

Electronic supplementary information

A novel all-solid-state asymmetric supercapacitor with superior electrochemical performance in a wide temperature range by using hydroquinone modified graphene xerogel as cathode and N-doped $\text{Ti}_3\text{C}_2\text{T}_x$ as anode

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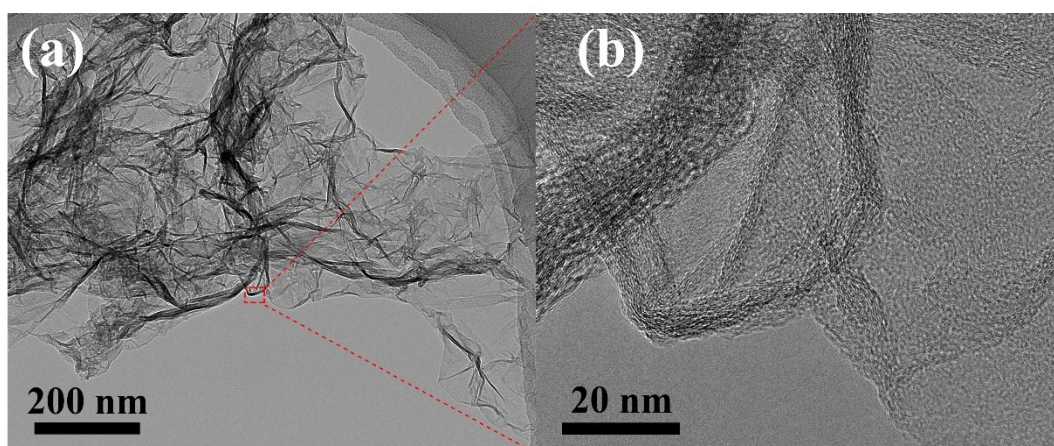


Figure S1 (a) TEM image and (b) HRTEM image of the HG2.

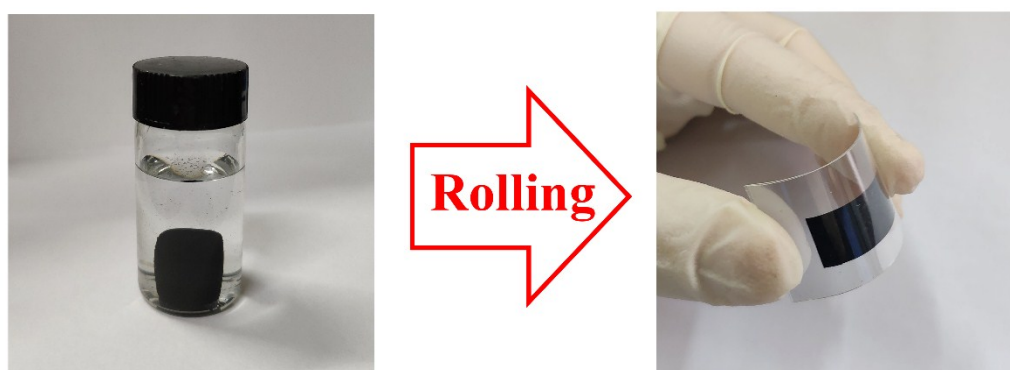


Figure S2: Prepared graphene hydrogel and graphene electrode prepared by rolling.

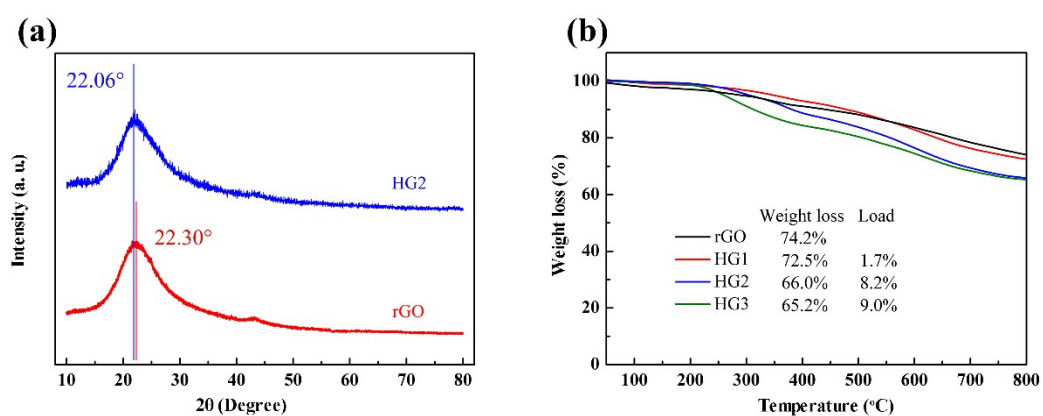


Figure S3 (a) XRD patterns of rGO and HG2. (b) TGA curves of rGO and HG2 samples in Ar at a heating rate of $5\text{ }^{\circ}\text{C min}^{-1}$.

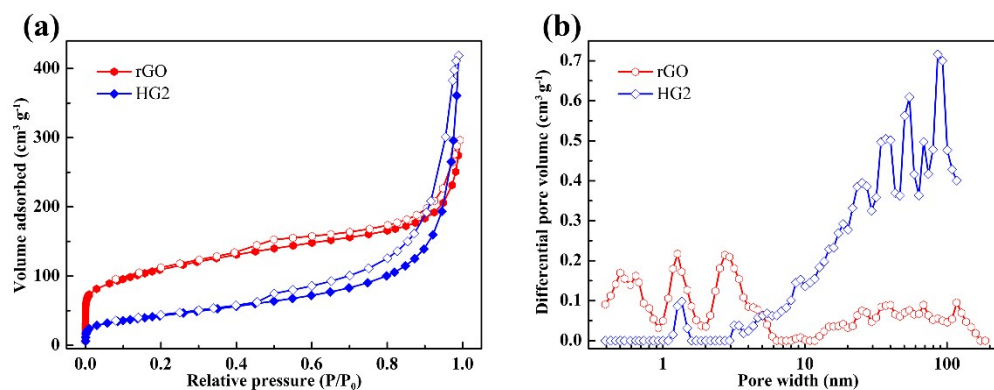


Figure S4 Nitrogen adsorption-desorption isotherm (a), and pore width distribution (b) for rGO and HG2.

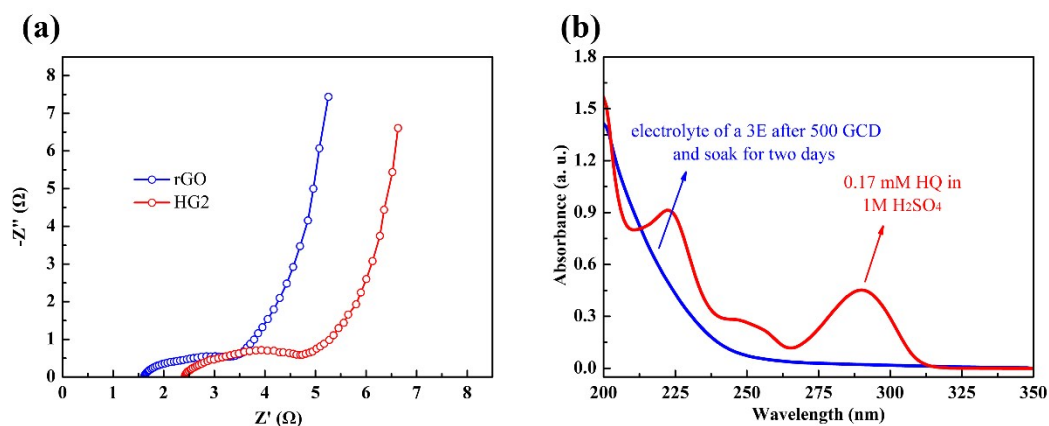


Figure S5: (a) Nyquist plots of rGO and HG2 electrodes at room temperature. (b) UV-vis absorption spectra of a 0.17 mM HQ in 1 M H₂SO₄ solution and the electrolyte which was after 500s CD at low current density of 2 A g⁻¹ and soak for two days in the three-electrode device.

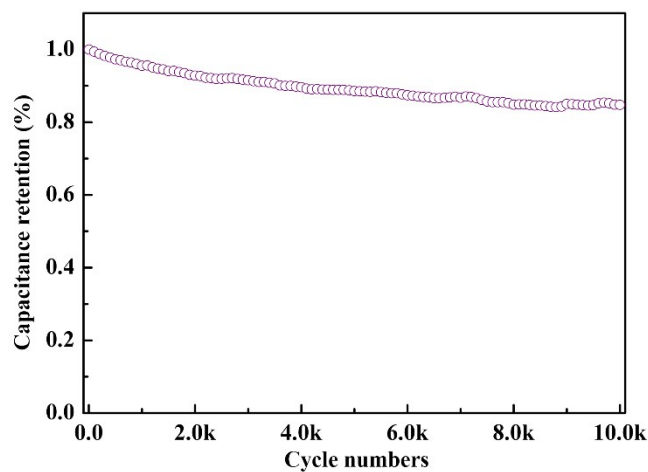


Figure S6 Cycling stability at 20 A g⁻¹ of the HG2 electrode at room temperature.

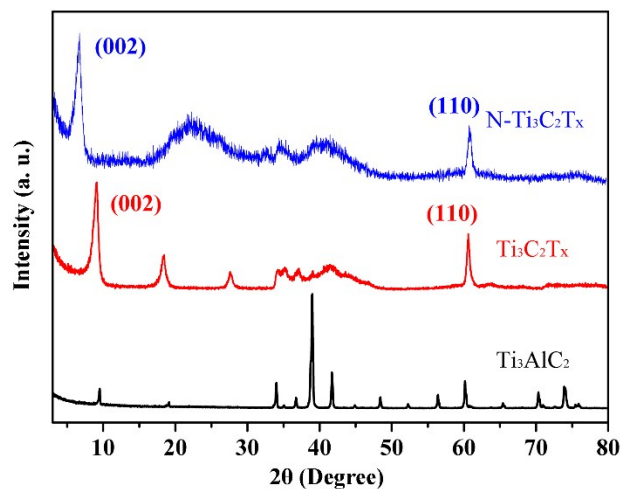


Figure S7 XRD patterns of Ti_3AlC_2 , $\text{Ti}_3\text{C}_2\text{T}_x$ and $\text{N-Ti}_3\text{C}_2\text{T}_x$.

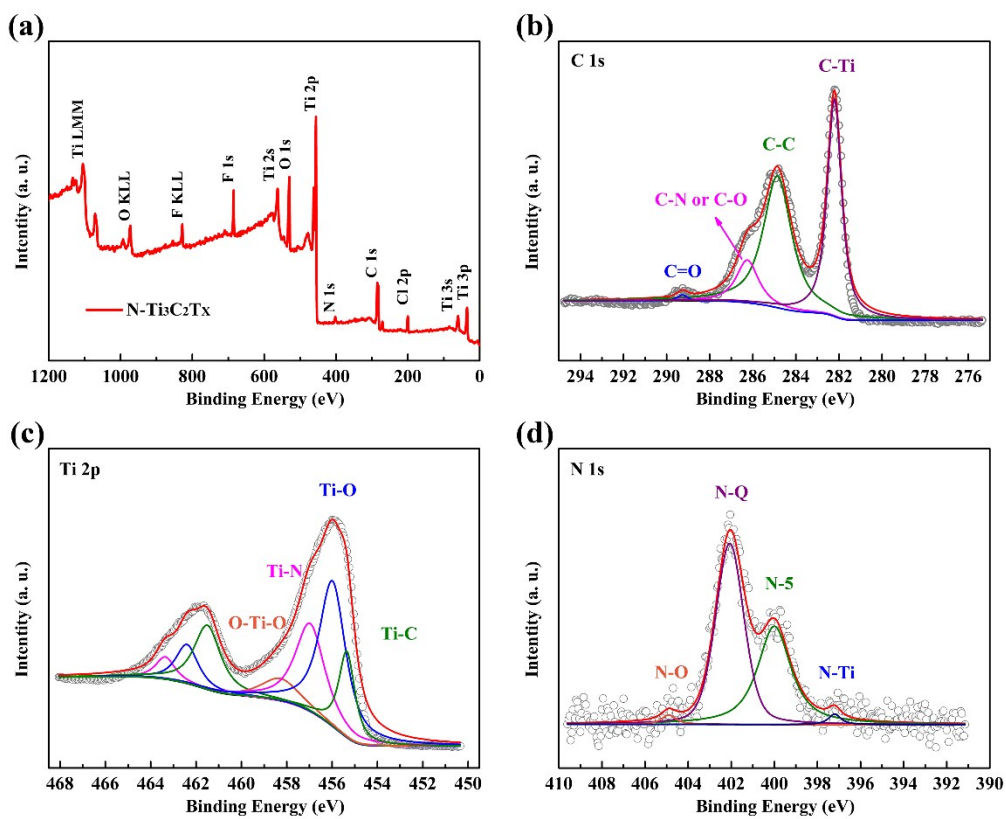


Figure S8 (a) XPS spectrum for $\text{N-Ti}_3\text{C}_2\text{T}_x$. (B) High-resolution XPS spectra of C 1s (b), Ti 2p (c) and N 1s (d) for $\text{N-Ti}_3\text{C}_2\text{T}_x$.

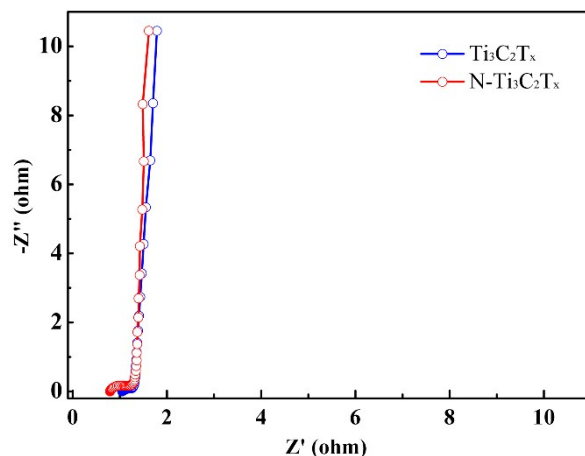


Figure S9 Nyquist plots of $\text{Ti}_3\text{C}_2\text{T}_x$ and $\text{N-Ti}_3\text{C}_2\text{T}_x$ at room temperature.

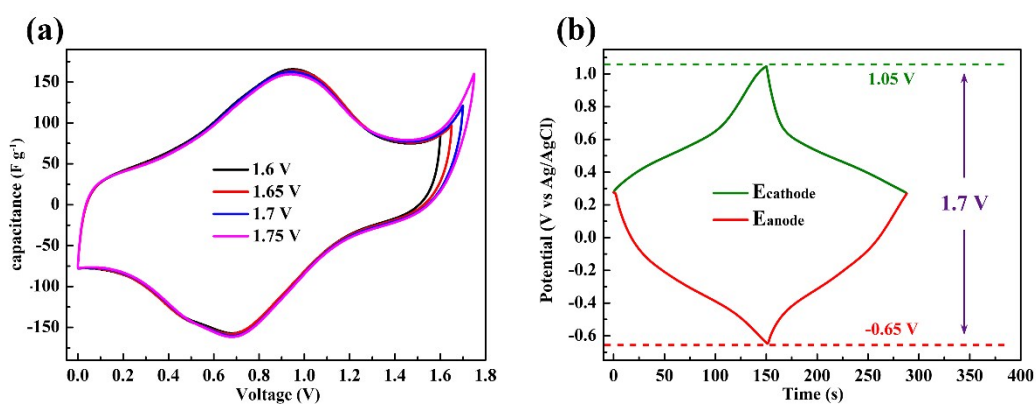


Figure S10 (a) CV curves under different voltage windows for asymmetric device. (b) In situ potential-time curves of cathode and anode at 1A g^{-1} at room temperature

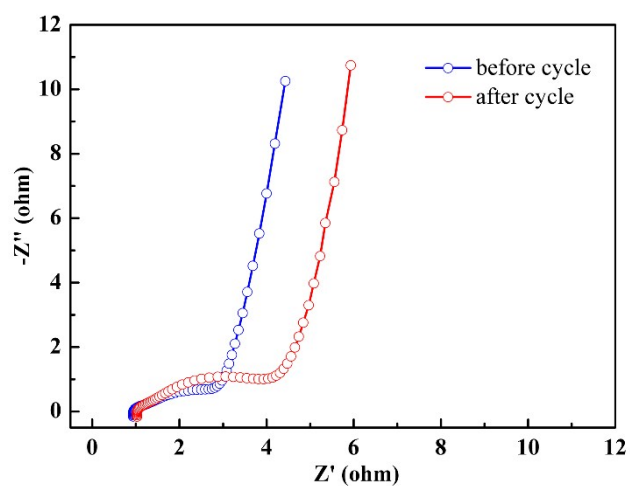


Figure S11 Nyquist plots for asymmetric device which is before and after the charge/discharge cycle at room temperature.

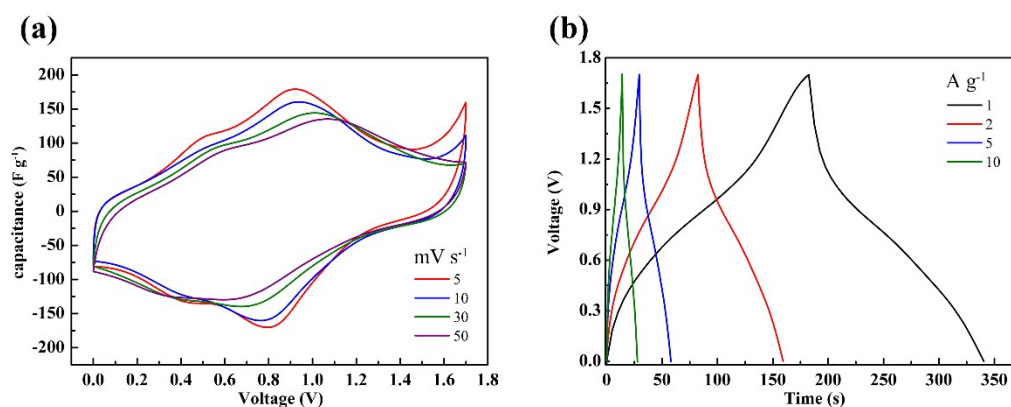


Figure S12 (a) CVs of the asymmetric device in gel electrolyte at different scan rate. (b) CDs of the asymmetric device in gel electrolyte at different current density at room temperature.

Table S1 Comparison of the electrochemical performance of supercapacitor recently reported at room temperature.

Materials	Electrolyte	Voltage window	Capacitance of cell (F g ⁻¹)	Current density	Energy density (Wh kg ⁻¹)	Power density (W kg ⁻¹)	Cycle numbers	Capacity retention	Ref
Ti ₃ C ₂ T _x /rGO	1 M H ₂ SO ₄	1.1 V	48	2 mV s ⁻¹	8	50	1000	76%	[1]
Ti ₃ C ₂ T _x @NC//Ti ₃ C ₂ T _x @NC	1 M H ₂ SO ₄	1.1 V	176.9	1 A g ⁻¹	29.7	582.3	5000	91.9%	[2]
G@MnO ₂ /porous graphene	1 M Na ₂ SO ₄	2 V	56	0.5 A g ⁻¹	30.6	197	10000	91.5%	[3]
Ti ₃ C ₂ T _x -Bi ₂ O ₃ //Ti ₃ C ₂ T _x -Bi ₂ O ₃	1 M KOH	1.2 V	76	0.5 A g ⁻¹	15.2	567	5000	85%	[4]
Ti ₃ C ₂ T _x //PEDOT@rGO	3 M H ₂ SO ₄	1.4 V	47	5 mV s ⁻¹	13	170	10000	80%	[5]
Ti ₃ C ₂ T _x //PPy@rGO	3 M H ₂ SO ₄	1.4 V	59	5 mV s ⁻¹	16	200	20000	75%	
Ti ₃ C ₂ T _x //PANI@rGO	3 M H ₂ SO ₄	1.45 V	57	5 mV s ⁻¹	17	200	20000	88.42%	
Ti ₃ C ₂ //TC-9	1 M KOH	1.5 V	49.3	1 A g ⁻¹	15.4	750.2	5000	82.4%	[6]
Ti ₃ C ₂ T _x //RuO ₂	1 M H ₂ SO ₄	1.5 V	93	5 mV s ⁻¹	29	3800	20000	86%	[7]
					24	26000			
PPy/rGO//NCs	3 M LiCl	1.6 V	43.2	20 mV s ⁻¹	15.8	140	10000	88.7%	[8]
MnCo ₂ S ₄ //rGO	3 M KOH	1.6 V	88	1 A g ⁻¹	31.3	800	5000	89%	[9]
400-KOH-Ti ₃ C ₂ //400-KOH-Ti ₃ C ₂	1 M H ₂ SO ₄	1.6 V	66.4	1 A g ⁻¹	23.6	300	5000	90.4%	[10]
Ti ₃ C ₂ T _x /rGO	3 M H ₂ SO ₄	1.8 V	76.5	1 A g ⁻¹	34.4	1000	10000	—	[11]
	+50 mM LiBr				23.7	22500			
N-Ti ₃ C ₂ T _x //HG2	1 M H ₂ SO ₄	1.7 V	102	0.5 A g ⁻¹	41	425	18000	83%	a
					26.7	42500			a

a This work

References

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