Supporting Information

Highly Efficient Planar Perovskite Solar Cells via Acid-assisted Surface Passivation

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Fig. S1 XPS spectra of (a) C 1s, (b) Sn 3d and (c) O 1s with marked specific peaks position.





Fig. S2 (a) Photographs of SnO_2 precursor solutions without (Left) and with (Right) adding acetic acid. (b) Chemical reaction of SnO_2 and acetic acid.



Fig. S3 Transmission spectra of pristine and Ac-modified SnO₂ films.



Fig. S4 The grain size distribution of perovskite films deposited on pristine and Acmodified SnO_2 substrates.



Fig. S5 Device structure of planar-type PSCs. For triple cation PSC, the active layer is $Cs_{0.05}(MA_{0.13}FA_{0.87})_{0.95}Pb(I_{0.87}Br_{0.13})_3$ and for dual cation PSC, the active layer is $(FAPbI_3)_{1-x}(MAPbBr_3)_x)$.

Table. S1 Photovoltaic parameters of triple cation PSCs based on different SnO₂/ITO substrates, the device architecture in this study is the regular planar structure of ITO/SnO₂/Perovskite/Spiro-OMeTAD/Au.

Sam	ples	$V_{oc}(\mathbf{V})$	$J_{sc}(\mathrm{mA}\cdot\mathrm{cm}^{-2})$	FF	PCE (%)
Pristine	Average	1.137 ± 0.004	21.83 ± 0.26	0.749 ± 0.006	18.57 ± 0.17
	Best	1.141	22.07	0.748	18.84
2.5%	Average	1.143 ±0.007	21.86 ± 0.11	0.781 ± 0.006	19.52 ± 0.15
	Best	1.141	21.92	0.787	19.68
5%	Average	1.153 ± 0.005	22.31 ± 0.09	0.791 ± 0.003	20.33 ± 0.17
	Best	1.158	22.39	0.793	20.56
7.5%	Average	1.145 ± 0.006	21.31 ± 0.25	0.779 ± 0.004	19.02 ± 0.18
	Best	1.145	21.36	0.784	19.17



Fig. S6 (a) Comparison of photovoltaic parameters of the dual cation PSCs based on pristine and Ac-modified SnO_2 . (b) The *J-V* curves of different devices with the highest FF (Device A is triple cation PSC and Device B is dual cation perovskite).

Sam	ples	$V_{oc}(\mathbf{V})$	$J_{sc} (\mathrm{mA} \cdot \mathrm{cm}^{-2})$	FF	PCE (%)
Pristine	Average	1.103 ± 0.009	23.23 ± 0.13	0.765 ± 0.011	19.52 ± 0.25
	Best	1.117	23.13	0.771	19.92
Ac- modified	Average	1.137 ±0.006	23.83 ± 0.15	0.798 ± 0.007	21.64 ± 0.15
	Best	1.139	23.68	0.814	21.95

Table. S2 Photovoltaic parameters of dual cation PSCs based on different SnO₂ with device architecture of ITO/SnO₂/Perovskite/Spiro-OMeTAD/Au.



Fig. S7 The J-V curves of the devices with pristine and Ac-modified SnO₂ ETL measured under reverse and forward scanning directions.

Sample	τ_1 (ns)	τ_2 (ns)
Pristine	4.83	22.15
Ac-modified	4.31	21.17
Glass	23.5	167.7

Table. S3 Parameters of the time-resolved PL spectra of triple cation perovskite based on pristine SnO_2 , Ac-modified SnO_2 and glass substrates.



Fig. S8 Time-resolved spectrum of glass/perovskite device with larger timescale.



Fig. S9 Nyquist plots of PSCs based on pristine and Ac-modified SnO_2 measured under constant light illumination of 100 mW cm² from 1 Hz to 1 MHz. The inset gives the equivalent circuit model, R_s is series resistance, R_{rec} is recombination resistance, R_{dr} is dielectric relaxation resistance, C_{con} is selective contact capacitance and C_{dr} is dielectric relaxation capacitance of the perovskite.

Sample	$R_{s}\left(\Omega ight)$	$R_{rec}\left(\Omega ight)$	$C_{con}\left(\mathrm{F}\right)$	$R_{dr}(\Omega)$	$C_{dr}\left(\mathrm{F}\right)$
Pristine	24.61	1208	9.23×10 ⁻⁹	112.80	1.22×10 ⁻⁵
Ac-modified	16.39	2068	3.51×10-9	88.39	8.40×10 ⁻⁶

Table. S4 EIS parameters of the pristine and Ac-modified PSCs obtained by Z-view software.



Fig. S10 (a) IPS spectrum of pristine SnO_2 . (b) IPS spectrum of Ac-modified SnO_2 . (c) Tauc plots for pristine and Ac-modified SnO_2 films. (d) Energy level diagram of the PSCs.