

Electronic Supplementary Materials

Visual Observation of Hydrogen Bubbles Generation from Monodisperse CoP QDs on Ultrafine G-C₃N₄ Fiber under Visible Light Irradiation

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1. Supplementary Figures:

Figure S1 SEM images of bare Co_3O_4 without CNF as a support (a, b).

Figure S2 XRD of bare Co_3O_4 and the Co_3O_4 standard card (JCPDS 42-1476).

Figure S3 XRD pattern (a) and EDS of 7- $\text{Co}_3\text{O}_4/\text{CNF}$ (b).

Figure S4 SEM images of CoP without CNF as the support (a, b).

Figure S5 XRD of CoP and CoP standard card (JCPDS 29-0497).

Figure S6 (A) XRD of bare CoP, CNF, 1-CoP/CNF, 3-CoP/CNF, 5-CoP/CNF, 7-CoP/CNF and 10-CoP/CNF, (B) FT-IR spectrum of CoP, CNF and 7-CoP/CNF, (C) standard curve of cobalt ions determined at characteristic $\lambda = 240.7 \text{ nm}$ by atomic absorption spectrum (AAS) and (D) mass content of CoP in the CoP/CNF composites.

Figure S7 Photocatalytic hydrogen evolution rate of Pt/CNF under visible light.

Figure S8 PL spectra of CNF and CoP/CNF composites (A) and magnified PL spectrum of CoP/CNF (B). a: CoP, b: 1-CoP/CNF, c: 3-CoP/CNF, d: 5-CoP/CNF, f: 7-CoP/CNF and g: 10-CoP/CNF.

Figure S9 Nyquist plots (a) and visible light photocurrent density of CoP, CNF and 7-CoP/CNF (b).

Figure S10 High-resolution C1 s XPS of CNF and 7-CoP/CNF.

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Figure S12 (a) TEM images of 7-CoP/CNF and (b) HR-TEM of CoP QDs in 7-CoP/CNF composite.

Figure S13 Co 2p XPS of 7-CoP/CNF before (a) and after reaction.

2. Supplementary Tables:

Table S1 C, N, P and Co atom and weight ratio of 7-CoP/CNF determined by energy dispersed X-ray spectrum (EDS).

Table S2 Comparison of various semiconductor composites for photocatalytic HER.

Table S3 Time-resolved photoluminescence (TR-PL) spectroscopy probed the specific charge carrier dynamics on the interface of CNF and 7-CoP/CNF.

Table S4 The charge-transfer resistance and ohmic resistance of CoP, CBF and 7-CoP/CNF.

Table S5 C, N, P and Co atom and weight ratio of 7-CoP/CNF after long-time cyclic HER determined by energy dispersed X-ray spectrum (EDS).

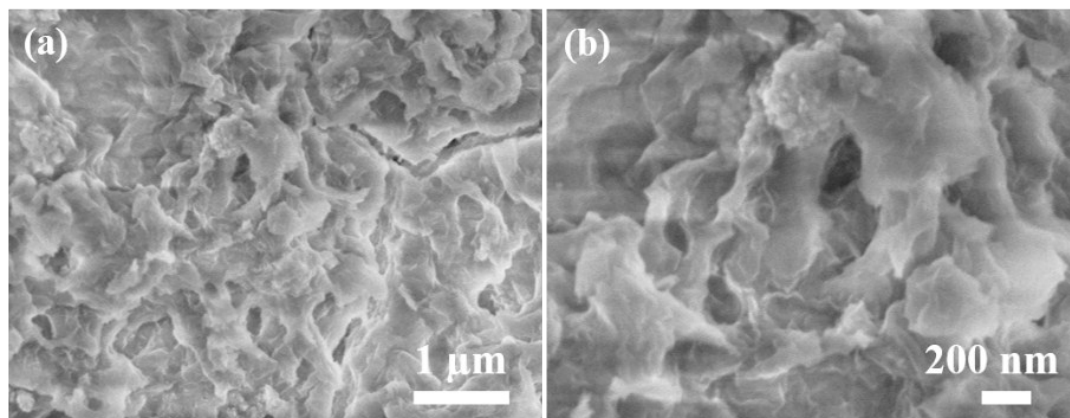


Figure S1 SEM images of bare Co_3O_4 without CNF as a support (a, b)

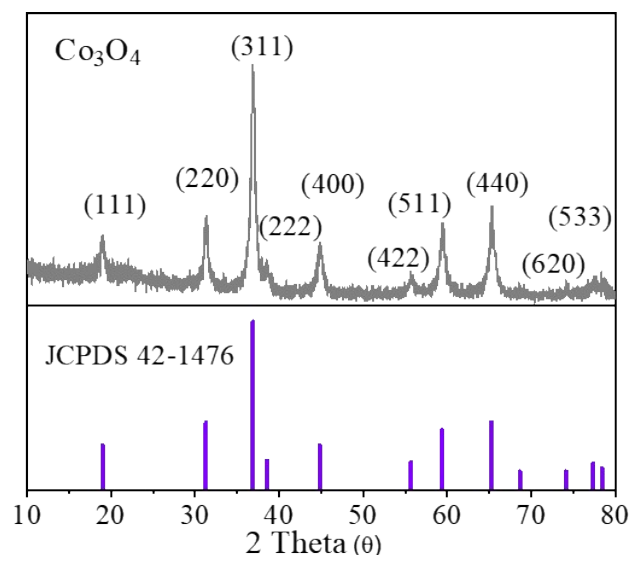


Figure S2 XRD of bare Co_3O_4 and the Co_3O_4 standard card (JCPDS 42-1476).

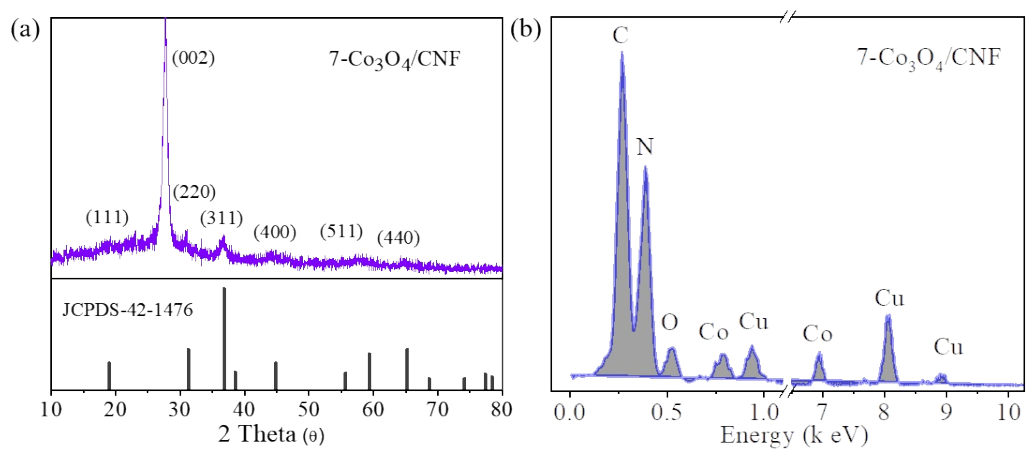


Figure S3 (a) XRD pattern and (b) EDS of 7- $\text{Co}_3\text{O}_4/\text{CNF}$.

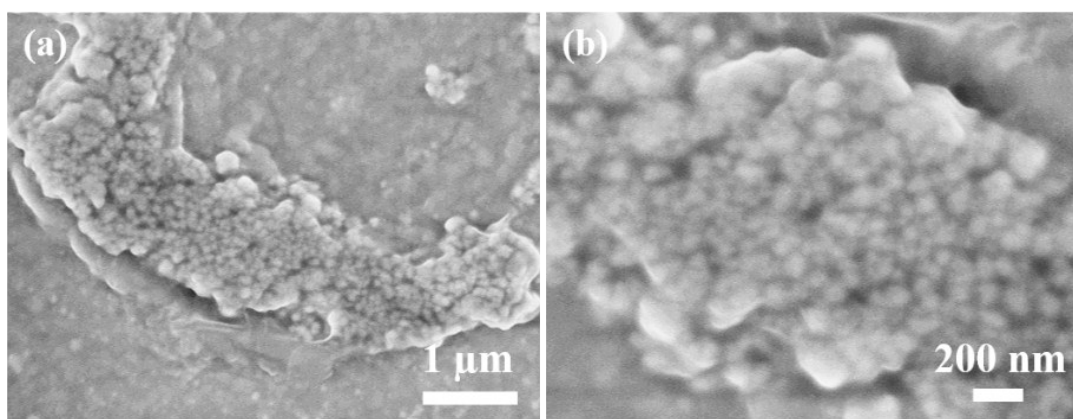


Figure S4 (a, b) SEM images of CoP without CNF as the support.

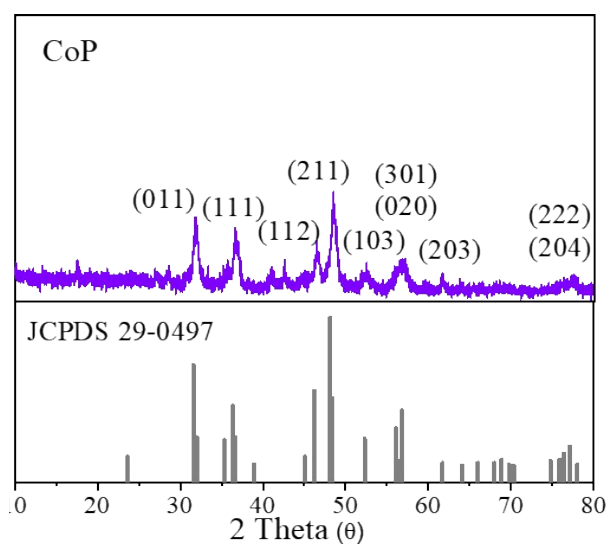


Figure S5 XRD of CoP and CoP standard card (JCPDS 29-0497).

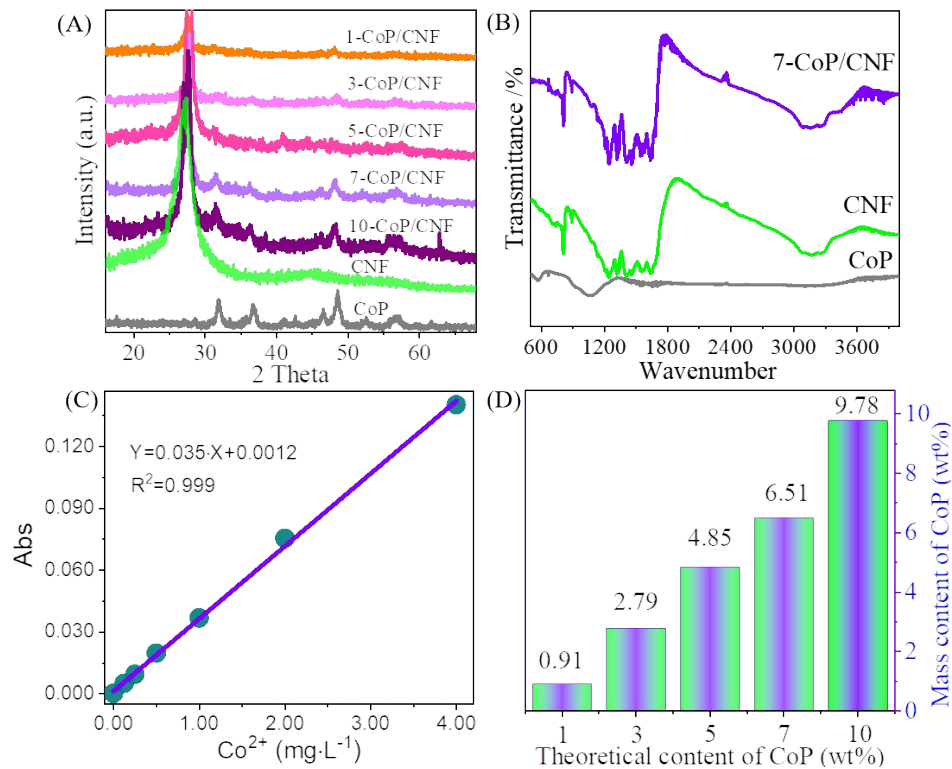


Figure S6 (A) XRD of bare CoP, CNF, 1-CoP/CNF, 3-CoP/CNF, 5-CoP/CNF, 7-CoP/CNF and 10-CoP/CNF, (B) FT-IR spectrum of CoP, CNF and 7-CoP/CNF, (C) standard curve of cobalt ions determined at characteristic $\lambda = 240.7 \text{ nm}$ by atomic absorption spectrum (AAS) and (D) mass content of CoP in the CoP/CNF composites.

Table S1 C, N, P and Co atom and weight ratio of 7-CoP/CNF determined by energy dispersed X-ray spectrum (EDS).

	C	N	P	Co	CoP
Weight (%)	34	58	3.9	4.1	6.32

Table S2 Comparison of various semiconductor composites for photocatalytic HER.

Photocatalysts	Light source Xe lamp (nm)	Reactant ^[a] solution	Activity (mmol/g/h)	AQE
Pt/g-C ₃ N ₄ network ^{S1}	$\lambda \geq 420$ nm	10 vol% TEOA	1.36	12% (420 nm)
Pt/g-C ₃ N ₄ nanofiber ^{S2}	$\lambda \geq 420$ nm	10 vol% TEOA	0.732	7% (420 nm)
Pt/O doped g-C ₃ N ₄ ^{S3}	$\lambda \geq 420$ nm	10 vol% TEOA	0.375	-
Pt/g-C ₃ N ₄ nanosheets ^{S4}	$\lambda \geq 420$ nm	10 vol% TEOA	1.15	-
Pt/g-C ₃ N ₄ nanosheets ^{S5}	300W Xe lamp $\lambda \geq 400$ nm	10 vol% TEOA	0.64	-
Pt/g-C ₃ N ₄ ^{S6}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.23	-
Pt/g-C ₃ N ₄ nanosheets ^{S7}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.24	-
Pt/g-C ₃ N ₄ ^{S8}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.098	0.6% (420 nm)
NiS/g-C ₃ N ₄ nanosheets ^{S9}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.482	4.4% (440 nm)
Pt/g-C ₃ N ₄ ^{S10}	300W Xe lamp $\lambda \geq 420$ nm	15 vol% TEOA	1.11	-
Pt/g-C ₃ N ₄ ^{S11}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.37	-
Ag ₂ O/g-C ₃ N ₄ ^{S12}	300W Xe lamp $\lambda \geq 420$ nm	10 vol% TEOA	0.33	-
Pt/g-C ₃ N ₄ ^{S13}	300W Xe lamp $\lambda \geq 400$ nm	10 vol% TEOA	0.604	20 %
Pt/g-C ₃ N ₄ ^{S14}	300W Xe lamp $\lambda \geq 400$ nm	10 vol% TEOA	1.510	5.7% (420nm)
Pt/g-C ₃ N ₄ ^{S15}	$\lambda \geq 420$ nm	10 vol% TEOA	1.751	-
Pt/g-C ₃ N ₄ ^{S16}	$\lambda \geq 420$ nm	15 vol% TEOA	3.446	32% (420 nm)
NiP/CdS ^{S17}	$\lambda \geq 420$ nm	NaSO ₃ /Na ₂ S	553	41% (450 nm)
MoS ₂ /CdS ^{S18}	$\lambda \geq 420$ nm	NaSO ₃ /Na ₂ S	17.2	24.2% (420 nm)
rGO/g-C ₃ N ₄ ^{S19}	UV-Vis light	30 vol% TEOA	0.273	-
COFs/g-C ₃ N ₄ ^{S20}	$\lambda > 420$ nm	10 vol% TEOA	10.1	20.7% (425 nm)
C/g-C ₃ N ₄ ^{S21}	$\lambda > 420$ nm	TEOA	3.53	10.9% (420 nm)
C/g-C ₃ N ₄ ^{S22}	$\lambda > 420$ nm	-	0.11 mol·h ⁻¹	-
P/g-C ₃ N ₄ ^{S23}	$\lambda > 420$ nm	10 vol% TEOA	¹	5.7% (420 nm)
ZnCdS/ZnO ^{S24}	$\lambda > 420$ nm	NaSO ₃ /Na ₂ S	0.929	-
P/g-C ₃ N ₄ ^{S25}	$\lambda > 400$ nm	TEOA	28.6	-
FeS/FeSe ₂ ^{S26}	UV-Vis light	NaSO ₃ /Na ₂ S	0.738	-

MoC/C ^{S27}	$\lambda > 400$ nm	Lactic acid	2.071	-
H _{0.53} WO ₃ /CdS ^{S28}	$\lambda > 400$ nm	Lactic acid	2.08	-
CdS ^{S29}	$\lambda \geq 420$ nm	Lactic acid	3.53	-
Pt/g-C ₃ N ₄ ^{S30}	$\lambda \geq 420$ nm	10 vol% TEOA	0.916	6.52 (420 nm)
Pt/g-C ₃ N ₄ ^{S31}	$\lambda \geq 420$ nm	20 vol% TEOA	1.596	3.56 (420 nm)
PDBTSO ^{S32}	$\lambda > 400$ nm	20 vol% TEOA	44.2	8% (420 nm)
MoSe ₂ ^{S33}	$\lambda > 400$ nm	10 vol% TEOA	1.672	5.2% (520 nm)
Polymer Dots ^{S34}	$\lambda > 420$ nm	ascorbic acid	0.84	0.9% (500 nm)
triazine polymers ^{S35}	$\lambda > 420$ nm	10 vol% TEOA	6.60	7.3% (420 nm)
CdS/Ag ₂ S ^{S36}	$\lambda \geq 420$ nm	10 vol% TEOA	0.376	-
g-C ₃ N ₄	$\lambda > 420$ nm	10 vol% TEOA	1.619	-
CoP/g-C ₃ N ₄ (this study)	$\lambda \geq 420$ nm	15 vol% TEOA	121	60% (420 nm)

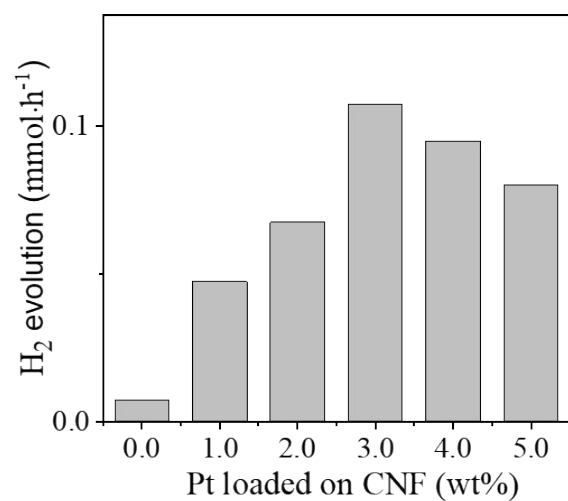


Figure S7 Photocatalytic hydrogen evolution rate of Pt/CNF under visible light.

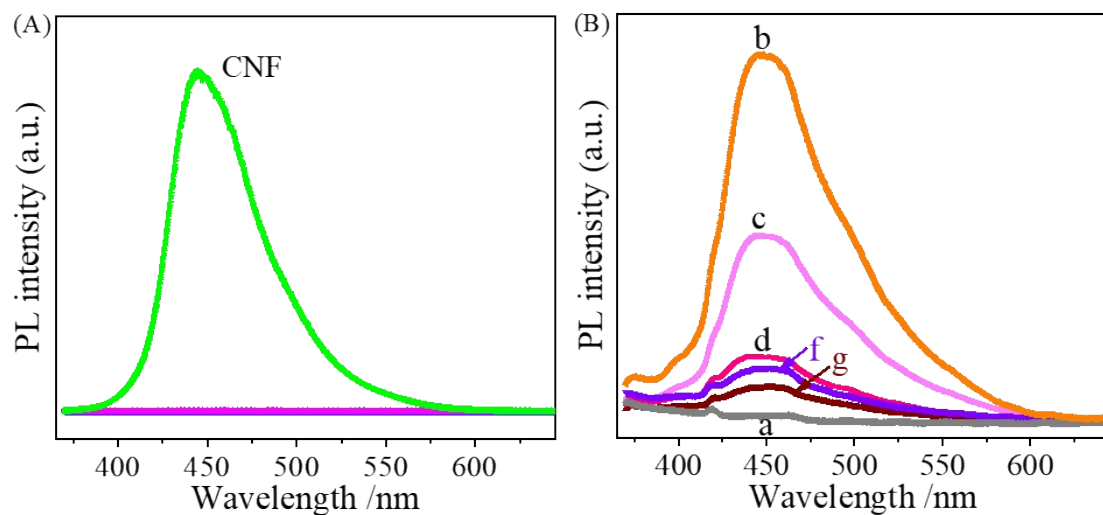


Figure S8 PL spectra of CNF and 7-CoP/CNF composites (A) and magnified PL spectrum of CoP/CNF (B). a: CoP, b: 1-CoP/CNF, c: 3-CoP/CNF, d: 5-CoP/CNF, f: 7-CoP/CNF and g: 10-CoP/CNF.

Table S3 Time-resolved photoluminescence (TR-PL) spectroscopy probed the specific charge carrier dynamics on the interface of CNF and 7-CoP/CNF.

Samples	A ₁ [%]	τ_1 [ns]	A ₂ [%]	τ_2 [ns]	A ₃ [%]	τ_3 [ns]	τ_A [ns]
7-CoP/CNF	49.13	2.15	39.27	5.69	11.26	11.25	6.39
CNF	32.87	1.98	50.13	6.40	17.05	12.89	8.28

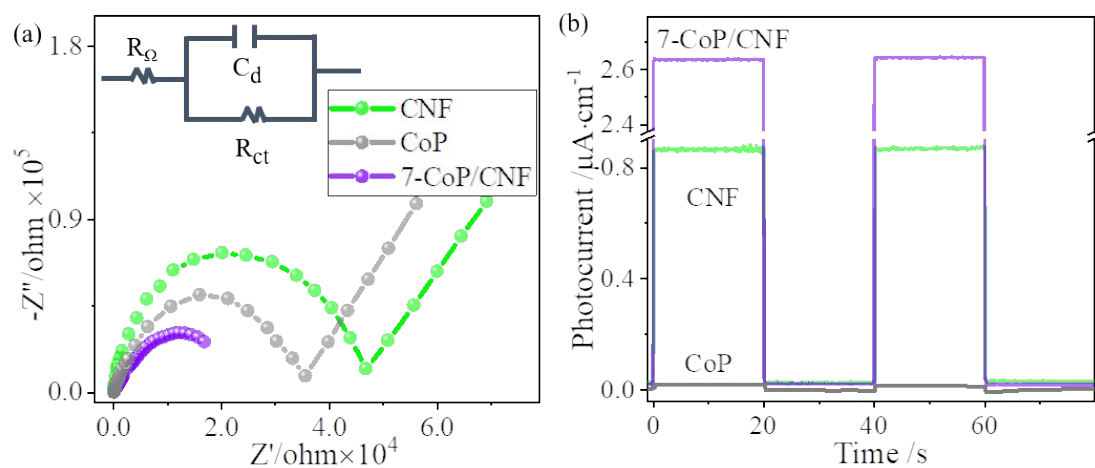


Figure S9 Nyquist plots (a) and visible light photocurrent density of CoP, CNF and 7-CoP/CNF (b).

Table S4 The charge-transfer resistance and ohmic resistance of CoP, CBF and 7-CoP/CNF fitted by Nyquist plots.

	CoP	CNF	7-CoP/CNF
R_{ct}	1.84 k Ω	23.9 k Ω	1.24 k Ω
R_s	0.16 k Ω	0.16 k Ω	0.16 k Ω

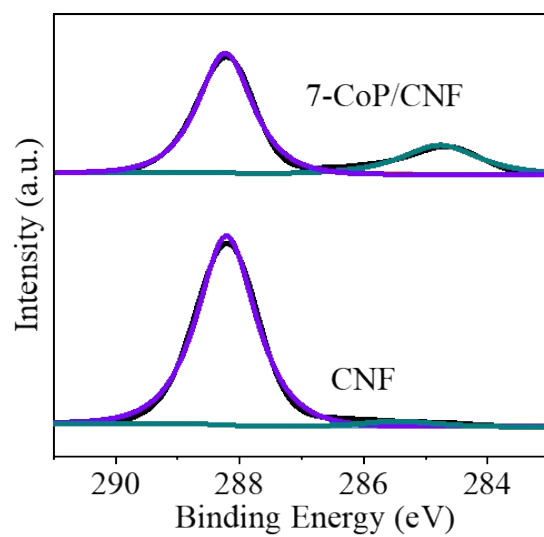


Figure S10 High-resolution C1 s XPS of CNF and 7-CoP/CNF.

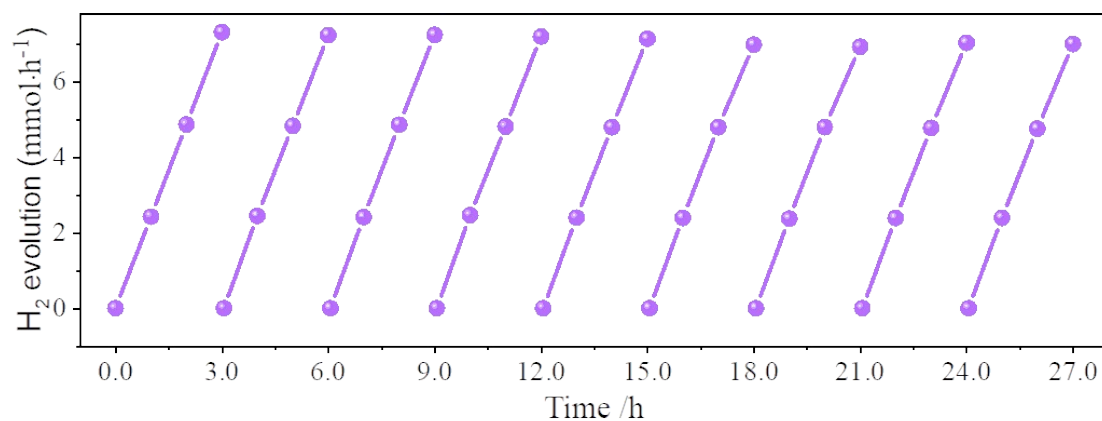


Figure S11 Photocatalytic HER rate comparison between Pt/CNF and CoP/CNF (a), HER activity of CNF loading various amount of Pt (b) and long-term HER stability of 7-CoP/CNF (c). Condition (a-c): 15 vol% TEOA as hole sacrificial molecules under visible light and room temperature ($\lambda \geq 420$ nm).

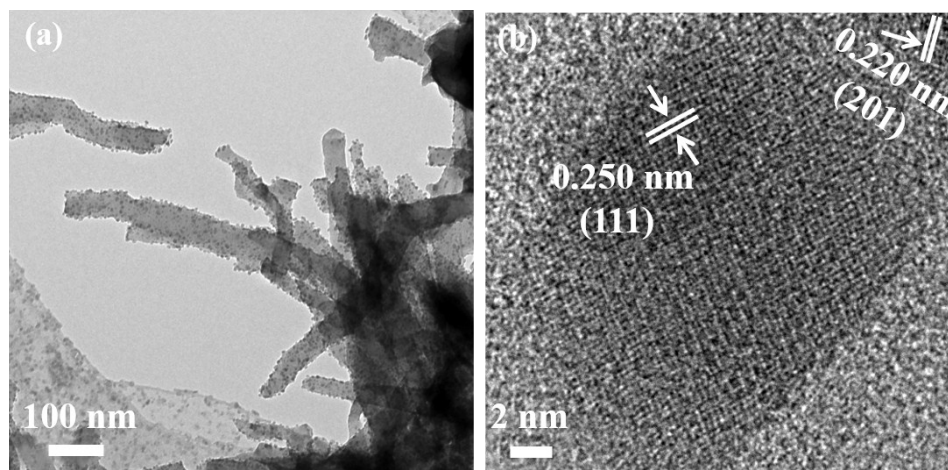


Figure S12 (a) TEM images of 7-CoP/CNF and (b) HR-TEM of CoP QDs in CoP/CNF composite.

Table S5 C, N, P and Co atom and weight ratio of 7-CoP/CNF after long-time cyclic HER determined by energy dispersed X-ray spectrum (EDS).

	C	N	P	Co	CoP
Weight (%)	35	57	3.8	4.0	6.2

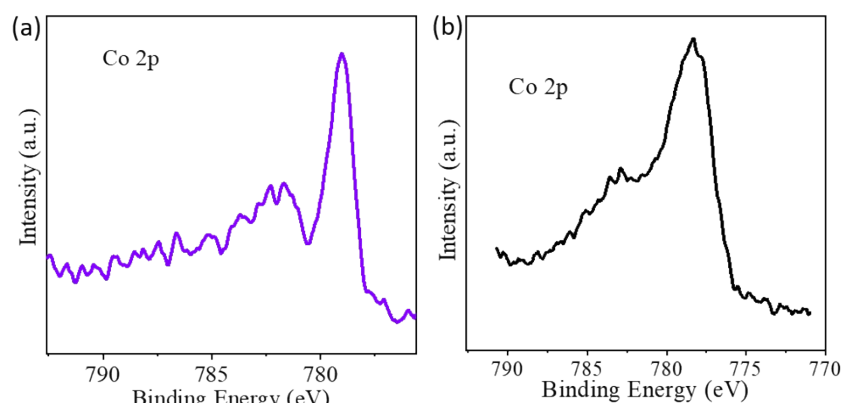


Figure S13 Co 2p XPS of 7-CoP/CNF before (a) and after reaction.

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