

Supporting Information for

**Bifunctional MoS<sub>2</sub>-based solar evaporator for both efficient water  
evaporation and clean freshwater collection**

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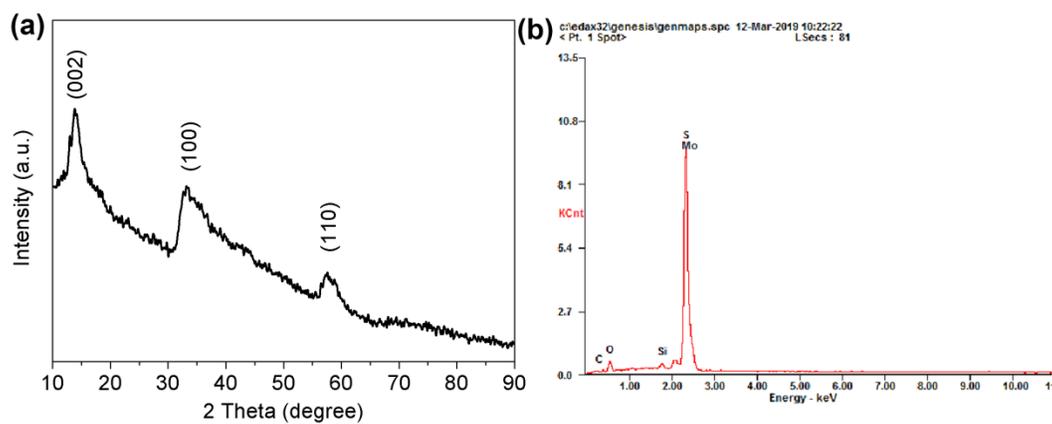
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**Number of figures: 6**

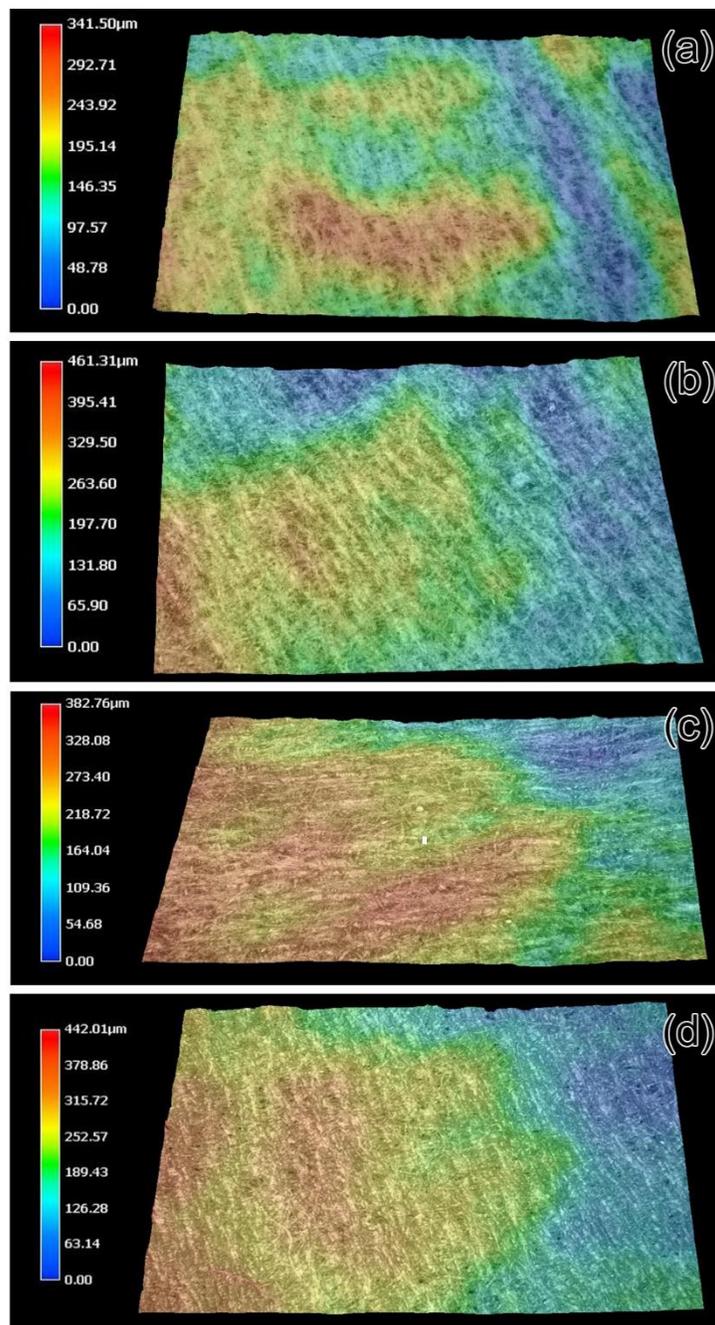
**Number of tables: 3**



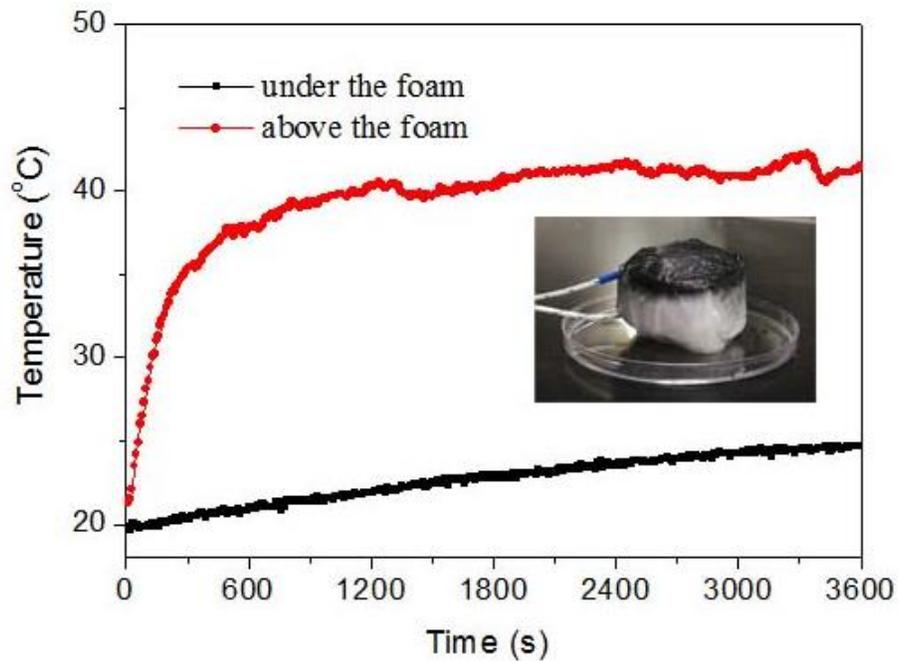
**Fig. S1** XRD pattern and EDS of the as-prepared MoS<sub>2</sub> nanoflowers.



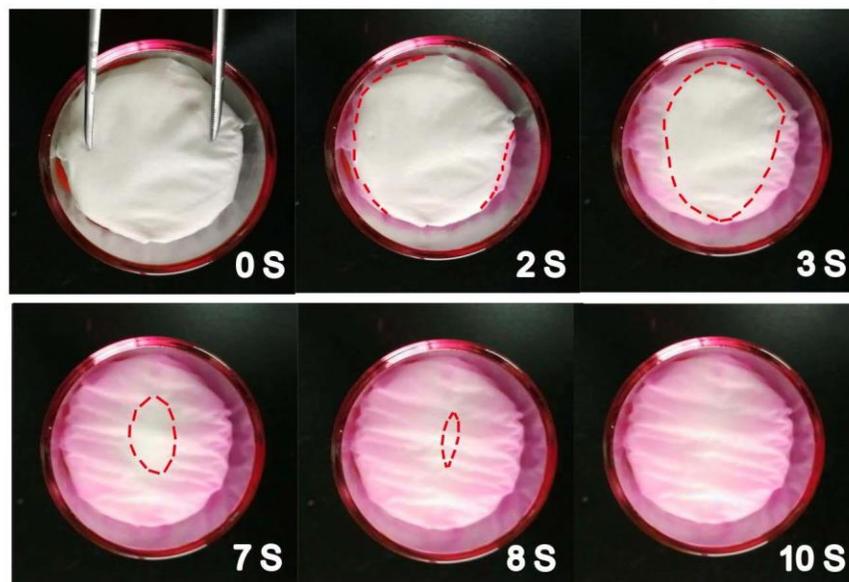
**Fig. S2** Air-laid paper (a) without and with (b) 0.004 kg·m<sup>-2</sup>, (c) 0.016 kg·m<sup>-2</sup>, (d) 0.040 kg·m<sup>-2</sup>, (e) 0.080 kg·m<sup>-2</sup>, (f) 0.160 kg·m<sup>-2</sup> MoS<sub>2</sub> deposition.



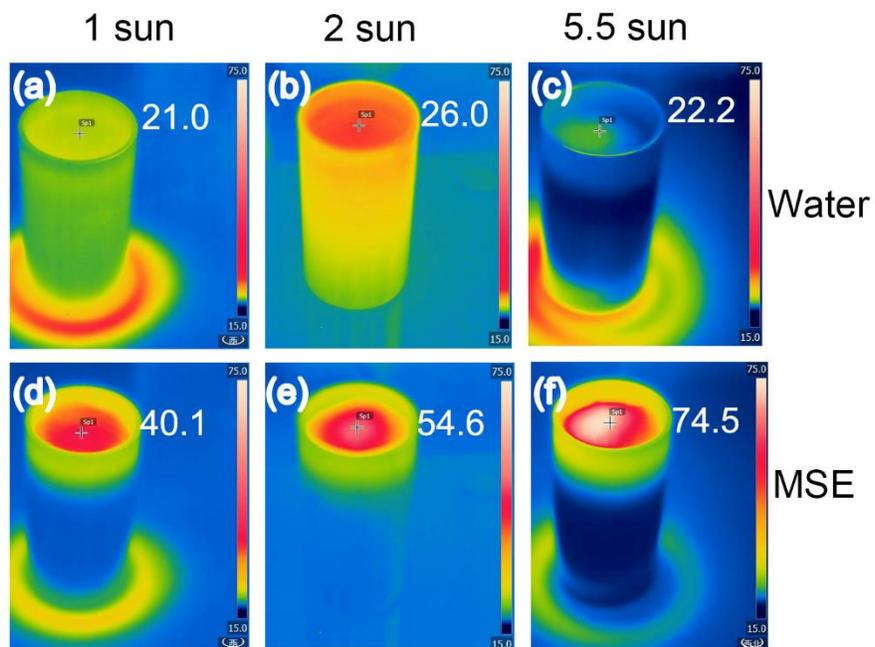
**Fig. S3** The 3D optical microscope images of air-laid paper with (a)  $0.004 \text{ kg}\cdot\text{m}^{-2}$ , (b)  $0.016 \text{ kg}\cdot\text{m}^{-2}$ , (c)  $0.080 \text{ kg}\cdot\text{m}^{-2}$ , and (d)  $0.160 \text{ kg}\cdot\text{m}^{-2}$  MoS<sub>2</sub> nanoflowers deposition.



**Fig. S4** The temperature above and below the foam under 2 sun solar irradiance.



**Fig. S5** Water-uptake behavior of the EPE foam warped with air-laid paper (EPE-ALP).



**Fig. S6** The surface temperatures at first 5 minutes solar irradiance with and without MSE under different intensity of 1, 2, and 5.5 sun.

**Table S1** The proportion of solar absorption of different samples in different spectrum regions.

Sample	MoS <sub>2</sub> (kg·m <sup>-2</sup> )	Total Absorption	UV 280-400 nm	Visible 400-760 nm	Infrared 760-2200 nm
Solar	–	100.00%	4.45%	49.37%	46.18%
ALP	0	3.68%±0.08%	1.08%±0.06%	1.54%±0.01%	1.06%±0.03%
Sample-1	0.004	82.66%±0.07%	3.72%±0.04%	40.82%±0.40%	38.12%±0.36%
Sample-2	0.016	95.39%±0.08%	4.21%±0.01%	46.91%±0.17%	44.27%±0.16%
Sample-3	0.040	96.64%±0.13%	4.24%±0.04%	47.72%±0.07%	44.68%±0.06%
Sample-4	0.080	96.02%±0.70%	4.28%±0.25%	47.26%±0.39%	44.48%±0.21%
Sample-5	0.160	96.15%±0.53%	4.22%±0.11%	46.72%±0.98%	45.21%±0.78%

\*ALP: Air-laid paper

**Table S2** The zero-order kinetic equations and evaporation efficiencies of different water evaporation processes.

Process	Zero-order kinetic equations	$k_1$ (kg·m <sup>-2</sup> ·min <sup>-1</sup> )	$k_2$ (kg·m <sup>-2</sup> ·h <sup>-1</sup> )	R <sup>2</sup>
1#	y=0.00602x-0.02363	0.00602	0.36	0.98106
2#	y=0.02531x-0.02910	0.02531	1.52	0.99905
3#	y=0.00731x-0.01341	0.00731	0.44	0.99764
4#	y=0.01091x-0.03367	0.01091	0.65	0.99311

\*  $k_1$ : water evaporation rate;  $k_2$ : water evaporation rate;  
 Water evaporation under two sun solar irradiation by different processes:  
 (1#) water itself evaporation  
 (2#) MSE with 0.04 kg·m<sup>2</sup> of MoS<sub>2</sub> and 2 cm of EPE foam  
 (3#) air-laid paper wrapped over the 2 cm of EPE foam (paper-foam)  
 (4#) 0.04 kg·m<sup>2</sup> of hydrophobic MoS<sub>2</sub> directly float on water surface

**Table S3** The vapor pressure and Henry's constant of three compounds.

Compound	Molecular Formula	Vapor Pressure mm Hg at 25 °C (est)	Henry's constant atm·m <sup>3</sup> /mole
Nitrobenzene	C <sub>6</sub> H <sub>5</sub> NO <sub>2</sub>	0.245 (PubChem)	2.44 × 10 <sup>-5</sup> (PubChem)
Carbamazepine	C <sub>15</sub> H <sub>12</sub> N <sub>2</sub> O	1.84 × 10 <sup>-7</sup> [1]	1.08 × 10 <sup>-10</sup> (PubChem)
Naproxen	C <sub>14</sub> H <sub>14</sub> O <sub>3</sub>	1.89 × 10 <sup>-6</sup> (PubChem)	3.39 × 10 <sup>-10</sup> (PubChem)

### Energy balance analysis:

According to previous report of J. Zhu's group and Qu's group [2,3], the main energy consumption under input heat flux of 1 kW·m<sup>-2</sup> could be caused by the approach as follows:

(1) Water evaporation consumption ( $\eta$ ):

The water evaporation consumption is 80%, which is equal to the evaporation efficiency.

(2) Reflection loss ( $\eta_{ref}$ ):

The solar absorption of black wood membrane is up to 96.64%. Thus,  $\eta_{ref}$  is about 3.36%.

(3) Conduction loss ( $\eta_{cond}$ ):

$$\eta_{cond} = Cm\Delta T \quad (1)$$

Where  $\eta_{cond}$  is the heat energy,  $C$  is the specific heat capacity of water (4.2 kJ °C<sup>-1</sup>

kg<sup>-1</sup>), m (140 g) is the weight of pure water used in this experiment.  $\Delta T$  (0.08 °C) is the average temperature difference of pure water after and before solar illumination under 1 sun after 1 h. The conduction loss is calculated ~4.7%.

(4) Radiation loss ( $\eta_{rad}$ ):

The radiation heat loss was calculated by the Stefan-Boltzmann equation.

$$\eta_{rad} = \varepsilon A \sigma (T_1^4 - T_2^4) \quad (2)$$

Where  $\eta_{rad}$  represents heat flux,  $\varepsilon$  is the emissivity (assumed a maximum emissivity of 1), and  $A$  is the surface area,  $\sigma$  is the Stefan-Boltzmann constant ( $5.67 \times 10^{-8}$  W/ m<sup>2</sup> K<sup>4</sup>),  $T_1$  is the average surface temperature ( $\approx 314.15$  K) of MSE at a steady state condition, and  $T_2$  is the ambient temperature ( $\approx 308.15$  K) upward the MSE under the illumination of constant 1 kW m<sup>2</sup> solar flux. According to the equation (2), the radiation heat loss is calculated ~6.52%.

(5) Convection loss ( $\eta_{conv}$ ):

$$\eta_{conv} = hA(T_1 - T_2) \quad (3)$$

Where  $\eta_{conv}$  represents heat energy,  $h$  is the convection heat transfer coefficient (assumed to 5 W m<sup>-2</sup> K as reported [4]), and  $A$  is the surface area,  $T_1$  and  $T_2$  are the same as mentioned above. The  $\eta_{conv}$  is 4.77%.

Therefore, the total energy consumption of the five main parts is about 99.35% (80% + 3.36% + 4.70% + 6.52% + 4.77%  $\approx$  99.35%), which is almost all input energy.

## References

- [1] PubChem, U.S. National Library of Medicine. Available online at:  
<https://pubchem.ncbi.nlm.nih.gov/search/>
- [2] N. Xu, X. Hu, W. Xu, X. Li, L. Zhou, S. Zhu and J. Zhu, *Adv. Mater.*, 2017, 29,1606762.
- [3] P. Zhang, Q. Liao, T. Zhang, H. Cheng, Y. Huang, C. Yang, C. Li, L. Jiang and L. Qu, *Nano Energy*, 2018, **46**, 415-422.
- [4] Y. Yang, H. Zhao, Z. Yin, J. Zhao, X. Yin, N. Li, D. Yin, Y. Li, B. Lei, Y. Du and W. Que, *Materials Horizons*, 2018, **5**, 1143-1150.