

Electronic Supplementary Information

Design of an ultralong-life Li-CO₂ batteries with IrO₂ nanoparticles highly dispersed onto nitrogen-doped carbon nanotubes

Gang Wu^a, Xue Li^a, Zhang Zhang^c, Peng Dong^a, Mingli Xu^a, Hongliang Peng^d, Xiaoyuan Zeng^{a,b,*},
Yingjie Zhang^{a,*}, and Shijun Liao^b

^a National and Local Joint Engineering Laboratory for Lithium-ion Batteries and Materials
Preparation Technology, Key Laboratory of Advanced Battery Materials of Yunnan Province, Faculty
of Materials Science and Engineering, Kunming University of Science and Technology, Kunming
650093, China.

^b The Key Laboratory of Fuel Cell Technology of Guangdong Province, School of Chemistry and
Chemical Engineering, South China University of Technology, Guangzhou 510641, China.

^c College of Chemistry and Chemical Engineering and Henan Province Key Laboratory of Utilization
of Non-Metallic Mineral, Xinyang Normal University, Xinyang 464000, China.

^d Guilin University of Electronic Technology, Guilin, 541004, P. R. China.

*Corresponding authors: E-mail addresses: zengxiaoyuan721@126.com; zyjkmust@126.com

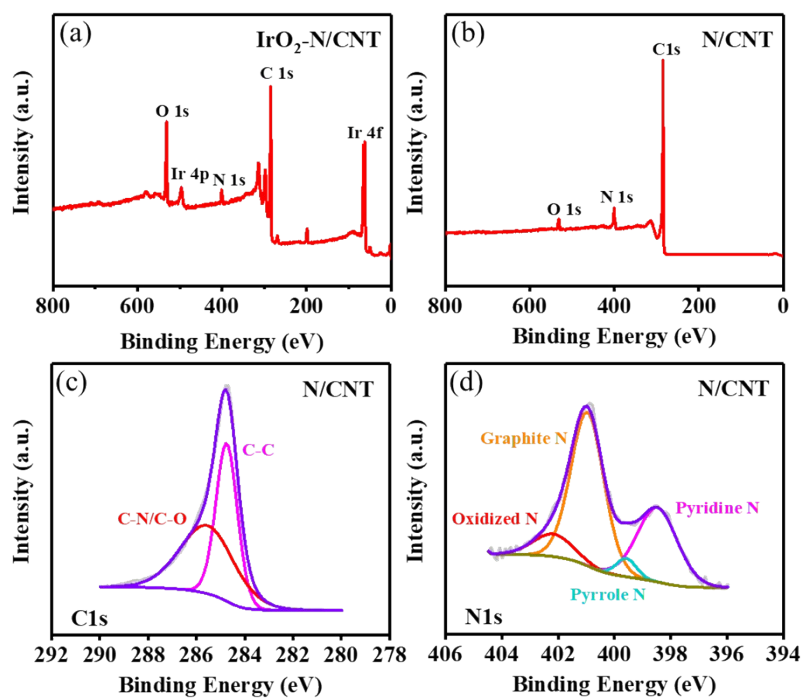


Figure S1. The full XPS spectrum of (a) IrO₂-N/CNT and (b) N/CNT; the high-resolution XPS spectra of (c) C1s and (d) N1s for N/CNT.

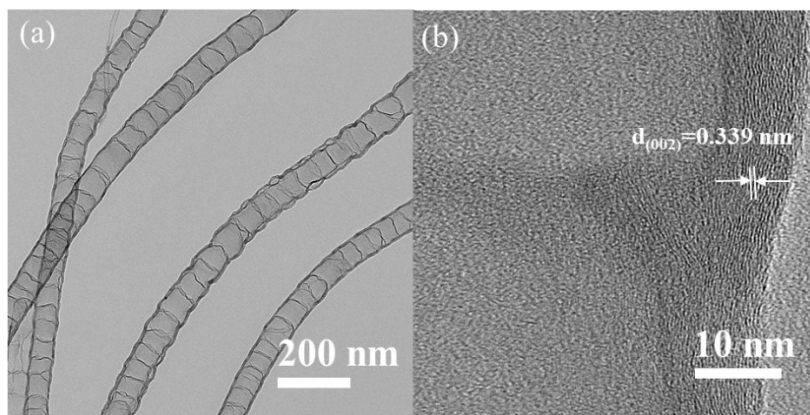


Figure S2. (a) TEM and (b) HRTEM images of N/CNT.

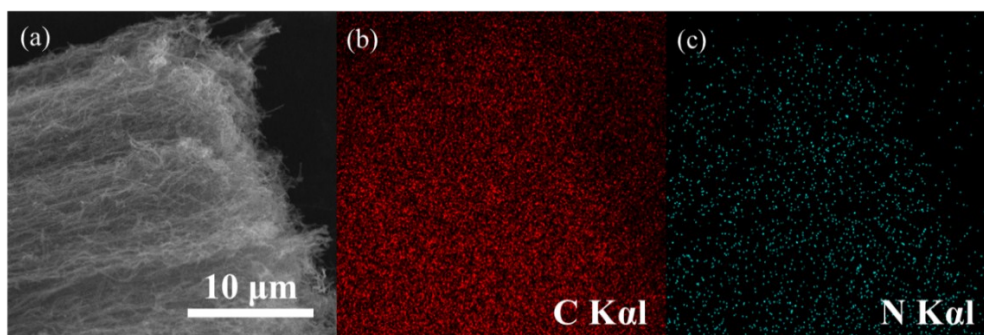


Figure S3. The EDS mapping of C and N in N/CNT.

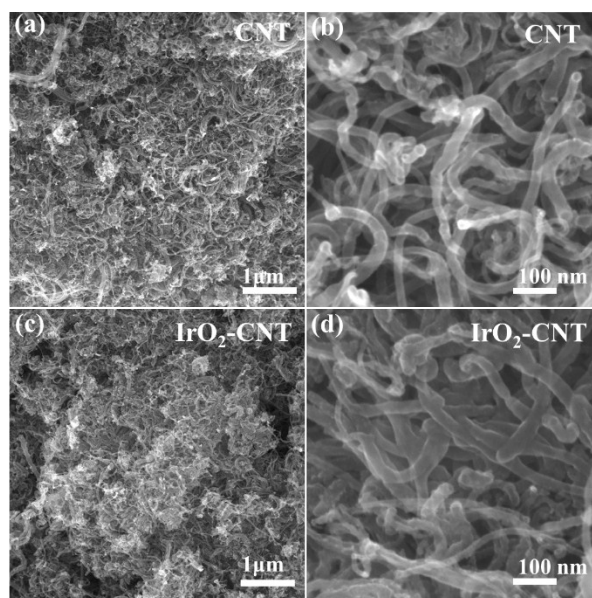


Figure S4. FESEM images of (a, b) CNT and (c, d) IrO₂-CNT.

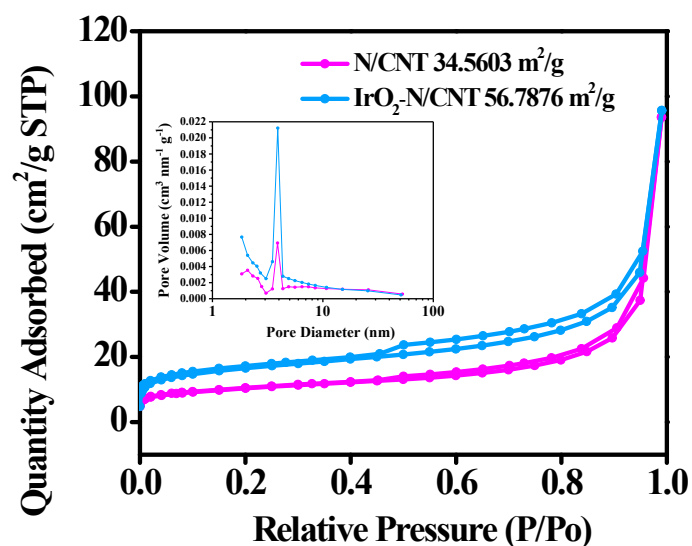


Figure S5. N_2 adsorption-desorption isotherms and the pore-size distribution (inset) of N/CNT and IrO_2 -N/CNT.

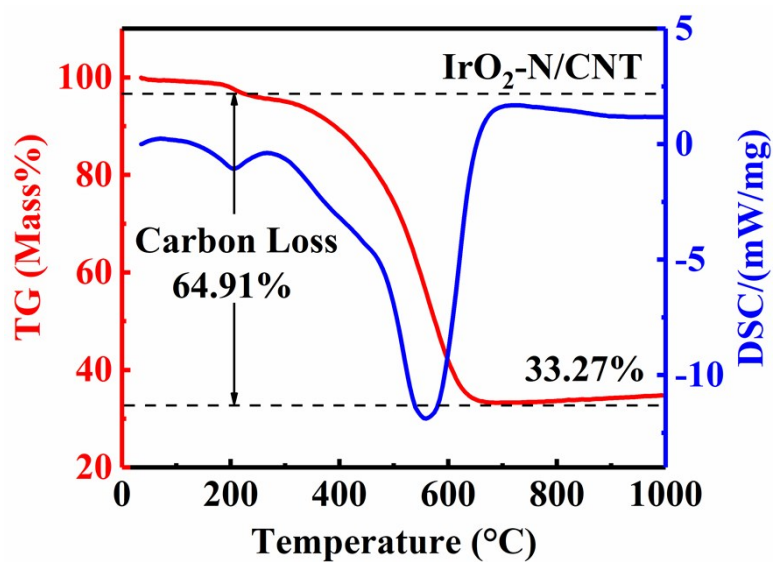


Figure S6. TGA-DTA curves of IrO_2 -N/CNT.

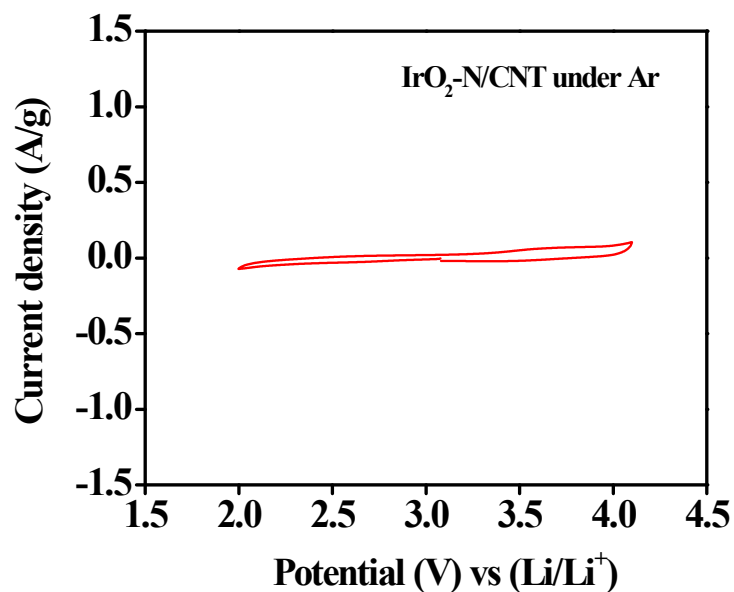


Figure S7. The CV curves of the Li-CO₂ batteries with IrO₂-N/CNT between 2.0-4.1 V at 0.5 mV/s under Ar atmosphere in TEGDME electrolyte.

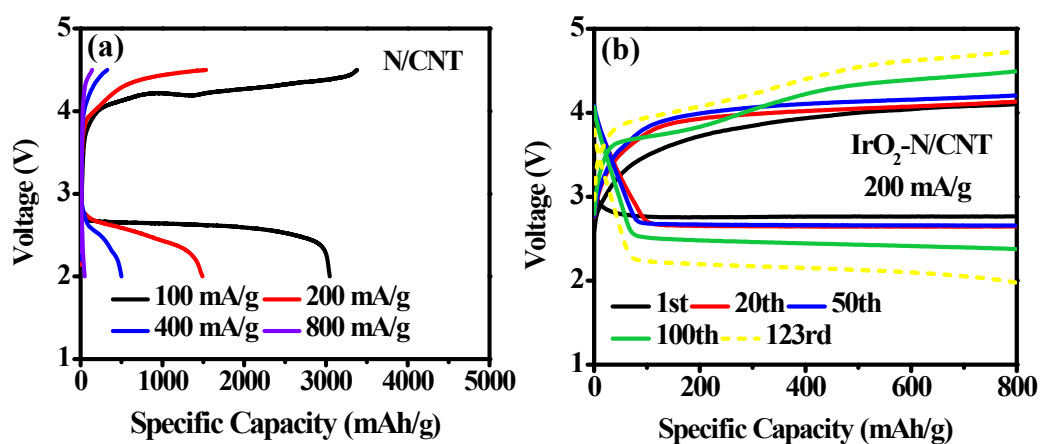


Figure S8. (a) The initial discharge-charge curves of Li-CO₂ batteries with N/CNT cathodes at different current density; (b) The cycling performance of IrO₂-N/CNT cathodes with a cutoff capacity of 800 mAh/g at a current density of 200 mA/g.

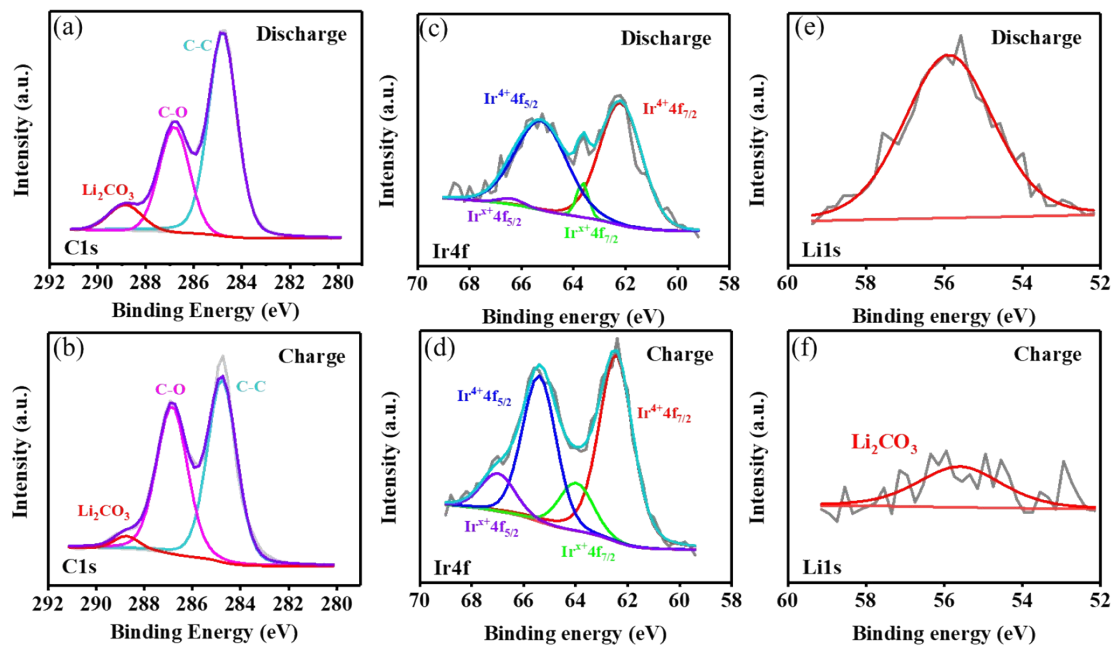


Figure S9. (a, b) C1s; (c, d) Ir4f and (e, f) Li1s XPS spectra of the cathodes after 20th discharge and charge with a curtailing capacity of 400 mAh/g at 100 mA/g.

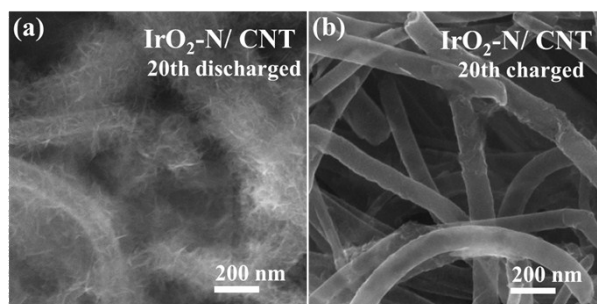


Figure S10. SEM of $\text{IrO}_2\text{-N/CNT}$ cathodes (a) after 20th discharged; (b) after 20th charged with a limited capacity of 400 mAh/g at 100 mA/g.

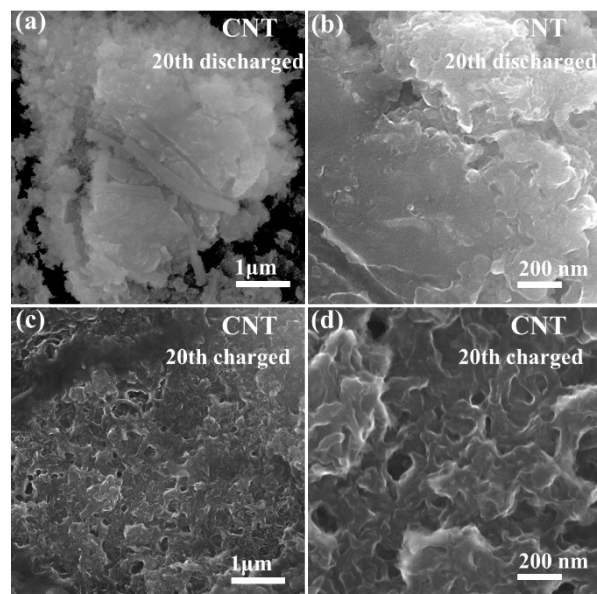


Figure S11. (a, b) SEM of CNT cathodes after 20th discharged (c, d) 20th charged with a limited capacity of 400 mAh/g at 100 mA/g.

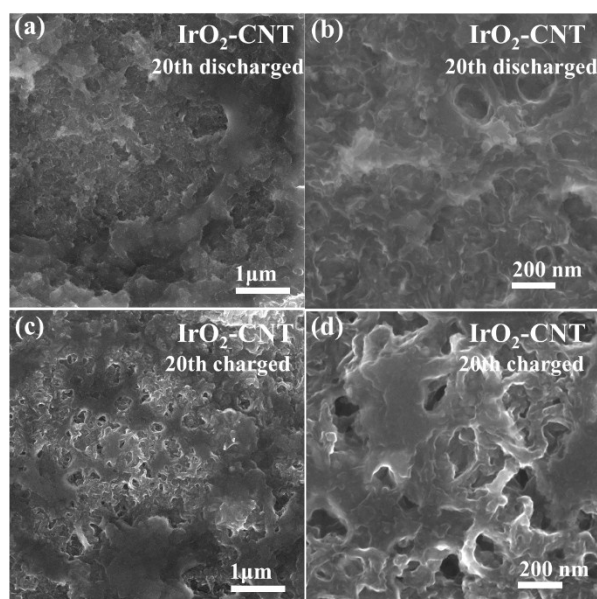


Figure S12. (a, b) SEM of IrO₂-CNT cathodes after 20th discharged (c, d) 20th charged with a limited capacity of 400 mAh/g at 100 mA/g.

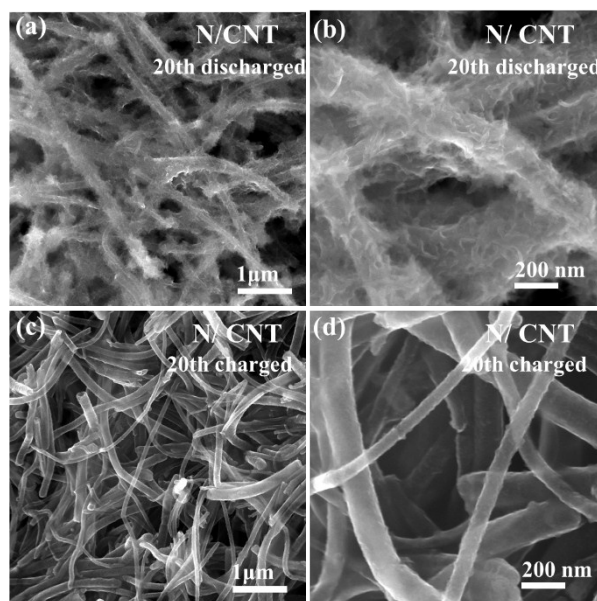


Figure S13. (a, b) SEM of N/CNT cathodes after 20th discharged (c, d) 20th charged with a limited capacity of 400 mAh/g at 100 mA/g.

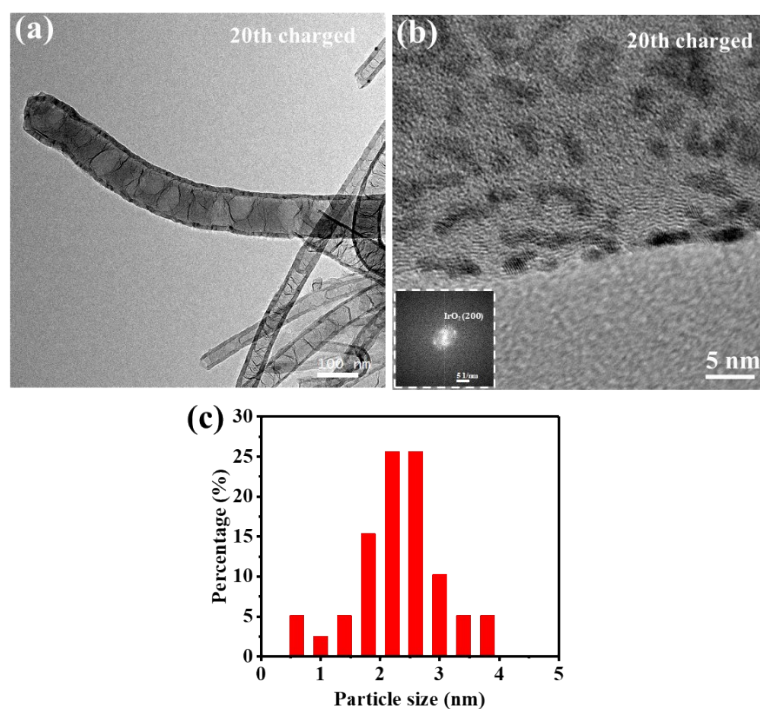


Figure S14. (a) TEM and (b) HRTEM of IrO₂-N/CNT cathodes after 20th charged with a limited capacity of 400 mAh/g at 100 mA/g. (c) The particle size of IrO₂ NPs on IrO₂-N/CNT surface.

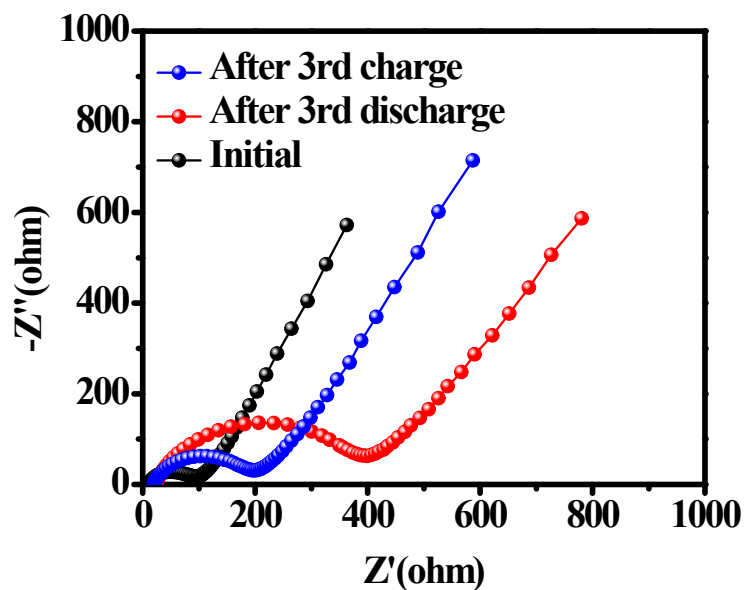


Figure S15. EIS curves of IrO₂-N/CNT cathode after 3rd discharge and charge at 100 mA/g.

Table S1. Comparison and summary of recent literatures on electrochemical performance of Li-CO₂ batteries with different cathodes at room temperature.

Materials	Overpotential (V)	Charge Plateau (V)	Battery operation time (h)	References
IrO₂-N/CNT	1.10 (100 mA/g, 400 mA/g)	3.69	2500	This work
N-CNT@Ti	1.37 (100 mA/g, 1000 mAh/g)	4.2	1000	Ref.S1
Ir NSs-CNFs	1.05 (500 mA/g, 1000 mAh/g)	3.8	1600	Ref.S2
MoS ₂ NFs	0.7 (100 mA/g, 500 mAh/g)	-	2000	Ref.S3
B-NCNT	1.33 (100 mA/g, 1000 mAh/g)	4.05	800	Ref.S4
CNT@RuO ₂	1.51 (100 mA/g, 500 mAh/g)	3.97	300	Ref.S5
Ru@Super P	1.71 (100 mA/g, 1000 mAh/g)	4.25	1400	Ref.S6
Ir/CNFs	1.38 (100 mA/g, 1000 mAh/g)	4.14	540	Ref.S7
NiO-CNT	1.40 (100 mA/g, 1000 mAh/g)	4.1	500	Ref.S8

IrO ₂ /MnO ₂	- (100 mA/g)	4.0	1890	Ref.S9
RuP ₂ -NPCF	1.30 (200 mA/g,1000 mAh/g)	4.0	2000	Ref.S10
Ru/NC	2.35 (100 mA/g,500 mAh/g)	4.1	1500	Ref.S11
CNTs	1.6 (100 mA/g,1000 mAh/g)	4.3	440	Ref.S12
Graphene	1.23 (100 mA/g,1000 mAh/g)	4.2	200	Ref.S13

References :

- 1 Y. Li, J. Zhou, T. Zhang, T. Wang, X. Li, Y. Jia, J. Cheng, Q. Guan, E. Liu, H. Peng and B. Wang, *Adv Funct Mater*, 2019, **29**, 1808117.
- 2 Y. Xing, Y. Yang, D. Li, M. Luo, N. Chen, Y. Ye, J. Qian, L. Li, D. Yang, F. Wu, R. Chen and S. Guo, *Adv mater*, 2018, **30**, e1803124.
- 3 R. E. W. Alireza Ahmadiparidari, Leily Majidi, Mohammad Asadi, J. R. J. Amir Chamaani, Sina Rastegar, Zahra Hemmat, Baharak Sayahpour,, B. N. Rajeev S. Assary, Pedram Abbasi, Paul C. Redfern, Anh Ngo, and J. G. Márton Vörös, Robert Klie, Larry A. Curtiss, and Amin Salehi-Khojin, *Adv Mater*, 2019, 1902518.
- 4 X. Li, J. Zhou, J. Zhang, M. Li, X. Bi, T. Liu, T. He, J. Cheng, F. Zhang, Y. L. X. Mu, J. Lu, B. Wang., *Adv Mater*, 2019, 1903852.
- 5 S. Bie, M. Du, W. He, H. Zhang, Z. Yu, J. Liu, M. Liu, W. Yan, L. Zhou and Z. Zou, *ACS appl mater inter*, 2019, **11**, 5146-5151.
- 6 S. Yang, Y. Qiao, P. He, Y. Liu, Z. Cheng, J. Zhu and H. Zhou, *Energy Environ. Sci.*, 2017, **10**, 972-978.
- 7 C. Wang, Q. Zhang, X. Zhang, X. G. Wang, Z. Xie and Z. Zhou, *Small*, 2018, **14**, e1800641.
- 8 X. Zhang, C. Wang, H. Li, X.-G. Wang, Y.-N. Chen, Z. Xie and Z. Zhou, *J. Mater Chem A*, 2018, **6**, 2792-2796.
- 9 Y. Mao, C. Tang, Z. Tang, J. Xie, Z. Chen, J. Tu, G. Cao and X. Zhao, *Energy Storage Materials*, 2019, **18**, 405-413.
- 10 Z. Guo, J. Li, H. Qi, X. Sun, H. Li, A. G. Tamirat, J. Liu, Y. Wang and L. Wang, *Small*, 2018, 1803246.
- 11 P. Zhang, Y.. Lu, Y. Wu, Z.. Yin, J. Li, Y. Zhou, Y. Hong, Y. Li, L. Huang and S. Sun, *Chem. Eng. J.*, 2019, **363**, 224-233.
- 12 X. Zhang, Q. Zhang, Z. Zhang, Y. Chen, Z. Xie, J. Wei, Z. Zhou, *Chem. Commun.*, 2015, **51**, 14636-14639
- 13 Z. Zhang, Q. Zhang, Y. Chen, J. Bao, X. Zhou, Z. Xie, J. Wei and Z. Zhou, *Angew. Chem. Int. Ed.* 2015, **54**, 6550-6553.