

Supporting Information

Alkali metal-selenium battery with wide temperature range and low self-discharge

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Figures

Fig. S1. SEM images of the as-prepared PEDOT in different magnifications.....	4
Fig. S2. Morphology characterization of SO-HPC _x (x=0, 1, 2, 3, 4, 5).....	5
Fig. S3. (a) Nitrogen adsorption/desorption isotherms and (b) micropore distribution of SO-HPC _x	6
Fig. S4. (a) TEM and (b) HRTEM images of SO-HPC ₃	7
Fig. S5. (a) Overall XPS spectrum of SO-HPC ₃ . High-resolution (b) C1s, (c) S 2p and (d) O1s XPS spectrum of SO-HPC ₃	8
Fig. S6. (a) Overall XPS spectrum of Se ₅₀ /SO-HPC ₃ . (b) High-resolution Se 3d XPS spectrum of pristine Se.....	9
Fig. S7. (a) Nitrogen (N ₂) adsorption-desorption isotherms, (b) micropore and mesopore distribution profiles of SO-HPC ₃ and Se ₅₀ /SO-HPC ₃	10
Fig. S8. Cycling performances of Se/SO-HPC _x (x=0, 1, 2, 3, 4, 5) cathodes in Li-Se battery.....	11
Fig. S9. (a) SEM images of Se ₆₀ /SO-HPC ₃ and (b) Se ₇₀ /SO-HPC ₃	12
Fig. S10. (a) XRD patterns and (b) Raman spectra of Se ₆₀ /SO-HPC ₃ and Se ₇₀ /SO-HPC ₃	13
Fig. S11. (a) Nitrogen adsorption-desorption isotherms, (b) micropore distribution, mesopore distribution and TGA curves of Se ₅₀ /SO-HPC ₃ , Se ₆₀ /SO-HPC ₃ and Se ₇₀ /SO-HPC ₃	14
Fig. S12. Eletrochemical performances of Se ₆₀ /SO-HPC ₃ and Se ₇₀ /SO-HPC ₃ cathode	15
Fig. S13. Eletrochemical performances of SO-HPC ₃ and Se cathode in Li-Se battery	16
Fig. S14. Eletrochemical performances of Se ₅₀ /SO-HPC ₃ cathode with different area coposite loading.....	17
Fig. S15. (a) TEM and (b) HRTEM imgaes of Se ₅₀ /SO-HPC ₃ after 1700 cycles.....	18
Fig. S16. SEM of the front and cross section of the lithium sheet from the cells after 1700 cycles with Se ₅₀ /SO-HPC ₃ (a and c) and Se (b and d) cathode	19
Fig. S17. The interaction of SO-HPC ₃ composite with LiSe ₂	20
Fig. S18. (a) Nitrogen adsorption-desorption isotherms, (b) micropore distribution and mesopore size distribution of AC and Se ₅₀ /AC. (c) Electrochemical performance of Se ₅₀ /AC in Li-Se batteries	21

Fig. S19. Cycling performances of $\text{Se}_{50}/\text{SO-HPC}_3$ (a and b), Se_{50}/AC (c and d) and Se (e and f) cathodes at 50 °C and 0 °C, respectively, with 0.5 C for Li-Se batteries	22
Fig. 20. Self-discharge behavior of Li-Se batteries with $\text{Se}_{50}/\text{SO-HPC}_3$ cathodes after an uninterrupted 30 cycles at 0.5 C for Li-Se batteries	23
Fig. S21. Eletrochemical performances of $\text{Se}_{50}/\text{SO-HPC}_3$ cathode in Na-Se batteries	24
Fig. S22. Cycling performance of (a) pristine SO-HPC_3 and (b) Se in sodium-ion battery	25
Fig. S23. Eletrochemical performances of $\text{Se}_{50}/\text{SO-HPC}_3$ and Se cathode in K-Se batteries.	26
Fig. S24. The cycle performances of $\text{Se}_{50}/\text{SO-HPC}_3$ cathodes in Na-Se (a and b) and K-Se (c and d) cells at 50 °C and 0 °C at a current density of 0.5 C.	27

Tables

Table S1. Electrode materials test temperature statistics table	28
Table S2. Specific surface areas and pore parameters of the SO-HPC_x	28
Table S3. Conductivity-related data recorded by four-probe method measures	28
Table S4. Comparison of the electrochemical performance of $\text{Se}_{50}/\text{SO-HPC}_3$ (this work) with that of previously reported Se cathodes in alkali metal-Se batteries.	29

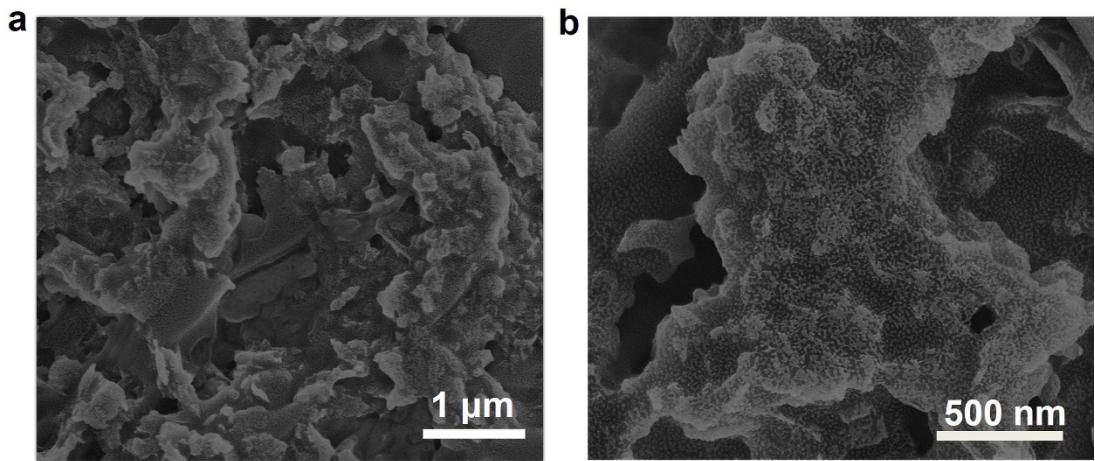


Fig. S1 SEM images of the as-prepared PEDOT in different magnifications.

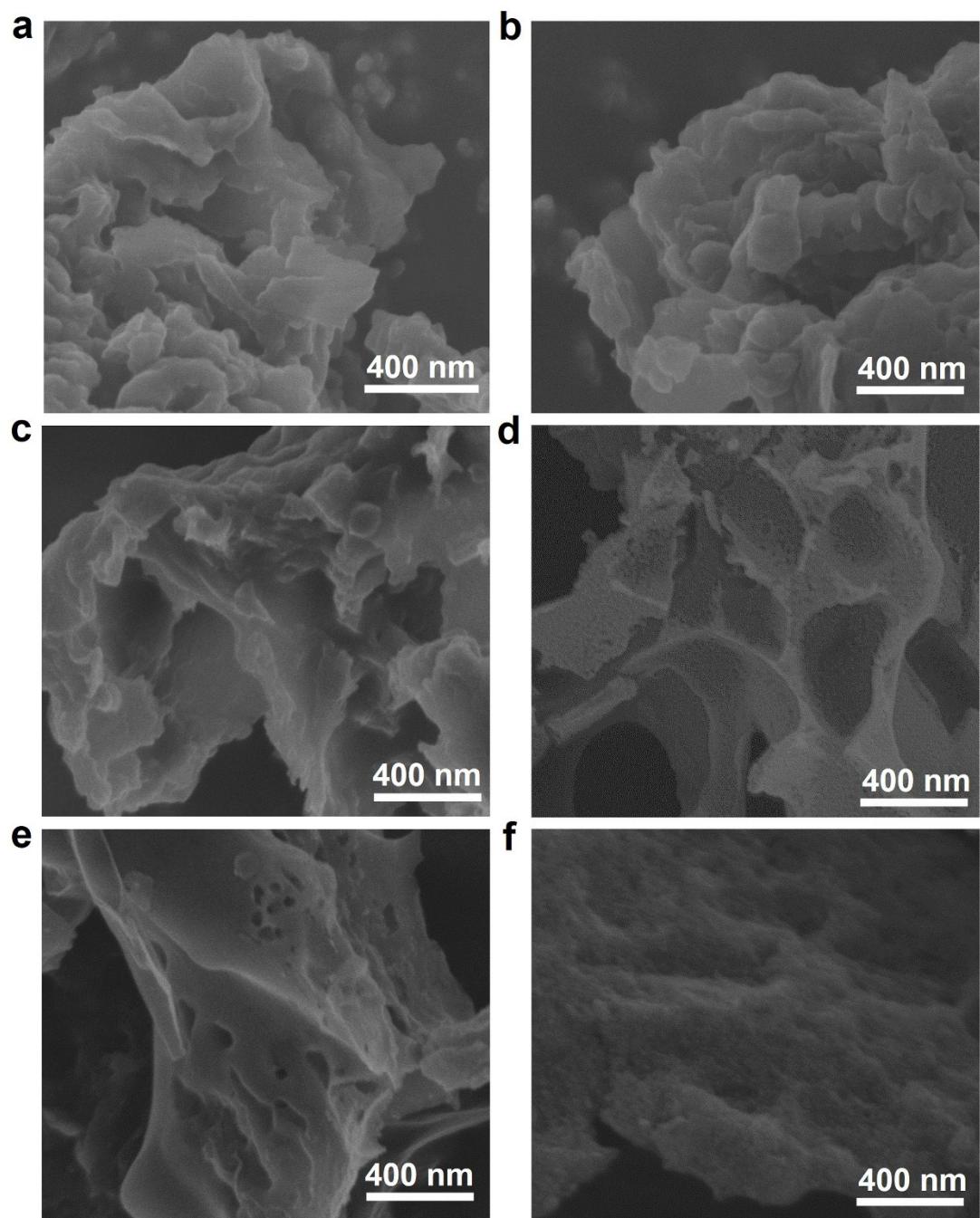


Fig. S2 Morphology characterization of SO-HPC_x. SEM images of SO-HPC samples obtained at various mass ratios of KOH to PEDOT: (a) SO-HPC₀; (b) SO-HPC₁; (c) SO-HPC₂; (d) SO-HPC₃; (e) SO-HPC₄; (f) SO-HPC₅.

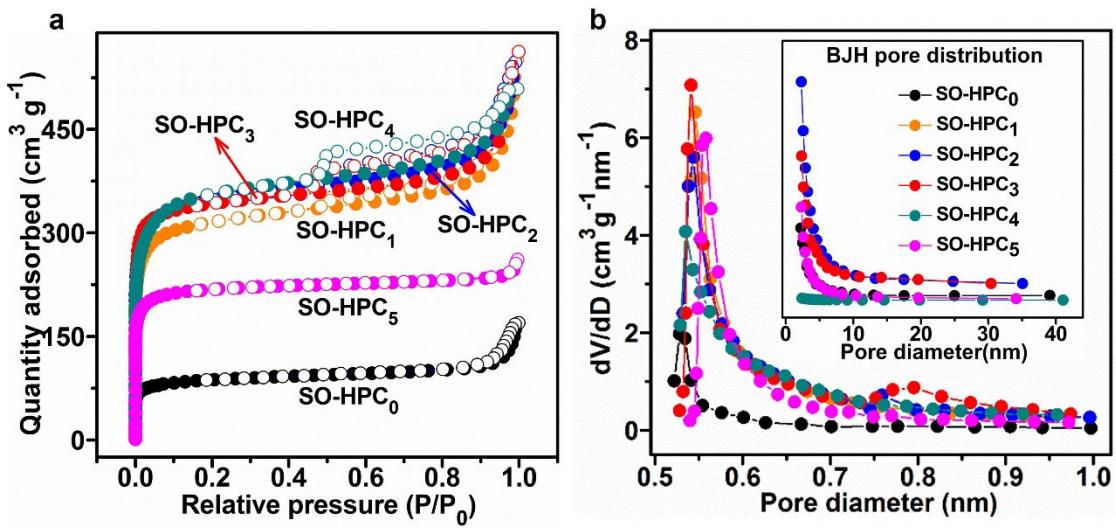


Fig. S3 (a) Nitrogen adsorption/desorption isotherms and (b) micropore distribution of SO-HPC_x ($x=0, 1, 2, 3, 4, 5$). Insert of Fig. S3b is the mesopore distribution of SO-HPC_x ($x=0, 1, 2, 3, 4, 5$).

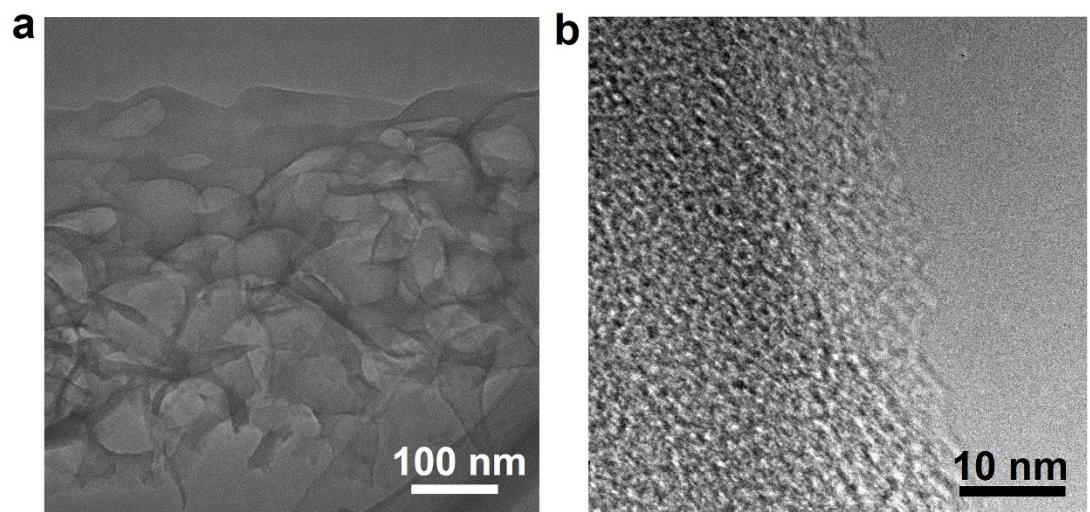


Fig. S4 (a) TEM and (b) HRTEM images of SO-HPC₃.

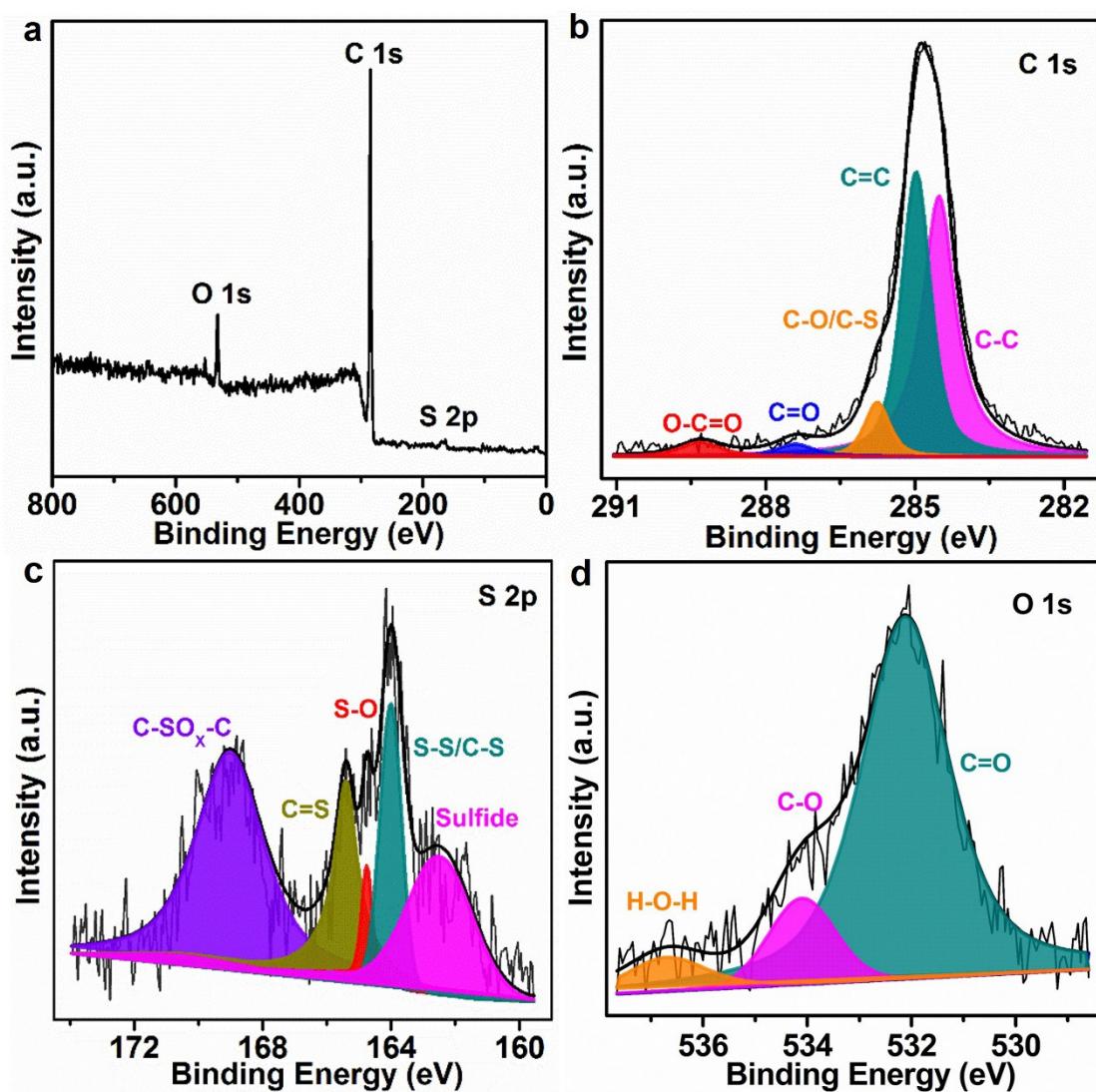


Fig. S5 (a) Overall XPS spectrum of SO-HPC₃. High-resolution (b) C1s, (c) S 2p and (d) O1s XPS spectrum of SO-HPC₃.

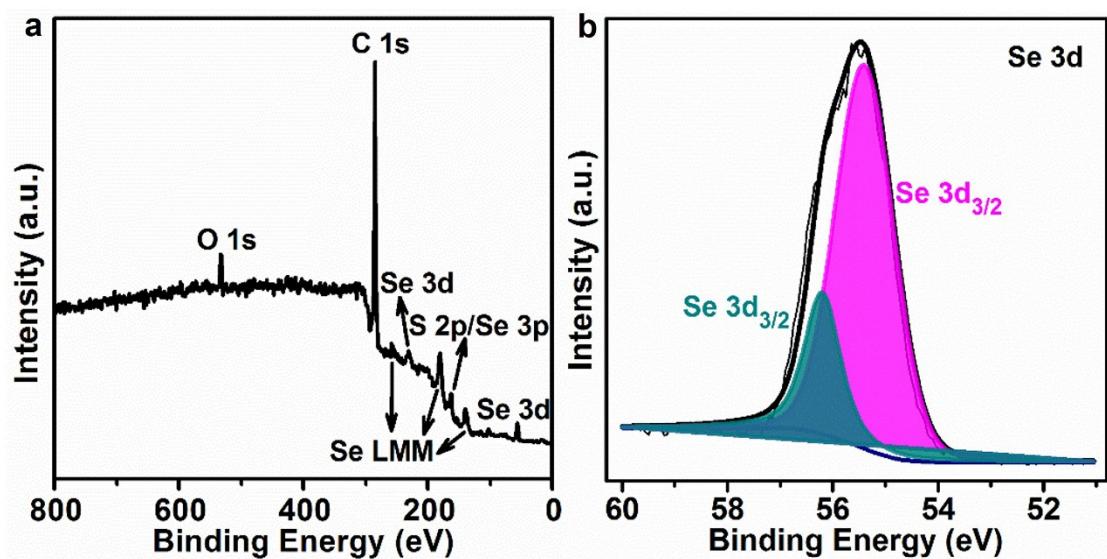


Fig. S6 (a) Overall XPS spectrum of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$. (b) High-resolution Se 3d XPS spectrum of pristine Se.

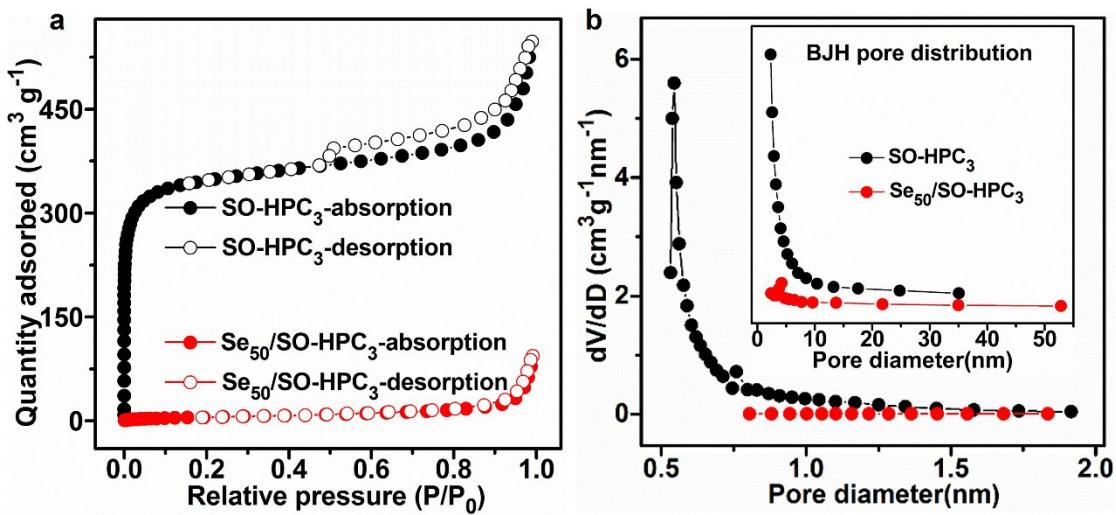


Fig. S7 (a) Nitrogen (N_2) adsorption-desorption isotherms and (b) micropore and mesopore (insert of Fig. S7b) distribution profiles of SO-HPC₃ and Se₅₀/SO-HPC₃.

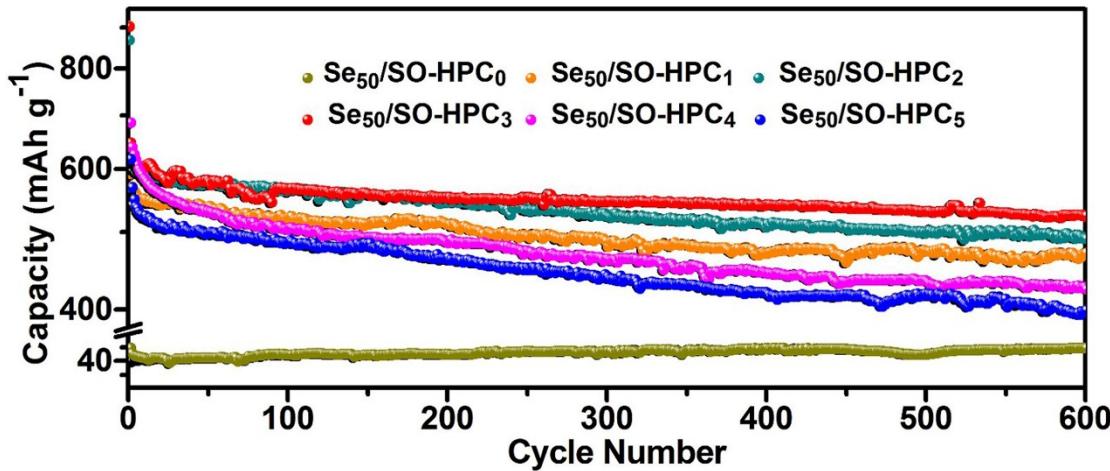


Fig. S8 Cycling performances of $\text{Se}_{50}/\text{SO}-\text{HPC}_x$ ($x=0, 1, 2, 3, 4, 5$) cathodes at a current density of 0.5 C with 50 wt.% Se-loading amount in the cut off voltage range of 1.0-3.0 V for Li-Se battery. The electrochemical performance of $\text{Se}_{50}/\text{SO}-\text{HPC}_x$ ($x=0, 1, 2, 3, 4, 5$) for Li-Se batteries were tested to evaluate the effect of KOH. The SO-HPC₃ derived from a suitable etch ratio with high specific surface area and high micropore volume not only could buffer the electrode volume change, but also prevent the side reaction between Se anions and electrolytes. Besides, the high mechanical strength (fragile framework) caused by moderate activation etching with appropriate amount of KOH and the high conductivities lead to satisfactory electrochemical properties of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$.

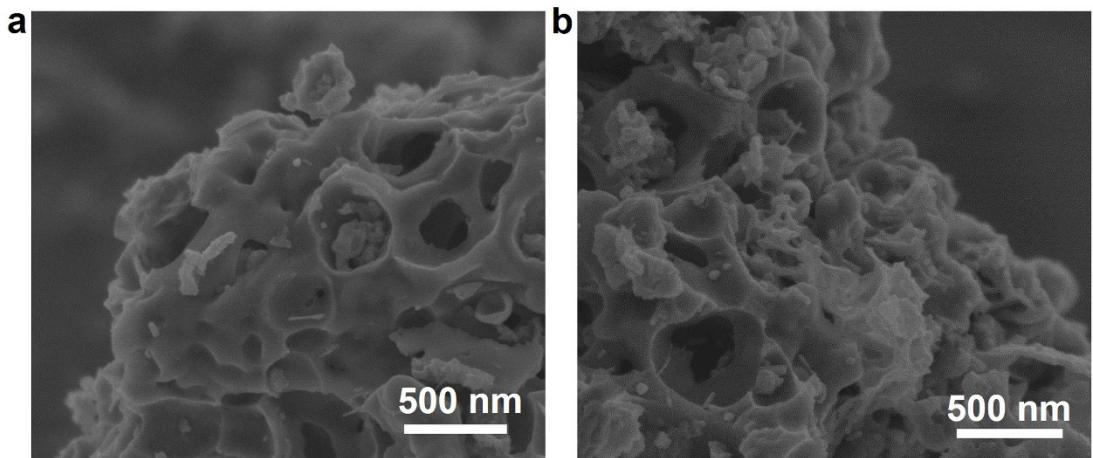


Fig. S9 (a) SEM images of $\text{Se}_{60}/\text{SO}-\text{HPC}_3$ and (b) $\text{Se}_{70}/\text{SO}-\text{HPC}_3$. The SEM image of $\text{Se}_{60}/\text{SO}-\text{HPC}_3$ (Fig. S9a) shows a similar morphology to that of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$, while a large amount of aggregated Se particles appear on the surface of $\text{Se}_{70}/\text{SO}-\text{HPC}_3$ (Fig. S9b).

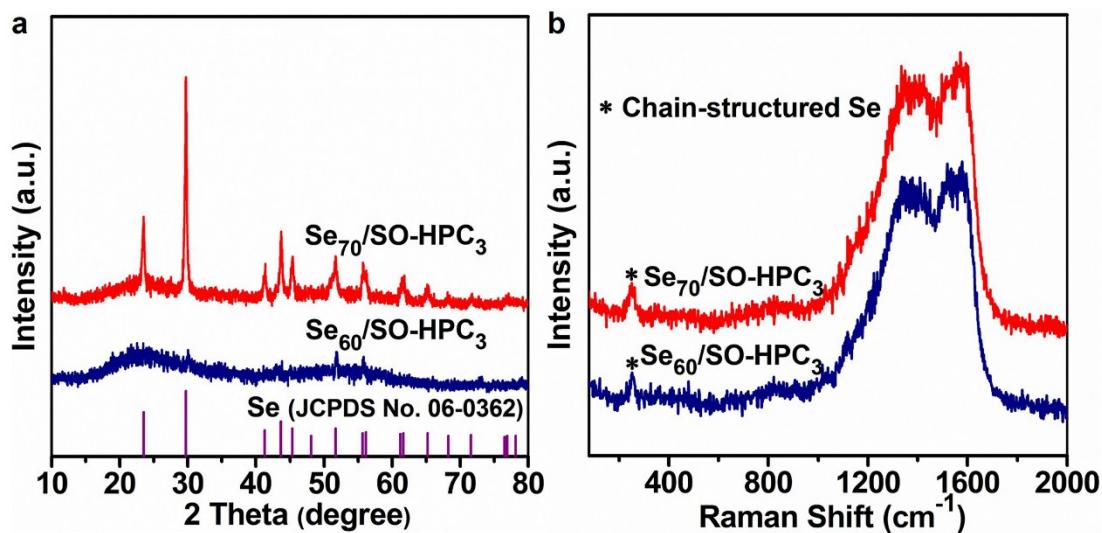


Fig. S10 (a) XRD patterns and (b) Raman spectra of $\text{Se}_{60}/\text{SO-HPC}_3$ and $\text{Se}_{70}/\text{SO-HPC}_3$. The reemerging characteristic diffraction peaks of Se in XRD (Fig. S10a) and Raman (Fig. S10b) confirm the presence of excessive Se in a crystalline state on the carbon matrix surface.

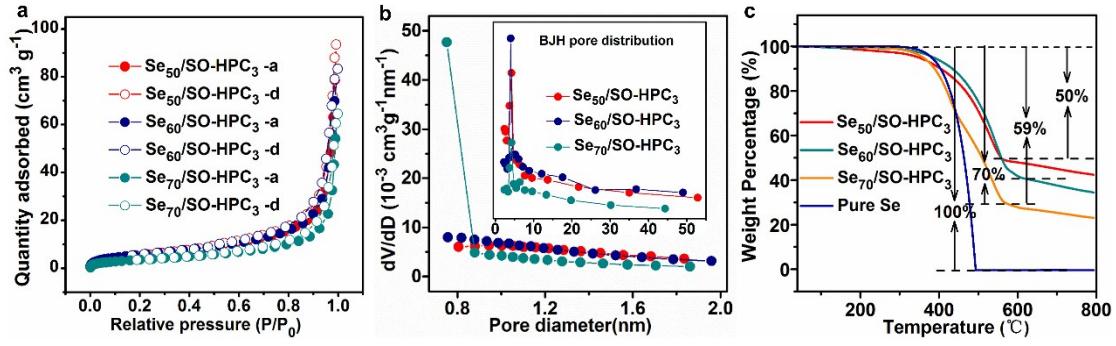


Fig. S11 (a) Nitrogen adsorption-desorption isotherms, (b) micropore distribution, mesopore (insert of Fig. S11b) distribution and TGA curves of $\text{Se}_{50}/\text{SO-HPC}_3$, $\text{Se}_{60}/\text{SO-HPC}_3$ and $\text{Se}_{70}/\text{SO-HPC}_3$. The TGA test results (Fig. S11c) confirm the Se loading amount and the different evaporation temperatures resulting from Se storage in different pore framework.

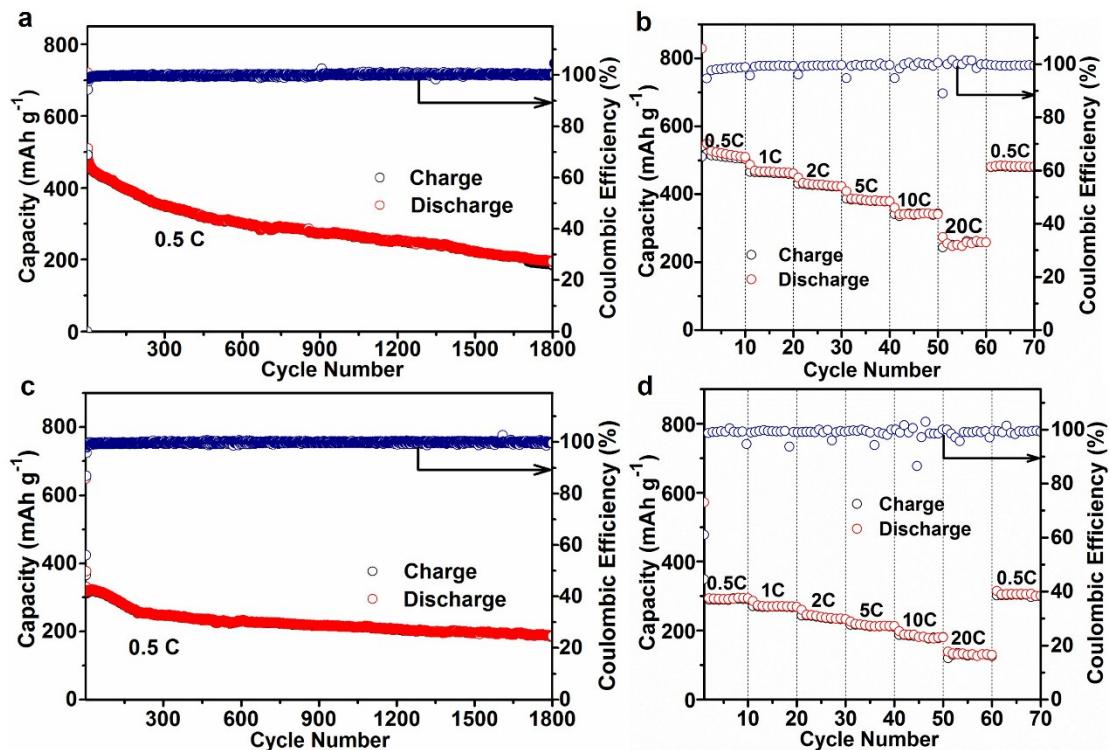


Fig. S12 Electrochemical performances of $\text{Se}_{60}/\text{SO-HPC}_3$ and $\text{Se}_{70}/\text{SO-HPC}_3$ cathode.

(a) Cycling stability of $\text{Se}_{60}/\text{SO-HPC}_3$ at a current density of 0.5 C in the cut off voltage range of 1.0-3.0 V and (b) rate capability of $\text{Se}_{60}/\text{SO-HPC}_3$ for Li-Se batteries. (c) Cycling stability of $\text{Se}_{70}/\text{SO-HPC}_3$ at a current density of 0.5 C in the cut off voltage range of 1.0-3.0 V and (d) rate capability of $\text{Se}_{70}/\text{SO-HPC}_3$ for Li-Se batteries.

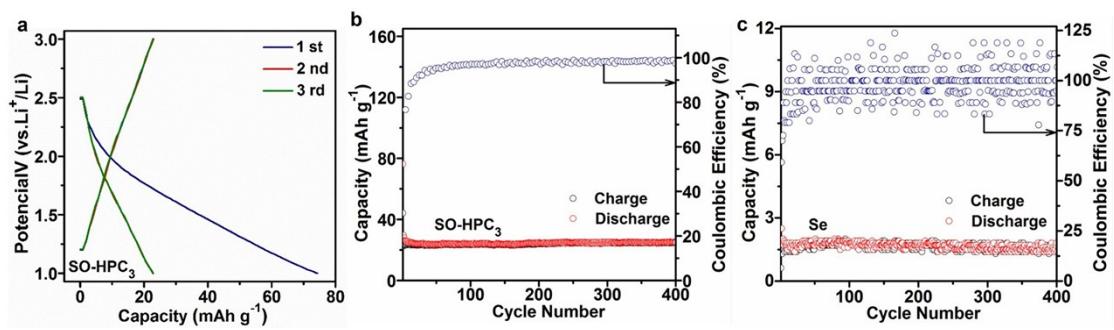


Fig. S13 Eletrochemical performances of SO-HPC₃ and Se cathode. (a) Galvanostatic discharge-charge curves of SO-HPC₃ at 0.5 C. Cycling performance of (b) pristine SO-HPC₃ and (c) Se at a current density of 0.5 C in the cut off voltage range of 1.0-3.0 V for Li-Se battery.

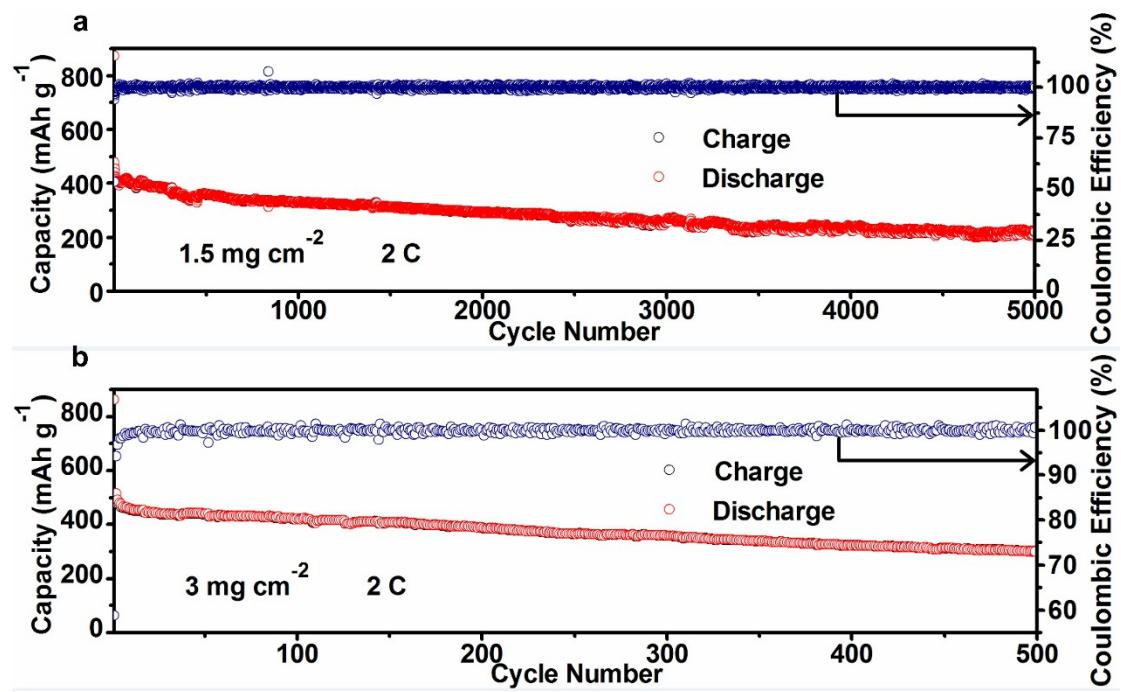


Fig. S14 Eletrochemical performances of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ cathode with different area coposite loading. (a) Cycle stability of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ in Li-Se battery at 2 C with an area loading of 1.5 mg cm^{-2} for 5000 cycles. (b) Cycle stability of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ in Li-Se battery at 2 C with an area composite loading of 3 mg cm^{-2} .

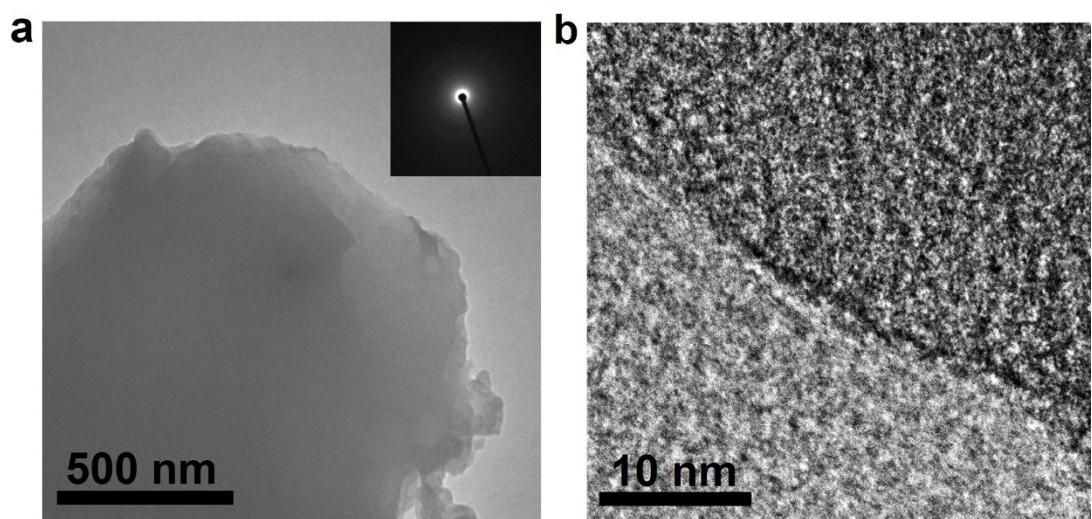


Fig. S15 (a) TEM and (b) HRTEM images of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ after 1700 cycles at 0.5 C. The insert of Fig. S15a is its SAED pattern.

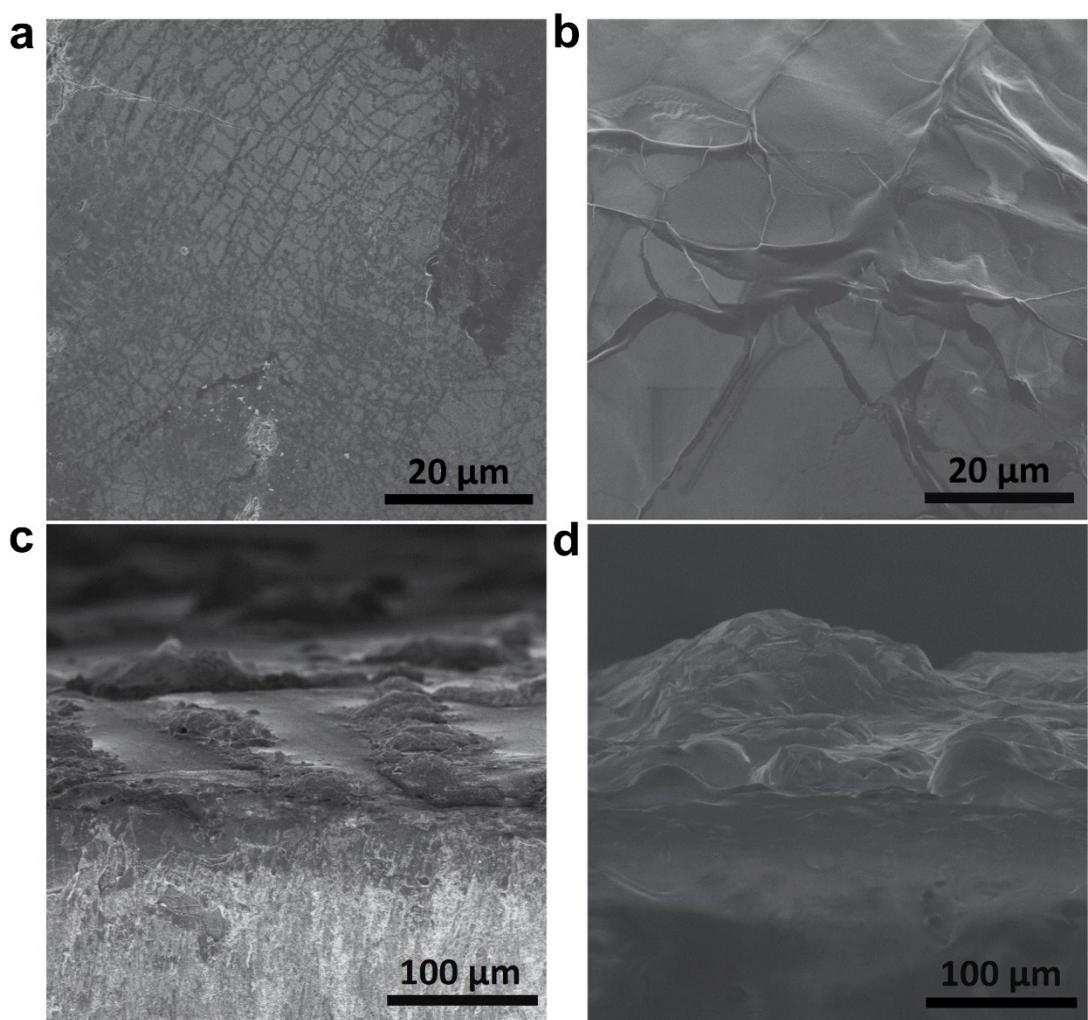


Fig. S16 SEM of the front and cross section of the lithium sheet from the cells after 1700 cycles at 0.5 C with $\text{Se}_{50}/\text{SO-HPC}_3$ (a and c) and Se (b and d) cathode.

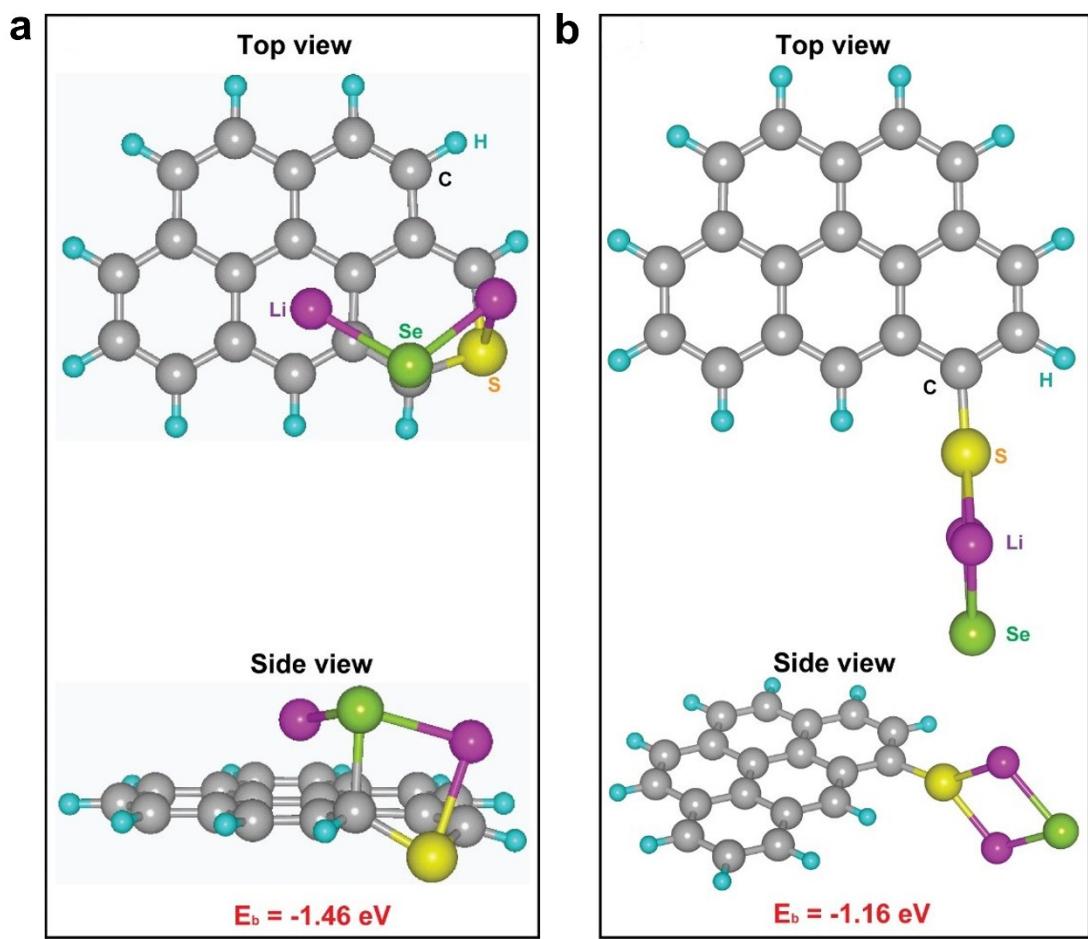


Fig. S17 The interaction of SO-HPC₃ composite with LiSe₂. Schematic adsorption configuration of Li₂Se on carbon only with S-doping. The calculated binding energies (in eV) are also presented.

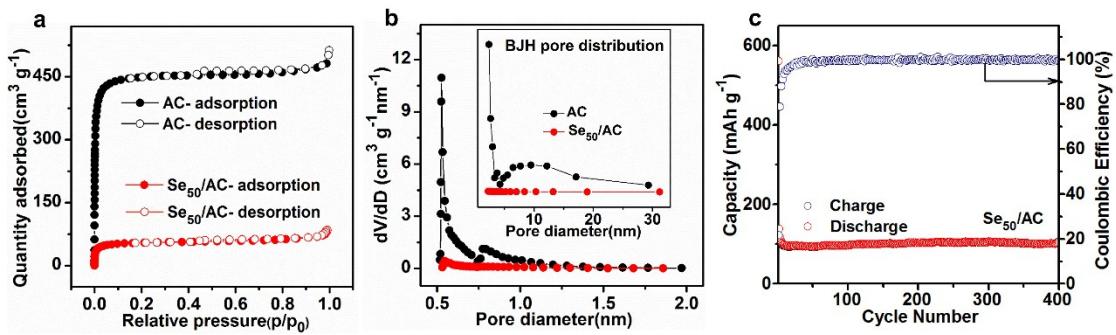


Fig. S18 (a) Nitrogen adsorption-desorption isotherms, (b) micropore distribution and mesopore (insert of Fig. S18b) size distribution of AC and Se_{50}/AC . (c) Electrochemical performance of Se_{50}/AC in Li-Se batteries at 0.5 C in the cut off voltage range of 1.0-3.0 V.

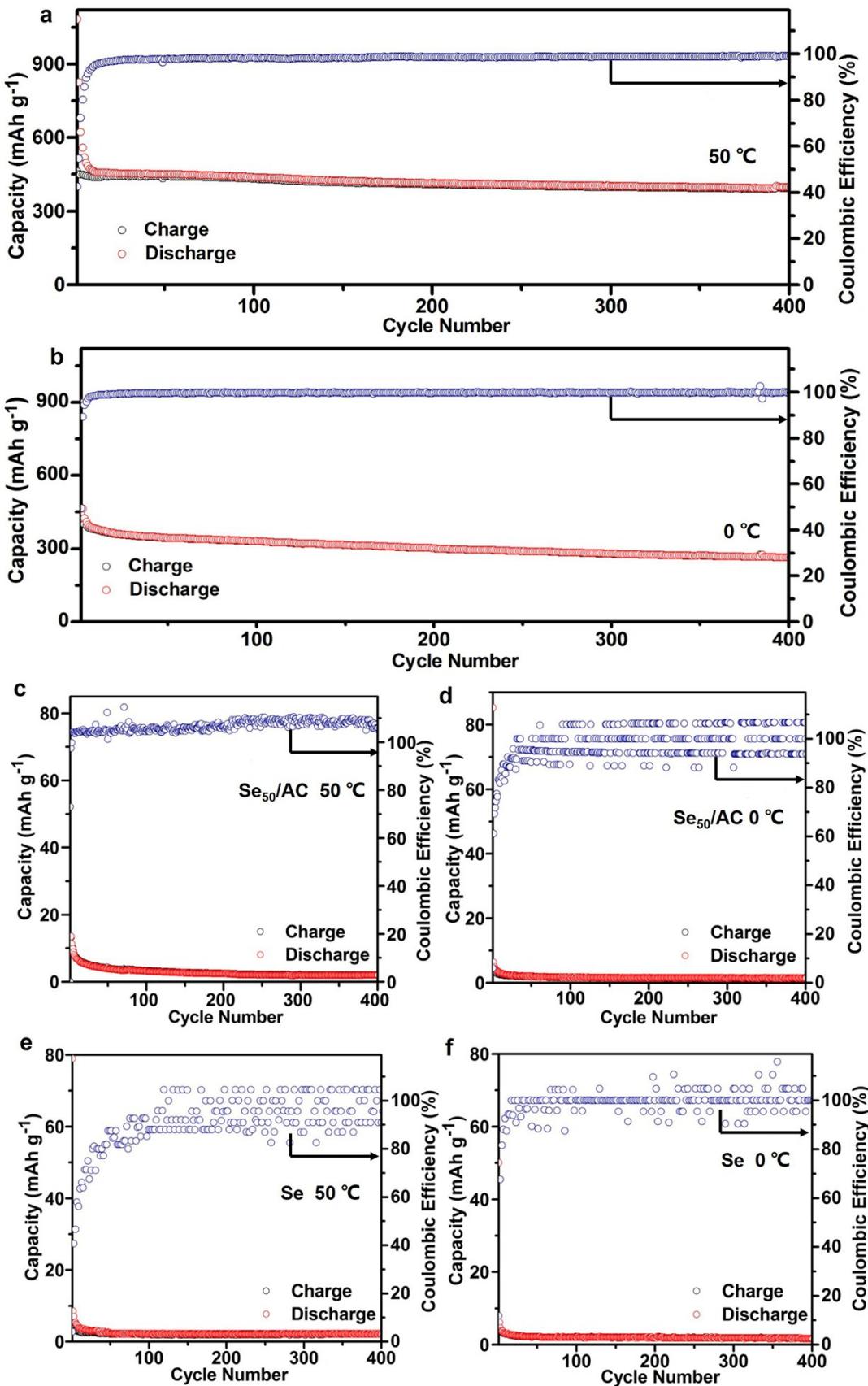


Fig. S19 Cycling performances of Se₅₀/SO-HPC₃ (a and b), Se₅₀/AC (c and d) and Se (e and f) cathodes at 50 °C and 0 °C, respectively, with 0.5 C for Li-Se batteries.

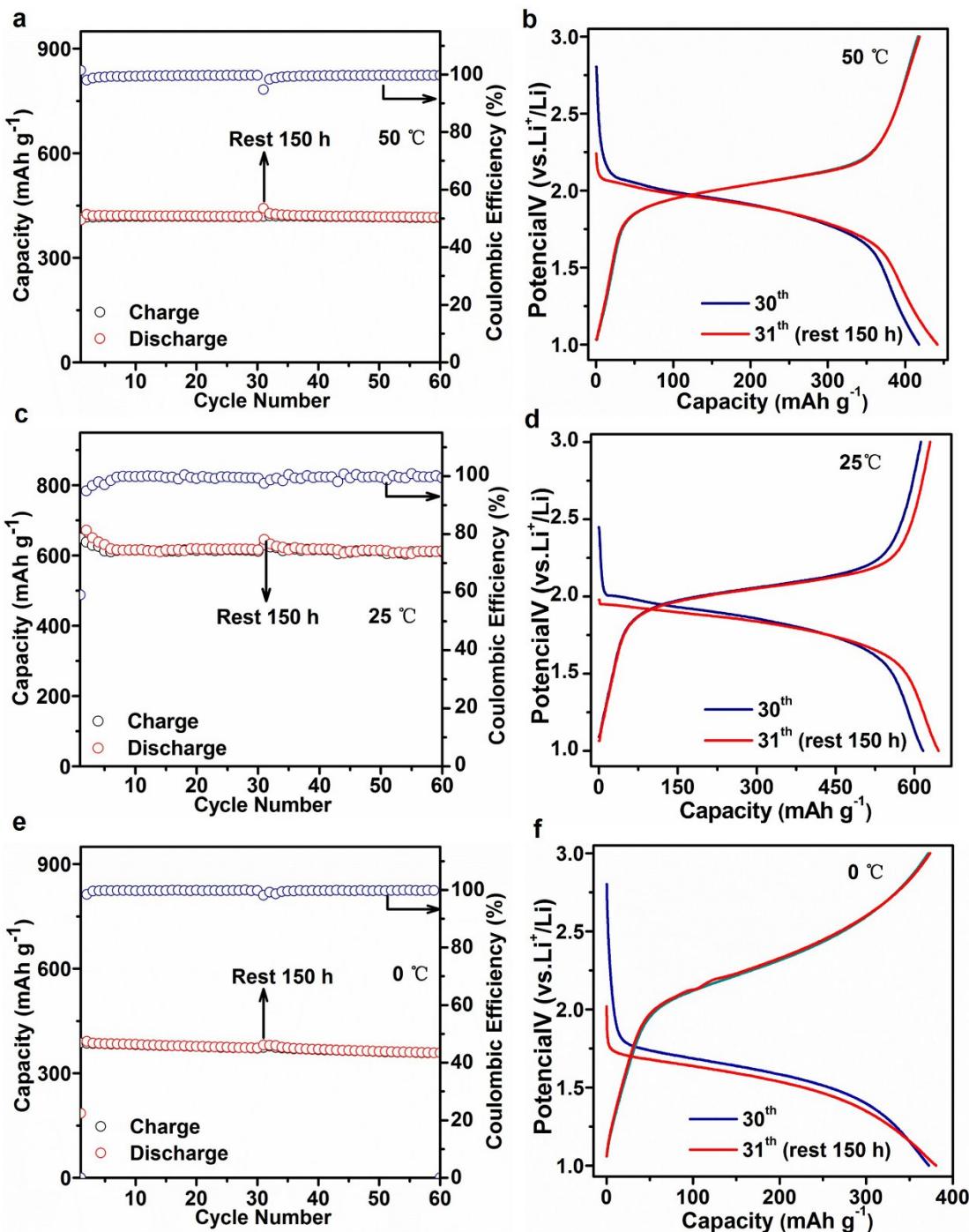


Fig. S20 Self-discharge behavior of Li-Se batteries with $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ cathodes after an uninterrupted 30 cycles at 0.5 C for Li-Se batteries. Cyclic performance and galvanostatic charge/discharge curves of the cathode materials before and after a rest period of 150 h at 50 °C (a and b), 25 °C (c and d) and 0 °C (e and f) for Li-Se batteries.

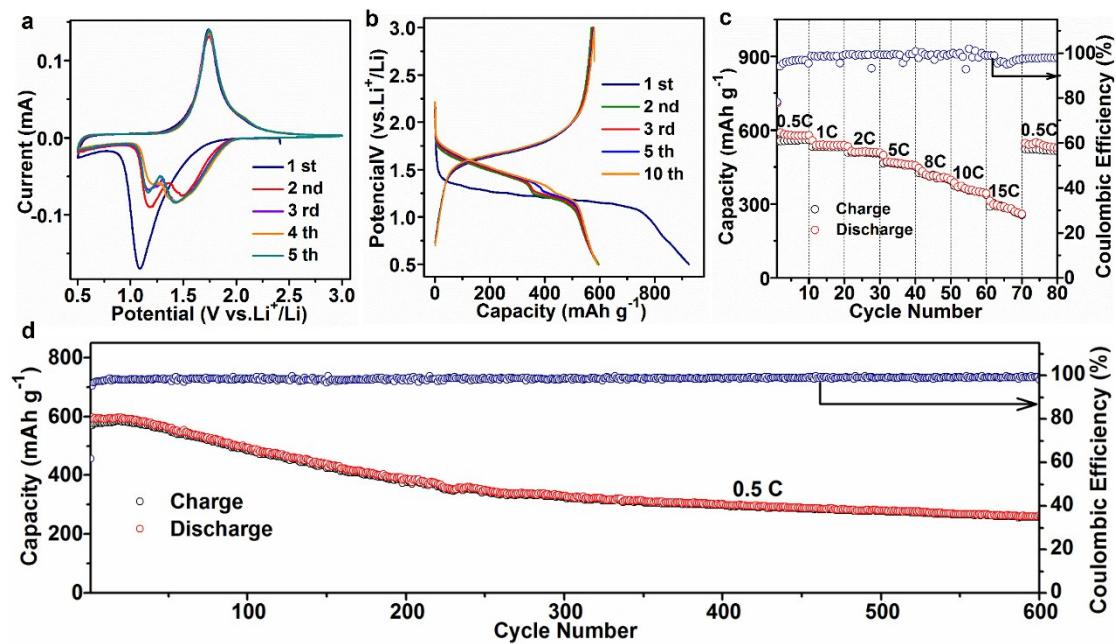


Fig. S21 Eletrochemical performances of Se₅₀/SO-HPC₃ cathode in Na-Se batteries.

(a) CV curves of Se₅₀/SO-HPC₃ at a scan rate of 0.1 mV s⁻¹ in the cut off voltage range of 0.5-3.0 V vs. Na/Na⁺. (b) Galvanostatic discharge-charge curves of Se₅₀/SO-HPC₃ for different cycles at 0.5 C for Na-Se batteries. (c) Rate capability and (d) cycling performance of Se₅₀/SO-HPC₃ at 0.5 C for Na-Se batteries.

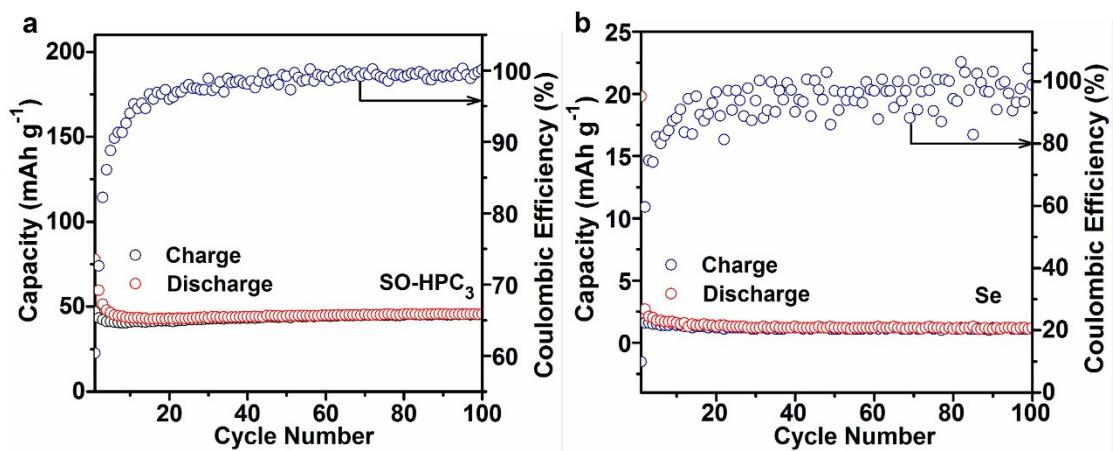


Fig. S22 Cycling performance of (a) pristine SO-HPC₃ and (b) Se for sodium battery at 0.5 C in the cut off voltage range of 0.5-3.0 V.

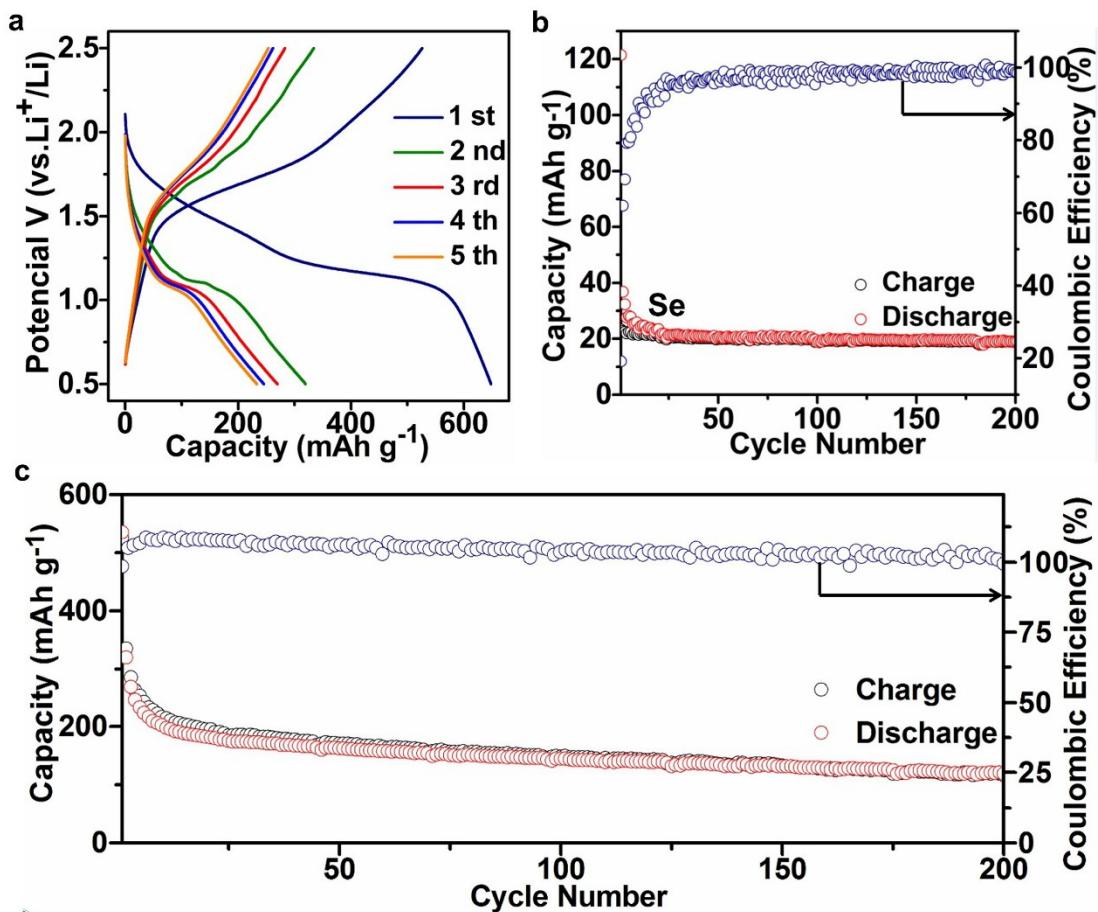


Fig. S23 Eletrochemical performances of Se₅₀/SO-HPC₃ and Se cathode in K-Se batteries. (a) Galvanostatic discharge-charge curves of Se₅₀/SO-HPC₃ for different cycles at 0.5 C in the cut off voltage range of 0.5-2.5 V vs. K/K⁺ for K-Se batteries. Cycling performance of (b) Se and (c) Se₅₀/SO-HPC₃ at 0.5 C for potassium batterie

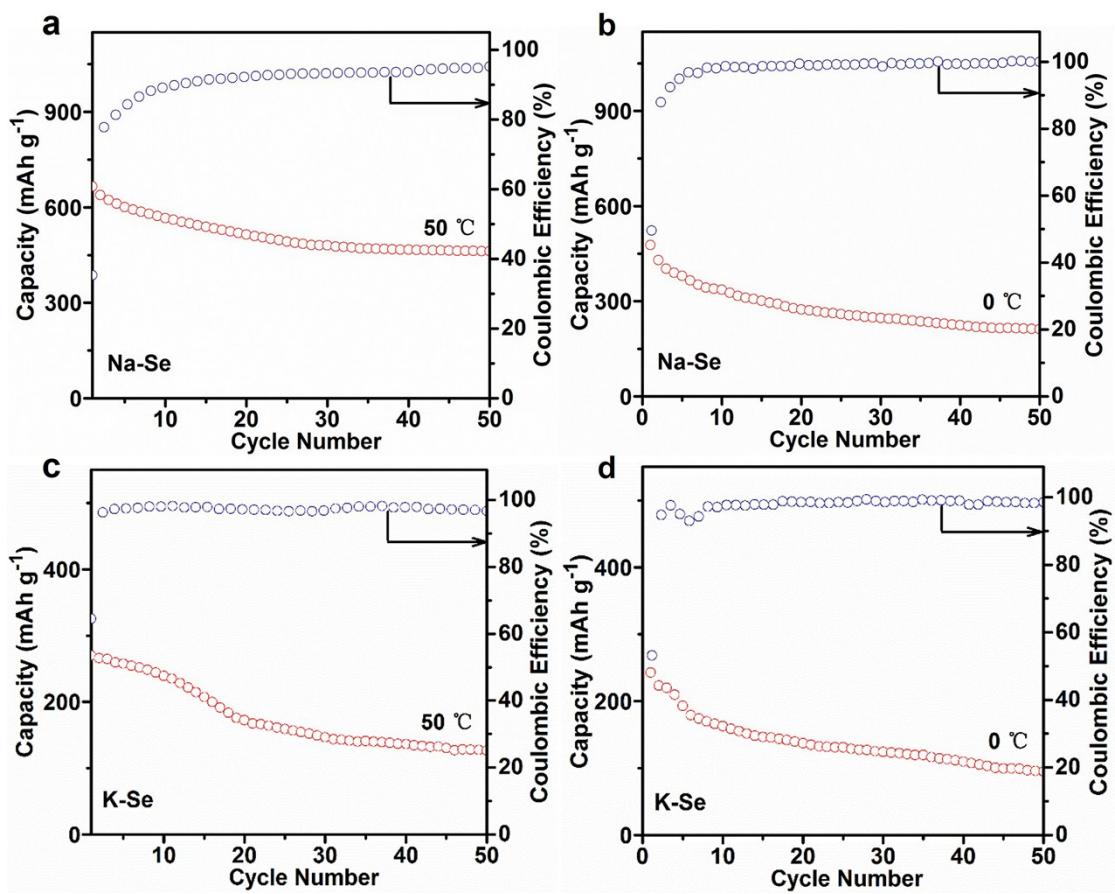


Fig. S24 The cycle performances of $\text{Se}_{50}/\text{SO}-\text{HPC}_3$ cathodes in Na-Se (a and b) and K-Se (c and d) cells at 50 °C and 0 °C at a current density of 0.5 C.

Table S1 Electrode materials test temperature statistics table.

System	Samples	0 °C	25 °C	50 °C
LIB	Se ₅₀ /SO-HPC _x (x=0, 1, 2, 4, 5)	√	/	/
	Se _n /SO-HPC3 (n=60, 70)	√	/	/
	Se ₅₀ /SO-HPC ₃	√	√	√
	Se and Se ₅₀ /AC	√	√	√
SIB	SO-HPC ₃	√	/	/
	Se ₅₀ /SO-HPC ₃	√	√	√
	Se and SO-HPC ₃	√	/	/
PIB	Se ₅₀ /SO-HPC ₃	√	√	√
	Se	√	/	/

Table S2 Specific surface areas and pore parameters of the SO-HPC_x (x=0, 1, 2, 3, 4, 5)

Sample	S_{BET} (m² g⁻¹)^a	A_{micro} (nm)^b	V (cm³ g⁻¹)^c	V_{micro} (cm³ g⁻¹)^d
SO-HPC₀	363.3	0.53	0.26	0.13
SO-HPC₁	1351.6	0.55	0.85	0.49
SO-HPC₂	1470.8	0.54	0.85	0.52
SO-HPC₃	1487.7	0.54	0.87	0.54
SO-HPC₄	1519.5	0.54	0.79	0.55
SO-HPC₅	941.8	0.56	0.40	0.34

^aS_{BET}: the specific surface area; ^bA_{micro}: the average micropore size; ^cV: the total pore volume; ^dV_{micro}: the micropore volume.

Table S3 Conductivity-related data recorded by four-probe method measures

Sample	t (m)^a	Sr (Ω□)^b	σ (S m⁻¹)^c
SO-HPC₀	/	/	/
SO-HPC₁	/	/	/
SO-HPC₂	2.7×10 ⁻⁵	6.2×10 ⁴	0.60
SO-HPC₃	4.3×10 ⁻⁵	5.8×10 ³	4.00
SO-HPC₄	3.5×10 ⁻⁵	2.1×10 ⁴	1.36
SO-HPC₅	2.7×10 ⁻⁵	3.2×10 ⁴	1.16

^at: thickness of material; ^bSr: sheet resistance; ^cσ: conductivity ($\sigma=1/(Sr \times t)$).

Table S4 Comparison of the electrochemical performance of $\text{Se}_{50}/\text{SO-HPC}_3$ (this work) with that of previously reported Se cathodes in alkali metal-Se batteries.

Sample	Loading density of Se (wt%)	Cycling performance (1C=675 mA g ⁻¹)		Rate performance	Ref
		Retained capacity [mAh g ⁻¹],	Capacity retention rate [%]		
Li-Se battery					
This work	50 %	450 (RT ^a) & 77.8% (0.5 C, 1700 cycles) 219 (RT ^a) & -- (2 C, 5000 cycles) 264 (0 °C) & 53.9% (0.5 C, 400 cycles) 394 (50 °C) & 86.1% (0.5 C, 400 cycles)		225 & 20 C	
Se/C	54 %	480 (RT ^a) & 100% (0.25 C, 1000 cycles)		229 & 5 C	[1]
PPC/Se	70 %	380 (RT ^a) & 63.3% (1 C, 60 cycles)		380 & 1 C	[2]
C/Se	30 %	430 (RT ^a) & 77% (0.15 C, 250 cycles)		280 & 1.8 C	[3]
UCNS-Se	64%	347 (RT ^a) & -- (2 C, 240 cycles)		--	[4]
Se@CNx	62.5 % ^a	453 (RT ^a) & 76% (1.2 C, 400 cycles)		474 & 2.4 C	[5]
MHPCS/Se	48.68 %	261 (RT ^a) & 44.3% (0.5 C, 500 cycles)		175.4 & 5 C	[6]
NCS/Se	50 %	443 (RT ^a) & -- (0.5 C, 200 cycles)		286.6 & 5 C	[7]
Se/PCNs	70.5 %	417 (RT ^a) & 64% (1 C, 1200 cycles)		386 & 20 C	[8]
Se/CMCs	50 %	425 (RT ^a) & -- (0.2 C, 100 cycles)		218.1 & 5 C	[9]
Se@PPy	60 %	399 (RT ^a) & -- (0.2 C, 100 cycles)		238 & 2 C	[10]
Se/CNTs-M	73.3 %	428 (RT ^a) & 80% (1 C, 500 cycles)		390 & 5 C	[11]
Se/C (PAN)	56.4 %	240 (RT ^a) & -- (0.15 C, 50 cycles)		187 & 0.74 C	[12]

TP-NCF/Se	60 %	463 (RT ^a) & 87% (0.5 C, 300 cycles)	235 & 10 C	[13]
Se/CMK-3	49 %	600 (RT ^a) & 89.5% (0.1 C, 50 cycles)	310.5 & 5 C	[14]
FLSC	63.5 %	508 (RT ^a) & 64% (0.074 C, 300 cycles)	389 & 1.2 C	[15]
Se/C-700	50 %	485.3 (RT ^a) & 88.2% (0.24 C, 80 cycles)	452.3 & 1.2 C	[16]
Se@HCS200	49 %	400 (RT ^a) & 80% (0.5 C, 400 cycles)	200 & 5 C	[17]
Se-NCHPC	56.2 %	305 (RT ^a) & 70% (2 C, 60 cycles)	261 & 5 C	[18]
Se/N-CSHPC	62 %	462 (RT ^a) & 94% (0.5 C, 200 cycles)	438 & 2 C	[19]
Na-Se battery				
This work	50 %	265 (RT ^a) & 44% (0.5 C, 600 cycles)	287 & 15 C	
Se/C	54 %	340 (RT ^a) & 70% (0.25 C, 380 cycles)	168 & 5 C	[1]
C/Se	30 %	280 (RT ^a) & 43% (0.15 C, 50 cycles)	138 & 1.8 C	[2]
Se-CCN	56.3 %	539 (RT ^a) & 88% (0.2 C, 500 cycles)	184 & 20 C	[20]
K-Se battery				
This work	50 %	120 (RT ^a) & 22.6% (0.5 C, 200 cycles)	--	
Se/MDPC	53 %	130 (RT ^a) & 39.8% (0.2 C, 100 cycles)	101 & 1 C	[21]
c-PAN-Se	40 %	396 (RT ^a) & -- (0.2 C, 100 cycles)	116 & 10 C	[22]

^aRT: Room Temperature (~25 °C).

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