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Supporting Information

# Exploring of high-voltage Mg<sup>2+</sup>/Na<sup>+</sup> co-intercalation reaction of

## Na<sub>3</sub>VCr(PO<sub>4</sub>)<sub>3</sub> in Mg-ion batteries

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Tabl	e S1: XPS	parameters	calculated	from the	deconvolution	of V2p	spectra	of NVC	P,
NVC	P-I <sub>2</sub> , NVC	P-Cl <sub>2</sub> and N	VCP-NO <sub>2</sub>	BF <sub>4</sub> .					

Sample	Assignments	B.E. /	FWHM	Contrib.
		eV	/ eV	/ %
NVCP	$V^{3+}2p_{3/2}$	516.5	2.7	30.6
	$V^{3+}2p_{1/2}$	522.6	3.3	53.4
	O1s (sat.)	520.4	1.2	16
NVCP-I <sub>2</sub>	$V^{3+}2p_{3/2}$	516.3	1.9	7.2
	$V^{4+}2p_{3/2}$	517.5	2.0	20.2
	$V^{3+}2p_{1/2}$	522.2	2.1	22.4
	$V^{4+}2p_{1/2}$	524.3	2.3	8.4
	O1s (sat.)	520.3	2.0	41.8
NVCP-Cl <sub>2</sub>	$V^{3+}2p_{3/2}$	516.1	1.6	5.7
	$V^{4+}2p_{3/2}$	517.4	1.5	9.9
	$V^{5+}2p_{3/2}$	518.6	1.4	4.4
	$V^{3+}2p_{1/2}$	522.3	1.7	15.6
	$V^{4+}2p_{1/2}$	523.6	1.7	17.4
	$V^{5+}2p_{1/2}$	525.0	1.7	10.9
	Ols (sat.)	520.6	2.1	36.1
NVCP-NO <sub>2</sub> BF <sub>4</sub>	$V^{3+}2p_{3/2}$	517.2	1.0	2.6
	$V^{4+}2p_{3/2}$	518.4	1.4	16.9
	$V^{5+}2p_{3/2}$	519.3	1.6	1.6
	$V^{3+}2p_{1/2}$	522.3	1.5	13.4
	$V^{4+}2p_{1/2}$	523.5	1.5	17.1
	$V^{5+}2p_{1/2}$	524.8	1.6	4.8
	Ols (sat.)	521.1	1.6	28.5

(B.E. Binding energy; FWHM: Full Width at Half Maximum; Contrib.: Relative contribution in percent).

Sample	Assignments	B.E. / eV	FWHM / eV	Contrib. /%
NVCP	Cr <sup>3+</sup> 2p <sub>3/2</sub>	577.8	2.6	52.3
	$Cr^{3+}2p_{3/2}$ (sat.)	580.4	4.7	17.0
	$Cr^{3+}2p_{1/2}$	587.4	2.8	26.2
	$Cr^{3+}2p_{1/2}$ (sat.)	590.3	2.6	4.5
NVCP-I <sub>2</sub>	$Cr^{3+}2p_{3/2}$	577.5	2.8	63.2
	Cr <sup>3+</sup> 2p <sub>3/2</sub> (sat.)	580.0	3.9	10.4
	$Cr^{3+}2p_{1/2}$	587.1	2.6	23.4
	Cr <sup>3+</sup> 2p <sub>1/2</sub> (sat.)	589.2	1.5	3.0
NVCP-Cl <sub>2</sub>	$Cr^{3+}2p_{3/2}$	577.2	3.3	53.8
	$Cr^{3+}2p_{3/2}$ (sat.)	580.2	3.6	12.7
	$Cr^{3+}2p_{1/2}$	586.8	3.9	29.5
	$Cr^{3+}2p_{1/2}$ (sat.)	590.0	3.3	4.0
NVCP-NO <sub>2</sub> BF <sub>4</sub>	$Cr^{3+}2p_{3/2}$	577.3	2.6	46.8
	Cr <sup>3+</sup> 2p <sub>3/2</sub> (sat.)	578.8	2.6	25.1
	$Cr^{3+}2p_{1/2}$	587.2	2.6	25.2
	$Cr^{3+}2p_{1/2}$ (sat.)	589.4	2.2	2.9

**Table S2:** XPS parameters calculated from the deconvolution of Cr2p spectra of NVCP,<br/>NVCP-I2, NVCP-Cl2 and NVCP-NO2  $BF_4$ .

(B.E. Binding energy; FWHM: Full Width at Half Maximum; Contrib.: Relative contribution in percent).

## Table S3. ICP-AES results of electrode samples.

Theoretical chemical formulae	Information of the electrodes	ICP-AES Na : Mg : V : Cr : P
Na <sub>3</sub> VCr(PO <sub>4</sub> ) <sub>3</sub>	Original NVCP material	2.96 : 0 : 1.01 : 0.97 : 3
Na <sub>1.5</sub> VCr(PO <sub>4</sub> ) <sub>3</sub>	After first charge in Na cell (1M NaClO <sub>4</sub> in PC)	1.54 : 0 : 0.98 : 0.96 : 3
Na <sub>1.5</sub> Mg <sub>0.75</sub> VCr(PO <sub>4</sub> ) <sub>3</sub>	After first discharge in Mg cell* (0.5 M Mg(TFSI) <sub>2</sub> in DME)	1.53 : 0.68 : 0.99 : 0.98 : 3

\*The NVCP electrode was charged in Na cell at  $-15^{\circ}$ C to remove the Na<sup>+</sup> ions. Afterwards, the electrode was washed with DME (1,2-dimethoxyethane) to remove all Na<sup>+</sup> ions from the electrolyte. Eventually, the electrode is assembled in Mg cell at  $-15^{\circ}$ C and discharged to reach 85 mAh/g.

### **ICP** Results:

Measurement data for original NVCP material

		Qu	antitative ar	alysis ICP-AES	
Sample ID	NVCP-X DIG	ESTION 1 (V Cr P Na)	0.0298 1_100		
Blank File	C:/Users/ICP-	AES\User\Documents\li	RISIntrepid\XSP	(ThermoElectron/Ortiz\blank	047
nitial San	ple Quantity (m	ng): 29.8			
Sample P	rep Volume (ml	.): 10.00			
Diluted to	volume (mL): 1	0.00			
Aliquot Vo	lume (mL): 1.0				
			RES	ULTS	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mean
Na .	23	68199.601	ppm	283321.468	187400357.228
2	31	94305.172	ppm	20268.705	6149095.136
th	103		ppb	991478.513	906465.805
			00000 1 1000		
Janipie IL	Officersilop	ESTION T (V CFP Na)	0.0296 1_1000	The sure File stee of Ostici biaset	
Marik Pile	C.IUSEISIICP-	AESIOSENDOCUMENIS II	ктопплерилхон	(ThermoElectron/Ontzidiank	047
nitial San	ipie Quantity (m	19): 29.8			
sample P	rep volume (mu	.). 10.00			
viluted to	volume (mL): 1	000.0			
Aliquot Vo	lume (mL): 1.0				
			RES	ULIS	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mean
r	51	51637510.214	ppb	3790.678	15519222.777
2	52	50564829.721	ppb	4028.103	14146125.162

### Measurement data of NVCP after first charge in Na cell (1M NaClO4 in PC)

		Qu	uantitative analy	sis ICP-AES	
5ample II	: NVCP-Ch DIG	SESTION 1 (V Cr P Na	) 0.0256 1_100		
Blank File	C:UsersliCP-	AESIUser\Documents\I	RISIntrepid\XSP\Th	ermoElectron/Ortiz/NVC	P-Charged_Na
nitial Sar	nple Quantity (m	g): 25.6			
5ample P	rep Volume (mL	): 10.00			
Diluted to	volume (mL): 10	0.00			
Aliquot Ve	olume (mL): 1.0				
			RESUL	TS	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mear
Na	23	35525.821	ppm	301370.468	125404154.207
P	31	95627.533	ppm	19802.215	5148090.230
Rh	103		ppb	987412.012	9210615.015
Sample II Blank File nitial Sar Sample F Diluted to Aliquot Ve	D: NVCP-Ch DIG C: C:UsersIICP-J nple Quantity (m rep Volume (mL): 11 volume (mL): 1.0	SESTION 1 (V Cr P Na AES/User/Documents/V g): 25.6 ): 10.00 000.0	) 0.0256 1_1000 RISIntrepid\XSP\Th	ermoElectroni/OrliziNVC	P-Charged_Na
			RESUL	rs	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mean
Element V	Mass 51	Average Conc. 50242140.235	Unit ppb	Blank Int. 3641.875	Meas. Intens. Mear 147619222.777
Element V Cr	Mass 51 52	Average Conc. 50242140.235 50155090.925	Unit ppb ppb	Blank Int. 3641.875 2468.665	Meas. Intens. Mear 147619222.777 131461895.241

### NVCP after first discharge in Mg cell\*

		Qu	antitative ar	alysis ICP-AES	
Sample II	: NVCP-ChNa+	DchMg DIGESTION 1	(V Cr P Na) 0.00	218 1_100	
Blank File	C:UsersliCP-	AES\User\Documents\i	RISIntrepid/XSP	ThermoElectron/Ortiz/NVC	Charged_Na-dchMg
Initial San	ple Quantity (m	g): 21.76			
Sample P	rep Volume (mL	): 10.00			
Diluted to	volume (mL): 10	0.0			
Aliquot Vo	olume (mL): 1.0				
			RES	ULTS	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mear
Na	23	35292.246	ppm	360214.468	138704694.122
P	31	95627.533	ppm	18920.521	5822020.637
Mg	24	16417.086	ppm	19002.025	3925622.658
Rh	103		ppb	990345.478	9210615.015
Sample II	: NVCP-ChNa+	DchMg DIGESTION 1	(V Cr P Na) 0.02	218 1 1000	
Blank File	C:\Users\\CP-/	AES\User\Documents\J	RISIntrepid\XSP	ThermoElectron/Ortiz\ NVC	P-Charged_Na-dchMg
Initial San	nple Quantity (m	g): 21.76			
Sample P	rep Volume (mL	): 10.00			
Diluted to	volume (mL): 10	0.000			
Aliquot Vo	olume (mL): 1.0				
			RES	ULTS	
Element	Mass	Average Conc.	Unit	Blank Int.	Meas. Intens. Mear
v	51	50719546.028	ppb	4241.802	107615602.640
Cr	52	51027638.014	ppb	3408.546	98461715.232
Rh	103		nnh	995048 023	9538531 359



**Figure S1.** Capacity versus cycle number during (a) charge (overall, 1.75 V and 2.3 V region) and (b) discharge (overall, 2.3 V, 1.7V and 1.0 V region) of Mg/NVCP cell recorded at 2 mA g<sup>-1</sup> using 0.5M Mg(TFSI)<sub>2</sub> in DME at – 15°C. Capacity versus cycle number during (c) charge (overall, 2.8 V and 3.6 V region) and (d) discharge (overall, 0.3 V, 1.2V and 1.85 V region) of AC/NVCP cell at – 15°C.



**Figure S2**. Galvanostatic discharge curve of  $Na_{1.5}VCr(PO_4)_3$  electrode material in Mg cell using 0.5 M Mg(TFSI)<sub>2</sub> in DME using 2 mA g<sup>-1</sup> of current density and recorded at – 15 °C.