

## Supporting Information

### One Minute from Pristine Carbon to Electrocatalyst for Hydrogen Peroxide Production

Yun-Lu Wang <sup>a</sup>, Shi-Song Li <sup>a</sup>, Xiao-Hong Yang <sup>b</sup>, Guo-Yong Xu <sup>c</sup>, Zi-Chun Zhu <sup>b</sup>,  
Ping Chen <sup>a,c,d\*</sup>, Shan-Qing Li <sup>b\*</sup>

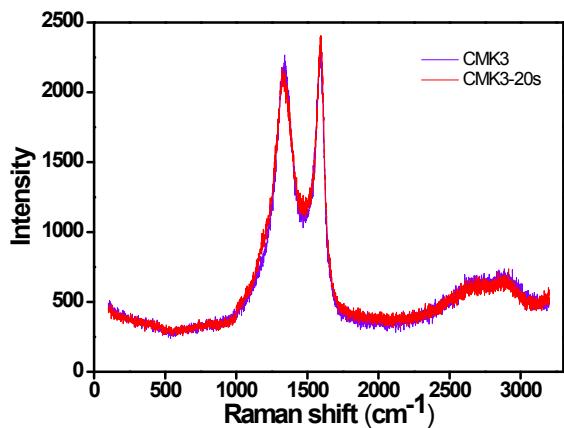
<sup>a</sup> School of Chemistry and Chemical Engineering, Anhui University, Hefei, Anhui,  
230601, P. R. China

<sup>b</sup> School of Chemistry and Materials Engineering, Chizhou University, Chizhou,  
Anhui, 247000, P. R. China

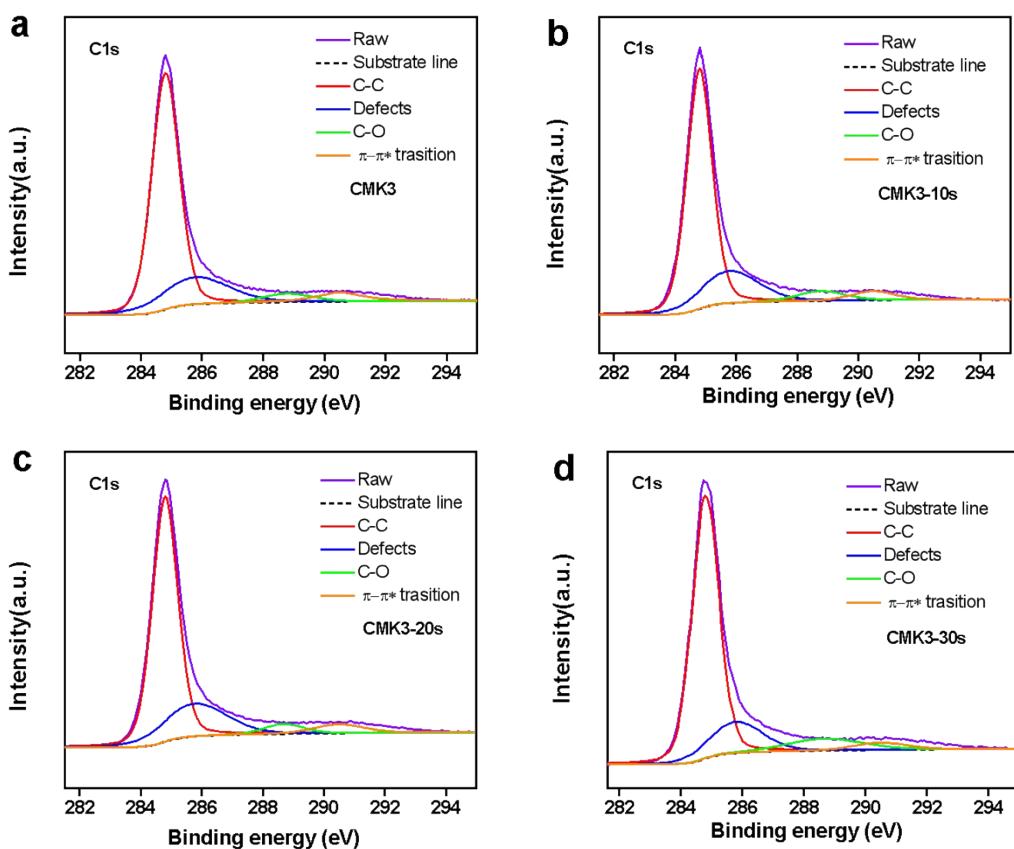
<sup>c</sup> Institute of Physical Science and Information Technology, Anhui University, Hefei,  
230601, P. R. China

<sup>d</sup> Anhui Province Key Laboratory of Chemistry for Inorganic/Organic Hybrid  
Functionalized Materials, Hefei, 230601, P. R. China

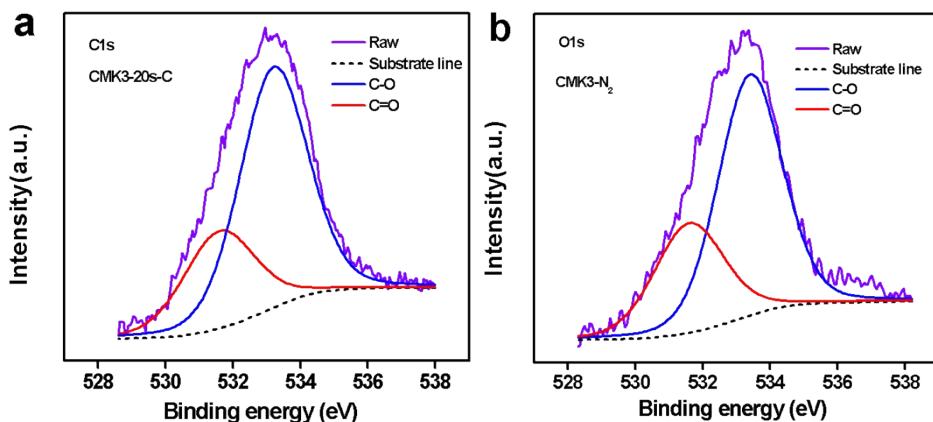
**\*Correspondence to:** Prof. Ping Chen (E-mail: [chenping@ahu.edu.cn](mailto:chenping@ahu.edu.cn)) or Dr. Shan-  
Qing Li (E-mail: [shanqingli@czu.edu.cn](mailto:shanqingli@czu.edu.cn))



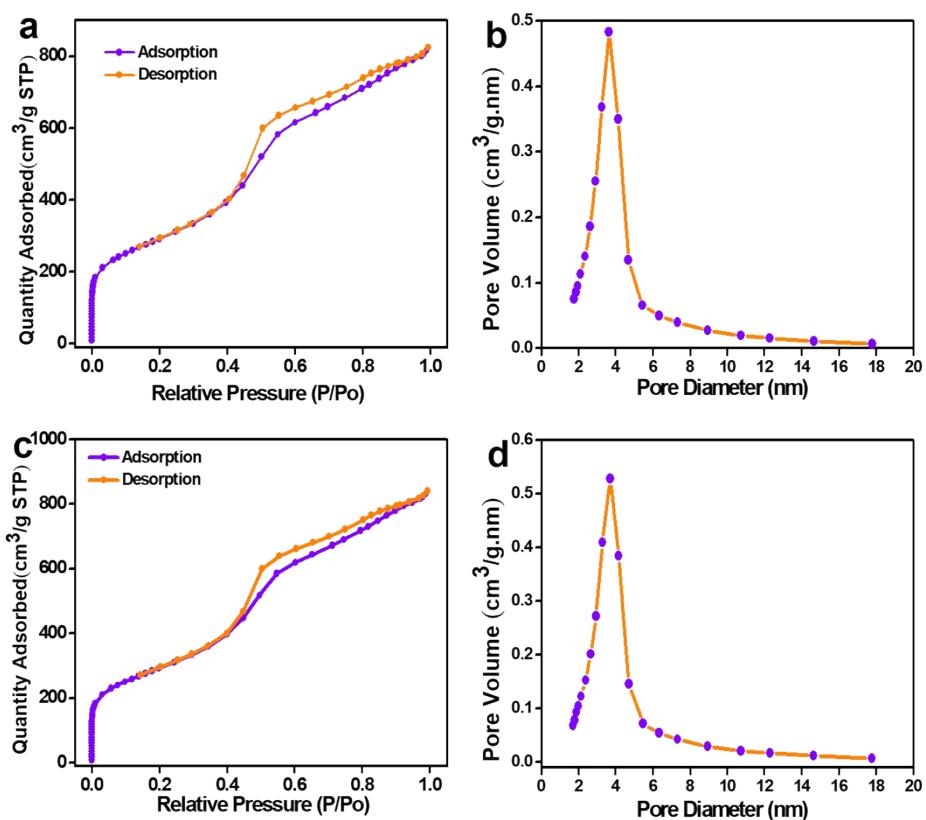
**Figure S1** Raman spectra of CMK3 and CMK3-20s.



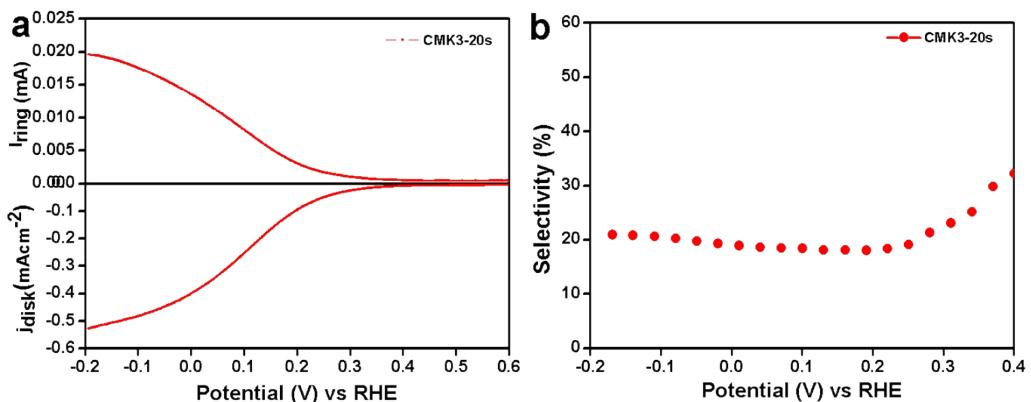
**Figure S2** (a-d) High resolution C1s spectra of CMK3, CMK3-10s, CMK3-20s and CMK3-30s.



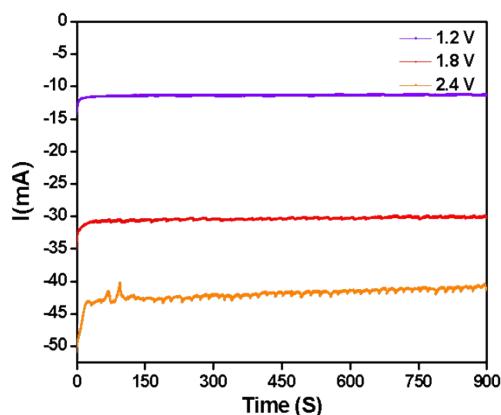
**Figure S3** (a) High resolution O1s spectrum of CMK3-20s-C; (b) High resolution O1s spectrum of CMK3-N<sub>2</sub>



**Figure S4** (a-b) Nitrogen adsorption-desorption isotherm curve and pore size distribution of CMK3-10s; (c-d) Nitrogen adsorption-desorption isotherm curve and pore size distribution of CMK3-30s



**Figure S5** (a) RRDE voltammograms of CMK3-20s in  $O_2$ -saturated 0.5 M  $H_2SO_4$  (at rotating speed of 1,600 rpm, sweep rate of 10 mV s<sup>-1</sup>); (b) The selectivity of the peroxide yield obtained from the RRDE curves for CMK3-20s in 0.5 M  $H_2SO_4$ .



**Figure S6** I-t curve of CMK3-20s measured at the different applied potentials the flow cell (0.1 M KOH).

**Table. S1** Content of carbon and oxygen elements in CMK3, CMK3-10s, CMK3-20s and CMK3-30s, and the ratio of functional groups containing oxygen element.

Sample	C at.%	O at.%	(C=O) %	(C-O) %
CMK3	96.2	2.85	33	67
CMK3-10s	95.8	3.22	28	72
CMK3-20s	96.2	2.97	24	76
CMK3-30s	96.1	2.93	21	79

**Table. S2** The value of BET surface area, pore volume and pore diameter of CMK3, CMK3-10s, CMK3-20s and CMK3-30s.

Sample	BET surface area (m <sup>2</sup> /g)	Total Pore Volume (cm <sup>3</sup> /g)	average pore diameter (nm)
CMK3	1,009.4	1.32	4.00
CMK3-10s	1,017.4	1.28	4.01
CMK3-20s	1,018.8	1.30	4.05
CMK3-30s	1,054.6	1.34	4.01

**Table. S3** Comparison of performance of different samples with the prepared CMK3-20s

Sample	pH	Potential (V vs. Ag/ AgCl)	[H <sub>2</sub> O <sub>2</sub> ] (mg L <sup>-1</sup> h <sup>-1</sup> )	Faradaic Efficiency (%)	Ref.
O-CNTs	14	-0.24	197.5	~95	[1]
o-GOMC-1	13	-0.23	~55	99	[2]
AGF1100	13	-0.7	~385	72.3	[3]
Nitrogen-doped mesoporous carbon	13	-0.85	~70	70	[4]
Vulcan carbon	13	-0.7	~55	41.0	[5]
CeO <sub>2</sub> /C	13	-0.7	~100	44	[5]
Biomass derived N/C catalyst	13	-0.67	2.04	40	[6]
Carbon supported MnO <sub>2</sub>	7	-0.7	100	none	[7]
N-O-P-Carbon-800	13	-0.7	320	95	[8]
rGO-PEI aerogel	13	-0.2	~160	90.7	[9]
CMK3-20s	13	-0.55	421	92.5	This work

## Supporting References

1. Z. Lu; G. Chen; S. Siahrostami; Z. Chen; K. Liu; J. Xie; L. Liao; T. Wu; D. Lin; Y. Liu; T. F. Jaramillo; J. K. Norskov; Y. Cui, *Nature Catalysis* 2018, *1*, 156-162
2. Y. J. Sa; J. H. Kim; S. H. Joo, *Angewandte Chemie-International Edition* 2019, *58*, 1100-1105

3. Z. Pan; K. Wang; W. Yi; P. Tsakarais; S. Song, *Applied Catalysis B Environmental* 2018, 237, S0926337318305113
4. Y. Sun; I. Sinev; W. Ju; A. Bergmann; S. Dresp; S. Kuehl; C. Spoeri; H. Schmies; H. Wang; D. Bernsmeier; B. Paul; R. Schmack; R. Krahnert; B. Roldan Cuenya; P. Strasser, *Acs Catalysis* 2018, 8, 2844-2856
5. F. Hasché; M. Oezaslan; P. Strasser; T. P. Fellinger, *Journal of Energy Chemistry* 2016, 25, 251
6. Y. Yang; F. He; Y. Shen; X. Chen; H. Mei; S. Liu; Y. Zhang, *Chemical Communications* 2017, 53, 9994-9997
7. L. R. Aveiro; A. G. M. D. Silva; V. S. Antonin; E. G. Candido; L. S. Parreira; R. S. Geonmonond; I. C. D. Freitas; M. R. V. Lanza; P. H. C. Camargo; M. C. Santos, *Electrochimica Acta* 2018, 268, S0013468618303682
8. H. X. Zhang; S. C. Yang; Y. L. Wang; J. C. Xi; J. C. Huang; J. F. Li; P. Chen; R. Jia, *Electrochimica Acta* 2019, 308, 74-82
9. X. Xiao; T. Wang; J. Bai; F. Li; T. Ma; Y. Chen, *Acs Applied Materials & Interfaces* 2018, 10, 42534-42541