

Supporting Information

Zinc niobate materials: crystal structures, energy-storage capabilities and working mechanisms

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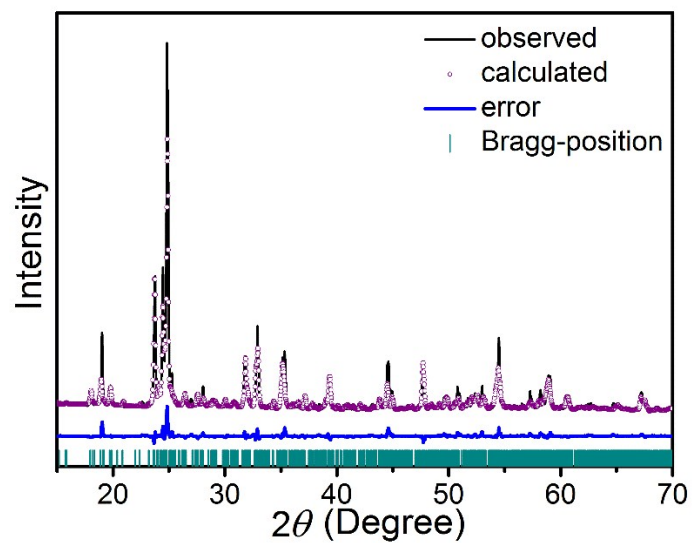


Fig. S1. XRD spectrum with Rietveld refinement of $W_5Nb_{16}O_{55}$.

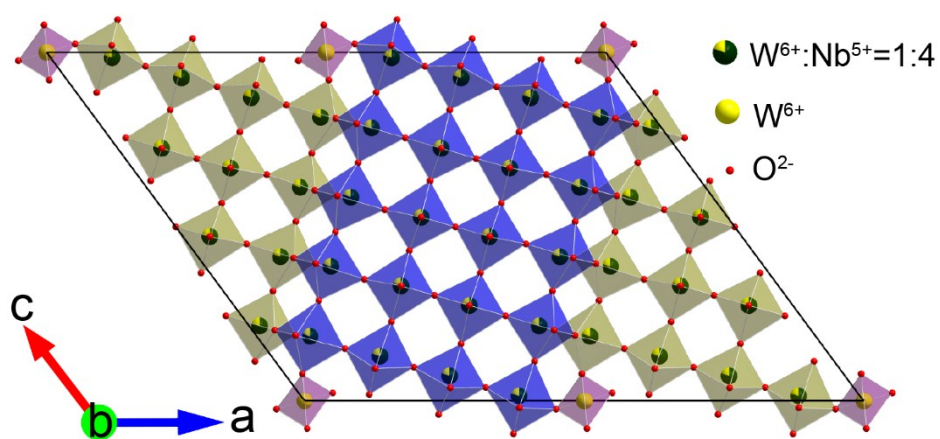


Fig. S2. Crystal structure of $W_5Nb_{16}O_{55}$.

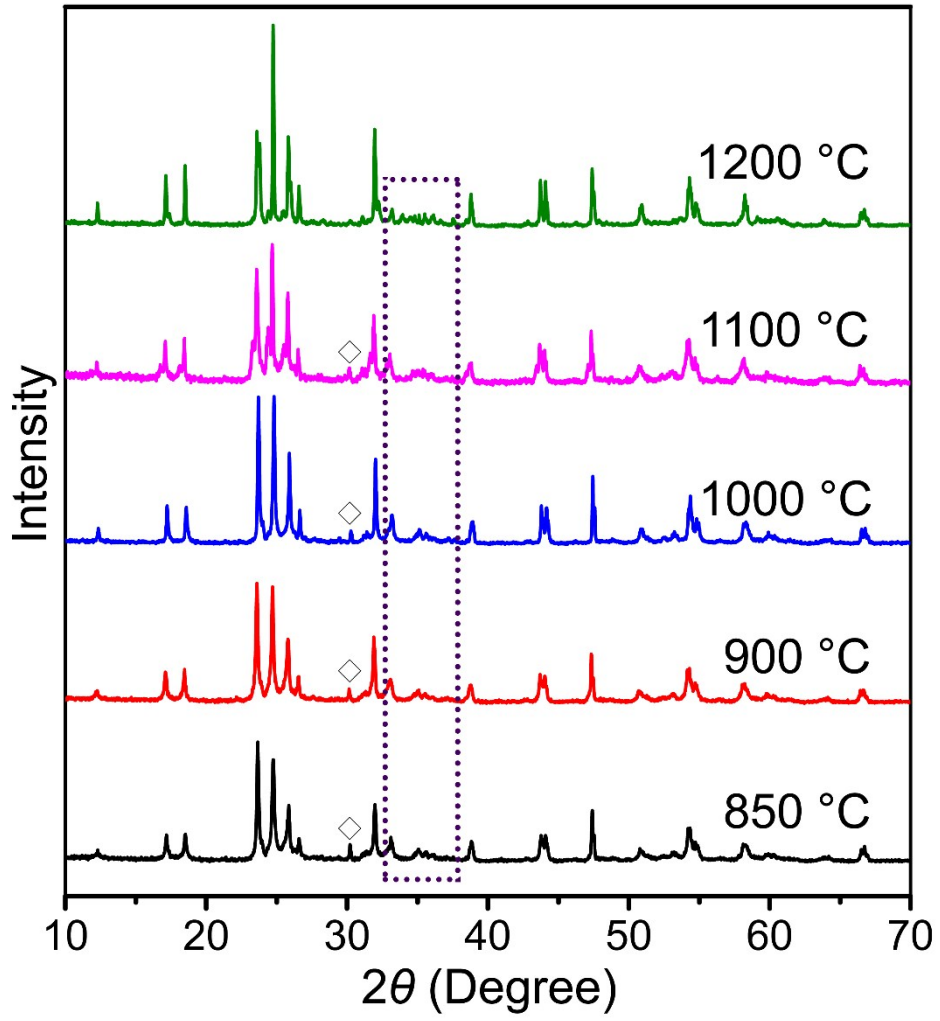


Fig. S3. XRD spectra of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ sintered at 850, 900, 1000, 1100 and 1200 °C

(◇: ZnNb_2O_6).

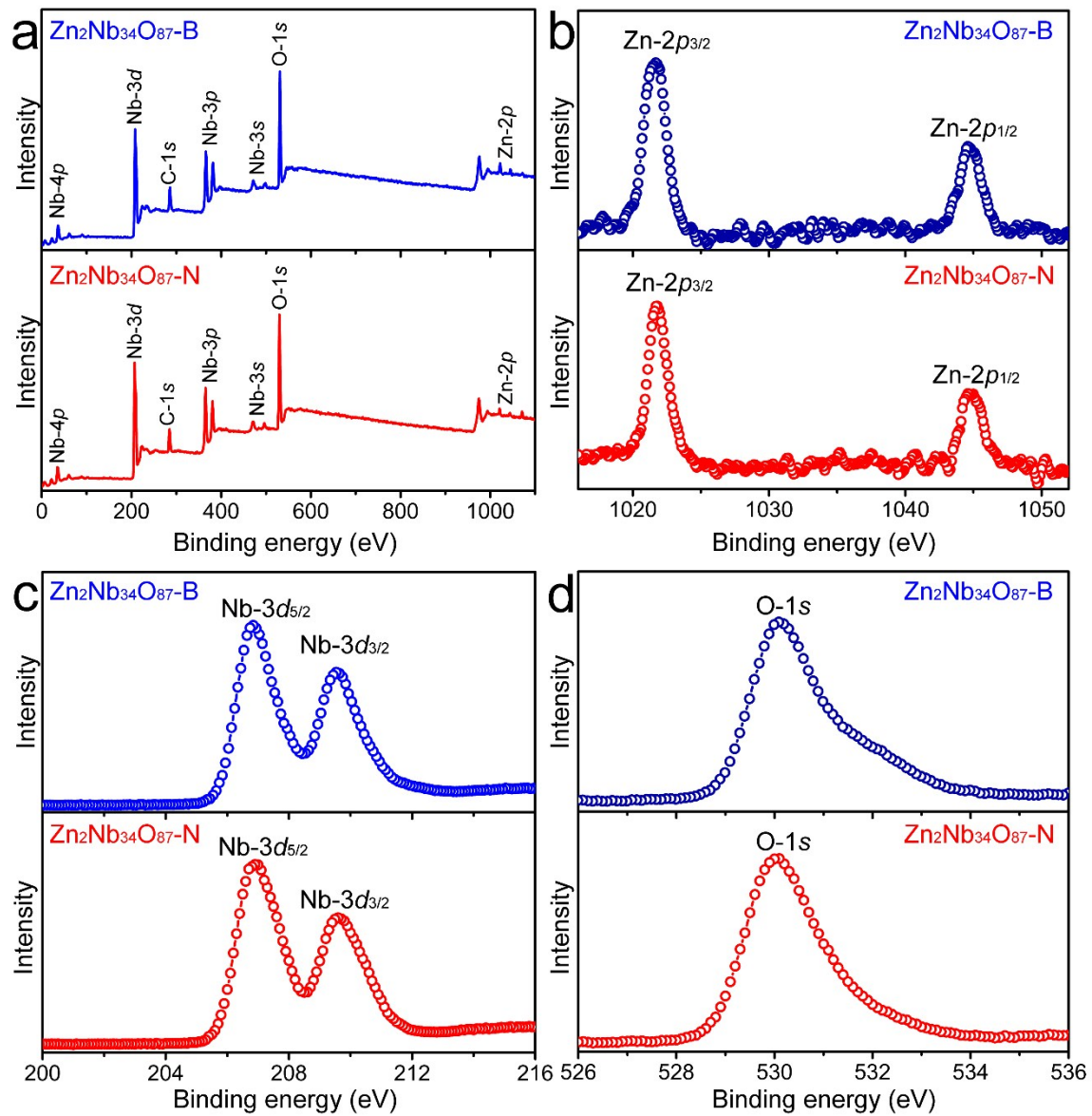


Fig. S4. (a) XPS survey spectra of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$. XPS spectra of (b) Zn, (c) Nb and (d) O elements in $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$.

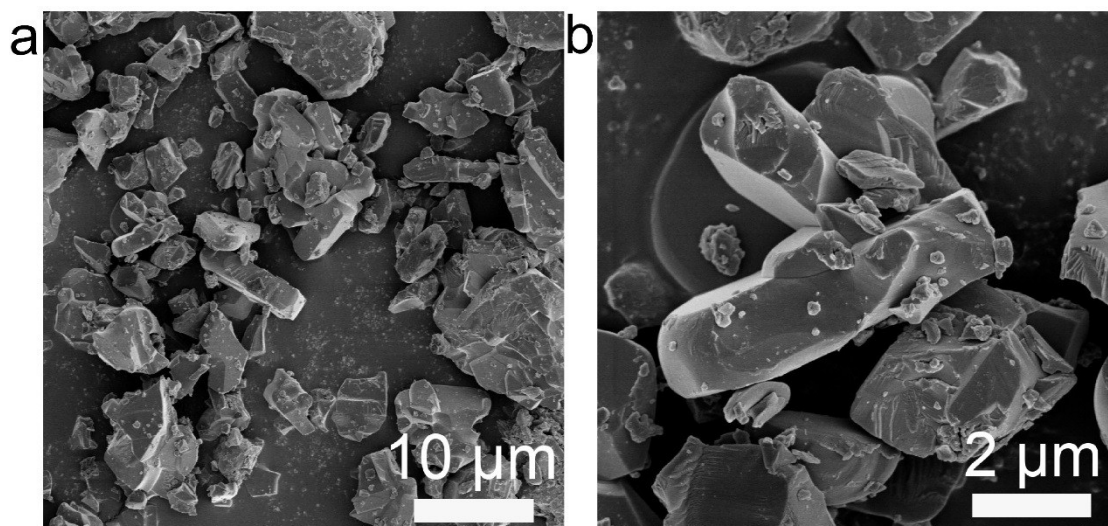


Fig. S5. (a, b) FESEM images of $Zn_2Nb_{34}O_{87}$ -B.

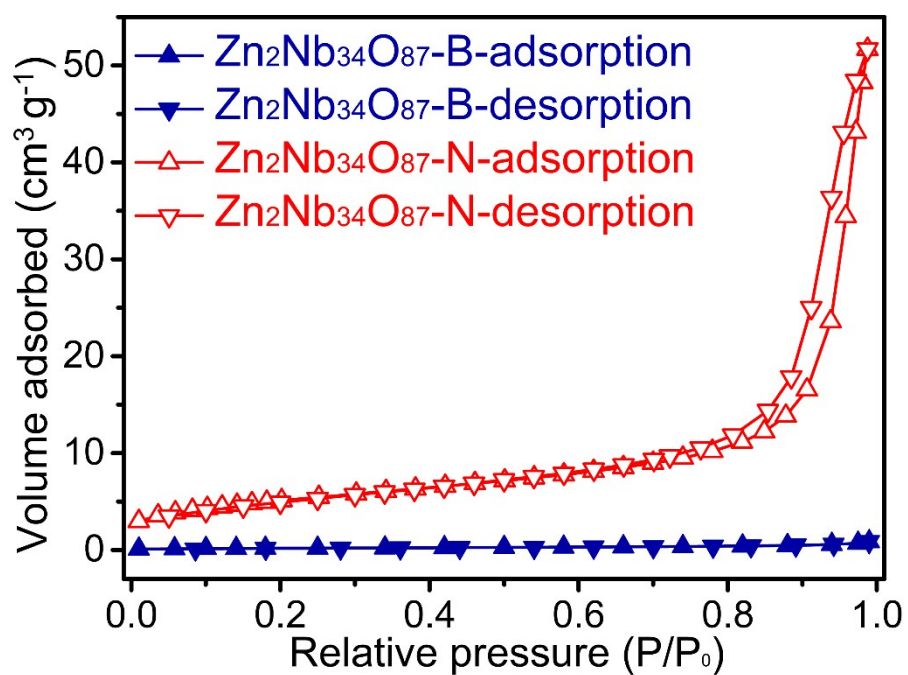


Fig. S6. N_2 adsorption–desorption isotherms of $Zn_2Nb_{34}O_{87}$ -B and $Zn_2Nb_{34}O_{87}$ -N.

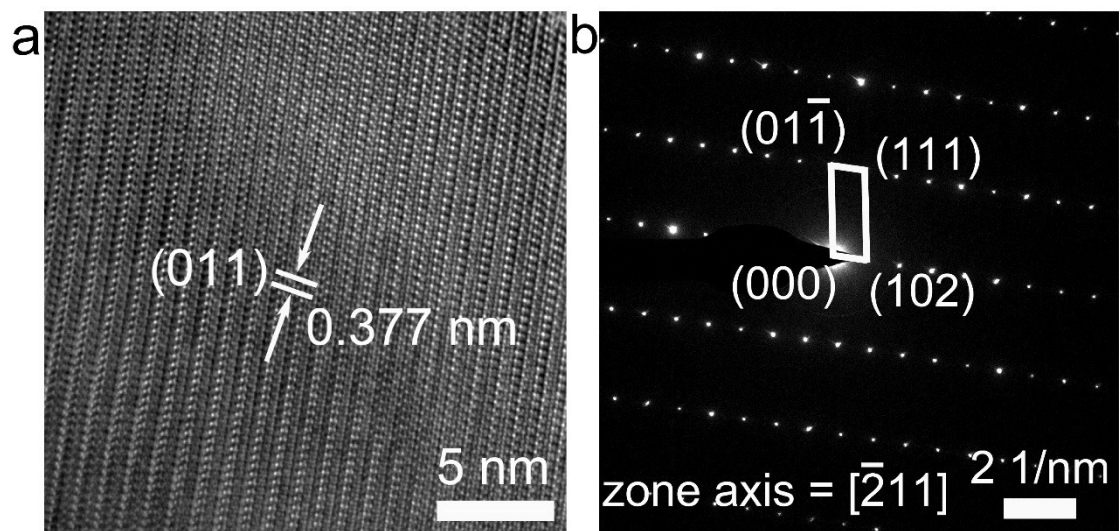


Fig. S7. (a) HRTEM image and (b) SAED pattern of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$.

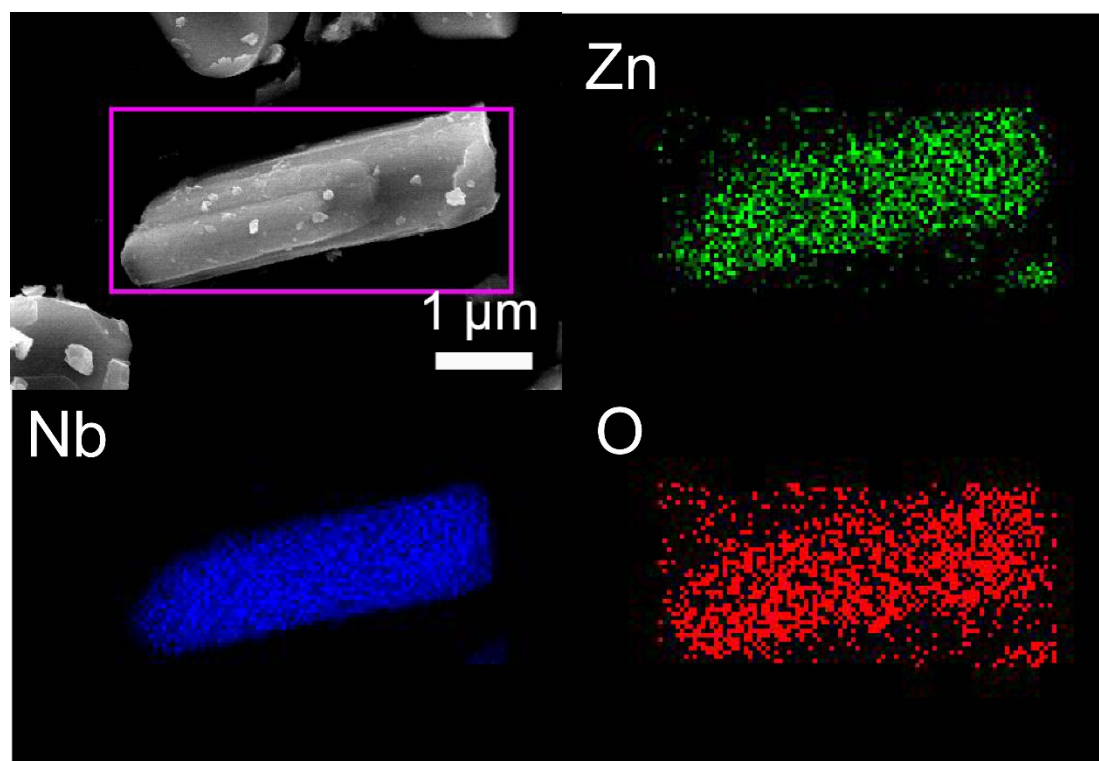


Fig. S8. EDX mapping images of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$.

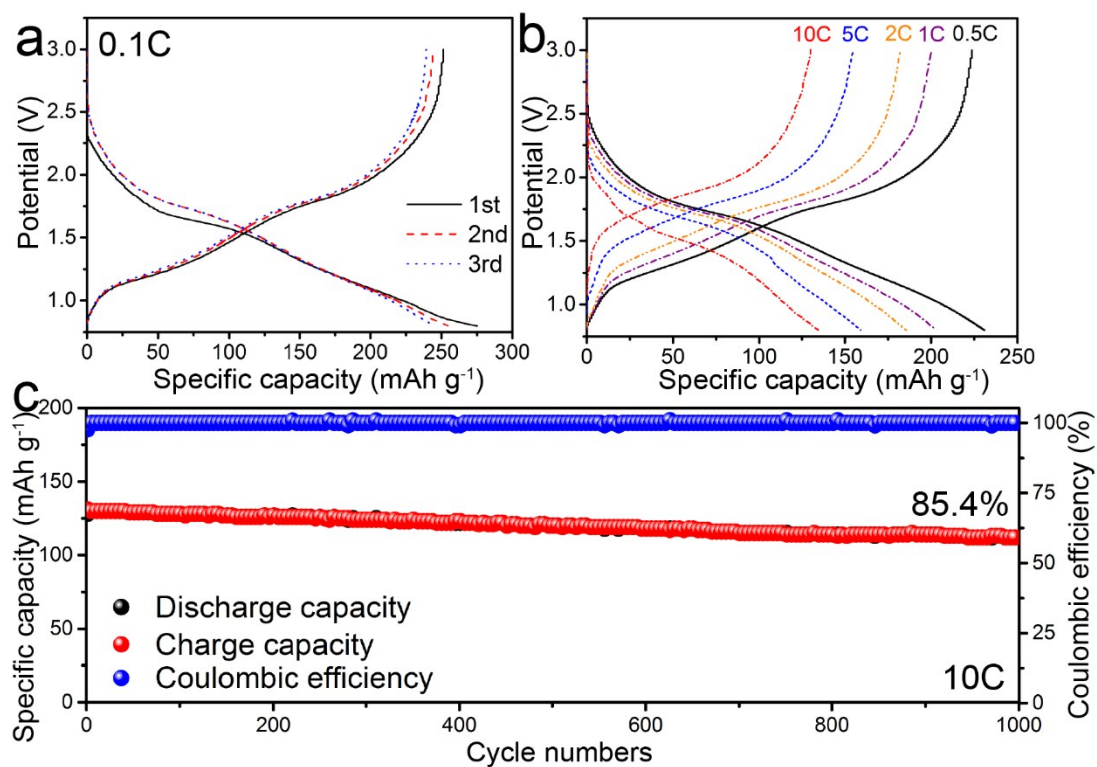


Fig. S9. Electrochemical performance of $W_5Nb_{16}O_{55}/Li$ cell (the fabrication process of the $W_5Nb_{16}O_{55}/Li$ cell is the same as that of the $Zn_2Nb_{34}O_{87}/Li$ cell): discharge–charge curves at (a) 0.1C and (b) different current rates, and (c) cyclability at 10C over 1000 cycles.

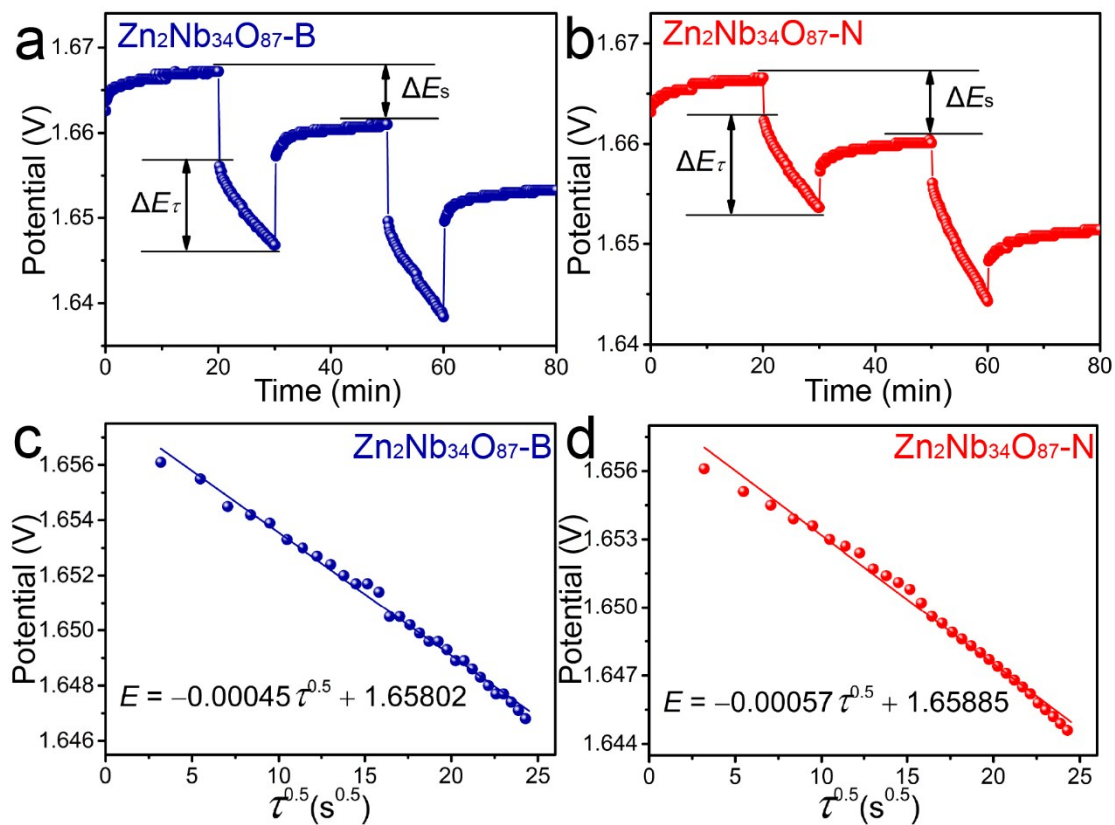


Fig. S10. E versus T curves for a single step in GITT experiment of (a) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and (b) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$. Linear behavior of E versus $\tau^{0.5}$ relationship during a typical titration in (c) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and (d) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$.

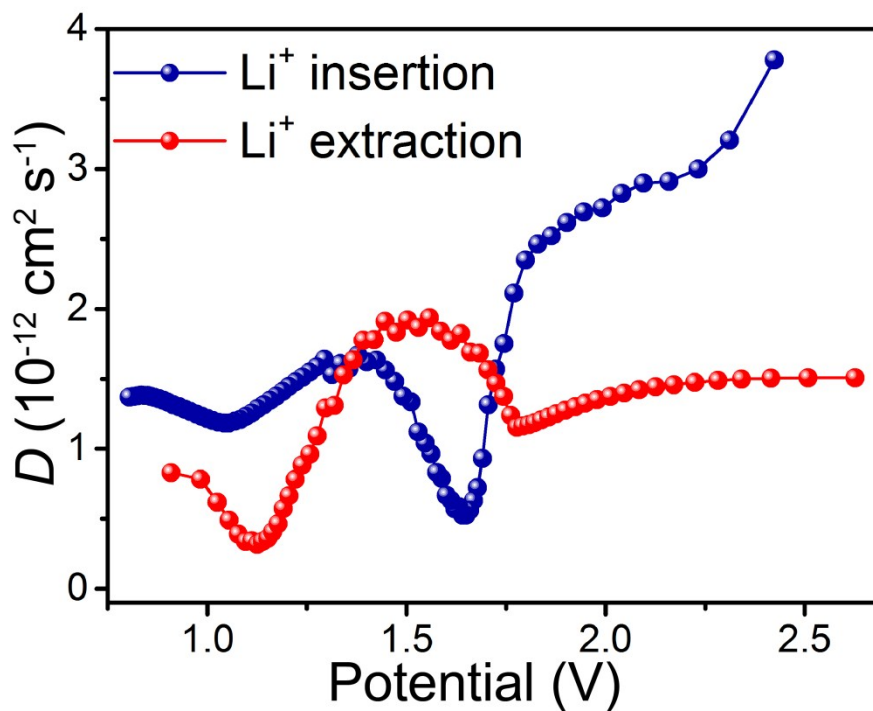


Fig. S11. Variation in Li^+ diffusion coefficient of $\text{W}_5\text{Nb}_{16}\text{O}_{55}$ calculated from GITT.

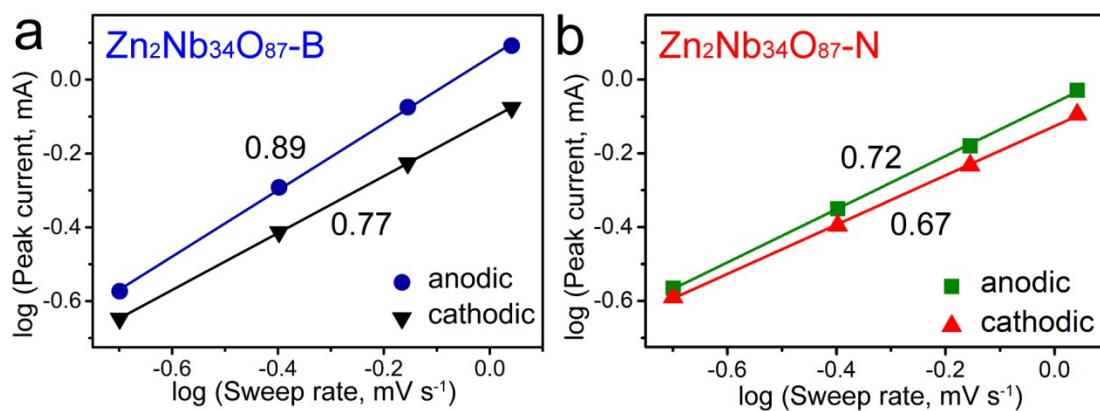


Fig. S12. Determination of b -values of (a) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and (b) $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ using relationship between logarithm of peak current and logarithm of sweep rate.

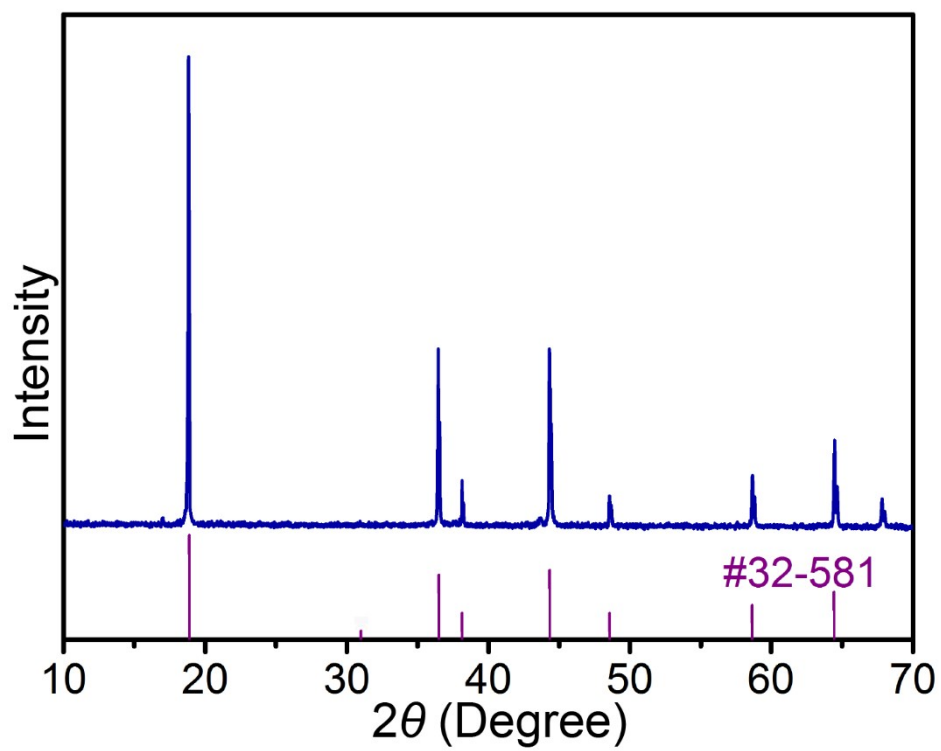


Fig. S13. XRD spectrum of $\text{LiNi}_{0.5}\text{Mn}_{1.5}\text{O}_4$.

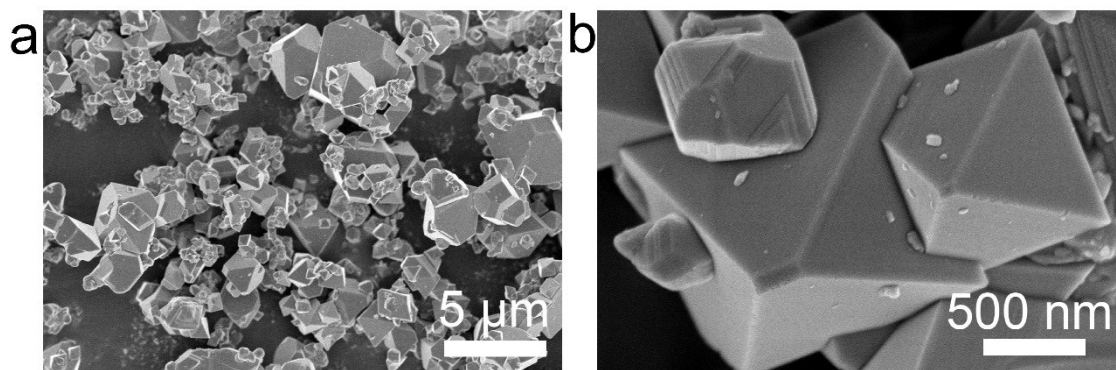


Fig. S14. (a) low-magnification and (b) high-magnification FESEM images of



Table S1. Results of crystal analyses by Rietveld refinements in $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ (orthorhombic) with *Amma* space group, $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ (monoclinic) with *A2/m* space group, and $\text{W}_5\text{Nb}_{16}\text{O}_{55}$ (monoclinic) with space group of *C2*.

material	<i>a</i> (nm)	<i>b</i> (nm)	<i>c</i> (nm)	α, γ (°)	β (°)	<i>V</i> (nm ³)
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$	2.871489 (11)	0.382780 (2)	2.065497 (8)	90	90	2.270295 (252)
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$	1.561179 (13)	0.383217 (2)	2.066574 (14)	90	113.089 (6)	1.137327 (177)
$\text{W}_5\text{Nb}_{16}\text{O}_{55}$	2.970832 (41)	0.381905 (4)	2.314088 (39)	90	126.546 (6)	2.109270 (547)

Table S2. Fractional atomic parameters of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ with space group of *Amma*.

atom	site	<i>x</i>	<i>y</i>	<i>z</i>
Zn1	8 <i>f</i>	0.046206	0	0.040711
Nb1	8 <i>f</i>	0.046206	0	0.040711
Nb2	8 <i>f</i>	0.050236	0	0.666035
Nb3	8 <i>f</i>	0.046842	0	0.856061
Nb4	8 <i>f</i>	0.183754	0	0.037185
Nb5	8 <i>f</i>	0.183215	0	0.668334
Nb6	8 <i>f</i>	0.181841	0	0.854427
O1	4 <i>c</i>	0.250000	0	0.031265
O2	4 <i>c</i>	0.250000	0	0.679758
O3	4 <i>c</i>	0.250000	0	0.853859
O4	8 <i>f</i>	0.056530	0	0.541091
O5	8 <i>f</i>	0.035402	0	0.144409
O6	8 <i>f</i>	0.036660	0	0.753260
O7	8 <i>f</i>	0.039926	0	0.334375
O8	8 <i>f</i>	0.024082	0	0.957767
O9	8 <i>f</i>	0.114317	0	0.033460
O10	8 <i>f</i>	0.111637	0	0.663379
O11	8 <i>f</i>	0.109285	0	0.849359
O12	8 <i>f</i>	0.188744	0	0.573336
O13	8 <i>f</i>	0.180203	0	0.143517
O14	8 <i>f</i>	0.173937	0	0.759216
O15	8 <i>f</i>	0.183017	0	0.349929
O16	8 <i>f</i>	0.175322	0	0.953713

Table S3. Fractional atomic parameters of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ with space group of $A2/m$.

atom*	site	x	y	z
M1	$4i$	0.094900	0	0.068931
M2	$4i$	0.093568	0	0.688770
M3	$4i$	0.095237	0	0.887346
M4	$4i$	0.374830	0	0.148083
M5	$4i$	0.364012	0	0.778313
M6	$4i$	0.356387	0	0.955579
O1	$2d$	0.500000	0	0
O2	$4i$	0.070218	0	0.166963
O3	$4i$	0.072770	0	0.354527
O4	$4i$	0.084555	0	0.582856
O5	$4i$	0.082229	0	0.784255
O6	$4i$	0.131786	0	0.992904
O7	$4i$	0.214648	0	0.099463
O8	$4i$	0.221544	0	0.730309
O9	$4i$	0.209007	0	0.920269
O10	$4i$	0.343510	0	0.062637
O11	$4i$	0.350924	0	0.253865
O12	$4i$	0.368872	0	0.387533
O13	$4i$	0.370454	0	0.664701
O14	$4i$	0.359728	0	0.860083
O15	$4i$	0.495672	0	0.192593

*M = $1/18 \text{Zn}^{2+} + 17/18 \text{Nb}^{5+}$

Table S4. Comparisons of reversible/theoretical capacity of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B/}$
 $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ with previously reported insertion negative electrode materials.

material	theoretical capacity (mAh g ⁻¹)	capacity in reports (mAh g ⁻¹)	reference
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$	389	310	this work
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$	389	284	this work
graphite	372	310	[S1]
$\text{Li}_4\text{Ti}_5\text{O}_{12}$	175	169	[S2]
TiNb_2O_7	388	281	[S3]
$\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$	396	247	[S4]
$\text{TiNb}_{24}\text{O}_{62}$	402	296	[S5]
Nb_2O_5	403	210	[S6]
$\text{FeNb}_{11}\text{O}_{29}$	400	270	[S7]
$\text{GaNb}_{11}\text{O}_{29}$	379	264	[S8]
$\text{GeNb}_{18}\text{O}_{47}$	386	217	[S9]
$\text{PNb}_9\text{O}_{25}$	381	200	[S10]
$\text{VNb}_9\text{O}_{25}$	416	220	[S11]
$\text{WNb}_{12}\text{O}_{33}$	302	228	[S12]
$\text{W}_5\text{Nb}_{16}\text{O}_{55}$	302	225	[S13]
$\text{BaNb}_{3.6}\text{O}_{10}$	306	264	[S14]
$\text{K}_2\text{Nb}_8\text{O}_{21}$	371	281	[S15]
$\text{W}_9\text{Nb}_8\text{O}_{47}$	289	238	[S16]
$\text{W}_{16}\text{Nb}_{18}\text{O}_{93}$	228	195	[S17]

Table S5. Comparisons of electrochemical performance of $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$ and $\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$ with previously reported niobium-based oxide negative electrode materials.

material	current rate	rate performance (mAh g ⁻¹)	reference
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-N}$	10C	~197 at 1000 th cycles	this work
$\text{Zn}_2\text{Nb}_{34}\text{O}_{87}\text{-B}$	10C	~140 at 1000 th cycles	this work
$\text{W}_5\text{Nb}_{16}\text{O}_{55}$ micron-sized particles	10C	~111 at 1000 th cycles	Fig. S9c
TiNb_2O_7 nanoparticles	10C	~123 at 500 th cycles	[S18]
TiNb_2O_7 nanofibers	5C	~170 at 500 th cycles	[S19]
TiNb_2O_7 nanorods	10C	~140 at 100 th cycles	[S20]
three-dimensional (3D) ordered macroporous TiNb_2O_7	10C	~87 at 100 th cycles	[S21]
$\text{Cu}_{0.02}\text{Ti}_{0.94}\text{Nb}_{2.04}\text{O}_7$	10C	~180 at 1000 th cycles	[S22]
$\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ hollow nanofibers	10C	~123 at 500 th cycles	[S23]
porous $\text{Ti}_2\text{Nb}_{10}\text{O}_{29}$ nanospheres	10C	~141 at 1000 th cycles	[S24]
porous $\text{TiNb}_{24}\text{O}_{62}$ microspheres	10C	~183 at 500 th cycles	[S5]
$\text{WNb}_{12}\text{O}_{33}$ nanowires	3C	~140 at 700 th cycles	[S12]
$\text{GeNb}_{18}\text{O}_{47}$ nanofibers	2C	~162 at 200 th cycles	[S9]
$\text{VNb}_9\text{O}_{25}$ nanoribbons	3C	~132 at 500 th cycles	[S11]
$\text{W}_9\text{Nb}_8\text{O}_{47}$ nanofibers	5C	~113 at 1000 th cycles	[S16]
$\text{GaNb}_{11}\text{O}_{29}$ nanowebs	10C	~153 at 1000 th cycles	[S8]

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