

## Supporting Information

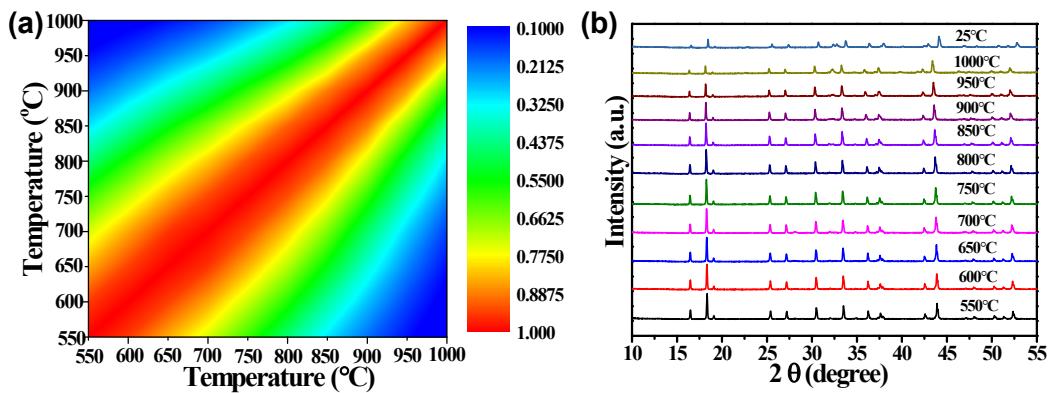
### **High cathode utilization efficiency through interface engineering in all-solid-state lithium metal battery**

*Xianwei Guo<sup>1</sup>, Liangwei Hao<sup>1</sup>, Yubo Yang<sup>1</sup>, Yongtao Wang<sup>1</sup>, Yue Lu<sup>2</sup>, Haijun Yu<sup>1\*</sup>*

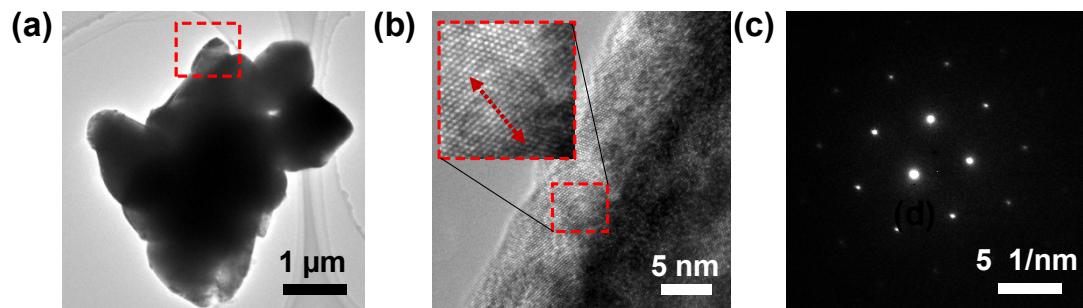
1 College of Materials Sciences and Engineering, Key Laboratory of Advanced Functional Materials, Education Ministry of China, Beijing University of Technology, Beijing 100124, P. R. China

2 Institute of Microstructure and Properties of Advanced Materials, Beijing University of Technology, Beijing 100124, P.R. China

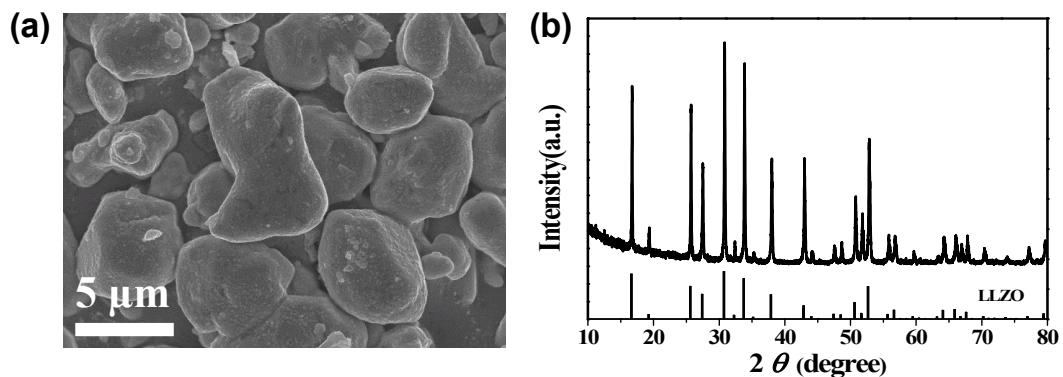
E-mail: hj-yu@bjut.edu.cn



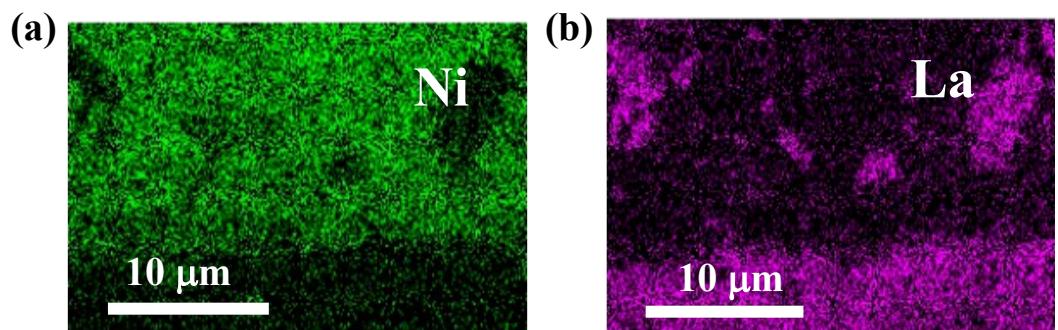
**Fig. S1** **a**, Contour plot of 2D covariance analysis of HTXRD patterns. **b**, The corresponding *in-situ* XRD patterns with the appearance of new phase from 550 to 1000 °C and cooling down to 25°C.



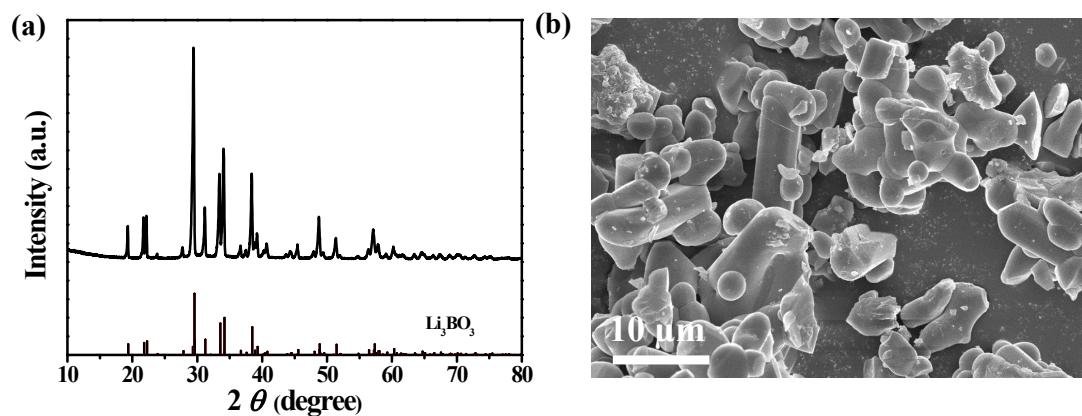
**Fig. S2** **a**, The typical particle of NCM622 MCG material. **b**, The corresponding high-resolution transmission electron microscopy (HRTEM) with the d spacing of 0.242 nm and **c**, selected area electron diffraction (SAED) pattern for the (101) plane of the red marked area of **a**.



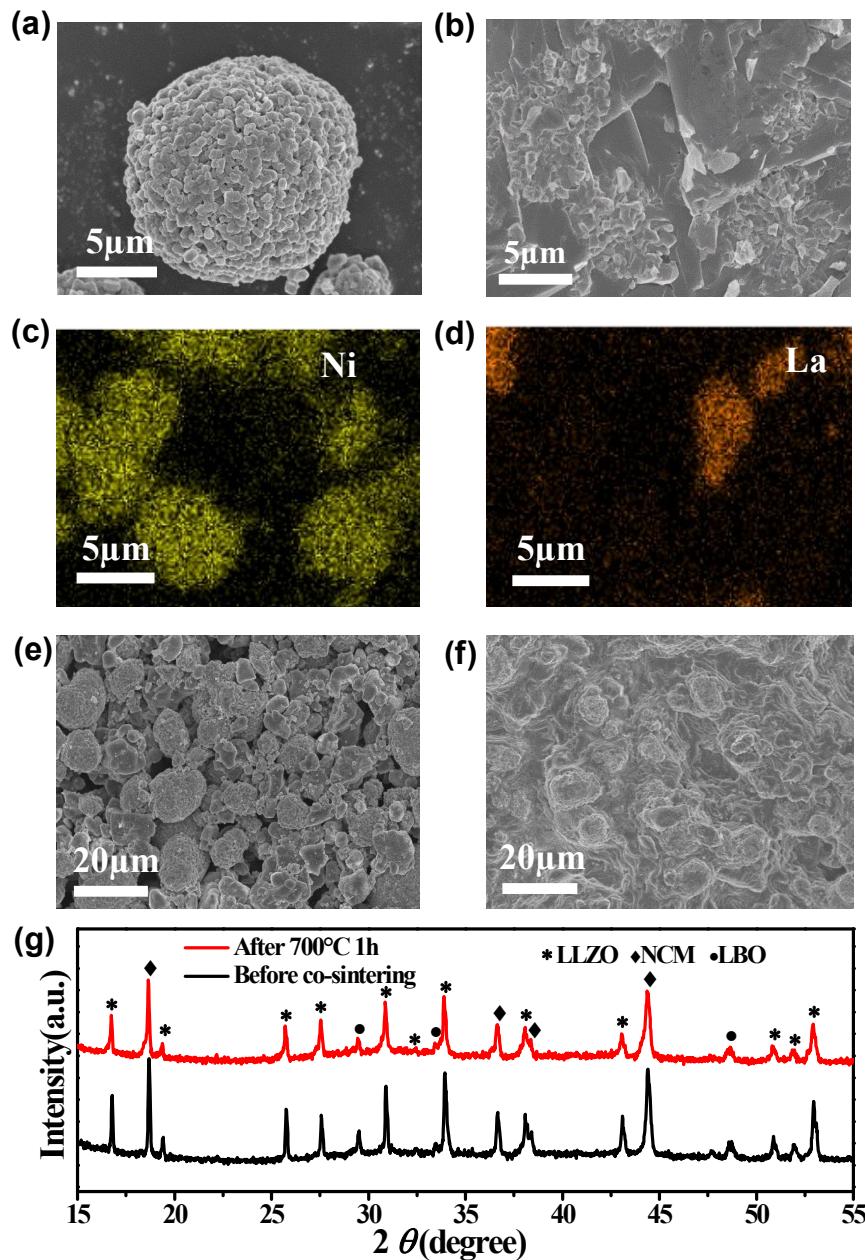
**Fig. S3** **a**, The SEM image and **b**, XRD pattern of the commercial LLZTO powder.



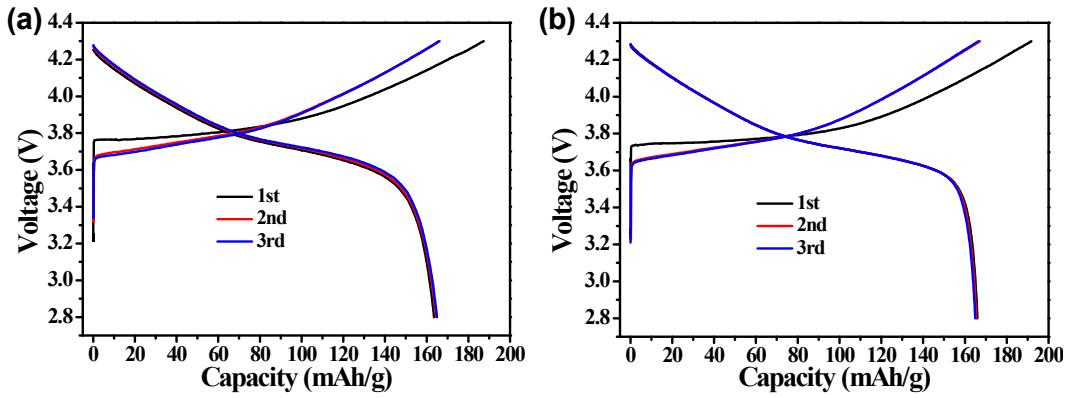
**Fig. S4** The elemental mappings of La and Ni in the NCM622 MCG based composite cathode after co-sintering.



**Fig. S5** **a**, The XRD pattern and **b**, SEM image of as-prepared LBO.



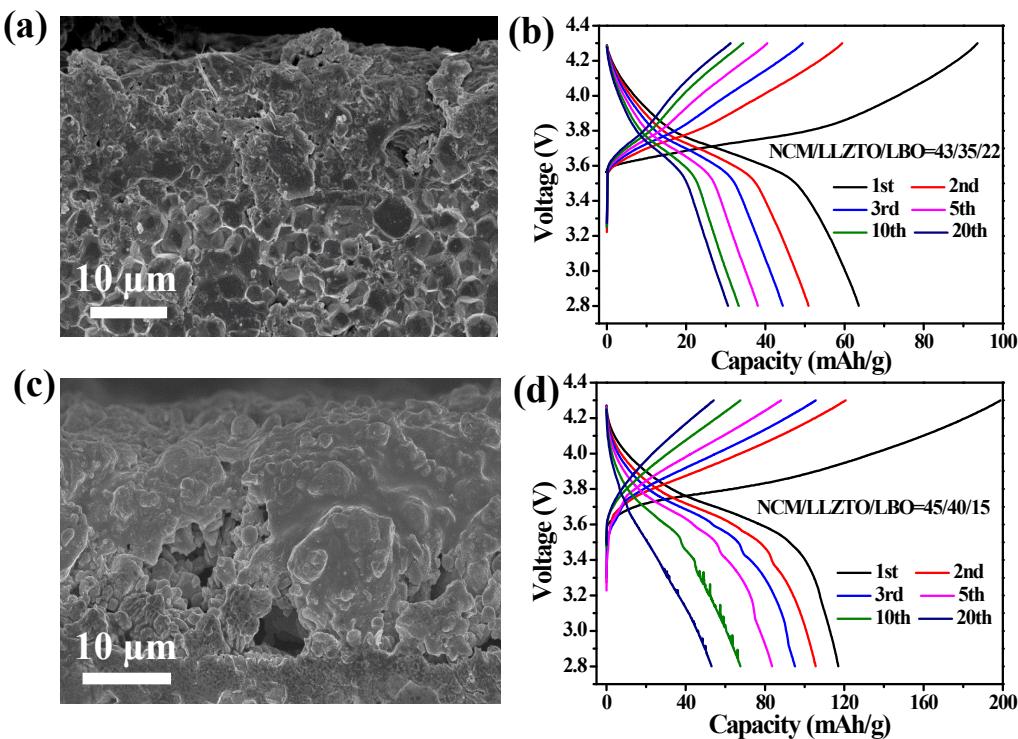
**Fig. S6** **a**, Morphology of NCM622 spherical agglomerates. **b**, The cross-sectional SEM images of the composite cathode with agglomerates and LLZTO particles after co-sintering. **c-d**, The elemental mappings of Ni and La in the composite cathode after co-sintering. **e-f**, The SEM images of the top surface of composite cathode before (insert) and after co-sintering. **g**, The XRD patterns of composite cathode with LBO additive before and after co-sintering.



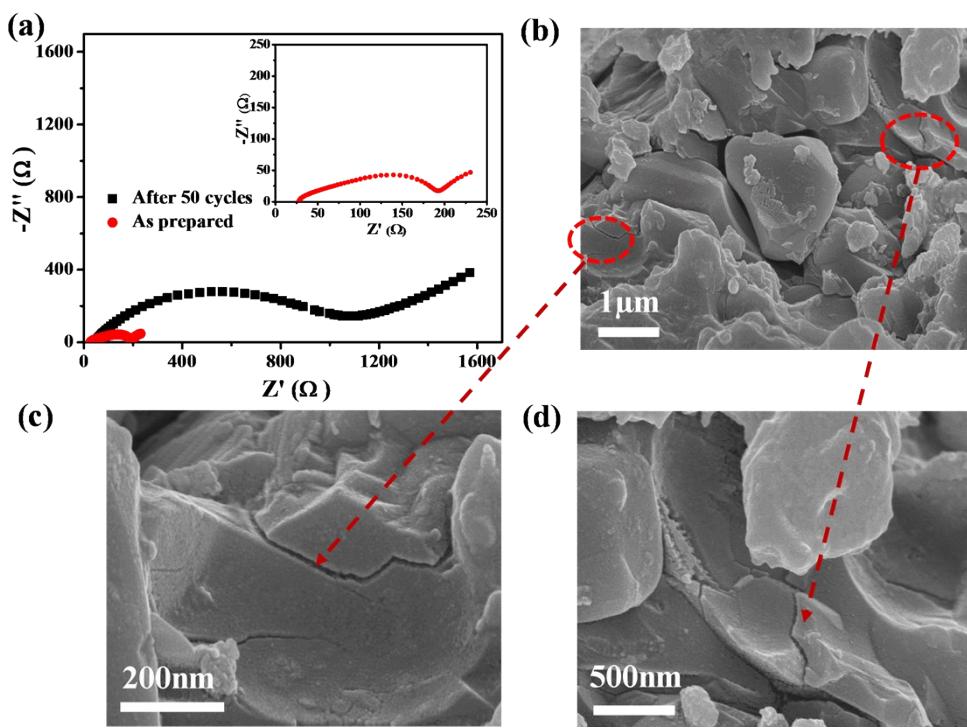
**Fig. S7.** The charge-discharge curves of NCM622 cathode material with **a**, MCG and **b**, agglomerates in the liquid electrolyte based batteries at the same current densities.

**Table S1** The comparisons of electrochemical performances with LLZO supporting bulk-type ASSLMBs with NCM or LCO cathode material.

Configuration of the solid state battery	Cathode	Temperature	Current density	Voltage Range (V)	Initial Capacity (mAh/g)	Cycles	Final Capacity (mAh/g)	Ref.
NCM523+LBO+ITO/LLZTO/Li	agglomerates	80 °C	5 $\mu$ A/cm <sup>2</sup>	3.0-4.6	123.3	5	76.6	[45]
NCM622+LBO+LLZTO/LLZTO/Gel/Li	agglomerates	RT	0.05 C (1C=115mA/g)	3.0-4.2	106	30	74.2	[46]
LCO+LLZTO+LCBO/LLZTO/Li	micro-sized particles	100°C	0.05 C (1C=115mA/g)	3.0-4.05	106	40	67	[47]
LCO+LBO+ITO/LLZTO/Li	micro-sized particles	80°C	5 $\mu$ A/cm <sup>2</sup>	2.8-4.3	69.6	1	69.6	[59]
<b>NCM622+LBO+LLZTO/LLZTO/Li</b>	<b>MCG</b>	<b>80°C</b>	<b>0.05 C (1C=180mA/g)</b>	<b>2.8-4.3</b>	<b>138.8</b>	<b>50</b>	<b>79</b>	<b>This work</b>



**Fig. S8.** **a, c,** The cross-sectional SEM images after co-sintering and **b,d,** the corresponding charge-discharge curves of the composite cathode by co-sintering the mixture of NCM622 MCG, LLZTO particles and LBO powder with a weight ratio of 43:35:22 and 45:40:15 at 700 °C for 1h, respectively.



**Fig. S9.** **a**, The comparisons of EIS spectra of ASSLMBs constructed with NCM622 MCG material at the initial state and after 50 cycles. **b-d**, Cracking occurs in the composite cathode after 50 cycles.