

Supporting Information

Synthesis and Water Oxidation Electrocatalytic and Electrochromic Behaviors of Mesoporous Nickel Oxide Thin Film Electrodes

Assel Amirzhanova,¹ Irmak Karakaya,¹ Can Berk Uzundal,¹ Gözde Karaoğlu,¹ Ferdi Karadas,^{1,2} Burak Ülgüt*,¹ Ömer Dag*^{1,2}.

¹Department of Chemistry, Bilkent University, 06800, Ankara, Turkey.

²UNAM — National Nanotechnology Research Center and Institute of Materials Science and Nanotechnology, Bilkent University, 06800, Ankara, Turkey.

Table S1. Composition of the clear solutions, used for the preparations of mesoporous NiO films.

Ni(II)/C ₁₂ E ₁₀ Mole Ratio		Amount of [Ni(OH) ₂] ₆ (NO ₃) ₂ (g)	Amount of CTAB (g)	Amount of C ₁₂ E ₁₀ (g)	Amount of Ethanol (ml)
2:1		0.4368	0.2908	0.500	5
4:1		0.9282	0.2908	0.500	5
6:1		1.3923	0.2908	0.500	5
8:1		1.8564	0.2908	0.500	5
10:1		2.3205	0.2908	0.500	5
12:1		2.7532	0.2908	0.500	5
15:1		3.4808	0.2908	0.500	10
20:1		4.6410	0.2908	0.500	10
25:1		5.8010	0.2908	0.500	10
30:1		6.9615	0.2908	0.500	10

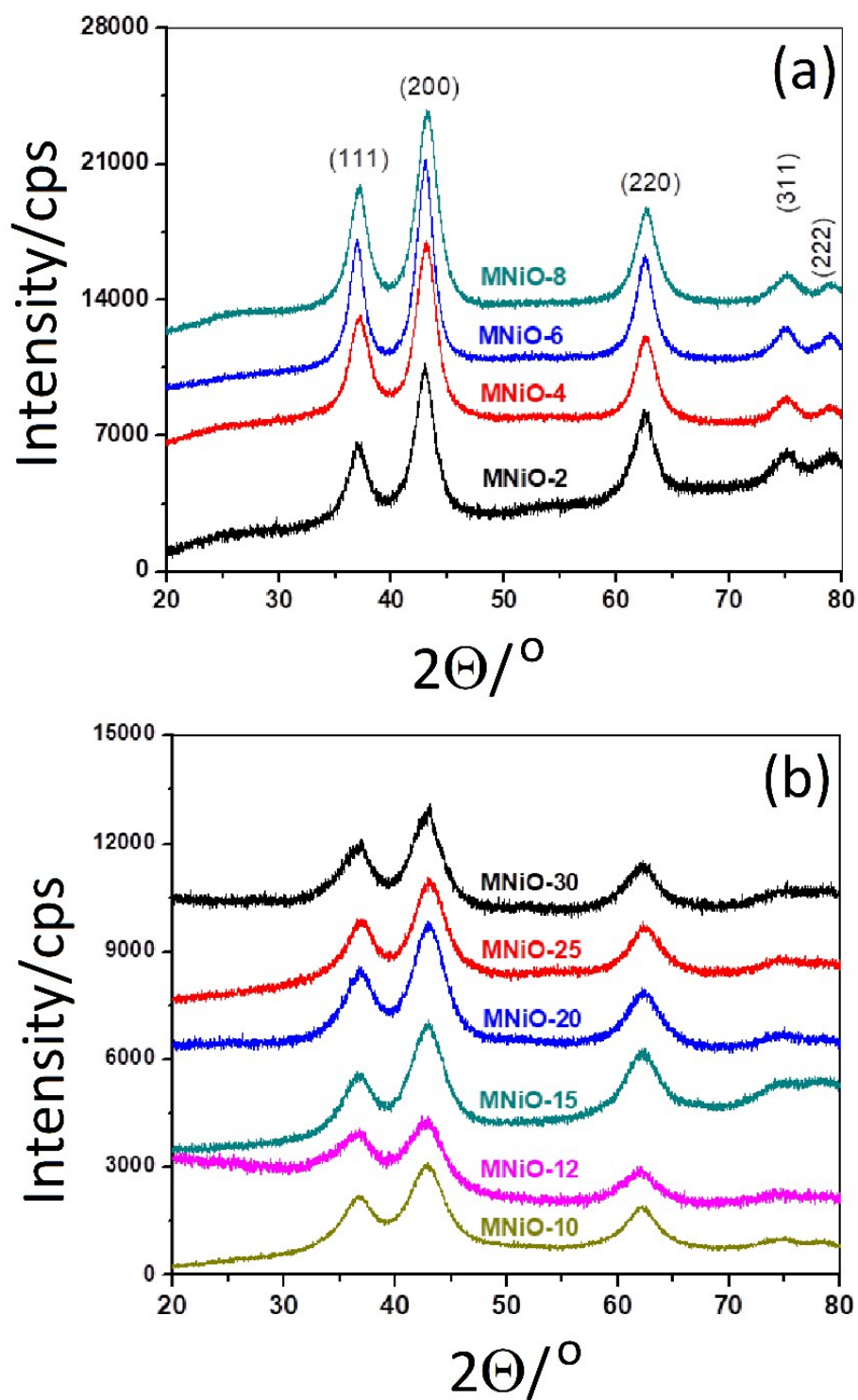


Figure S1. XRD patterns of mesoporous NiO, prepared using (a) small and (high) Ni(II)/C₁₂E₁₀ mole ratio solutions and calcination at 300 °C.

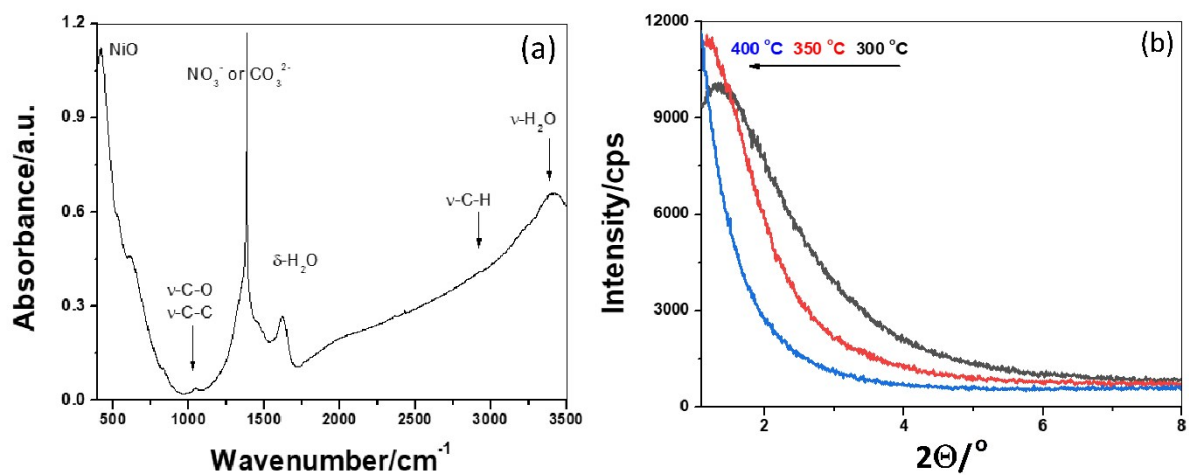


Figure S2. (a) FTIR spectrum of m-NiO-10-300 and (b) XRD patterns of m-NiO-10-300, m-NiO-10-350, and m-NiO-10-400.

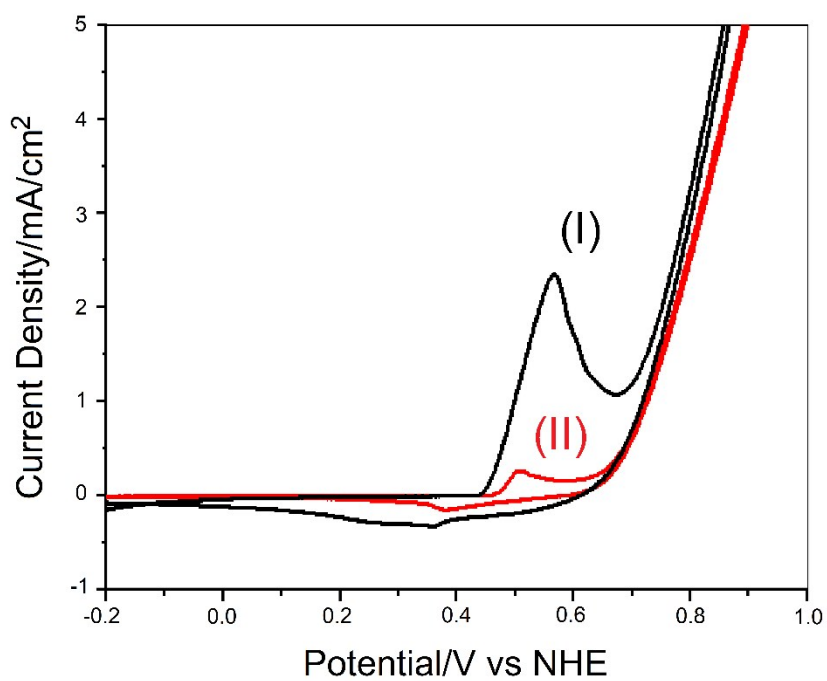


Figure S3. CV of fresh (I) and aged (II) mesoporous Ni(OH)₂ in 1 M KOH solution.

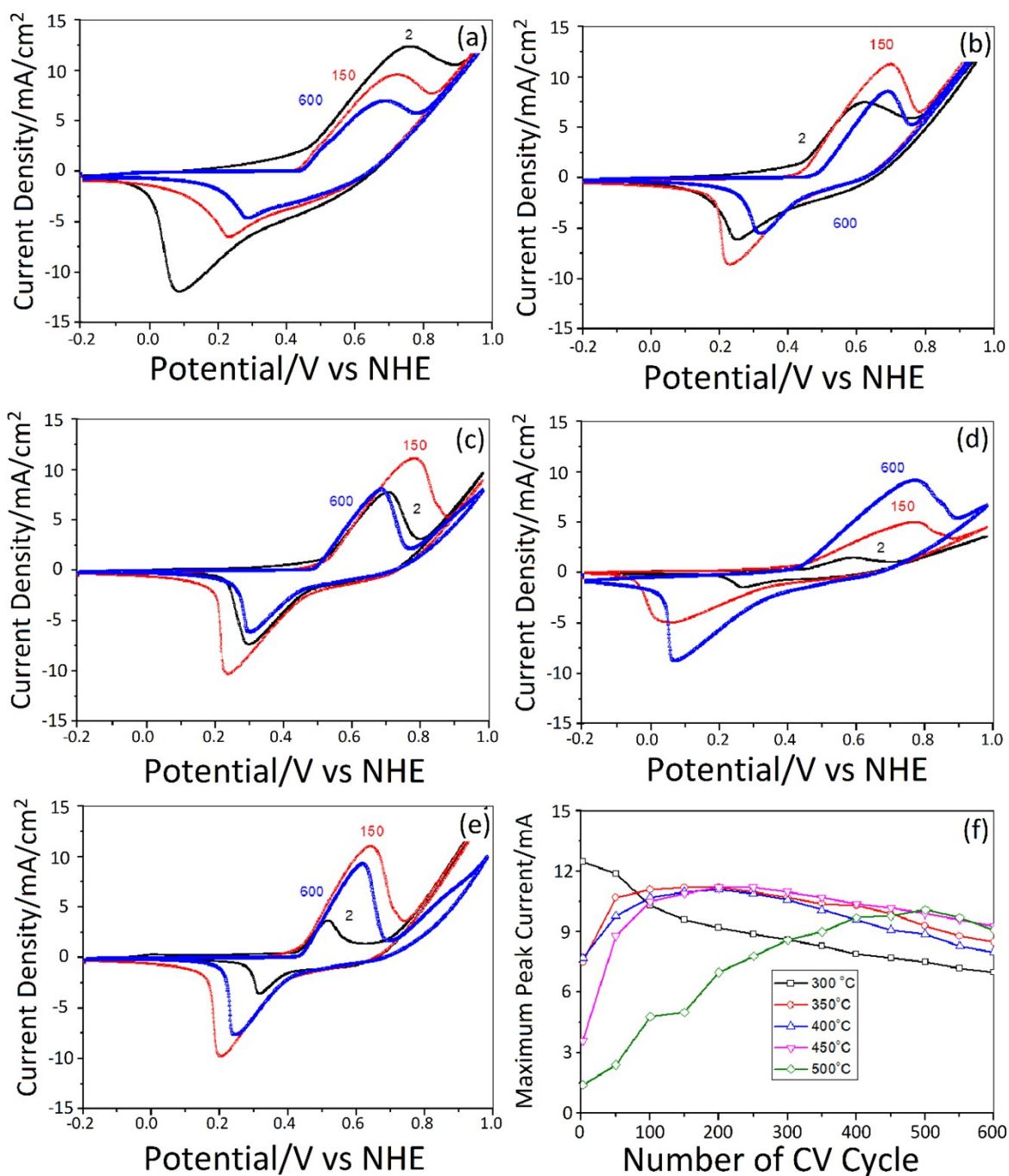


Figure S4. Time dependent CV cycles (2nd, 150th and 600th cycles) of m-NiO-10 calcined/annealed at (a) 300, (b) 350, (c) 400, (d) 450, and (e) 500 °C and (f) overall change in peak current at around 0.6 V over cycling.

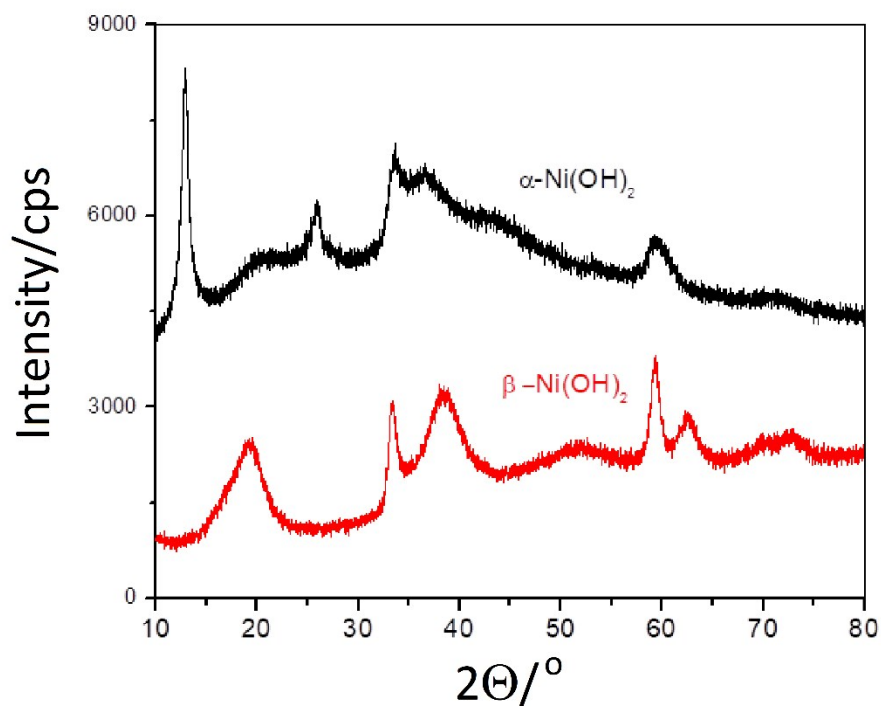


Figure S5. XRD of Ni(OH)₂ (top) the fresh and (bottom) aged in 1M KOH solution for 1 day.

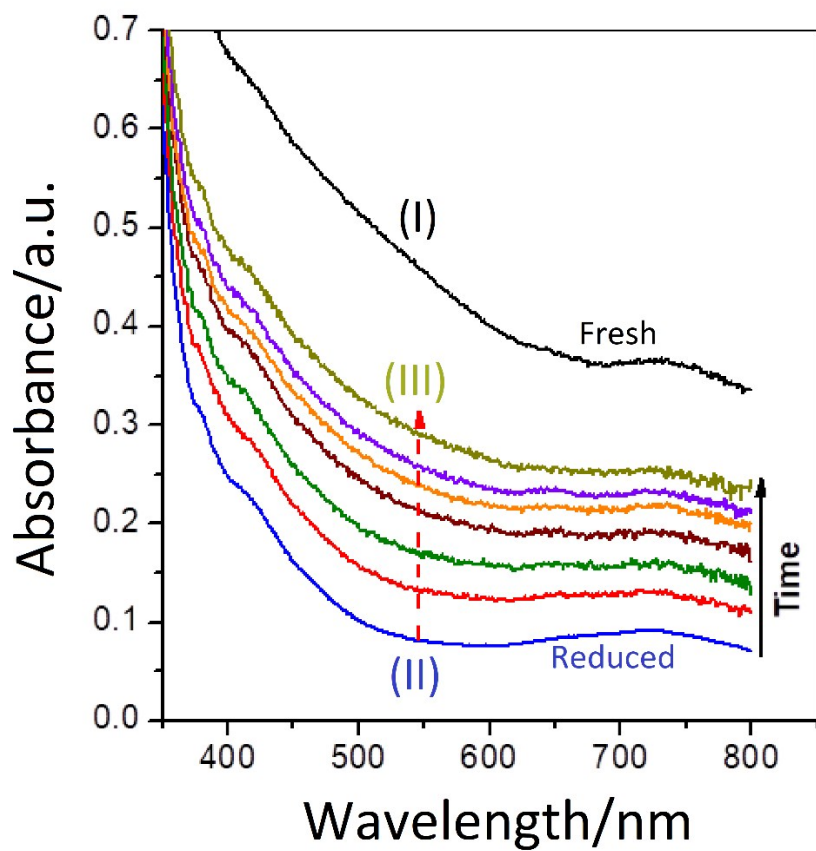


Figure S6. UV-Vis spectra of m-NiO calcined at 300 °C, (I) fresh sample, (II) the same sample treated with NaBH₄ solution and (II) to (III) washed and aged under ambient condition for bottom to top 5, 10, 25, 45, 60, and 300 min.

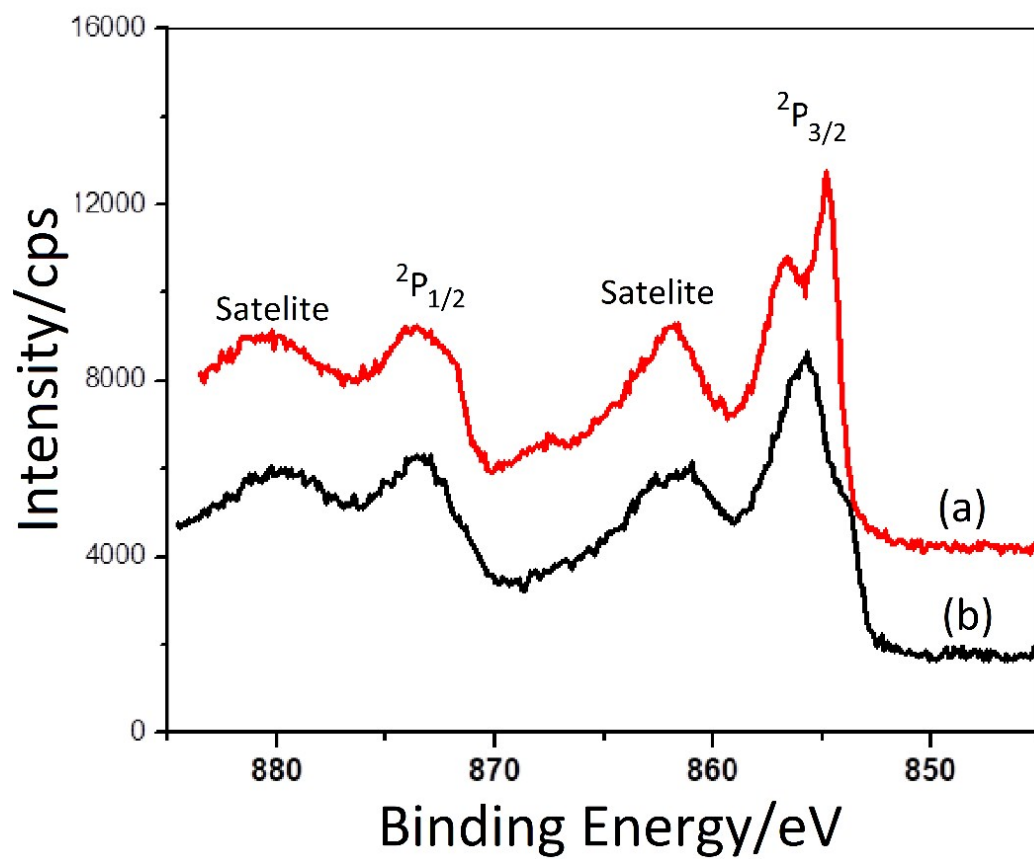


Figure S7. XPS spectra of m-NiO before and after CV (1000 cycle).

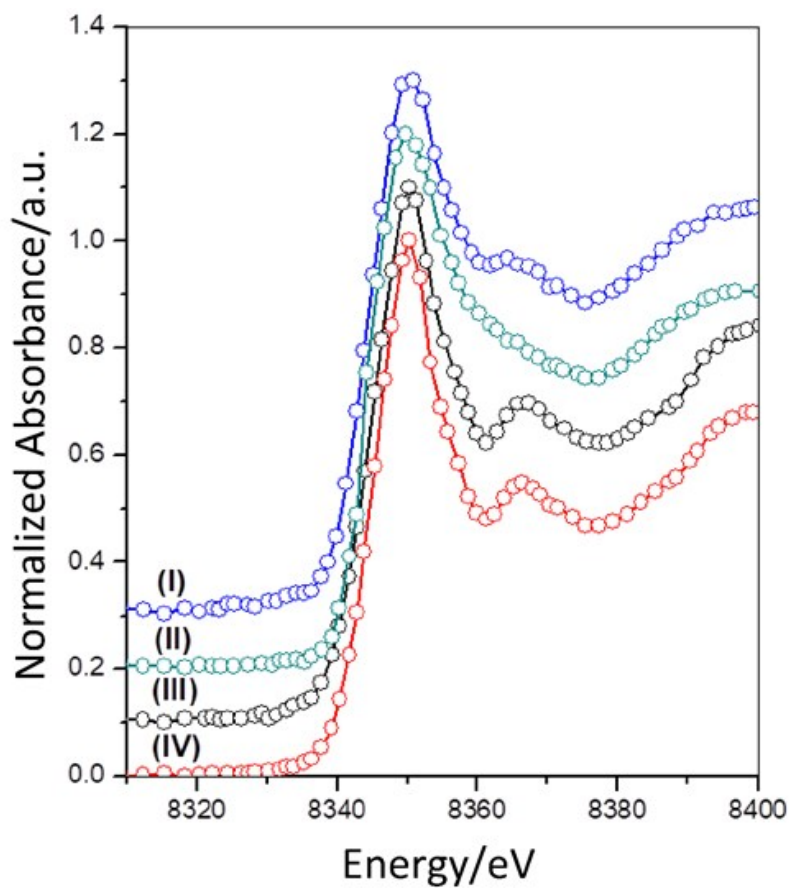


Figure S8. Ni K-edge XANES spectra of (I) *m*-Ni(OH)₂, (II) *m*-NiO-10-300, (III) *m*-NiO-10-400. (IV) *m*-NiO-10-400 after 1000 CV cycles.

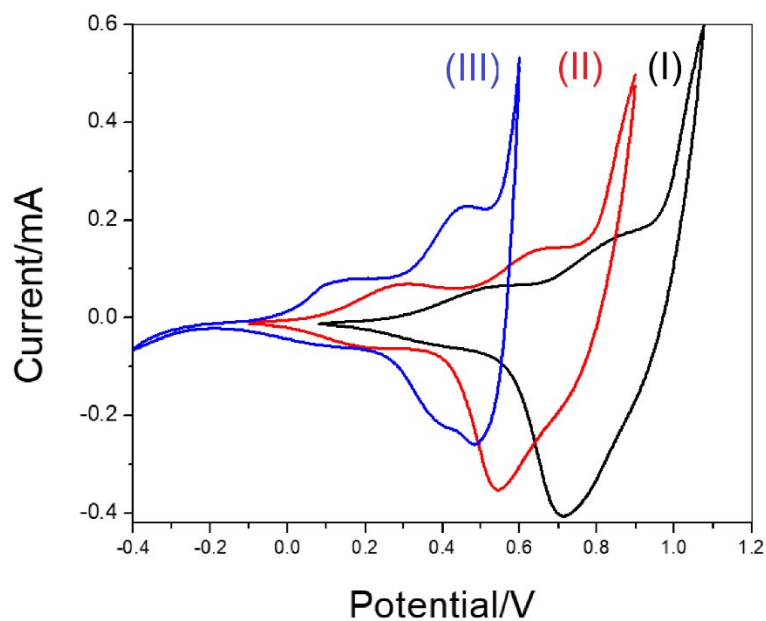


Figure S9. CVs of *m*-NiO-10-300 at 3 different pH values of (I) 7, (II) 10.2, and (III) 12.85.

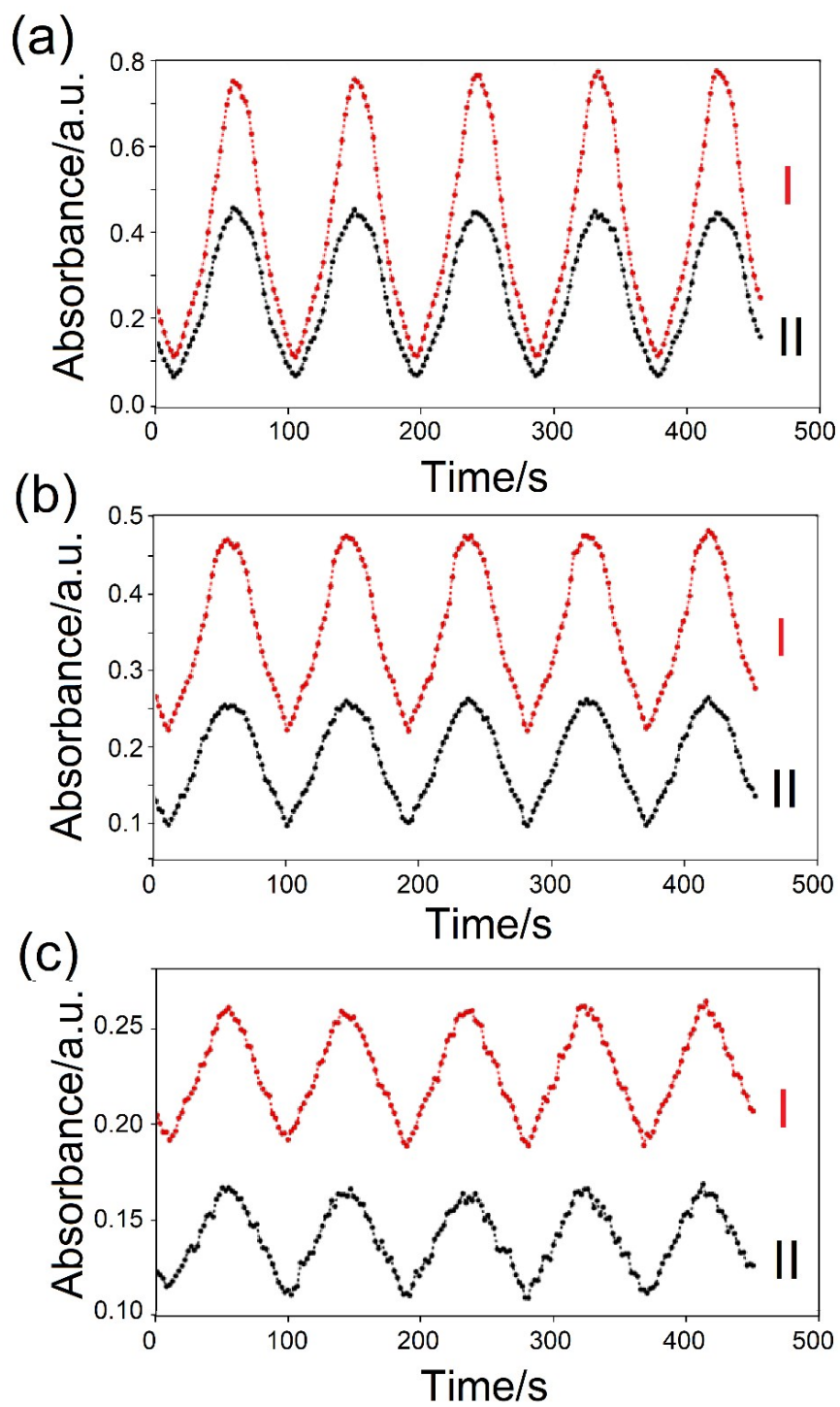


Figure S10. Electrochromic (absorbance vs time plot) behavior of *m*-NiO-10, calcined at (a) 300, (b) 400, and (c) 500 °C. Absorbance values: (I) 585 nm and (II) 886 nm.

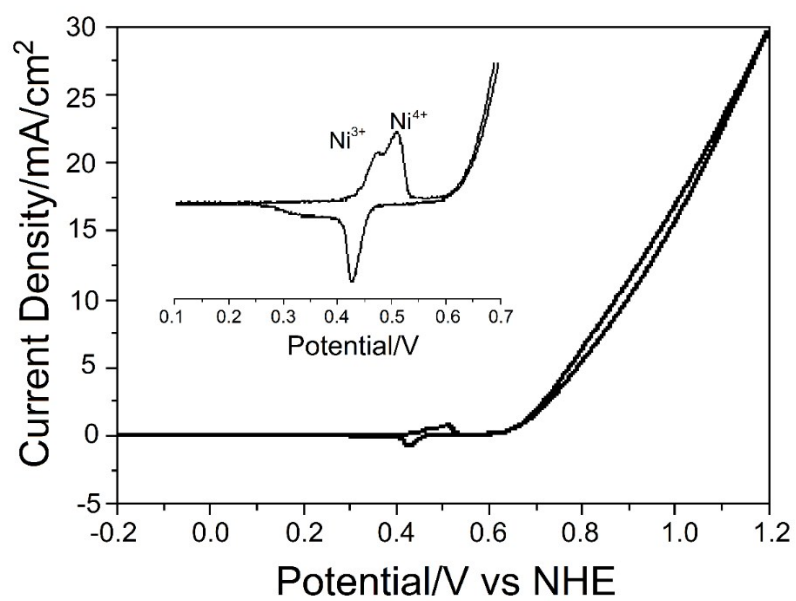


Figure S11. CV of m-NiO-10-400 with a scan rate of 1 mV/S.

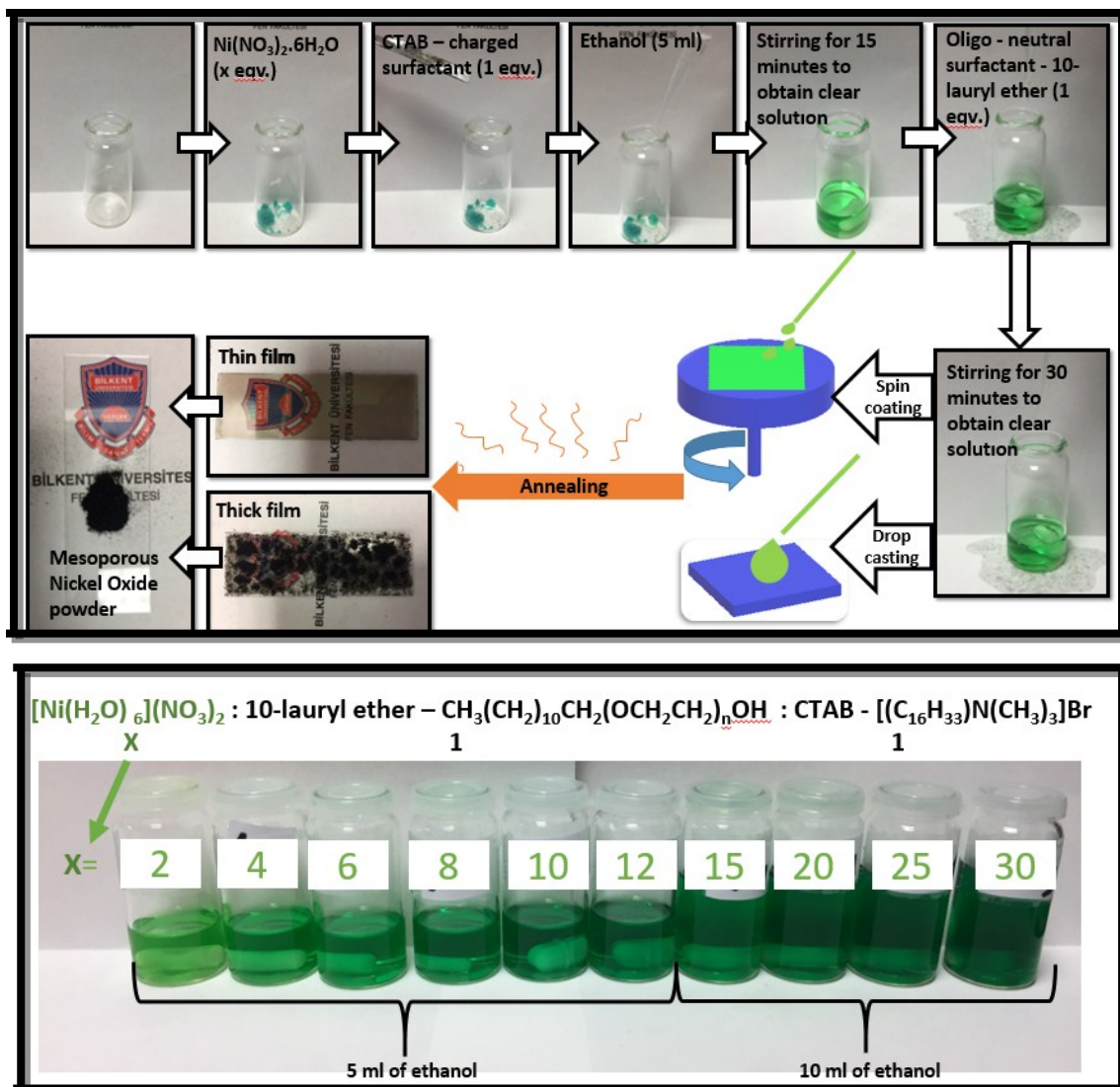


Figure S12. Schematic representation of the procedure (top) and photographs of the clear solutions at various concentrations (bottom).