

Fe-N-Doped Hierarchical Mesoporous Carbon Nanomaterials as Efficient Catalysts for Oxygen Reduction in Both Acidic and Alkaline Media

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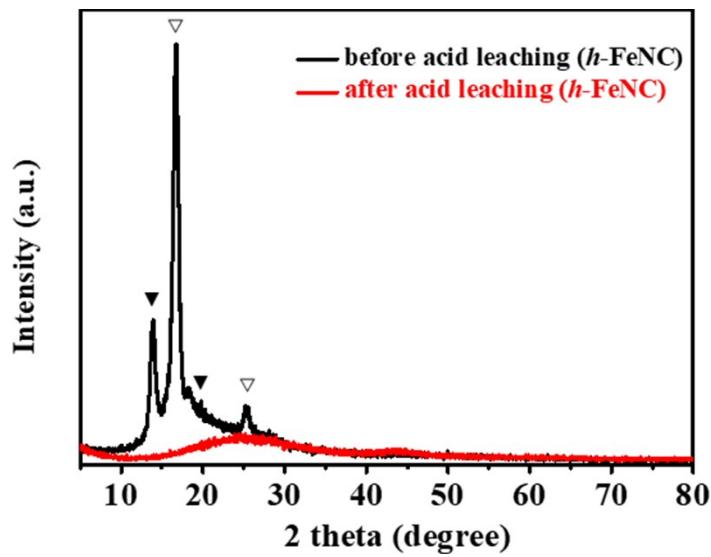


Fig. S1 XRD test of *h*-FeNC with and without acid leaching.

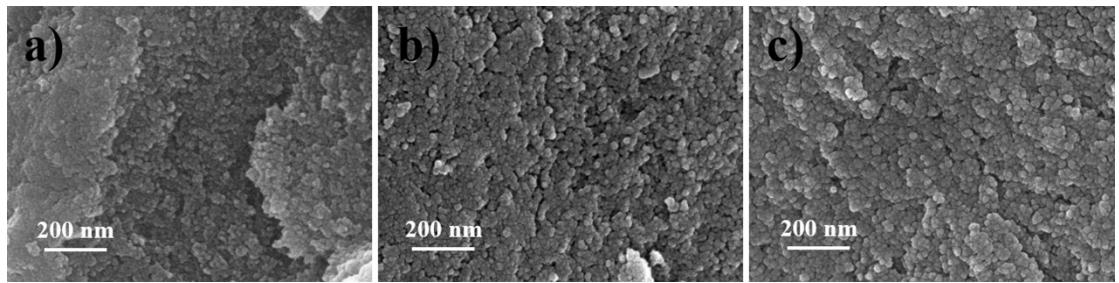


Fig. S2 SEM of *p*-FeNC, *r*-FeNC and *c*-FeNC.

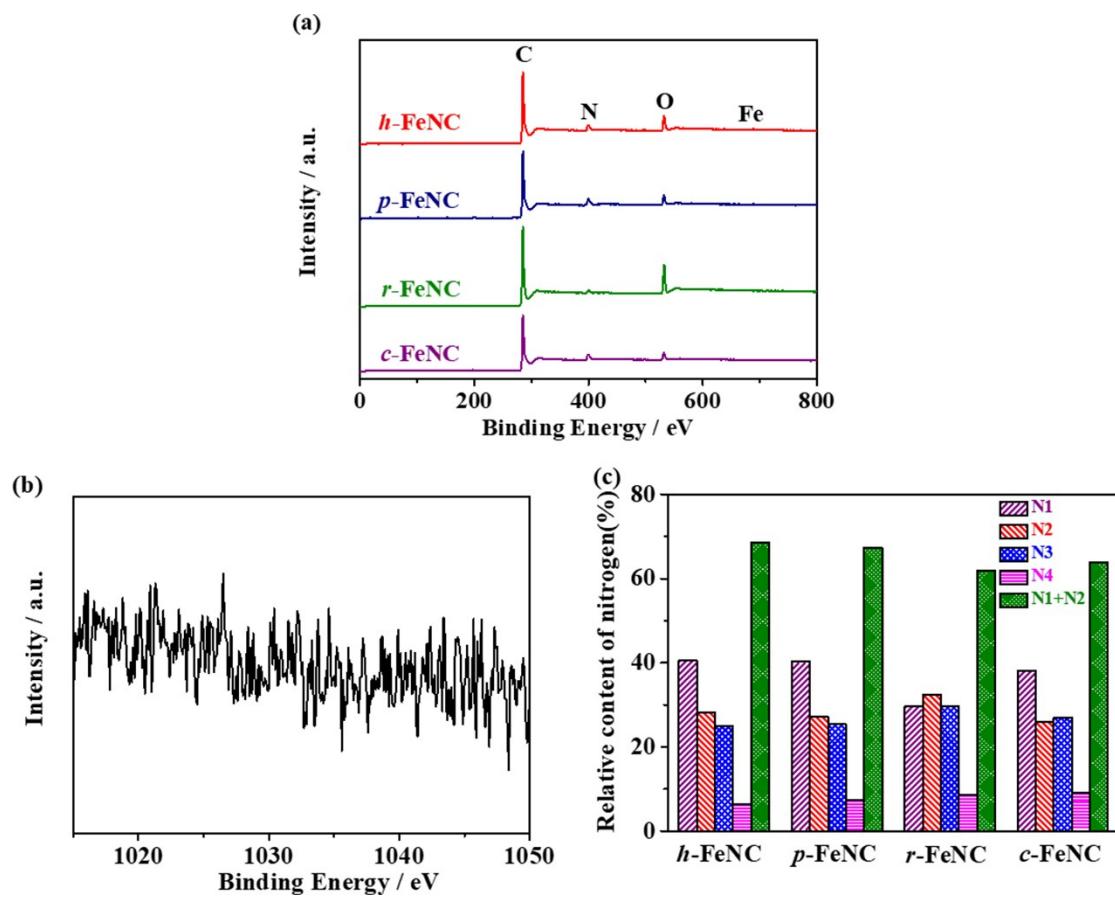


Fig. S3 (a) Full XPS spectrum of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC. (b) Zn 2p XPS spectrate of *h*-FeNC and (c) the ratio of different N species of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

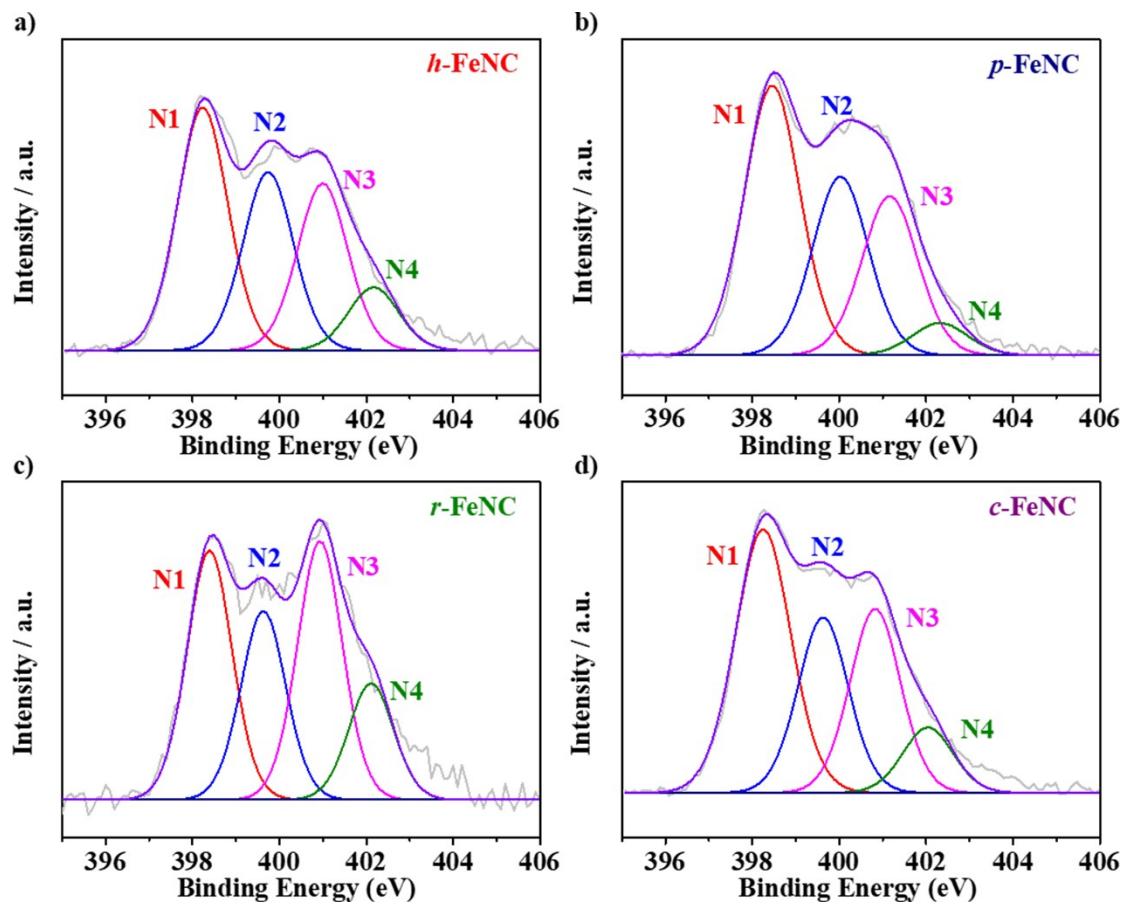


Fig. S4 N1s XPS deconvolution results of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

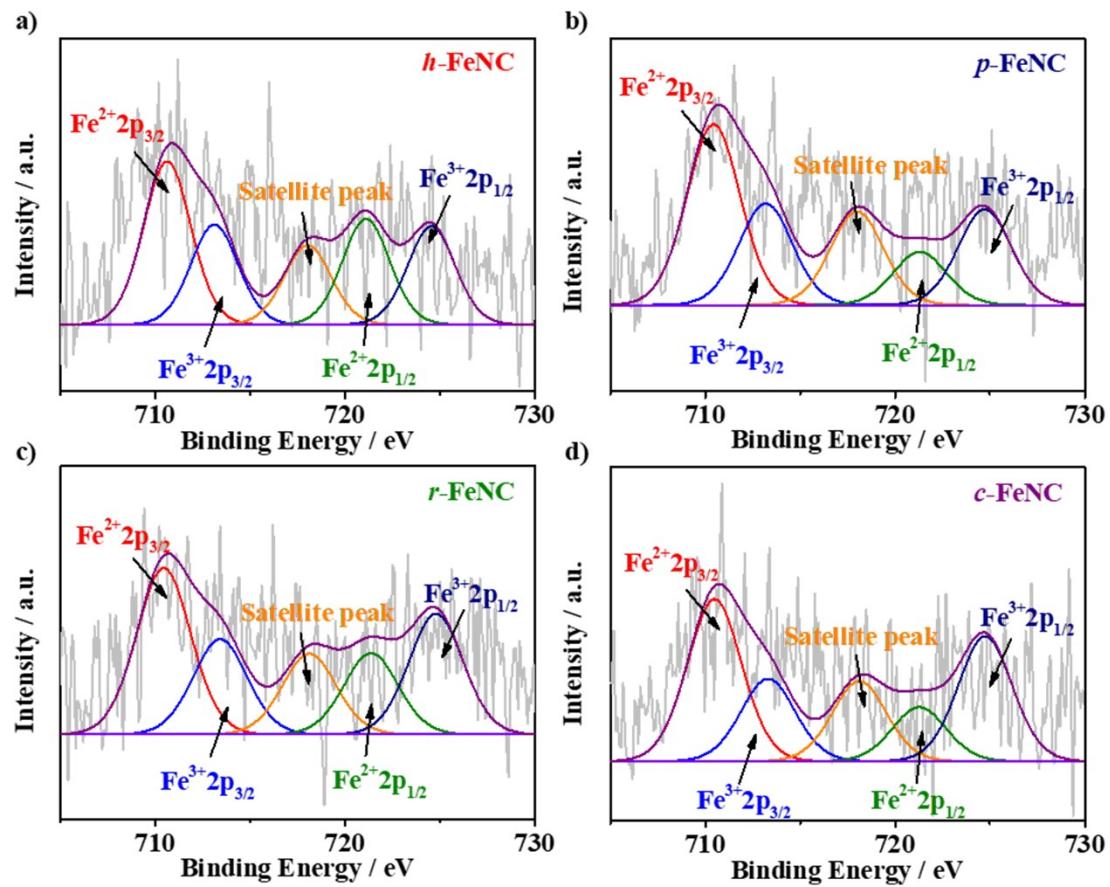


Fig. S5 Fe 2p XPS deconvolution results of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

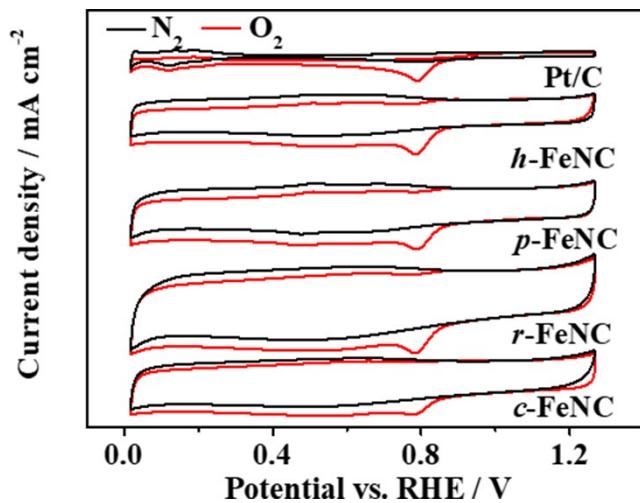


Fig. S6 CVs of Pt/C, *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC in N₂- and O₂-saturated 0.1 M HClO₄ solution at a scan rate of 10 mV s⁻¹.

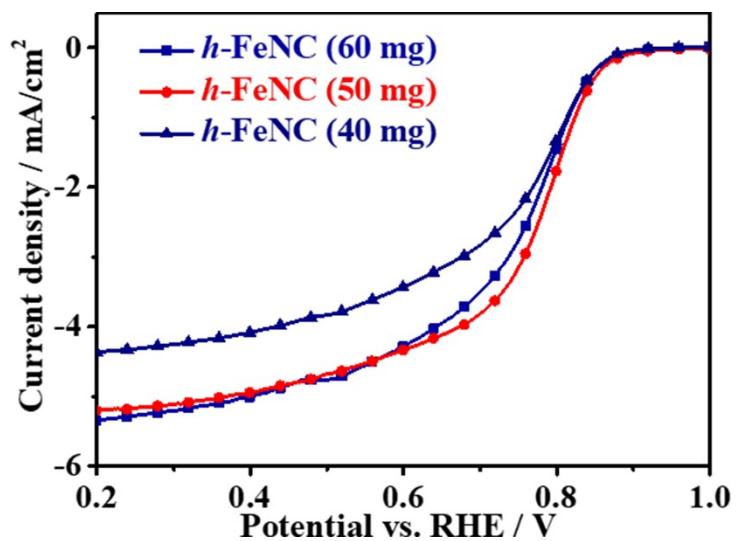


Fig. S7 LSV of *h*-FeNC with different Fe_2O_3 content in 0.1 M HClO_4 solution at a scan rate of 10 mV s⁻¹.

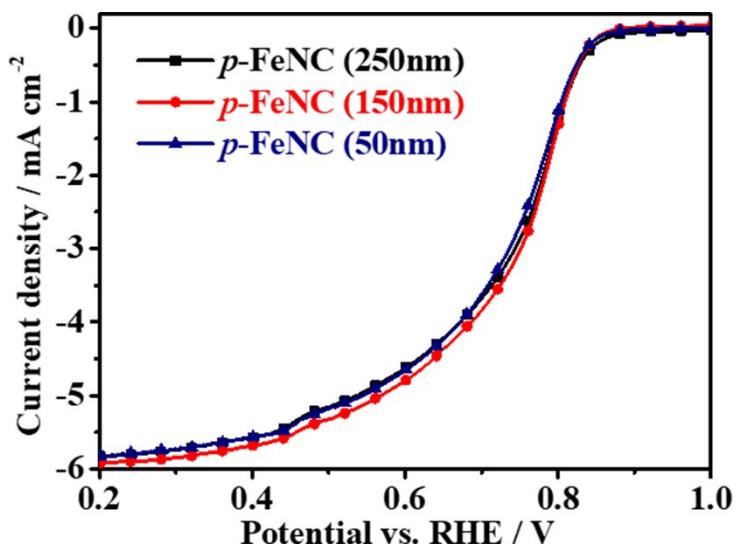


Fig. S8 RDE polarization curves of *p*-FeNC (250 nm), *p*-FeNC (150 nm) and *p*-FeNC (50 nm) in O₂-saturated 0.1 M HClO₄ solution at 1600 rpm, with a scan rate of 10 mV s⁻¹.

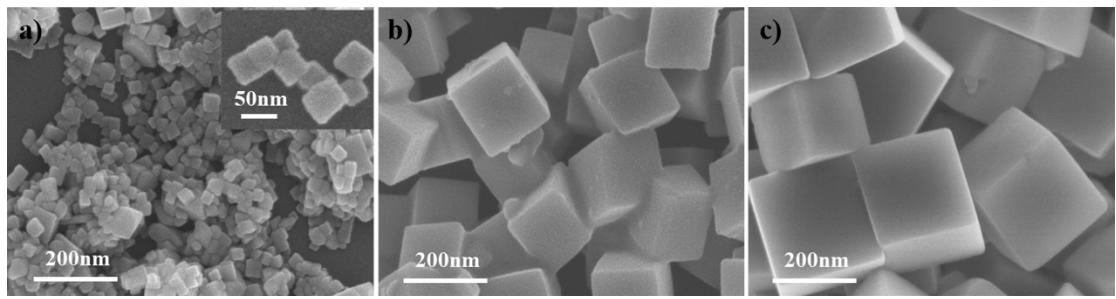


Fig. S9 SEM of *p*-Fe₂O₃ with different sizes (a) 50 nm, (b) 150nm and (c) 250 nm.

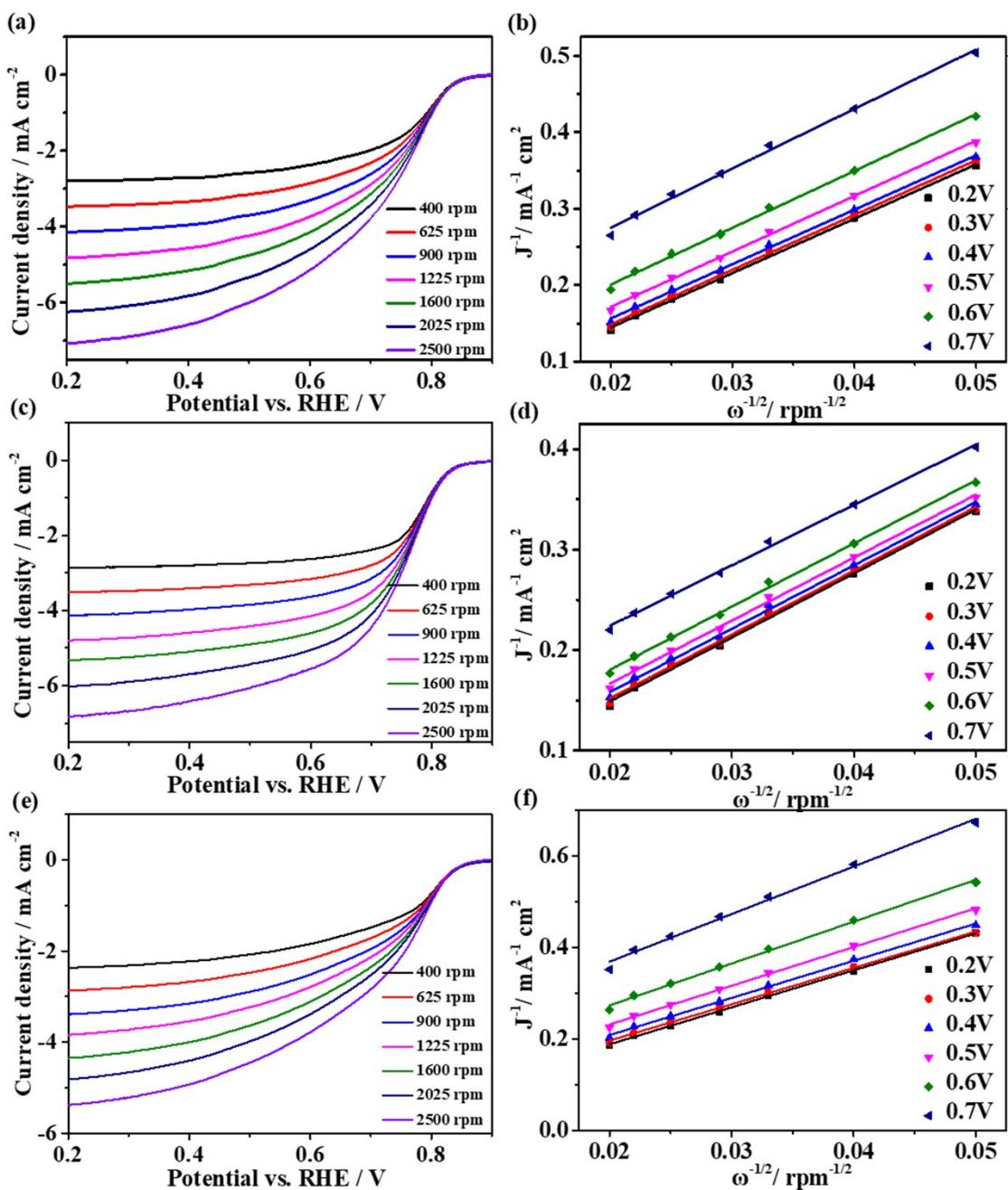


Fig. S10 (a, c, e) LSV curves of *p*-FeNC, *r*-FeNC and *c*-FeNC at different rotating speeds with a scan rate of 10 mV s^{-1} in O_2 -saturated 0.1 M HClO_4 solution, respectively. (b, d, f) the corresponding Koutecky-Levich plots of *p*-FeNC, *r*-FeNC and *c*-FeNC, respectively.

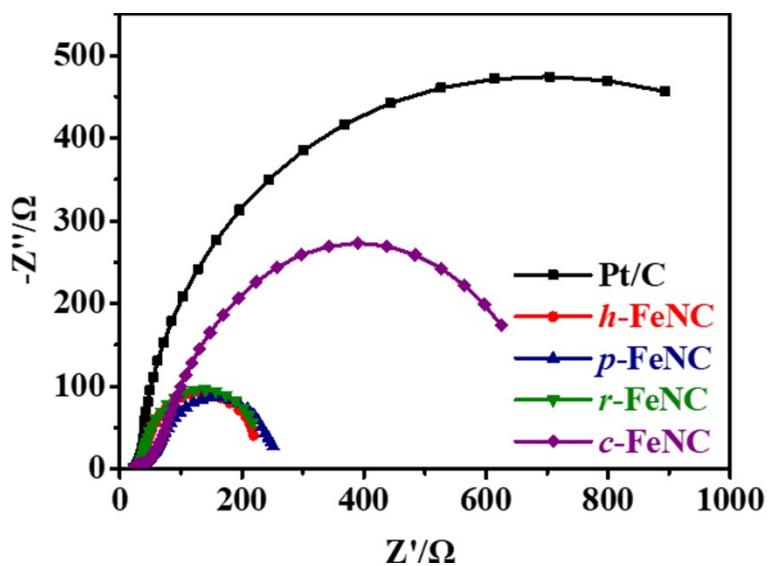


Fig. S11 The electrochemical impedance spectra of Pt/C, *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

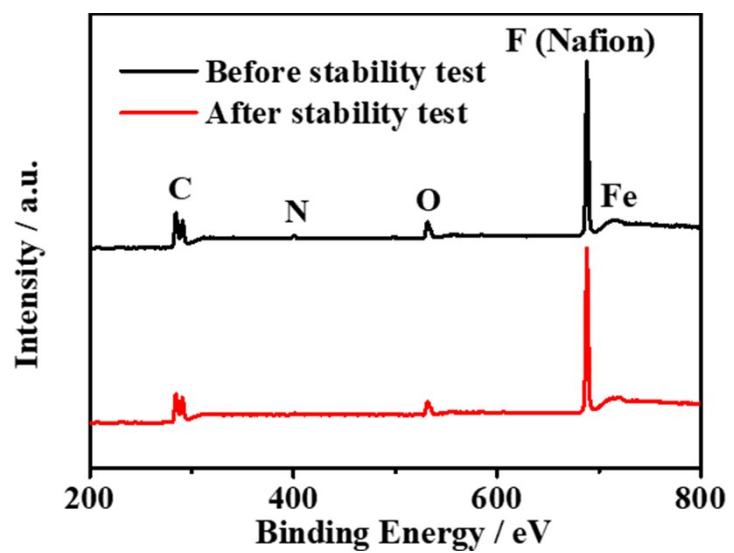


Fig. S12 Full XPS spectrum of *h*-FeNC before and after stability test.

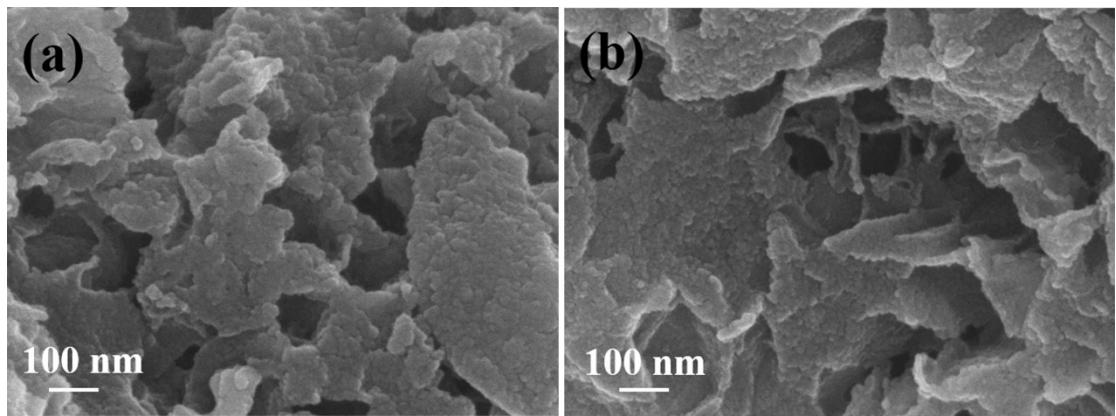


Fig. S13 SEM of *h*-FeNC before and after stability test.

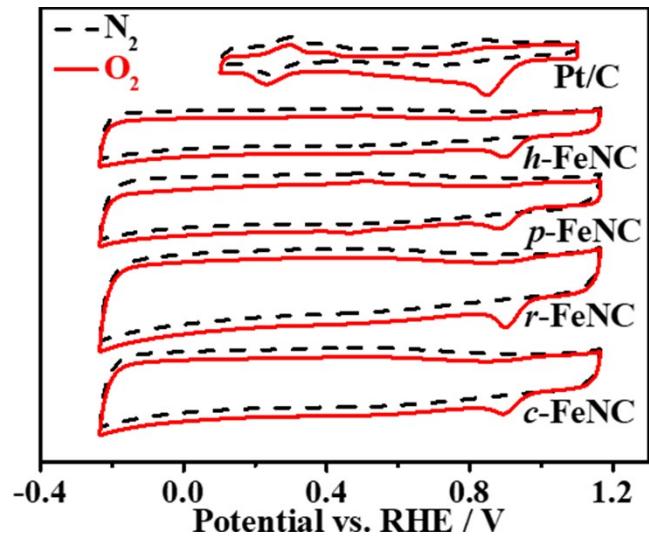


Fig. S14 CVs of Pt/C, *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC in N_2 - and O_2 -saturated 0.1 M KOH solution at a scan rate of 10 mV s^{-1} .

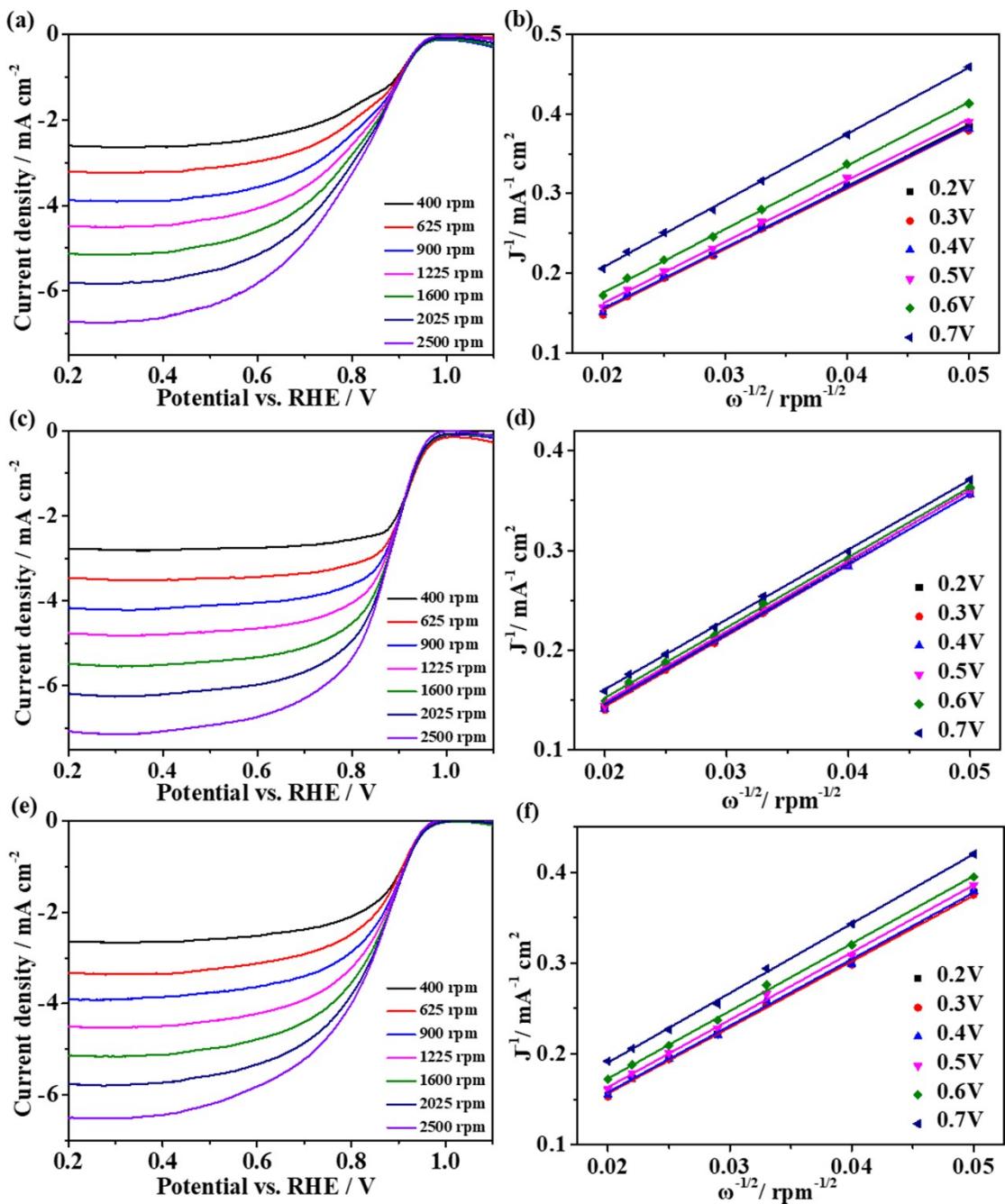


Fig. S15 (a, c, e) LSV curves of *p*-FeNC, *r*-FeNC and *c*-FeNC at different rotating speeds with a scan rate of 10 mV s^{-1} in O_2 -saturated 0.1 M KOH solution, respectively. (b, d, f) the corresponding Koutecky-Levich plots of *p*-FeNC, *r*-FeNC and *c*-FeNC, respectively.

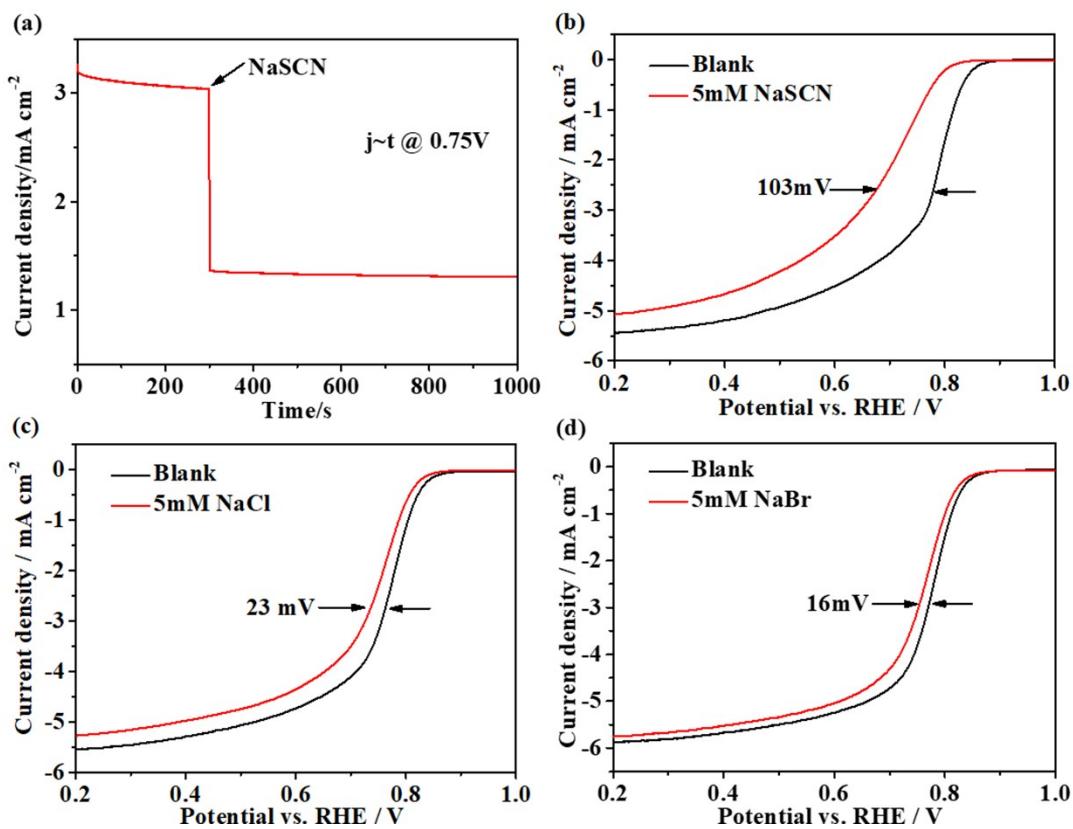


Fig. S16 (a) Current-time (I - t) curves and (b) RDE results of h -FeNC in O_2 -saturated 0.1 M HClO_4 solution (without and with SCN^-). (c, d) RDE results of h -FeNC in O_2 -saturated 0.1 M HClO_4 solution (without and with Cl^- , without and with Br^-). All tests on a rotating disk electrode (1600 rpm) with a scan rate of 10 mV s^{-1} .

Table S1 Nitrogen sorption analysis results of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

Sample	S _{BET} (m ² g ⁻¹)	V _{tol} (cm ³ g ⁻¹)	D _{av} (nm)
<i>h</i> -FeNC	2454	1.9	3.4
<i>p</i> -FeNC	2248	1.7	3.3
<i>r</i> -FeNC	1934	1.6	3.2
<i>c</i> -FeNC	1864	1.6	3.2

Table S2 The XPS atomic ratio of *h*-FeNC, *p*-FeNC, *r*-FeNC and *c*-FeNC.

Sample	XPS atomic ratio (%)				
	N1	N2	N3	N4	N1+N2
<i>h</i> -FeNC	40.51	28.14	24.95	6.40	68.65
<i>p</i> -FeNC	40.22	27.17	25.36	7.25	67.39
<i>r</i> -FeNC	29.53	32.38	29.52	8.57	61.91
<i>c</i> -FeNC	38.04	25.84	26.97	9.15	63.88

Table S3 The electrocatalytic activities of *h*-FeNC and some recently reported NMP catalysts for ORR in acidic media.

Catalyst	Electrode rotation speed (rpm)	Half-wave Potential (mV)	Catalyst loading (mg cm ⁻²)	Reference
<i>h</i> -FeNC	1600	777	0.4	This work
Co-N-C	1200	761	0.6	S1
FePPyC-900	1600	740	0.4	S2
Fe ₃ C/C-700	900	730	0.6	S3
PANI-Co-C	900	750	0.6	S4
Fe-N-GC-900	1600	740	0.6	S5
Fe-Nx/HPC	1600	760	0.2	S6
Fe-g-C ₃ N ₄ @C	900	750	0.6	S7
FeNC-900	1600	720	0.6	S8

Table S4 The change of C, N, O and Fe content of *h*-FeNC before and after the stability test.

Sample	XPS			
	$\Delta C/C$ (at %)	$\Delta N/N$ (at %)	$\Delta O/O$ (at %)	$\Delta Fe/Fe$ (at %)
<i>h</i> -FeNC	-0.71	+6.41	+2.43	+1.20

Table S5 The electrocatalytic activities of *h*-FeNC and some recently reported NMP catalysts for ORR in 0.1 M KOH solution.

Catalyst	Electrode rotation speed (rpm)	Half-wave potential (mV)	Catalyst loading (mg cm ⁻²)	Reference
<i>h</i> -FeNC	1600	883	0.4	This work
NiCo/NLG-270	1600	820	0.4	S9
Co/N-CNTs	1600	840	0.2	S10
Fe-N/C-800	1600	809	0.1	S11
PFA-Fe20-900-ALP	1500	830	0.8	S12
Fe@Aza-PON	1600	839	0.239	S13
Fe-Phen-N-800	1600	860	0.1	S14
Fc-F/Co@N-C800	1600	860	0.2	S15
NCNT/Co _x Mn _{1-x} O	1600	840	0.2	S16

References

- 1 B. You, N. Jiang, M. Sheng, W. S. Drisdell, J. Yano and Y. Sun, *ACS Catal.*, 2015, **5**, 7068-7076.
- 2 T. N. Tran, M. Y. Song, K. P. Singh, D. S. Yang and J. S. Yu, *J. Mater. Chem. A*, 2016, **4**, 8645-8657.
- 3 Y. Hu, J. O. Jensen, W. Zhang, L. N. Cleemann, W. Xing, N. J. Bjerrum and Q. Li, *Angew. Chem. Int. Ed.*, 2014, **53**, 3675-3679.
- 4 G. Wu, K. L. More, C. M. Johnston and P. Zelenay, *Science*, 2011, **332**, 443-447.
- 5 A. Kong, X. Zhu, Z. Han, Y. Yu, Y. Zhang, B. Dong and Y. Shan, *ACS Catal.*, 2014, **4**, 1793-1800.
- 6 Z. Zhang, X. Gao, M. Dou, J. Ji and F. Wang, *J. Mater. Chem. A*, 2017, **5**, 1526-1532.
- 7 M. Wang, W. Yang, H. Wang, C. Chen, Z. Zhou and S. Sun, *ACS Catal.*, 2014, **4**, 3928-3936.
- 8 Q. Zuo, P. Zhao, W. Luo and G. Cheng, *Nanoscale*, 2016, **8**, 14271-14277.
- 9 X. Wang, J. Liu, Z. Liu, W. Wang, J. Luo, X. Han, X. Du, S. Qiao and J. Yang, *Adv. Mater.*, 2018, **30**, 1800005.
- 10 Y. Liu, H. Jiang, Y. Zhu, X. Yang and C. Li, *J. Mater. Chem. A*, 2016, **4**, 1694-1701.
- 11 L. Lin, Q. Zhu and A. Xu, *J. Am. Chem. Soc.*, 2014, **136**, 11027-11033.
- 12 L. C. P. Pérez, N. R. Sahraie, J. Melke, P. Elsässer, D. Teschner, X. Huang, R. Krahnert, R. J. White, S. Enthaler, P. Strasser and A. Fischer, *Adv. Funct. Mater.*, 2018, **28**, 1707551.
- 13 S. J. Kim, J. Mahmood, C. Kim, G. Han, S. W. Kim, S. M. Jung, G. Zhu, J. J. D. Yoreo, G. Kim and J. B. Baek, *J. Am. Chem. Soc.*, 2018, **140**, 1737-1742.
- 14 Z. Yang, Z. Zhao, K. Liang, X. Zhou, C. Shen, Y. Liu, X. Wang and A. Xu, *J. Mater. Chem. A*, 2016, **4**, 19037-19044.
- 15 M. Tan, T. He, J. Liu, H. Wu, Q. Li, J. Zheng, Y. Wang, Z. Sun, S. Wang and Y. Zhang, *J. Mater. Chem. A*, 2018, **6**, 8227-8232.
- 16 X. Liu, M. Park, M. G. Kim, S. Gupta, X. Wang, G. Wu and J. Cho, *Nano Energy*, 2016, **20**, 315-325.