

# Single-crystalline Melem ( $C_6N_{10}H_6$ ) Nanorods: A Novel Stable Molecular Crystal Photocatalyst with Modulated Charge Potentials and Dynamics

Renbo Lei <sup>a</sup>, Bingsheng Du <sup>a</sup>, Xiaofang Lai <sup>a</sup>, Jing Wu <sup>a</sup>, Zhihua Zhang <sup>b</sup>, Shengwei Liu <sup>c,\*</sup>, Rong Wu <sup>d</sup>, Xin Li <sup>e</sup>, Bo Song <sup>f,\*</sup> and Jikang Jian <sup>a,\*</sup>

<sup>a</sup> School of Physics and Optoelectronic Engineering, Guangdong University of Technology, Guangzhou 510006, P. R. China

<sup>b</sup> Liaoning Key Materials Laboratory for Railway, School of Materials Science and Engineering, Dalian Jiao tong University, Dalian 116028, P. R. China

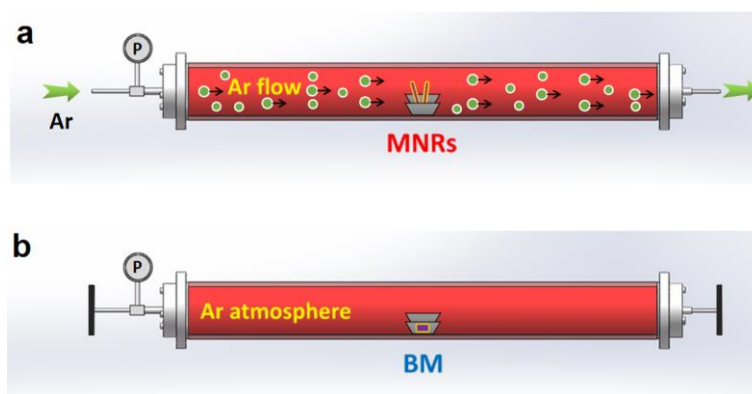
<sup>c</sup> School of Environmental Science and Engineering, Sun Yat-sen University, Guangzhou 510006, P. R. China

<sup>d</sup> Key Laboratory of Solid-state Physics and Devices, School of Physical Science and Technology, Xinjiang University, Urumqi 830046, P. R. China

<sup>e</sup> College of Forestry and Landscape Architecture, Key Laboratory of Energy Plants Resource and Utilization, Ministry of Agriculture, South China Agricultural University, Guangzhou 510642, P. R. China

<sup>f</sup> National Key Laboratory of Science and Technology on Advanced Composites in Special Environments, Harbin Institute of Technology, Harbin 150080, P. R. China

\*E-mail addresses: [liushw6@mail.sysu.edu.cn](mailto:liushw6@mail.sysu.edu.cn) (S. Liu); [songbo@hit.edu.cn](mailto:songbo@hit.edu.cn) (B. Song); [jianjiekang@gdut.edu.cn](mailto:jianjiekang@gdut.edu.cn) (J. Jian)



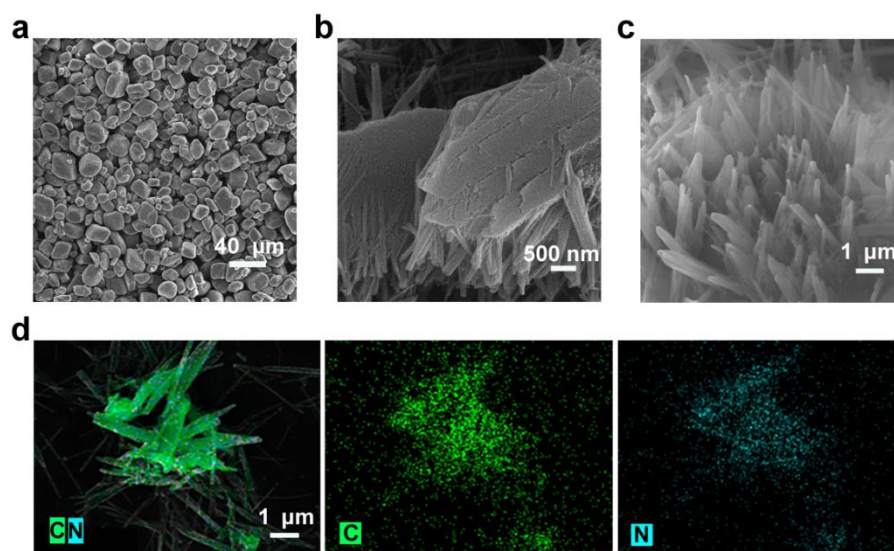
**Fig. S1** Schematic illustrations of synthetic routes for (a) MNRs and (b) BM.

**Table S1** Observed and calculated distances of crystalline planes of melem nanorods (MNRs). The calculation is based on the CIF file of melem reported in Ref .1.

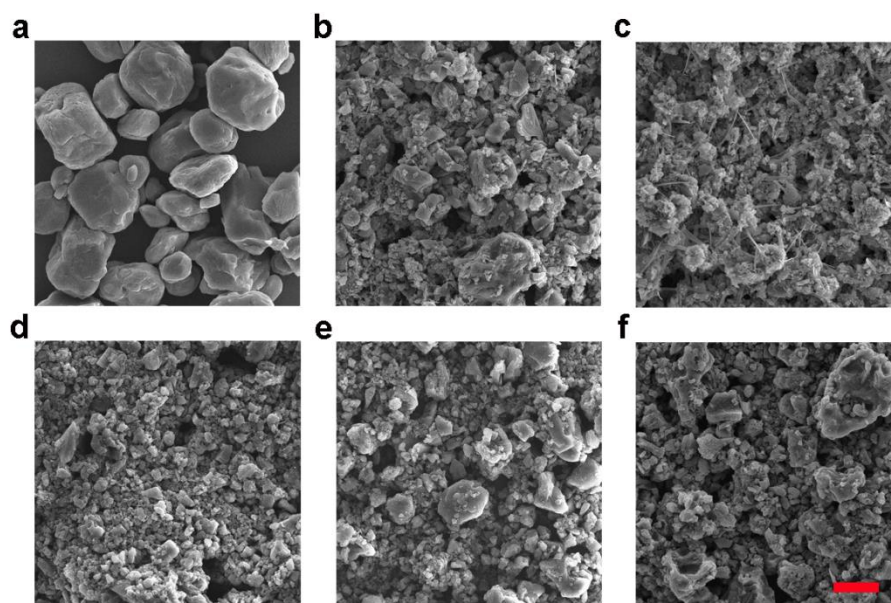
<b>Formula</b>		$H_6N_7(NH_2)_3$		
<b>Space group</b>		$p2_1/c$ (no. 14)		
<b>H K L</b>	<b>DOBS</b>	<b>DCAL</b>	<b>2TH.OBS</b>	<b>2TH.CAL</b>
	<b>(Å)</b>	<b>(Å)</b>	<b>(degree)</b>	<b>(degree)</b>
1 0 0	7.3255	7.2887	12.072	12.133
0 1 1	7.1393	7.2336	12.388	12.226
0 0 2	6.6873	6.5909	13.229	13.423
-1 0 2	5.3481	5.3701	16.562	16.494
1 1 1	4.8459	4.9077	18.292	18.061
-1 1 2	4.5415	4.5628	19.53	19.439
1 1 2	4.0373	4.0043	21.998	22.182
1 2 0	3.738	3.7204	23.784	23.899
1 2 1	3.5343	3.501	25.177	25.421
-2 -1 1	3.4049	3.3857	26.15	26.302
0 0 4	3.2821	3.2955	27.147	27.035
1 1 3	3.2361	3.2491	27.54	27.428
-1 1 4	3.0035	3.0161	29.72	29.594
-1 2 3	2.9353	2.971	30.427	30.054
1 0 4	2.8533	2.8257	31.324	31.638
1 2 3	2.716	2.7237	32.952	32.856
-1 -3 1	2.6684	2.6612	33.557	33.65
1 3 1	2.5941	2.5962	34.547	34.52
1 2 4	2.3631	2.3658	38.048	38.003

(\* 2Theta < 40°)

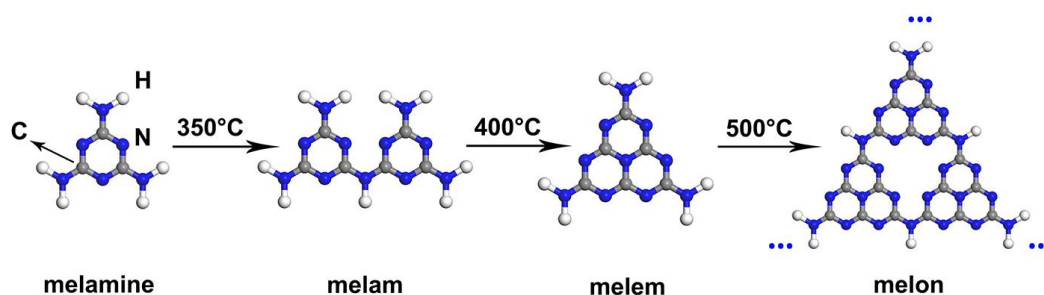
**Fig. S2a** and **S3e** show the SEM images of melamine and g-C<sub>3</sub>N<sub>4</sub> synthesized at 550 °C for 2 h, respectively. **Fig. S3** also shows the SEM images of carbon nitride samples obtained at different temperatures. When the raw materials were annealed at 300 °C, massive crystals were obtained. **Fig. S3a** shows that melamine could be stable under 300 °C. **Fig. S3b** and **S3c** show that the original morphology of melamine gradually broken at a processing temperature of 350 °C, which then transformed into thick plates mixed with some rods at 400 °C. As the annealing temperature was increased to 500 °C, the nanorods disappeared and micro-sized particles with irregular shape were obtained (**Fig. S3d**).



**Fig. S2** SEM images of (a) raw material melamine, (b) products synthesized at 450 °C for 10 min, and (c) products prepared at 450 °C for 120 min. (d) EDS elemental mapping images of MNRs.

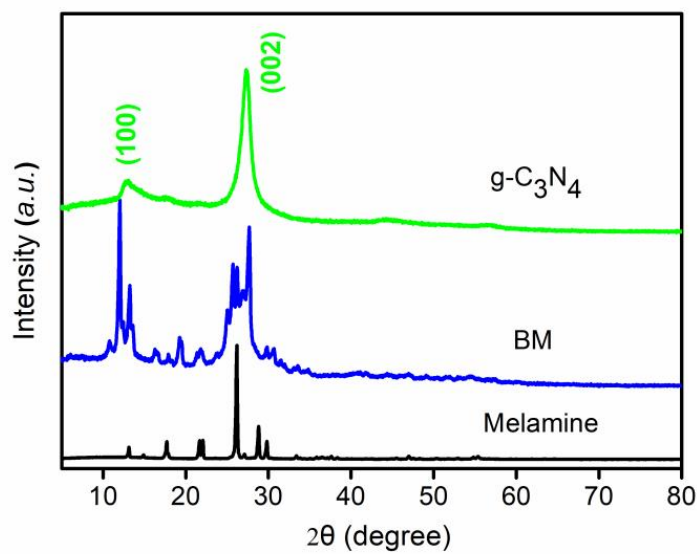


**Fig. S3** SEM images of carbon nitride samples obtained at different temperatures, (a) 300 °C, (b) 350 °C, (c) 400 °C, and (d) 500 °C. SEM images of (e) g-C<sub>3</sub>N<sub>4</sub> synthesized at 550 °C and (f) BM synthesized at 450 °C. The scale bar represents 10 μm.

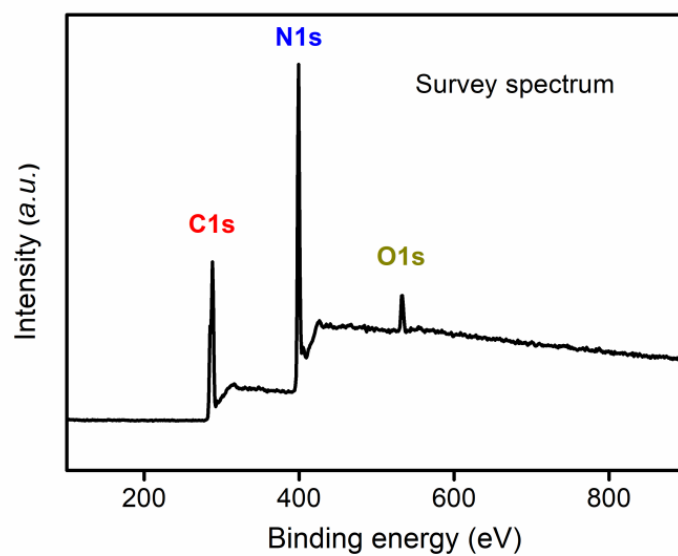


**Scheme S1** Formation of carbon nitrides during the thermal polymerization of melamine, proposed by Ref. 2.

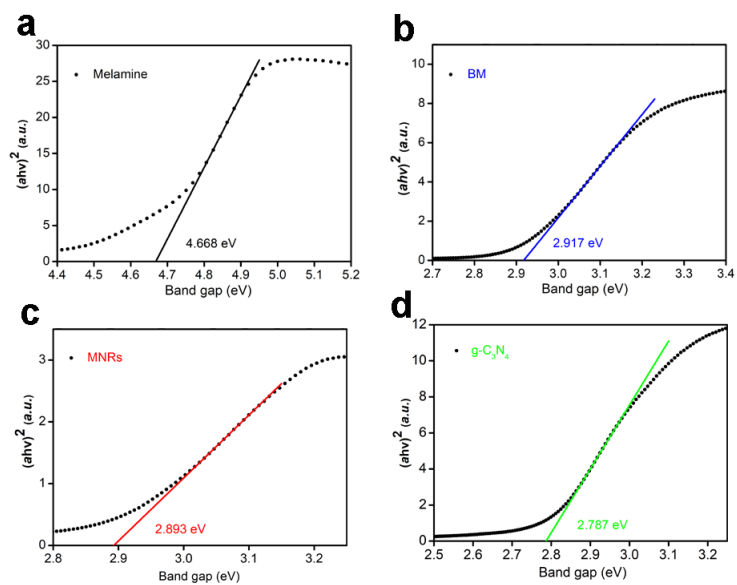
The thermal polymerization of melamine is shown in **Scheme S1**,<sup>[2]</sup> indicating that as the pyrolysis temperature was gradually increased, the C–N rings became connected to yield melam, melem, and melon. However, C–N hexagonal ring remained undestroyed, exhibiting a higher stability.



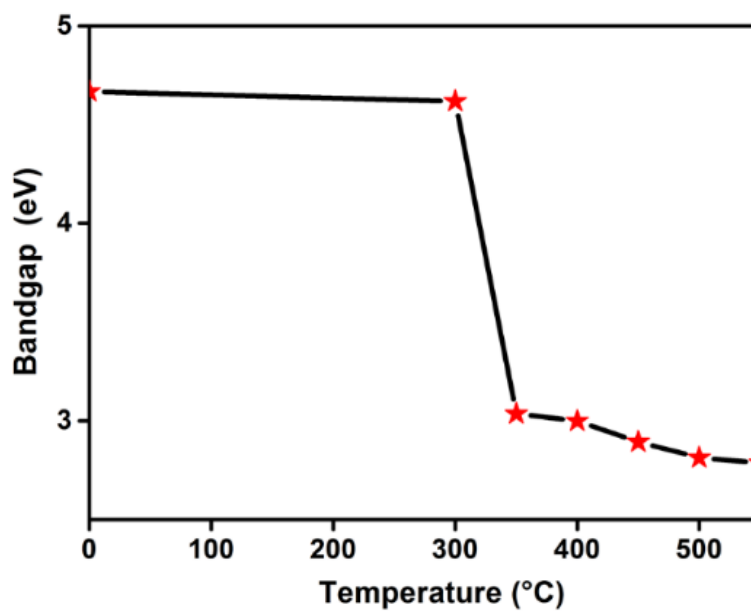
**Fig. S4** XRD patterns of the as-synthesized g-C<sub>3</sub>N<sub>4</sub>, BM, and the raw melamine.



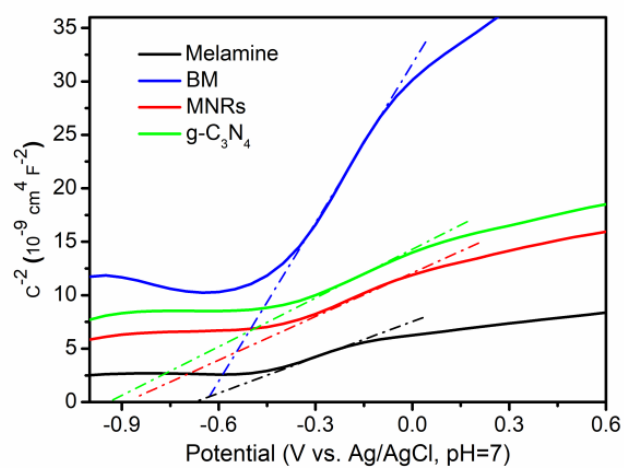
**Fig. S5** XPS survey spectrum of MNRs.



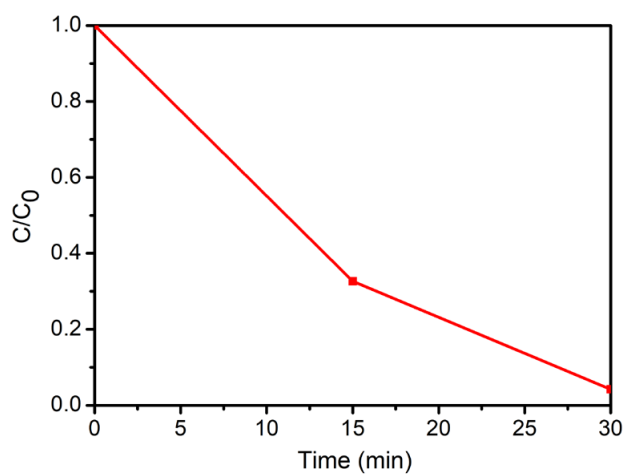
**Fig. S6** Fitted Tauc plots of: (a) melamine, (b) BM, (c) MNRs, and (d) g-C<sub>3</sub>N<sub>4</sub>.



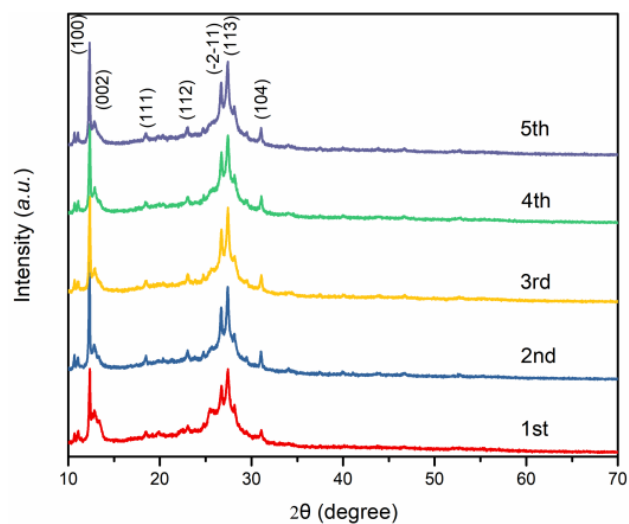
**Fig. S7** Temperature-dependent bandgap of carbon nitride products via the pyrolysis of melamine.



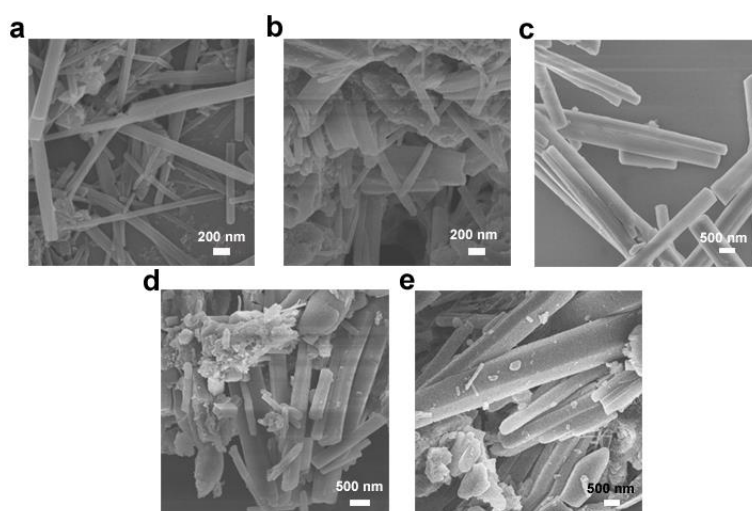
**Fig. S8** Mott–Schottky plots of melamine, BM, MNRs, and g-C<sub>3</sub>N<sub>4</sub>.



**Fig. S9** Time-dependent photocatalytic degradation of RhB on MNR photocatalysts under low-temperature environment (0–3 °C).

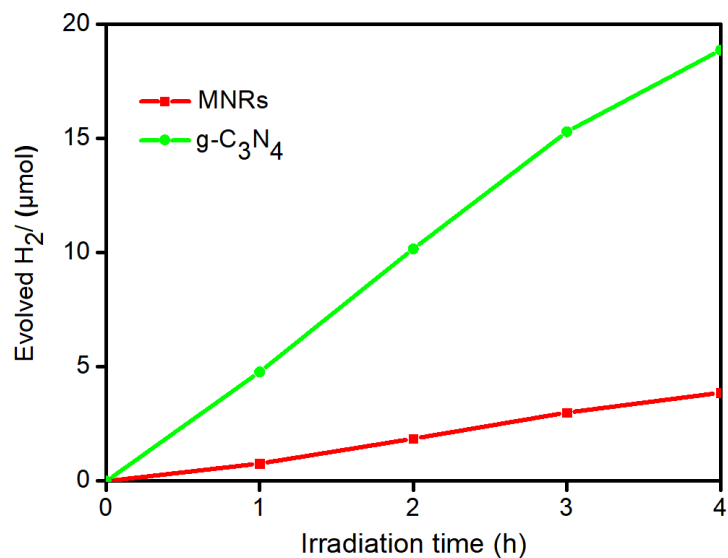


**Fig. S10** XRD patterns of MNRs after each cycling test of photocatalytic degradation of RhB.



**Fig. S11** SEM images of MNRs after each cycling test of photocatalytic degradation of RhB. (a) first, (b) second, (c) third, (d) fourth, and (e) fifth cycle.





**Fig. S12** Time course of H<sub>2</sub> evolution from a 10 vol% aqueous triethanolamine solution by 0.1 wt% Pt-loaded MNRs and g-C<sub>3</sub>N<sub>4</sub> under visible-light irradiation.

### References

- [1] B. Jürgens, E. Irran, J. Senker, P. Kroll, H. Müller and W. Schnick, *J. Am. Chem. Soc.*, 2013, **125**, 10288-10300.
- [2] X.C. Wang, K. Maeda, A. Thomas, K. Takanabe, G. Xin, J. M. Carlsson, K. Domen and M. Antonietti, *Nat. Mater.*, 2009, **8**, 76-80.