.61	-	-	1.9	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.95} (MAPbBr ₃) _{0.05} /HTM/Au	0.16	Li-TFSI + <i>t</i> BP + FK209	1.14	24.2	71	19.84	6
MC8-9-NPC	Central core	-5.51	-	-	-	ITO/HTM/MAPbI ₃ /PC ₆₁ PM/Al/LiF	0.05	-	0.90	20.80	74	13.85 (11.90)	7

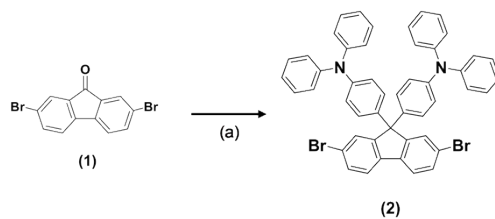
HT1	Central core	-	-	-	1.12	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.85} (MAPbBr ₃) _{0.15} /HTM/Au	0.16	Li-TFSI + <i>t</i> BP + FK209	1.12	21.91	70	17.18 (18.27)	8
HT2	Central core	-	-	-	1.04	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.85} (MAPbBr ₃) _{0.15} /HTM/Au	0.16	Li-TFSI + <i>t</i> BP + FK209	1.11	22.26	73	18.04 (18.27)	8
SDF-OMeTAD	Central core	-5.28	400	-	-	FTO/SnO ₂ /MAPbI ₃ /HTM/Au	0.09	Li-TFSI + <i>t</i> BP	1.10	19.06	62	13.01 (17.13)	9
SDF-OMeTAD	Central core	-5.28	400	-	-	FTO/SnO ₂ /MAPbI ₃ /HTM/Au	0.09	-	1.01	19.02	25	4.83 (4.27)	9
HTM4	Central core	-	413	89	0.22	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.95} (MAPbBr ₃) _{0.05} /HTM/Au	0.16	Li-TFSI + <i>t</i> BP + FK209	1.052	21.3	75	16.5 (17.88)	10
WY-1	Central core	-5.16	401	122	-	FTO/c-TiO ₂ / mp-TiO ₂ /MAPbI ₃ /HTM/Au	0.1	Li-TFSI + <i>t</i> BP	1.053	19.48	69.1	14.19 (14.84)	11
BTPA-6	Central core	-5.34	400	143	-	FTO/c-TiO ₂ / mp-TiO ₂ /MAPbI ₃ /HTM/Au	1.02	Li-TFSI + <i>t</i> BP	1.036	20.61	64.7	13.81 (13.25)	12
mp-SFX-2PA	Central core	-5.06	410	-	0.3	FTO/c-TiO ₂ / mp-TiO ₂ /MAPbI ₃ /HTM/Au	0.1	Li-TFSI + <i>t</i> BP	1.023	21.17	77.44	16.77 (15.45)	13
mp-SFX-2PA	Central core	-5.06	410	-	0.3	c-TiO ₂ /mp-TiO ₂ /(FAPbI ₃) _{0.95} (MAPbBr ₃) _{0.05} /HTM/Au	0.1	Li-TFSI + <i>t</i> BP	1.083	20.87	78.31	17.70 (17.64)	13

^{a)} Photovoltaic parameters of the solar cells with HTMs evaluated under 1 sunlight illumination (100 mW cm⁻²) condition.

^{b)} The best PCE of the HTMs based devices

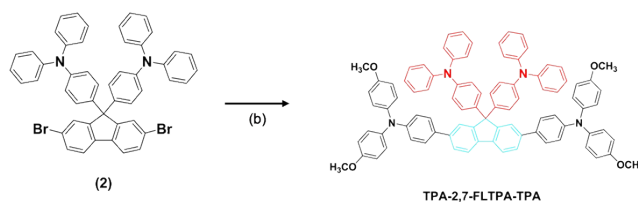
^{c)} The best PCE of standard conventional HTMs

Table S3. Materials quantities and cost evaluation for the synthesis of 4,4'-(2,7-dibromo-9H-fluorene-9,9-diyl)bis(*N,N*-diphenylaniline) (**2**)



Chemical Name	Weight reagent [g]	Weight solvent [mL]	Weight of workup [g or mL]	Price of chemical [AUD/g or AUD/mL]	Material Cost [AUD/g product]	Cost per step [AUD/step]
2,7-dibromo-FN	1.33			180	2.39	30.38
TPA	15.45			194	5.99	
methane sulfonic acid	0.44			147	0.13	
DCM		250		15.5	1.55	
Na ₂ SO ₄			1	10.9	0.02	
Na ₂ CO ₃			2	67	0.27	
SiO ₂			500	704	14.08	
Hexane			400	16.05	2.57	
DCM			400	15.5	2.48	
Acetone			150	14.85	0.89	

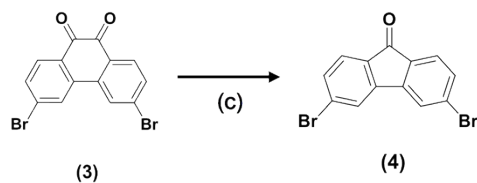
Table S4. Materials quantities and cost evaluation for the synthesis of 4,4'-(2,7-bis(4-(bis(4-methoxyphenyl)amino)phenyl)-9*H*-fluorene-9,9-diyl)bis(*N,N*-diphenylaniline) (TPA-2,7-FLTPA-TPA)



Chemical Name	Weight reagent [g]	Weight solvent [mL]	Weight of workup [g or mL]	Price of chemical [AUD/g or AUD/mL]	Material Cost [AUD/g product]	Cost per step [AUD/step]
K ₂ CO ₃	16.60			171	2.84	29.16
Tetrakis(triphenylphosphine)palladium	0.22			120	5.28	
Toluene		100		71.5	7.15	
CF		250		18.5	1.85	
Na ₂ SO ₄			1	10.9	0.02	
SiO ₂			250	704	7.04	
Hexane			200	16.05	1.28	
DCM			200	15.5	1.24	
Methanol			150	41	2.46	

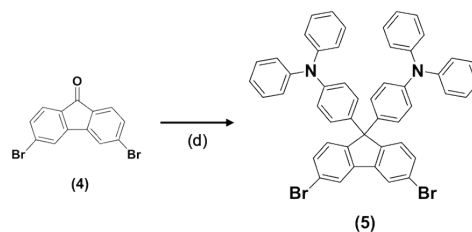
Total cost is ~119.83 AUD/g. It is equal to ~92.89 \$/g.

Table S5. Materials quantities and cost evaluation for the synthesis of 3,6-dibromo-9*H*-fluoren-9-one (**4**)



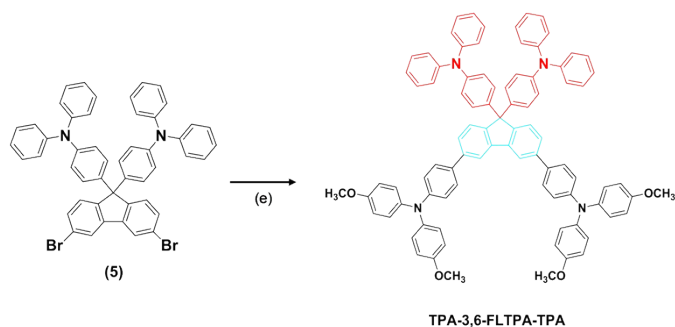
Chemical Name	Weight reagent [g]	Weight solvent [mL]	Weight of workup [g or mL]	Price of chemical [AUD/g or AUD/mL]	Material Cost [AUD/g product]	Cost per step [AUD/step]
Compound 3	2.39			542	12.95	16.59
KOH	4.80			11	0.11	
KMnO ₄	2.04			103	0.42	
CF		300		18.5	2.22	
Acetone			150	14.85	0.89	

Table S6. Materials quantities and cost evaluation for the synthesis of 4,4'-(3,6-dibromo-9*H*-fluorene-9,9-diyl)bis(*N,N*-diphenylaniline) (**5**)



Chemical Name	Weight reagent [g]	Weight solvent [mL]	Weight of workup [g or mL]	Price of chemical [AUD/g or AUD/mL]	Material Cost [AUD/g product]	Cost per step [AUD/step]
TPA	18.42			194	7.15	29.16
methane sulfonic acid	0.53			147	0.16	
DCM		250		15.5	1.55	
Na ₂ SO ₄			1	10.9	0.02	
Na ₂ CO ₃			2	67	0.27	
SiO ₂			500	704	14.08	
Hexane			400	16.05	2.57	
DCM			400	15.5	2.48	
Acetone			150	14.85	0.89	

Table S7. Materials quantities and cost evaluation for the synthesis of 4,4'-(3,6-bis(4-(bis(4-methoxyphenyl)amino)phenyl)-9*H*-fluorene-9,9-diyl)bis(*N,N*-diphenylaniline) (TPA-3,6-FLTPA-TPA)



Chemical Name	Weight reagent [g]	Weight solvent [mL]	Weight of workup [g or mL]	Price of chemical [AUD/g or AUD/mL]	Material Cost [AUD/g product]	Cost per step [AUD/step]
K ₂ CO ₃	16.60			171	2.84	30.12
Tetrakis(triphenylphosphine)palladium	0.26			120	6.24	
Toluene		100		71.5	7.15	
CF		250		18.5	1.85	
Na ₂ SO ₄			1	10.9	0.02	
SiO ₂			250	704	7.04	
Hexane			200	16.05	1.28	
DCM			200	15.5	1.24	
Methanol			150	41	2.46	

Total cost is ~136.17 AUD/g. It is equal to ~105.56 \$/g.

References

1. N. J. Jeon, H. Na, E. H. Jung, T.-Y. Yang, Y. G. Lee, G. Kim, H.-W. Shin, S. Il Seok, J. Lee and J. Seo, *Nat. Energy*, 2018, DOI: 10.1038/s41560-018-0200-6.
2. W. Yu, J. Zhang, X. Wang, X. Liu, D. Tu, J. Zhang, X. Guo and C. Li, *Sol. RRL*, 2018, DOI: 10.1002/solr.201800048.
3. A. Krishna, D. Sabba, J. Yin, A. Bruno, L. J. Antila, C. Soci, S. Mhaisalkar and A. C. Grimsdale, *J. Mater. Chem. A*, 2016, **4**, 8750-8754.
4. S. Daskeviciu Te, N. Sakai, M. Franckevicius, M. Daskeviciene, A. Magomedov, V. Jankauskas, H. J. Snaith and V. Getautis, *Adv. Sci.*, 2018, **5**, 1700811.
5. Y. C. Chen, S. K. Huang, S. S. Li, Y. Y. Tsai, C. P. Chen, C. W. Chen and Y. J. Chang, *ChemSusChem*, 2018, **11**, 3225-3233.
6. B. Xu, D. Bi, Y. Hua, P. Liu, M. Cheng, M. Grätzel, L. Kloo, A. Hagfeldt and L. Sun, *Energy Environ. Sci.*, 2016, **9**, 873-877.
7. L. Bai, Z. Wang, Y. Han, Z. Zuo, B. Liu, M. Yu, H. Zhang, J. Lin, Y. Xia, C. Yin, L. Xie, Y. Chen, Z. Lin, J. Wang and W. Huang, *Nano Energy*, 2018, **46**, 241-248.
8. Y. Hua, J. Zhang, B. Xu, P. Liu, M. Cheng, L. Kloo, E. M. J. Johansson, K. Sveinbjörnsson, K. Aitola, G. Boschloo and L. Sun, *Nano Energy*, 2016, **26**, 108-113.
9. Z. Li, J. Chen, H. Li, Q. Zhang, Z. Chen, X. Zheng, G. Fang, H. wang and Y. Hao, *RSC Adv.*, 2017, **7**, 41903-41908.
10. R. Tiazkis, S. Paek, M. Daskeviciene, T. Malinauskas, M. Saliba, J. Nekrasovas, V. Jankauskas, S. Ahmad, V. Getautis and M. Khaja Nazeeruddin, *Sci. Rep.*, 2017, **7**, 150.
11. Y. Wang, T. S. Su, H. Y. Tsai, T. C. Wei and Y. Chi, *Sci. Rep.*, 2017, **7**, 7859.
12. G. Wu, Y. Zhang, R. Kaneko, Y. Kojima, K. Sugawa, T. H. Chowdhury, A. Islam, Q. Shen, M. Akhtaruzzaman, T. Noda and J. Otsuki, *Sol. RRL*, 2017, **1**.
13. K. Liu, Y. Yao, J. Wang, L. Zhu, M. Sun, B. Ren, L. Xie, Y. Luo, Q. Meng and X. Zhan, *Mater. Chem. Front.*, 2017, **1**, 100-110.