

Cu-Fe-Ni Nano Alloy Particles Obtained by Exsolution from Cu(Ni)Fe₂O₄ for Active Anode of SOFCs

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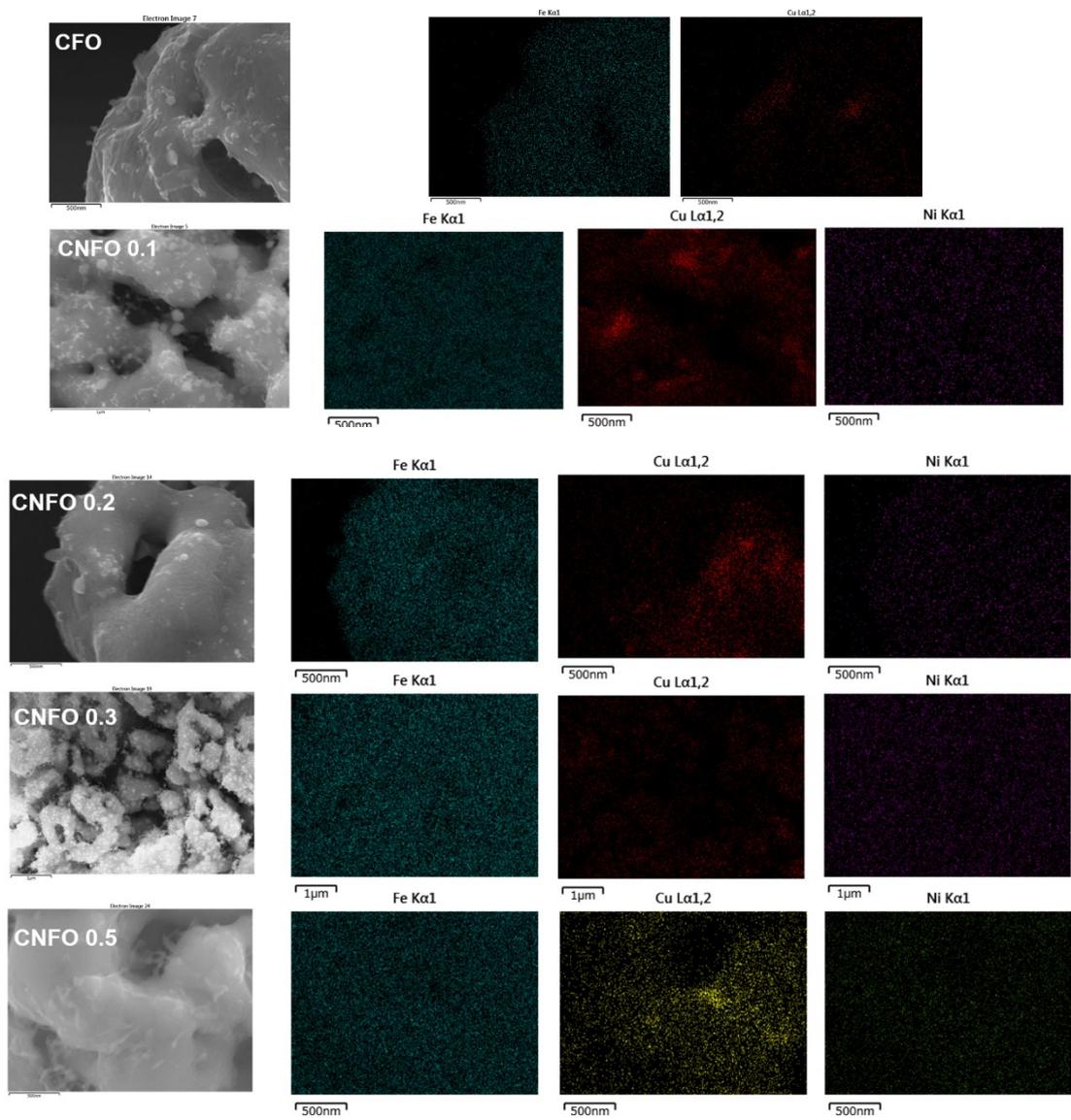


Fig. S1 EDX Mapping images for different composition of CNFO after cell test measured by SEM-EDX (a) CFO, (b) CNFO 0.1, (c) CNFO 0.2, (d) CNFO 0.3, and (e) CNFO 0.5.

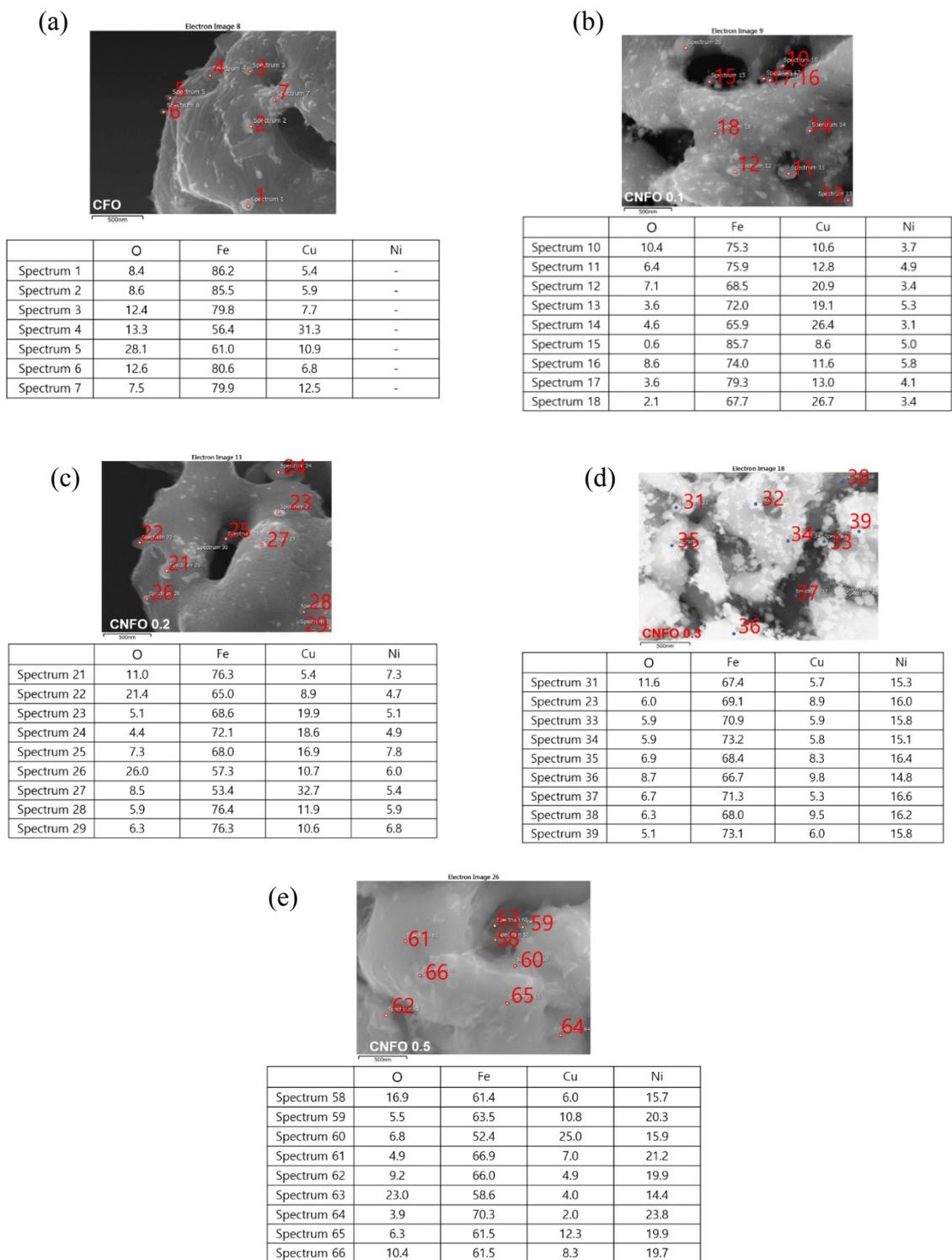


Fig. S2 Point EDX images and the related composition for different composition of CNFO after cell test measured by SEM-EDX (a) CFO, (b) CNFO 0.1, (c) CNFO 0.2, (d) CNFO 0.3, and (e) CNFO 0.5.

CNF0.1 before test

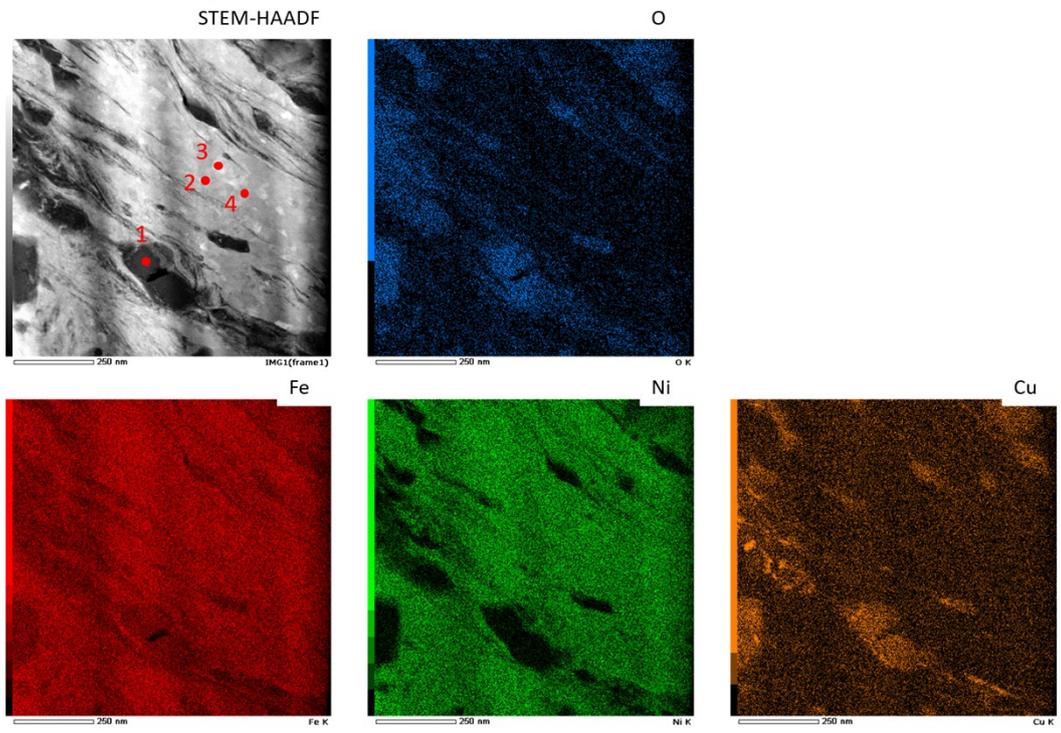


Fig. S3 STEM-EDX images of CNFO 0.1 before reduction.

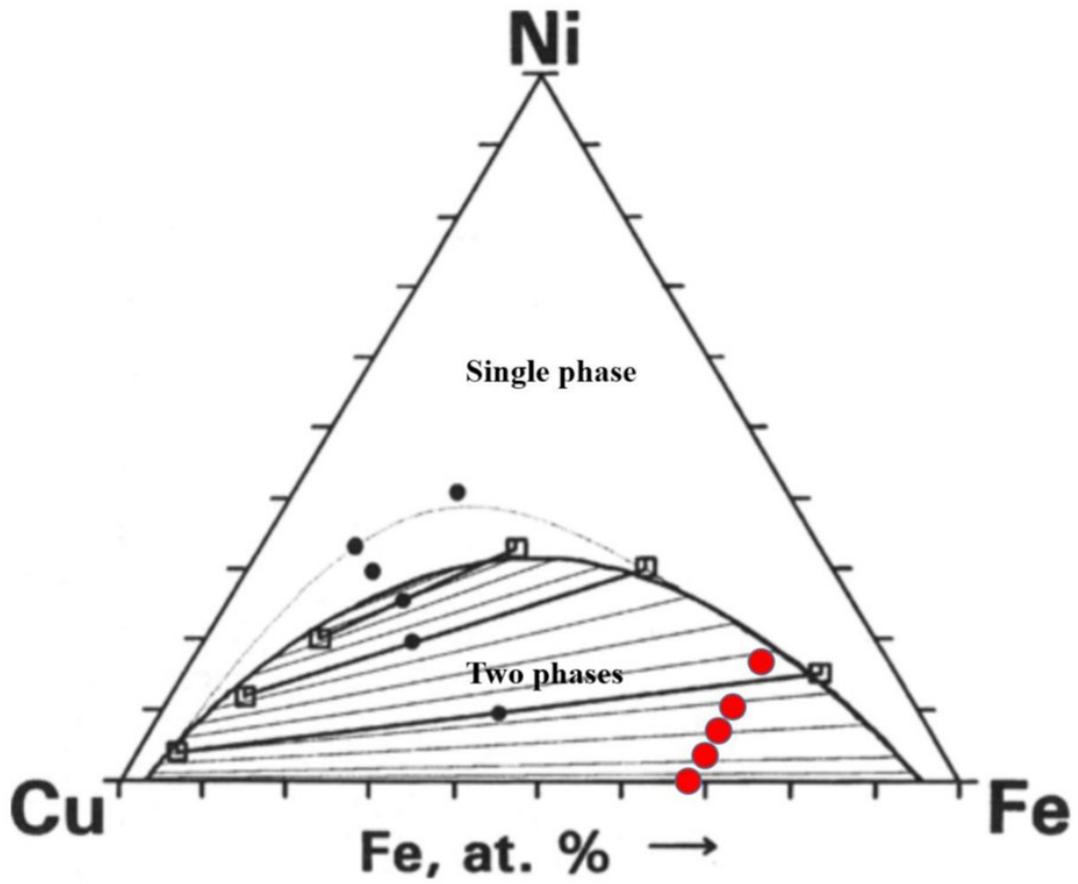


Fig. S4 Phase diagram of the Cu-Fe-Ni-system at 1273 K based on the experimental results and the thermodynamic reassessment [1] with CNFO composition, tested in this study (red circle).

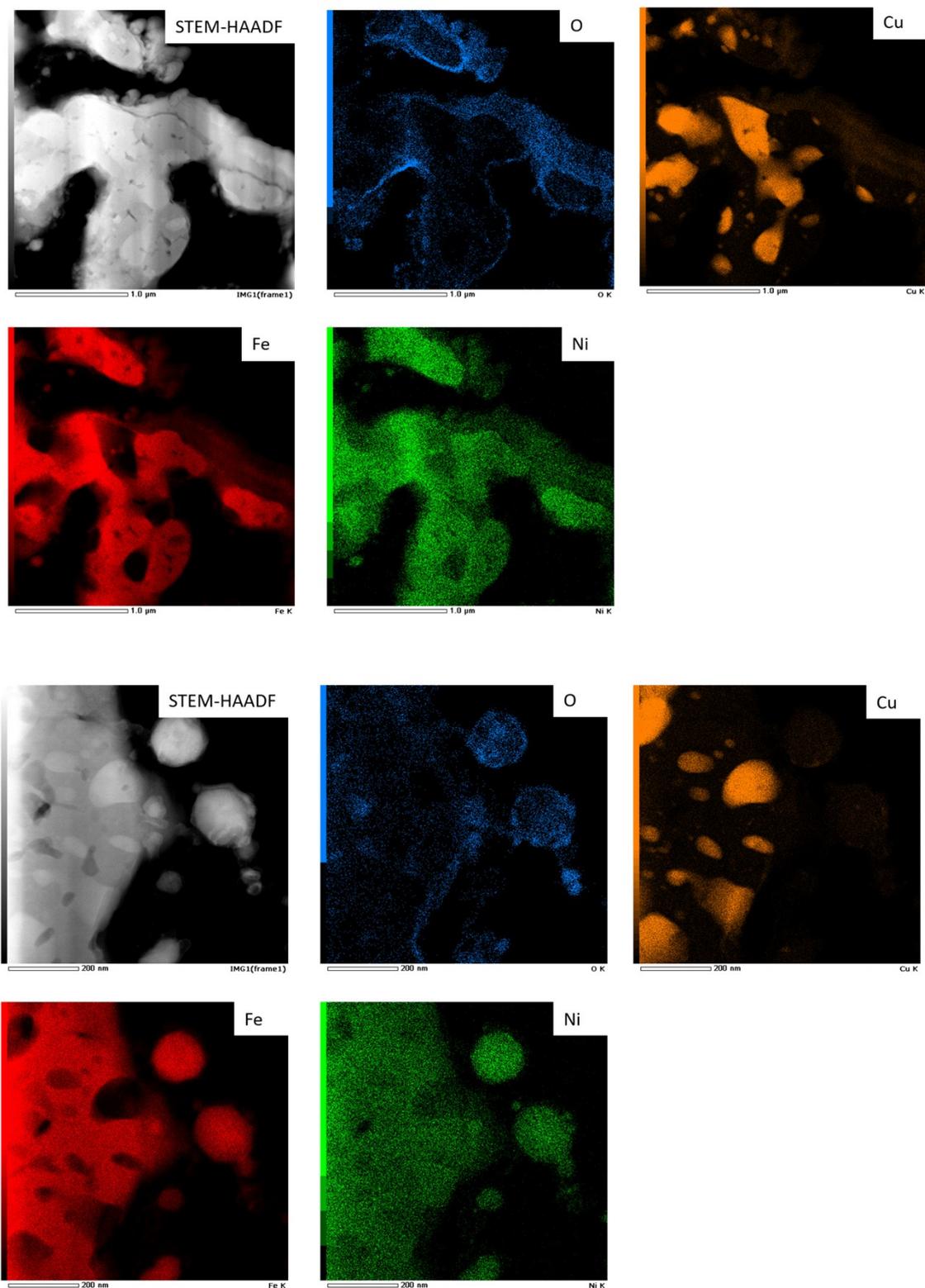


Fig. S5 STEM-EDX images of CNFO 0.1 after test.

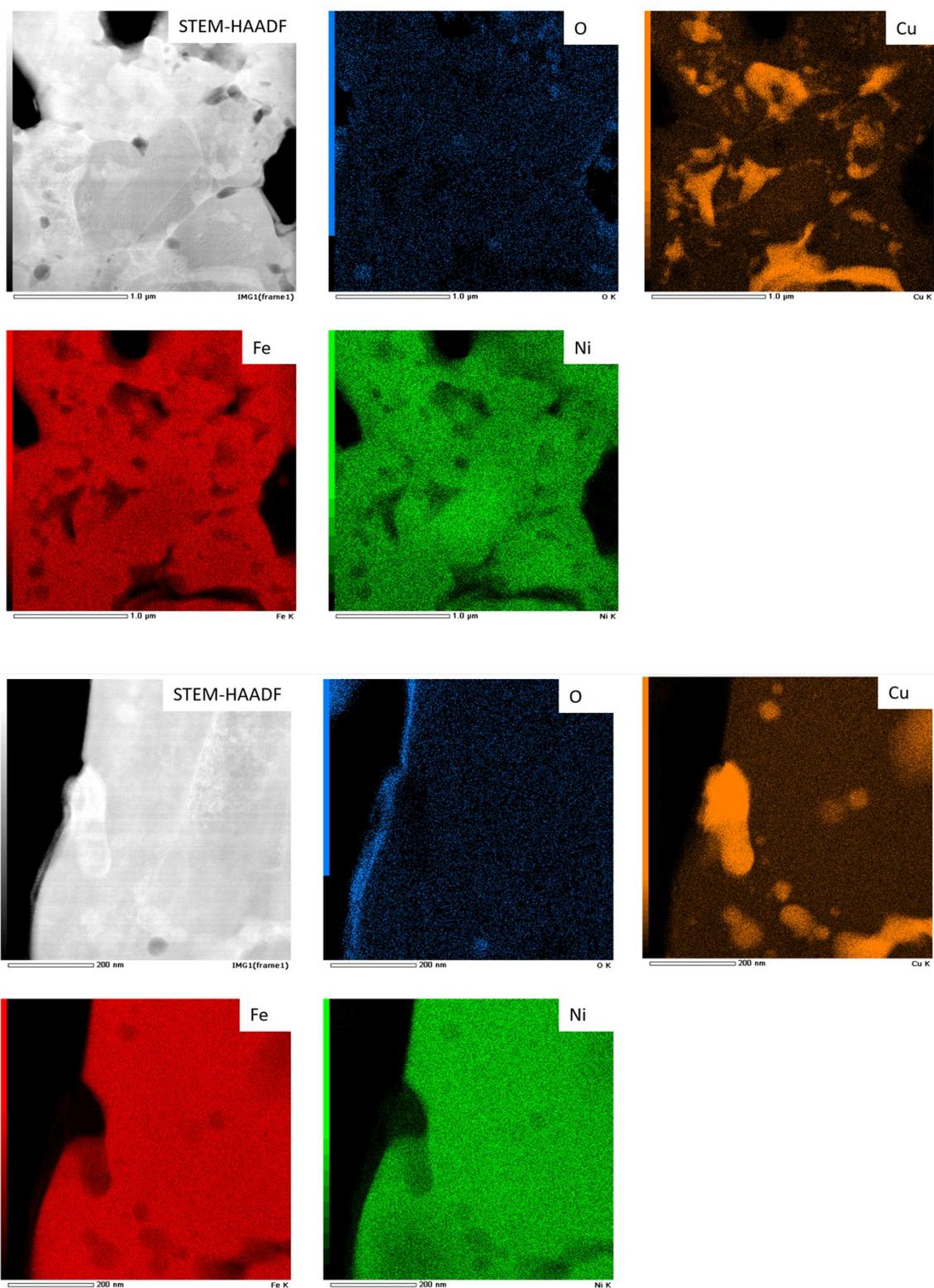


Fig. S6 STEM-EDX images of CNFO 0.5 after reduction.

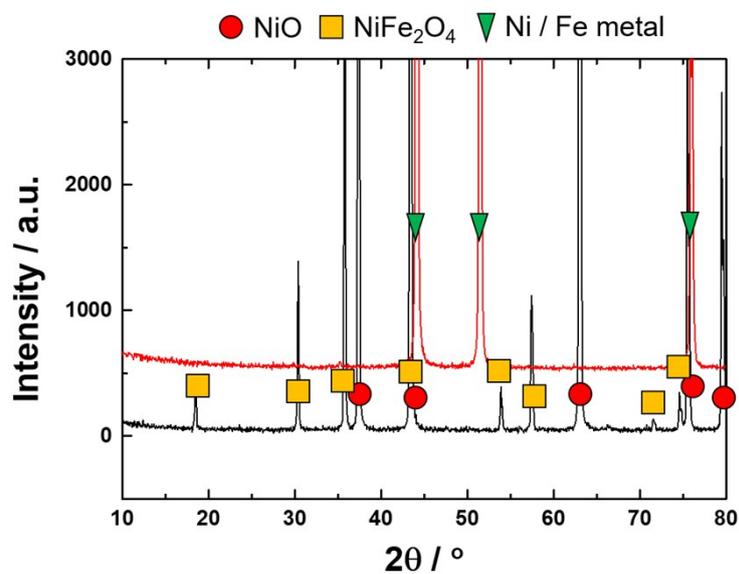


Fig. S7 XRD patterns of NiFe oxide before and after reduction at 1073 K.

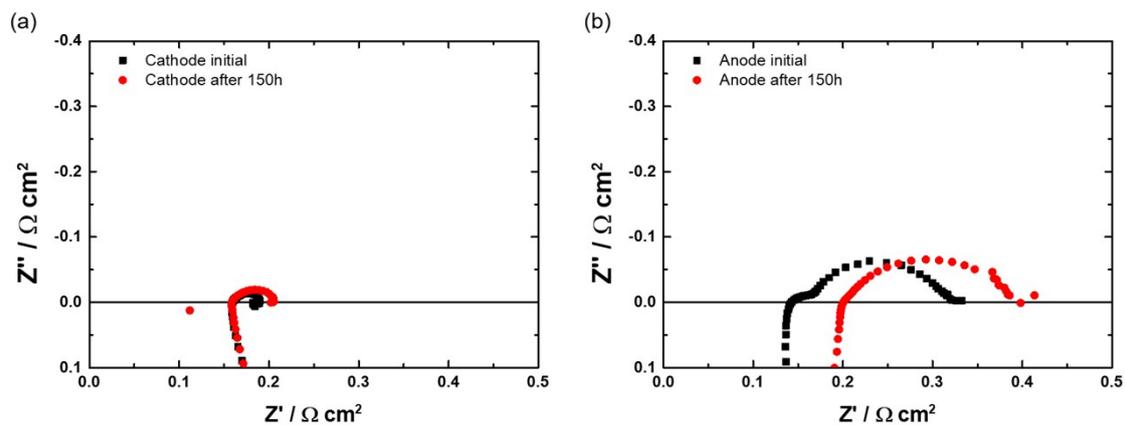


Fig. S8 Impedance plot of CNFO 0.1 initial and after operation under 0.8V for 150 h at 1073K, (a) cathode and (b) anode.

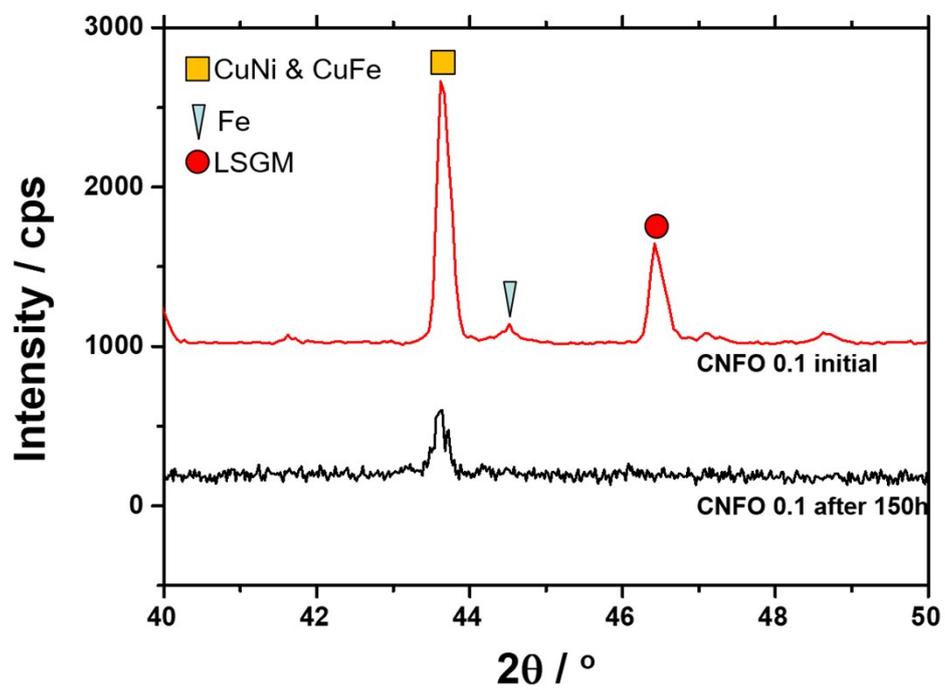


Fig. S9 XRD patterns of CNFO 0.1 initial and after operation under 0.8V for 150 h at 1073K.

[1] K. J. Rönkä, et. al., Metall. And Materials Transactions A, 27, (1996), 2229-2238.