Supporting Information

"Fishnet-like" Ion-selective Nanochannels in Advanced Membranes

for Flow Batteries

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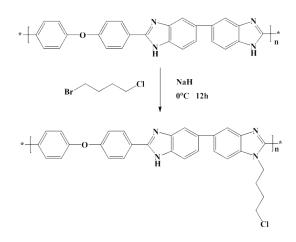


Figure S1. Synthesis of CPBI.

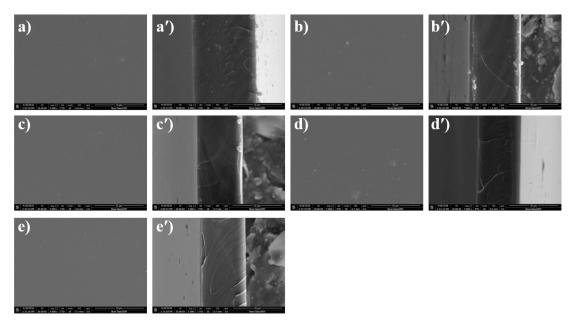
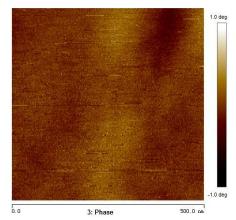


Figure S2. SEM images of membranes: PBI (a: surface, a': cross section); CPBI (b: surface, b': cross section); CPBI-PEI-10 (c: surface, c': cross section); CPBI-PEI-15 (d: surface, d': cross section); CPBI-PEI-20 (e: surface, e': cross section).





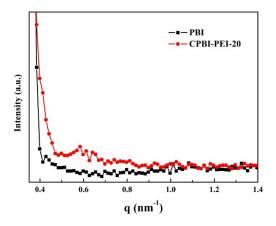


Figure S4. SAXS of PBI and CPBI-PEI-20 membrane.

Membrane	Proton conductivity ^a	VO ²⁺ permeability	Ion selectivity
	mS cm ⁻¹	10 ⁻¹⁰ cm ² s ⁻¹	10 ¹⁰ mS s cm ⁻³
PBI	35.2	No detected	-
CPBI	39.7	No detected	-
CPBI-PEI-10	63.1	0.5	118.1
CPBI-PEI-15	78.0	2.7	28.8
CPBI-PEI-20	115.9	9.2	12.6
Nafion 212	182.7	552.6	0.3

 Table S1. Proton conductivity, VO²⁺ permeability and ion selectivity of the prepared membranes.

^aThe proton conductivity is calculated from area resistances of prepared membranes as follow: proton conductivity = L/Area resistance, where L is thickness of membrane.

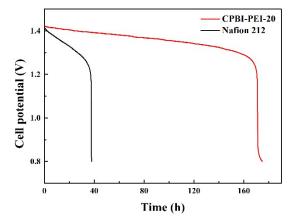


Figure S5. The self-discharge curves using CPBI-PEI-20 and Nafion 212 membranes.



Figure S6. Digital photograph of CPBI-PEI-20 membrane after 2000 charge-discharge cycles at 120 mA

cm⁻².

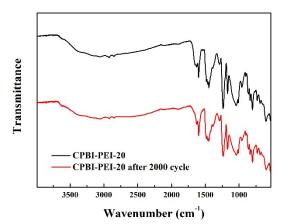


Figure S7. FTIR spectra of CPBI-PEI-20 membranes before and after 2000 charge-discharge cycles at 120

mA cm⁻².