

## Supplementary Information

### Dopamine Modified $\text{Li}_{6.4}\text{La}_3\text{Zr}_{1.4}\text{Ta}_{0.6}\text{O}_{12}$ /PEO Solid-State Electrolyte: Enhanced Thermal and Electrochemical Properties

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#### Part I: Characterizations of LLZTO bulk electrolyte, powder and composite electrolytes

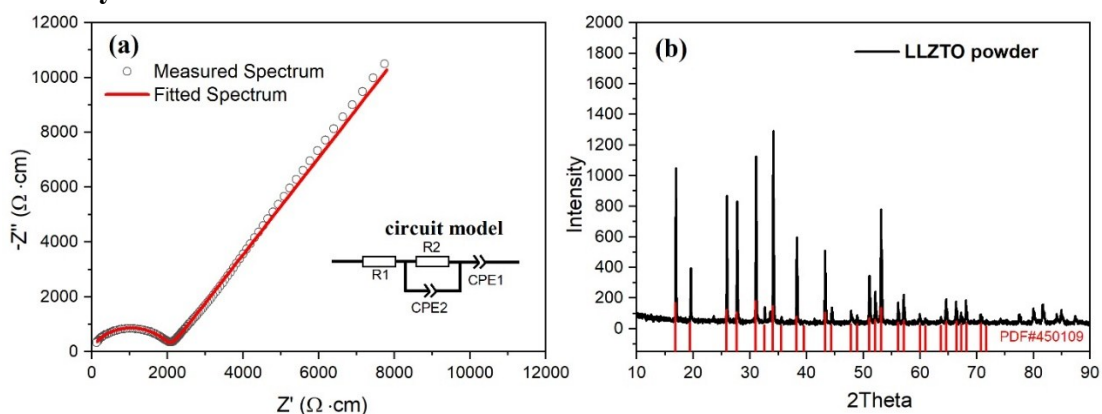


Fig. S1 (a) Electrochemical Impedance Spectrum of the synthesized bulk LLZTO; (b) XRD of LLZTO powder

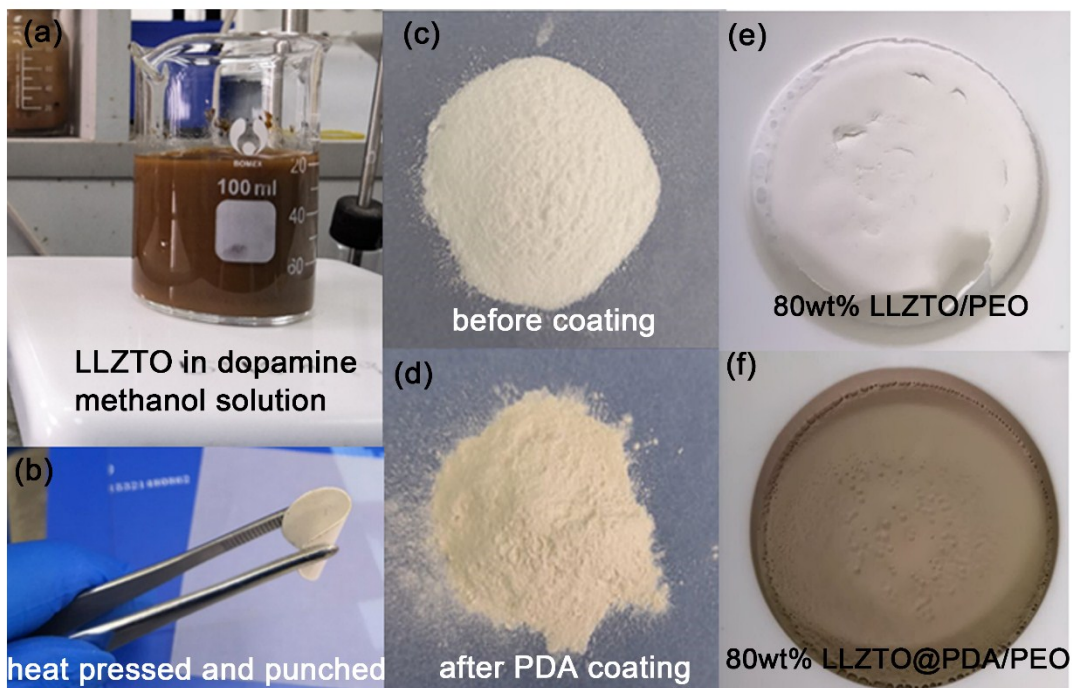


Fig. S2. (a) LLZTO powder dispersed in methanol solution dissolved dopamine and after a while the color of solution changed to brown. (b) The composite solid-state electrolyte after heat pressed

and cut. (c) The color of pristine LLZTO is white; (d) after polydopamine coating, the color of LLZTO change to light brown. (e) 80 wt% LLZTO@PDA/PEO and (f) 80 wt% LLZTO/PEO after dried in PTFE mold;

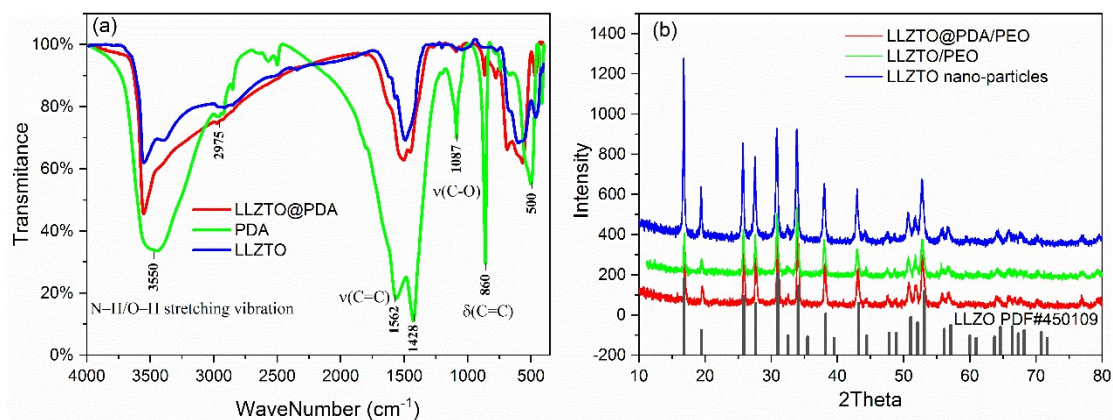


Fig. S3. (a) Infrared spectra of LLZTO, LLZTO@PDA and the pure PDA. The IR peaks near 3550  $\text{cm}^{-1}$ , 1562  $\text{cm}^{-1}$ , 1428  $\text{cm}^{-1}$  indicate that polydopamine was coated on the LLZTO surface; (b) XRD of LLZTO nano-particles, LLZTO@PDA/PEO and LLZTO/PEO composite solid-state electrolyte.

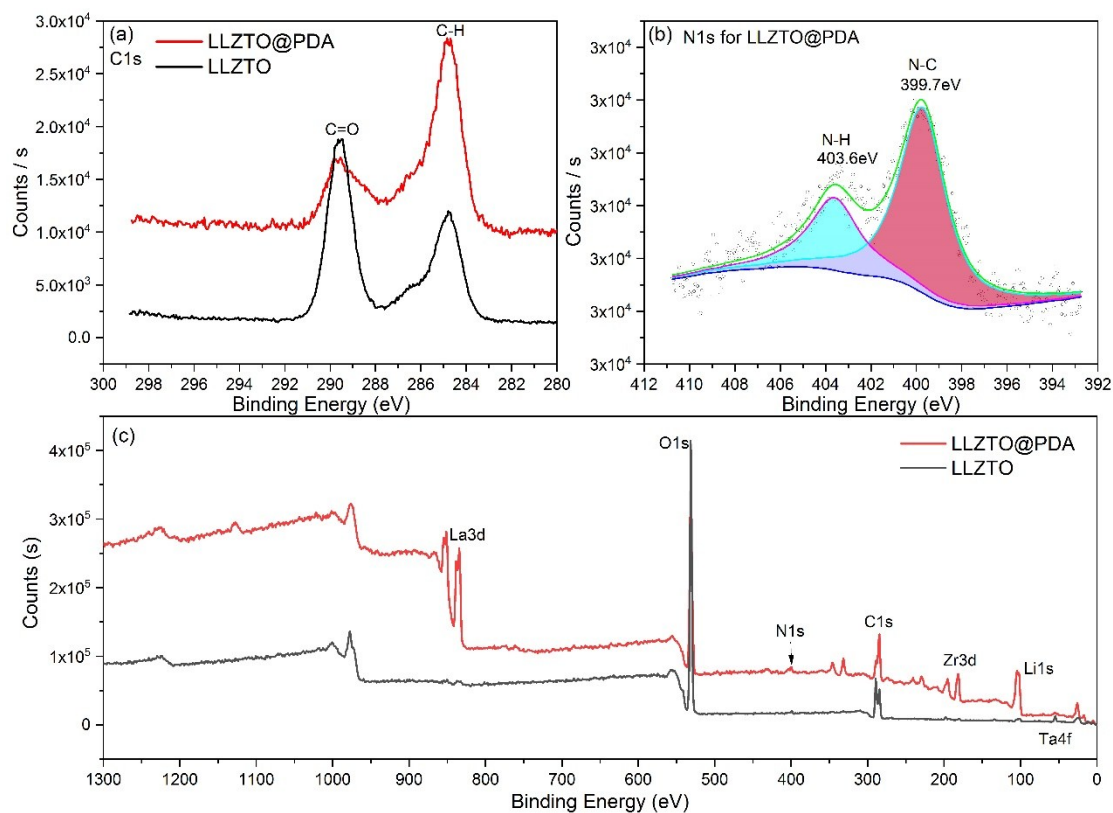


Fig. S4. XPS spectra of LLZTO and LLZTO@PDA powders. (a) High- resolution XPS spectra of C1s region; (b) High-resolution XPS spectra of N1s region of LLZTO@PDA (c) The full XPS spectra of LLZTO and LLZTO@PDA

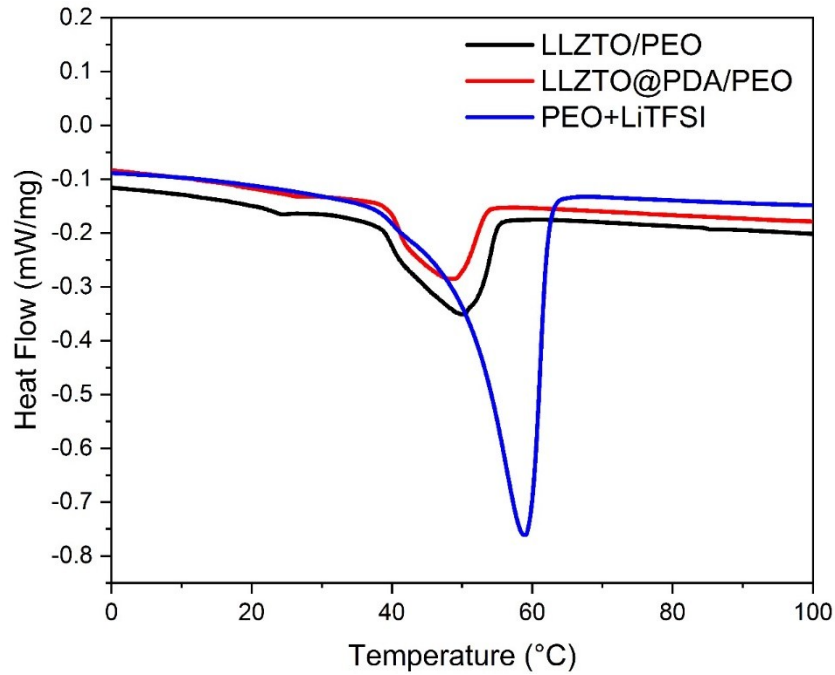


Fig. S5. DSC curves of PEO/LiTFSI, LLZTO in PEO and LLZTO@PDA in PEO

**Part II: Characterizations of polydopamine coated nano-ZrO<sub>2</sub> particles and composite electrolyte based on ZrO<sub>2</sub> and PEO**

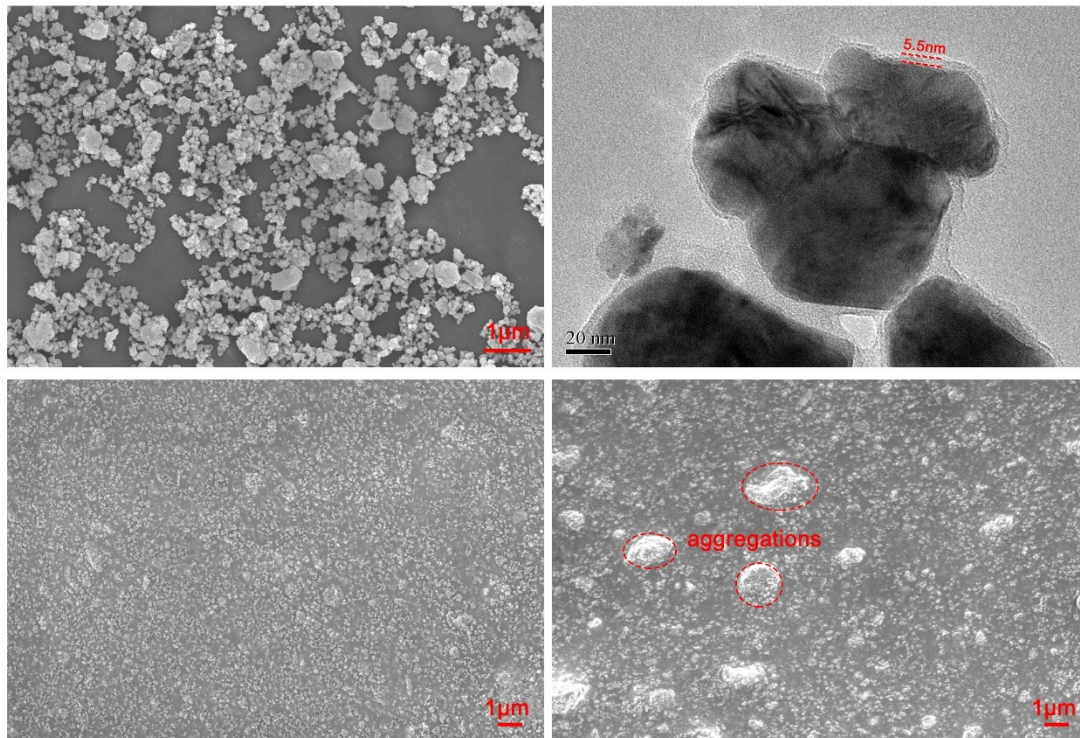


Fig. S6 (a) SEM image of nano-sized ZrO<sub>2</sub> powder (b)TEM photograph of PDA coated nano-ZrO<sub>2</sub>, which indicates that the nano-ZrO<sub>2</sub> was coated with 5~6nm polydopamine. (c) Cross-section image of ZrO<sub>2</sub>@PDA/PEO composite electrolyte, where 80wt% ZrO<sub>2</sub>@PDA dispersed well in the PEO matrix. (d) Cross-section image of ZrO<sub>2</sub>/PEO composite electrolyte, where 80wt% ZrO<sub>2</sub> dispersed in PEO and aggregations formed in the matrix.

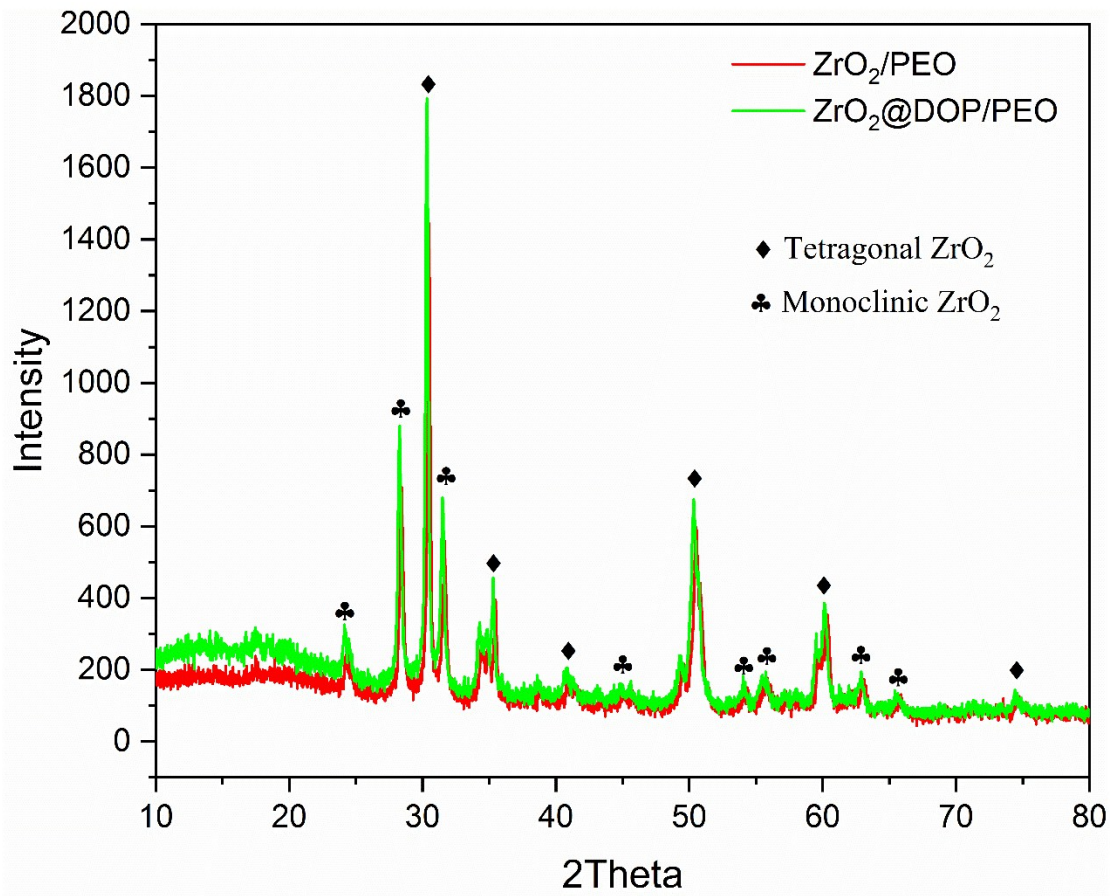


Fig. S7 XRD spectrums of  $\text{ZrO}_2@\text{PDA}/\text{PEO}$  and  $\text{ZrO}_2/\text{PEO}$  composite electrolyte. The spectrum indicates that  $\text{ZrO}_2$  powders combined tetragonal and monoclinic phases. No peaks assigned to crystalline PEO are found.

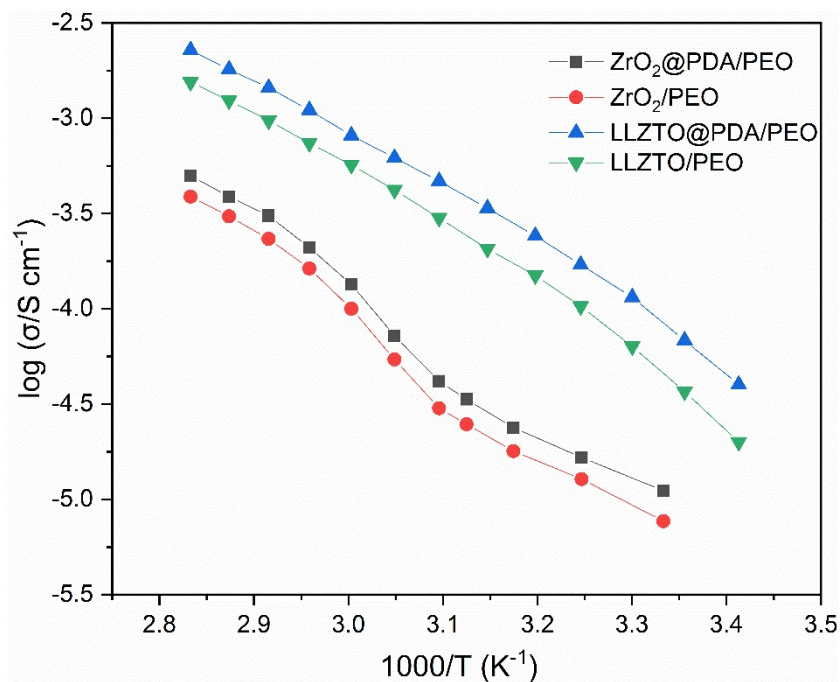


Fig. S8 The  $\log\sigma \sim 1000/T$  plots of  $ZrO_2@PDA/PEO$  and  $ZrO_2/PEO$  electrolyte in the range from room temperature to 80 °C. The conductivities of  $LLZTO@PDA/PEO$  and  $LLZTO/PEO$  electrolyte are also plotted in the graph as comparison.

### Part III: Polydopamine coated LLZTO tablets and $Li^+$ transporting through the interface

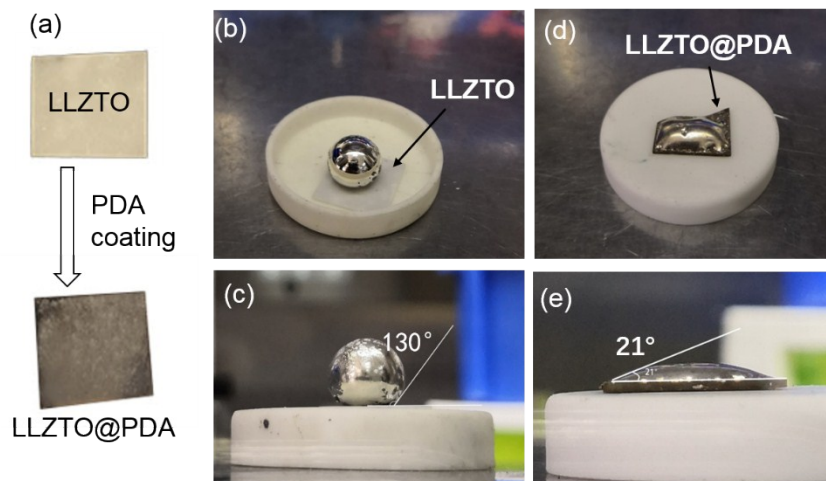


Fig.S9 Demonstrations of contact angles of LLZTO tablet or polydopamine coated LLZTO tablet with molten lithium droplet. (a) The color change of LLZTO after polydopamine coating; (b) and (c) the contact angle of molten lithium droplet on pristine LLZTO tablet; (d) and (e) the contact angle of molten lithium droplet on LLZTO@PDA tablet; The reduced contact angle from 130° to 21° indicates that molten lithium has good wetting ability with polydopamine modified LLZTO surface.

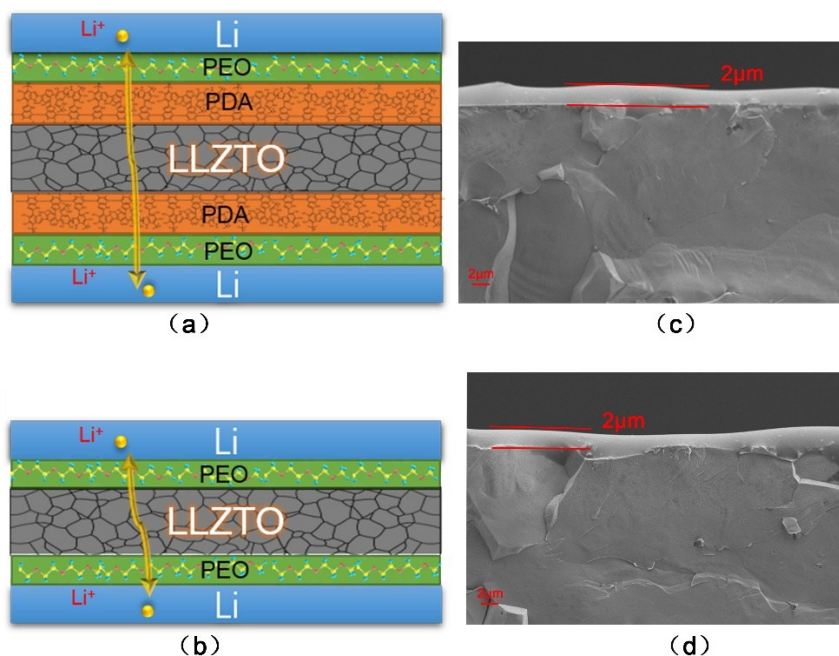


Fig.S10 schematic diagrams of Li-Li symmetrical cells and the interface between Li electrode and LLZTO was buffered with PEO/LiTFSI electrolyte. The PEO buffer layer was spin-coated onto the LLZTO surface. (a) The LLZTO tablet was first coated with polydopamine layer before PEO spin-coating, and the cell was notated as  $Li|PEO|PDA|LLZTO|PDA|PEO|Li$ . (b) No polydopamine layer was formed between LLZTO and PEO, and the cell was notated as  $Li|PEO|LLZTO|PEO|Li$ . The

height ratio of structural layers in the schematic diagrams do not describe the true thickness ratio. (c) and (d), the cross-section images show the thickness of PEO buffer layer are both about  $2\mu\text{m}$  in those Li-Li symmetrical cells.

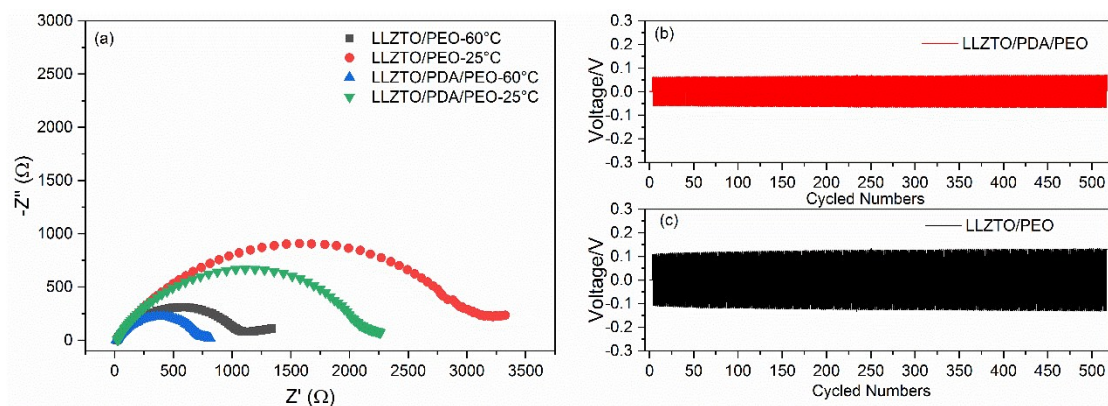


Fig.S11 (a) The total electrochemical impedance spectrums of Li|PEO|LLZTO|PEO|Li and Li|PEO|PDA|LLZTO|PDA|PEO|Li cells at 25 °C and 60 °C. (b) The symmetrical cells were cycled at 60 °C with a current density of  $25\mu\text{A cm}^{-2}$  to demonstrate the  $\text{Li}^+$  transporting ability through polydopamine layer. Each cycle cost 20 minute. The cell cycled over 500 times without evident polarization increase, which indicates a good stability of interface.