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A safe and fast-charging lithium-ion battery anode using MXene supported Li_3VO_4

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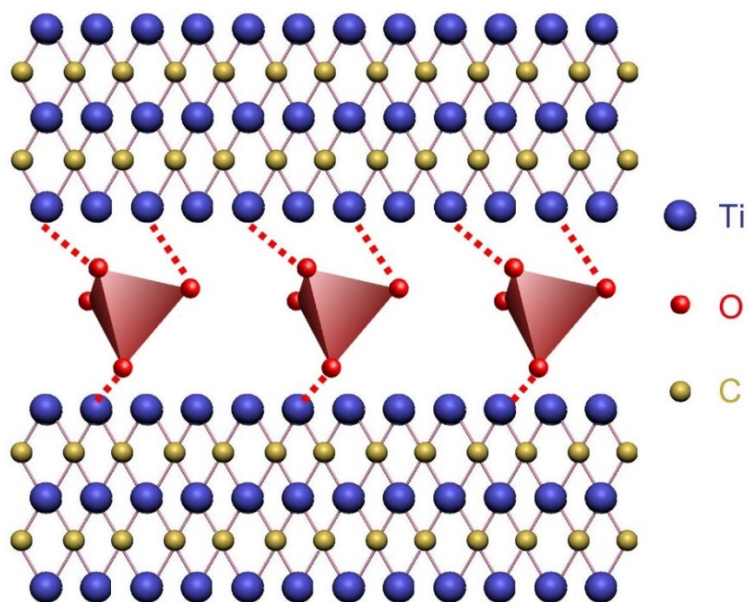


Figure S1. The binding between Ti atomic of MXene and O atomic of LVO.

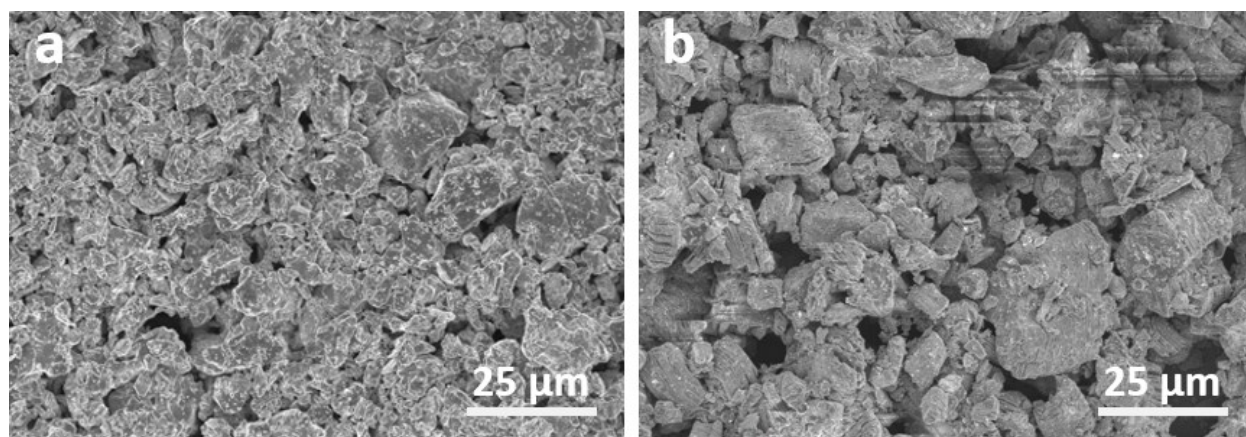


Figure S2. SEM images of (a) MAX and (b) MXene.

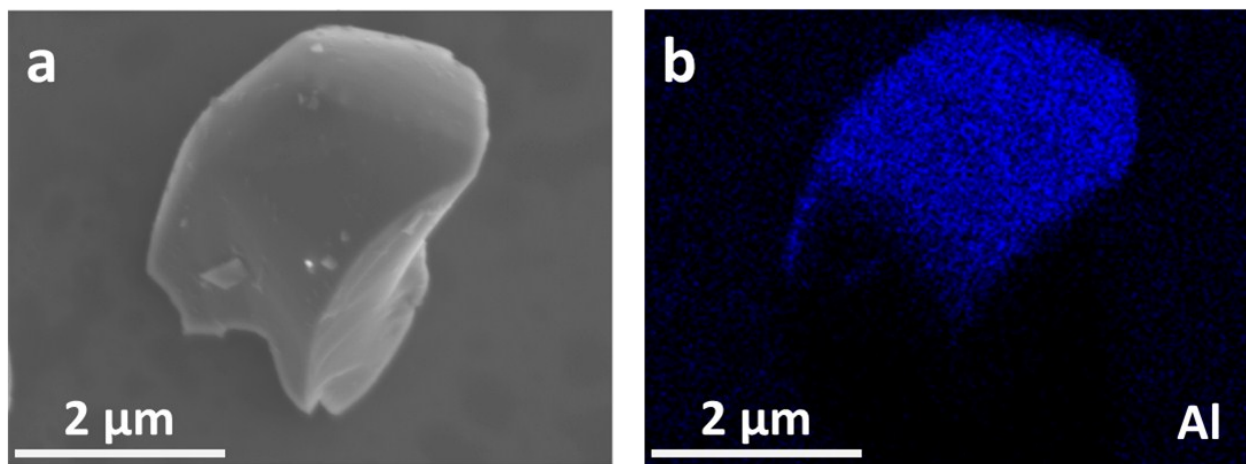


Figure S3. (a) SEM image and (b) EDX mapping of MAX.

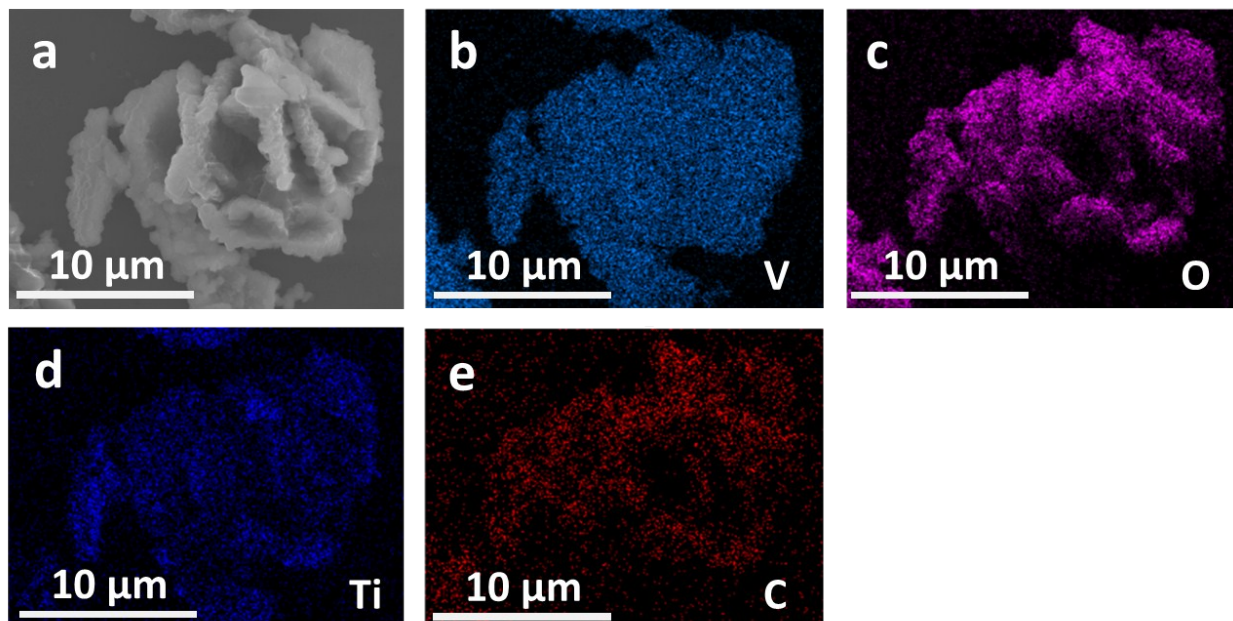


Figure S4. (a) SEM image and corresponding EDX mapping of (b) V, (c) O, (d) Ti and (e) C in the LVO/Ti₃C₂T_x MXene composite.

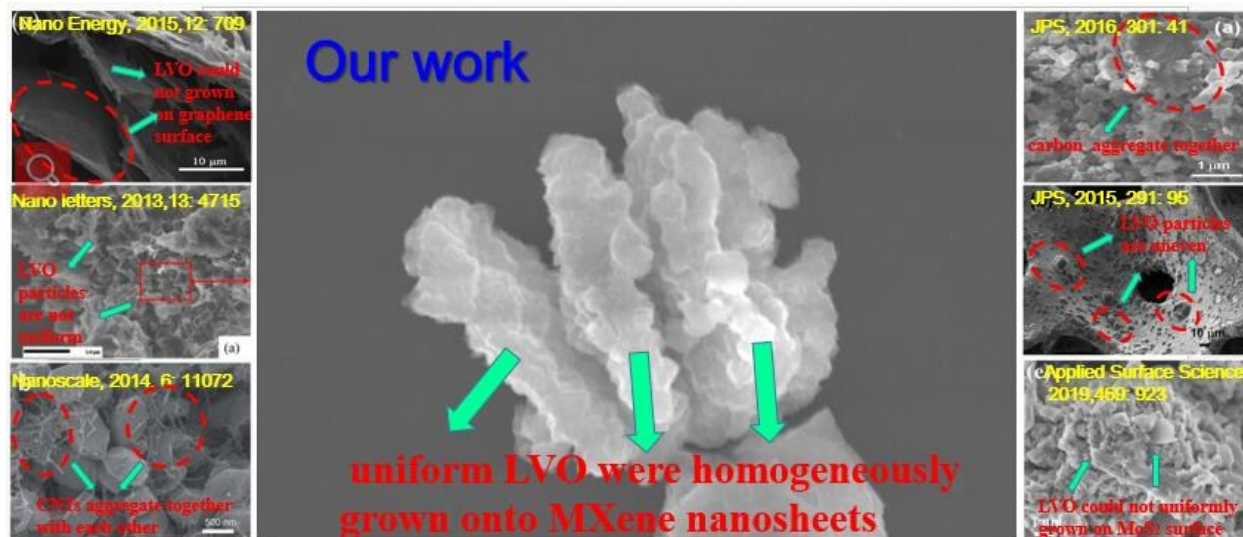


Figure S5. The morphology of LVO composites from past works (six images on the two sides) and our work (center). The intrinsic electrical conductivity of LVO is low. Compositing LVO with conductive materials such as graphene^{1, 2}, carbon nanotubes³, graphite⁴, Ni⁵ and MoS₂⁶ have been tried before. However, most of the past works still contain a large amount of LVO clusters on the surface of MXene nanosheets (**Figure S5**). In our work, the uniform LVO nanoparticles were uniformly grown onto MXene nanosheets. We believe the reason is that MXene is composed of metal and carbon, and the metal ion termination gives it strong affinity with LVO, which allows the uniform growth of LVO on MXene, as illustrated in **Figure S1**.

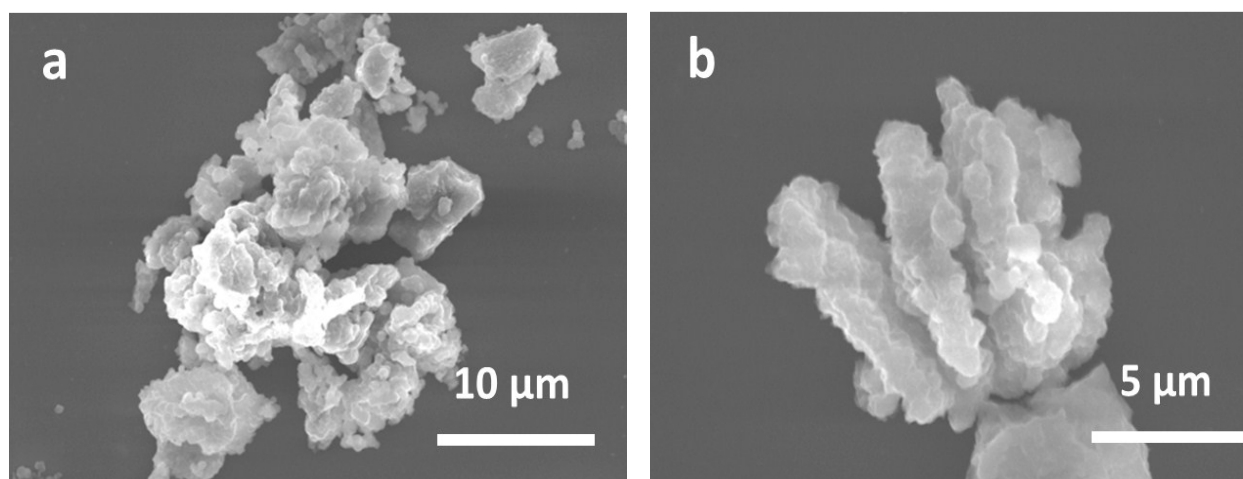


Figure S6. SEM images of LVO/Ti₃C₂T_x MXene composite after 10 minutes sonication.

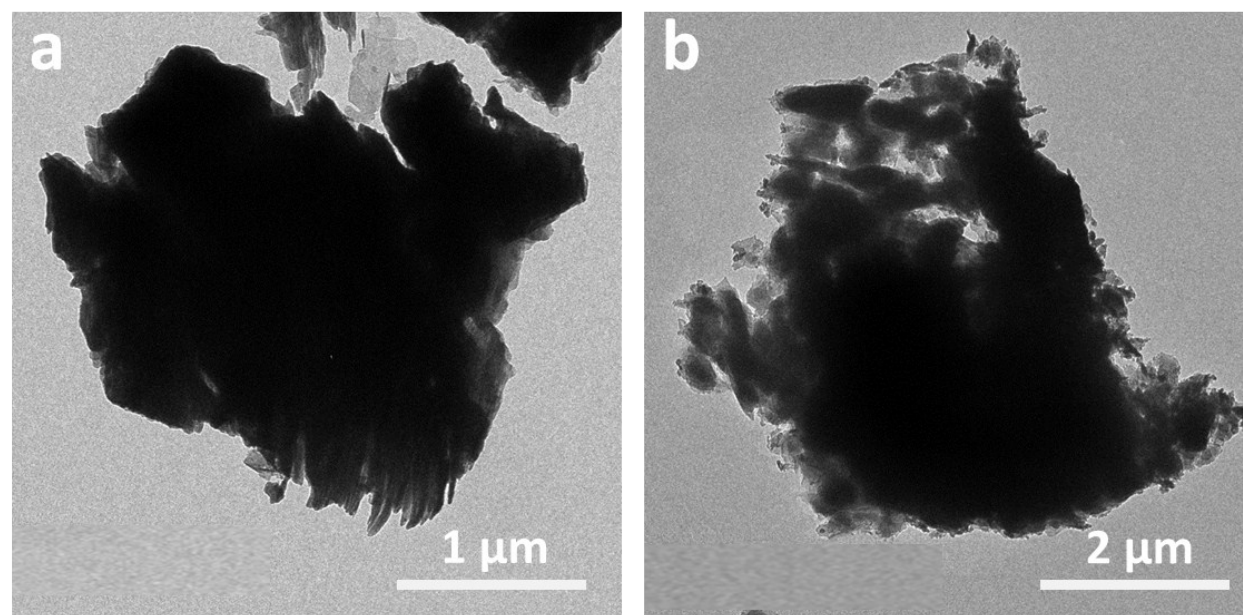


Figure S7. Low-magnification TEM images of (a) Ti₃C₂T_x MXene and (b) LVO/Ti₃C₂T_x MXene.

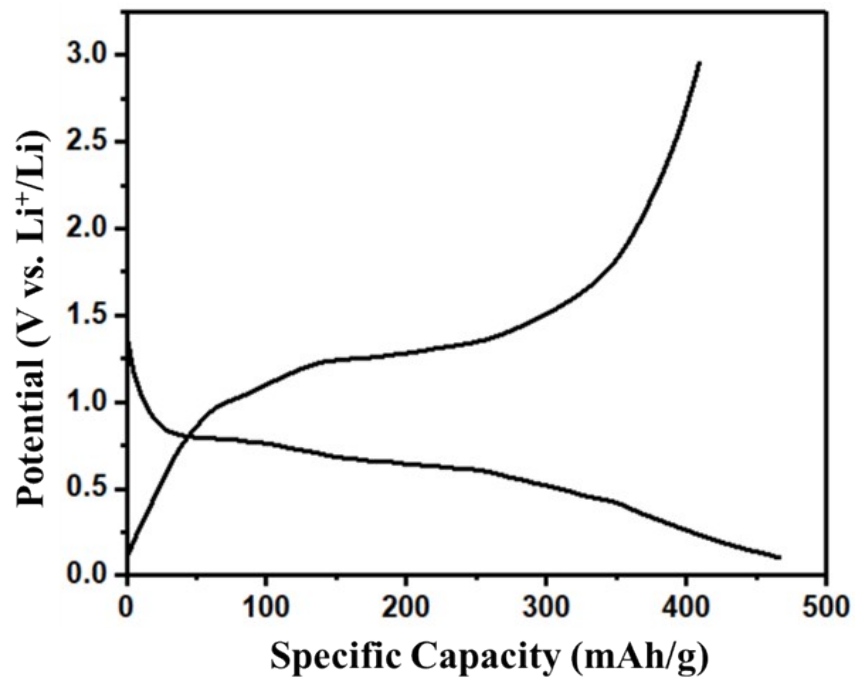


Figure S8. The first cycle charge/discharge curves of LVO/Ti₃C₂T_x MXene composite anode in the voltage window of 0.1 V to 3.0 V at 0.1 C.

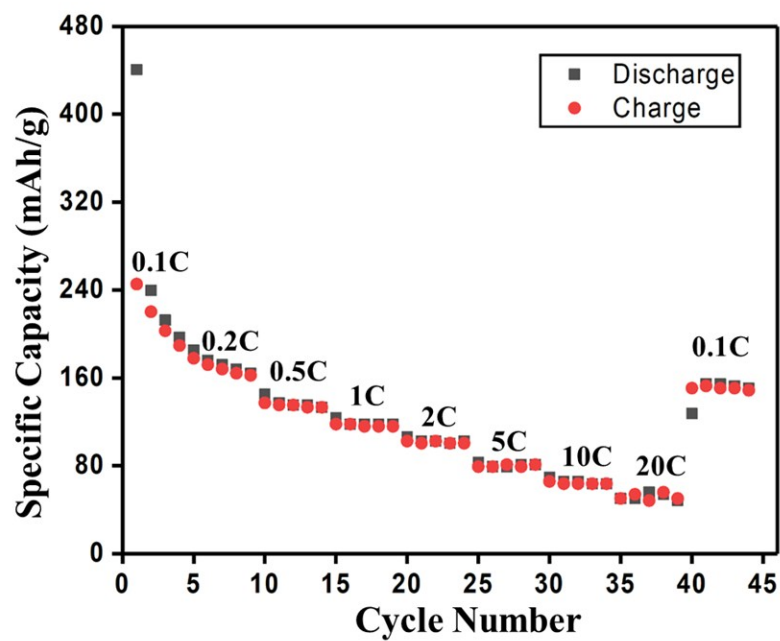


Figure S9. Rate performance of pure Ti₃C₂T_x MXene.

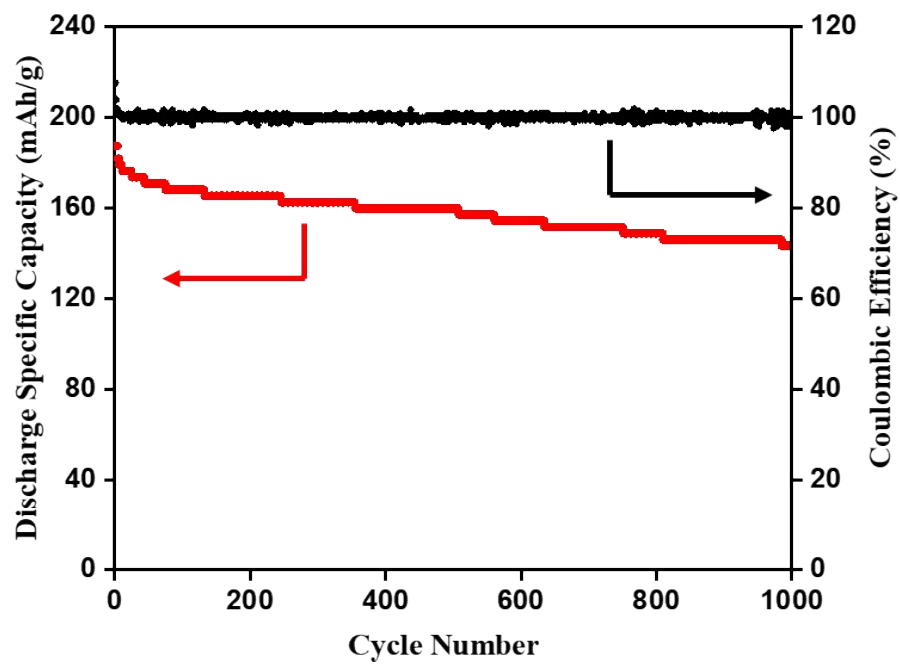
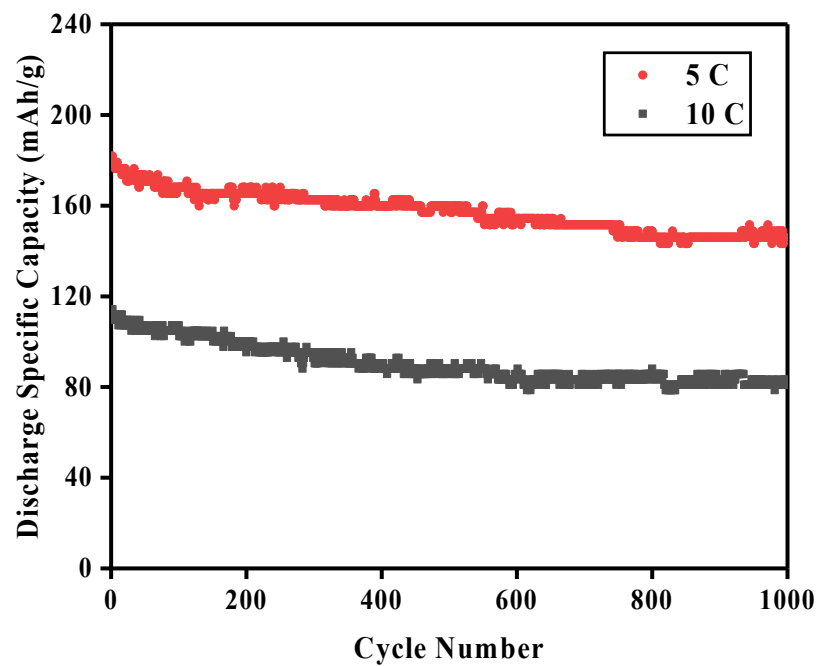


Figure S10. Discharge specific capacity and Coulombic efficiency of LVO/Ti₃C₂T_x MXene anode



cycled at 5 C.

Figure S11. Capacity retention of LVO/Ti₃C₂Tx MXene cycled at 5 C and 10 C.

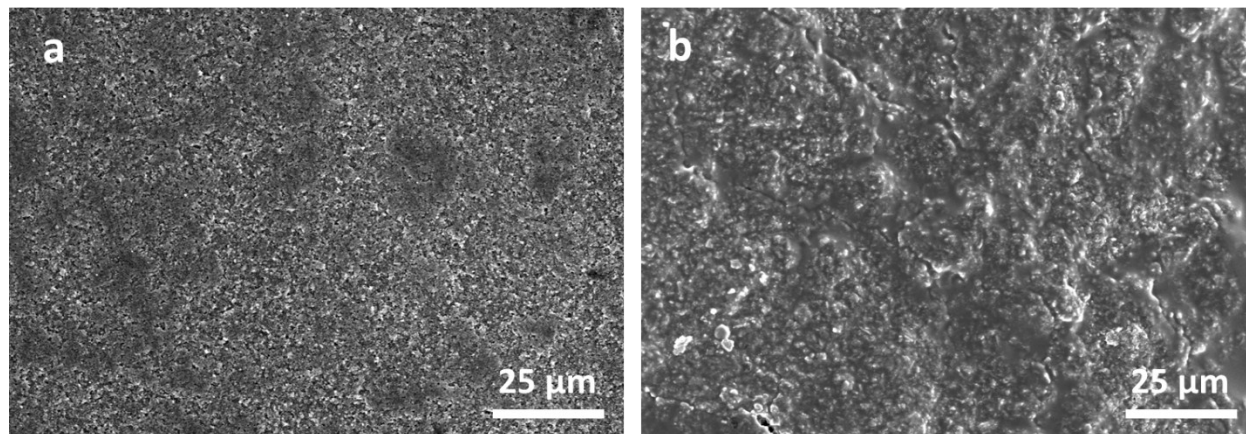


Figure S12. SEM images of LVO/Ti₃C₂Tx MXene composite anode (a) before and (b) after 1000 cycles at 5 C.

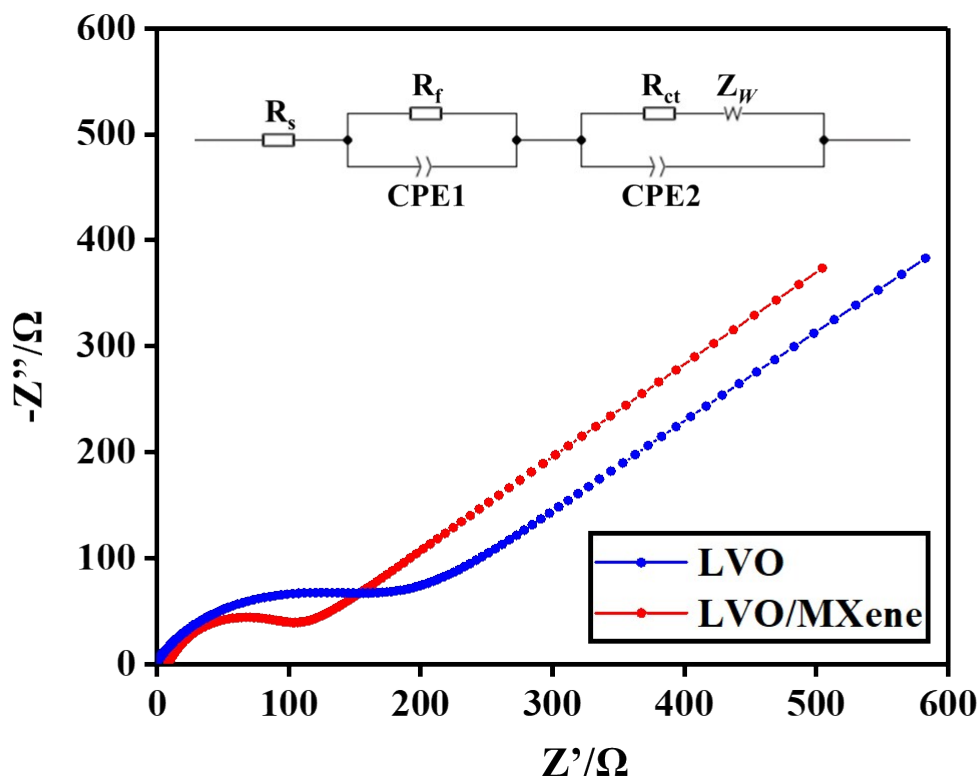


Figure S13. Nyquist plots of bare LVO and LVO/Ti₃C₂Tx MXene anodes. Inset: the equivalent circuit model used to simulate the impedance spectrum.

Table S1. Comparison of electrical conductivity between MXene, Graphene, Graphene Oxide (GO) and Reduced Graphene Oxide (RGO)

	Mxene ⁷	Graphene ^{8,9}	GO ¹⁰	RGO ¹¹
Electrical Conductivity (S/m)	240000	500	Insulation	1.14

Table S2. Performance comparison of our LVO/MXene anode with other similar anodes.

Anodes	High rate capacity	High rate cycle stability	Reference
LVO@C ¹²	147 mAh g ⁻¹ at 0.75C	Not reported	<i>J Mater Chem A</i> , 2015 , 3, 11253
O-Deficient LVO ¹³	90 mAh g ⁻¹ at 5C (20% acetylene black)	No reported	<i>Adv. Sci.</i> 2015 , 2, 1500090

LVO/CNT ¹⁴	87 mAh g ⁻¹ at 5C 62 mAh g ⁻¹ at 10C	Not reported	<i>ACS Nano</i> 2016 , 10, 5398
LVO nanoparticle ¹⁵	107 mAh g ⁻¹ at 5C (20% acetylene black)	Not reported	<i>ACS Appl Mater Inter</i> , 2016 , 8, 23739
LVO-Rod ¹⁶	150 mAh g ⁻¹ at 5C 110 mAh g ⁻¹ at 5C	Not provided	<i>J. Mater. Chem. A</i> , 2018 , 6, 456
LVO@C nanofibers ¹⁷	141 mAh g ⁻¹ at 5C 98 mAh g ⁻¹ at 10C	Not reported	<i>Sci. Bull.</i> 2017 , 62, 1081
hollow-structured LVO ¹⁸	140 mAh g ⁻¹ at 5C 110 mAh g ⁻¹ at 10C	100 mAh g ⁻¹ after 1000 cycles at 10C (20% carbon black)	<i>Chem. Eur. J.</i> 2014 , 20, 5608-5612
LVO nanoparticle ¹⁹	86 mAh g ⁻¹ at 10C	71 mAh g ⁻¹ after 1000 cycles at 10C (20% carbon black)	<i>J. Solid State Electrochem.</i> 2017 , 21, 2547
LVO@GNs ²⁰	250 mAh g ⁻¹ at 5C 210 mAh g ⁻¹ at 10C	163 mAh g ⁻¹ after 5000 cycles at 5C	<i>Chem. Commun.</i> 2015 , 51, 229-231
LVO@N-doped C ²¹	141 mAh g ⁻¹ at 5C 60 mAh g ⁻¹ at 10C	103 mAh g ⁻¹ after 1000 cycles at 10C (20% acetylene black)	<i>J. Alloy. Comp.</i> 2018 , 767, 657
LVO/MXene	187 mAh g ⁻¹ at 5C 114 mAh g ⁻¹ at 10C	146 mAh g ⁻¹ after 1000 cycles at 5C 81 mAh g ⁻¹ after 1000 cycles at 10C (only 10% carbon black)	This work

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