Supporting Information

Enhanced Stability and Photovoltage for Inverted Perovskite Solar Cells via Precursor Engineering

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Experimental Section

Patterned ITO glass with a sheet resistance of 15 Ω sq⁻¹ was purchased from Ying Kou You Xuan Trade Co. Ltd. PbI₂ (99.9985%) was obtained from Alfa Aesar. TAPC (97%), CsI, PTAA (Mn=3200, Mw=4900), MAI, MABr, FAI, PbBr₂ (99%), PbCl₂ (99%), C₆₀ (99%), and BCP (>99%) were purchased from Xi'an Polymer Light Technology Corp. DMF (99.8%), DMSO (\geq 99.9%), and chlorobenzene (99.8%) were purchased from Sigma Aldrich.

Perovskite precursor preparation: FAPbI₃ (A) precursor was prepared by mixing 1M FAI and 1.05M PbI₂ in 1 mL DMF/DMSO (v/v, 4:1). MAPbBr₃ (B) precursor was prepared by mixing 1M MABr and 1M PbBr₂ in 1 mL DMF/DMSO (v/v, 4:1). MAPbI_{3-x}Cl_x (C) precursor was prepared by mixing 1.2M MAI, 1.1M PbI₂, and 0.12M PbCl₂ in 1 mL DMF/DMSO (v/v, 4:1). Then 0.83 mL A and 0.17 mL B was mixed to form the control precursor $A_{0.83}B_{0.17}$. For the alloyed perovskite precursors, certain amount of C (0 µL, 93 µL, 147 µL, or 208 µL) was added to the control precursor to form the alloyed precursor ($A_{0.83}B_{0.17}$)_{1-y}C_y with y = 0, 0.1, 0.15, or 0.2, respectively. For the Cs_{0.05}($A_{0.83}B_{0.17}$)_{0.95} precursor, 1M FAI, 1.05M PbI₂, 0.2M MABr, and 0.2M PbBr₂ were dissolved in 1 mL DMF/DMSO (v/v, 4:1), then 5% vol CsI (1.5M stock solution in DMSO) was added to the mixed precursor as described elsewhere.¹

Device fabrication: Patterned ITO-coated glass substrates were cleaned ultrasonically with detergent, deionized water, acetone, and isopropanol in sequence for 20min each. After UV-ozone cleaning for 30min, ITO glass was moved into N₂-purged glovebox. TAPC+PTAA HTL were deposited on the ITO substrate, as reported previously.² The alloyed precursor $(A_{0.83}B_{0.17})_{1-y}C_y$ was spin coated onto the TAPC-coated ITO substrates at 1000 rpm for 10s and 5000 rpm for 30s, and

100 μ L of chlorobenzene were rapidly dropped on the substrates after 15 s of the spin coating. Then the samples were annealed at 100 °C for 1h. Finally, C₆₀ (30 nm), BCP (8 nm), and Ag (100 nm) were sequentially deposited onto the perovskite layer by thermal evaporation through a shadow mask with an active area of 0.10 cm².

Characterization: The morphologies of perovskite and HTLs films were characterized by scanning electron microscopy (SEM, TESCAN MIRA3) at a 5kV accelerating voltage. Atomic force microscope (AFM) images of HTLs films were collected on a multimode SPM (Bruker) with tapping mode. UPS analysis was conducted to measure the energy level of HTLs with an unfiltered He I (21.22 eV) gas discharge lamp and a hemispherical analyzer. Steady-state and time-resolved PL spectra were measured using an Edinburgh FLS920 spectroscopy system, using pulsed laser excitation at 405 nm. The time-resolved PL measurements were carried out by a Deltaflex TCSPC system (Horiba) at 950 nm with a 479 nm laser. Electrochemical impedance spectroscopy (EIS) measurements were performed on an electrochemical workstation (Zahner). Photocurrent density-voltage (*J-V*) curves were measured under AM 1.5 G one sun illumination (100 mW/cm²) with a solar simulator (Enlitech SS-F7-3A) equipped with a 300 W Xenon lamp and a Keithley 2400 source meter. The light intensity was adjusted by an NREL-calibrated Si solar cell. During measurement, the cell was covered by a mask with an aperture of 0.1 cm².

Year	Method	Perovskite	Device type	$V_{\rm oc}$ increment	Ref.
2014	MAI doping	FAPbI ₃	Regular	From 0.93V to 1.03V	3
2015	MACl doping	MAPbI ₃	Regular	From 0.93V to 1.08V	4
2016	Pb(SCN)2 doping	MAPbI ₃	Regular	From 1.09V to 1.11V	5
2017	Cu(Tu)I doping	MAPbI _{3-x} Cl _x	Inverted	From 1.00V to 1.11V	6
2018	F4TCNQ doping	MAPbI _{3-x} Cl _x	Inverted	From 1.00V to 1.10V	7
2019	Solvent replacement for DMF	MAPbI ₃	Inverted	From 1.00V to 1.06V	8

Table S1. Typical method of precursor engineering toward increased V_{oc} .



Figure S1. High contrast SEM images of Figure 1. (a) y = 0, (b) y = 0.1, (c) y = 0.15, and (d) y = 0.2. Scale bar, 300 nm.

Table 52. I fitting parameters from TREE spectra.								
	A_1	τ_1 (ns)	A_2	$ au_2$ (ns)	$\tau_{\rm average} ({\rm ns})$			
y = 0	640	4.5	1340	206	203.9			
y = 0.1	800	7.0	2260	210	207.6			
y = 0.15	620	7.5	1750	238	235.5			
y = 0.2	1300	4.3	2630	190	187.9			

Table S2. Fitting parameters from TRPL spectra.^{a)}

^{a)}The TRPL spectra were fitted by a bi-exponential decay function, $I(t) = A_1 e^{-t/\tau_1} + A_2 e^{-t/\tau_2}$ and the average PL lifetime is calculated by

$$\tau_{average} = \frac{A_1 \tau_1^2 + A_2 \tau_2^2}{A_1 \tau_1 + A_2 \tau_2}$$



Figure S2. TRPL spectra of perovskite/HTL samples. Inset is the magnification of the region with longer decay time. The decay lifetime is 196.7 ns, 164.5 ns, 137.5 ns, and 146.2 ns for y = 0, 0.1, 0.15, and 0.2, respectively.



Figure S3. EQE spectra of devices with y = 0.1 and y = 0.2, of which the integrated J_{sc} is 21.7 mA/cm² and 21.3 mA/cm², respectively.



Figure S4. Steady state J_{sc} and PCE upon light exposure for (a) control device, y = 0 and (b) optimized device, y = 0.15.



Figure S5. XRD of perovskite films, $Cs_{0.05}(A_{0.83}B_{0.17})_{0.95}$ and $(A_{0.83}B_{0.17})_{0.85}C_{0.15}$, after stored in ambient air under 70% RH for 300h in dark. Insets show the appearance of corresponding films.



Figure S6. EIS spectra of devices based on control and alloyed perovskites. Inset shows the equivalent circuit.

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