

Supporting Information

Polypyrrole Nanostructures Modified with Mono- and Bimetallic Nanoparticles for Photocatalytic H₂ Generation

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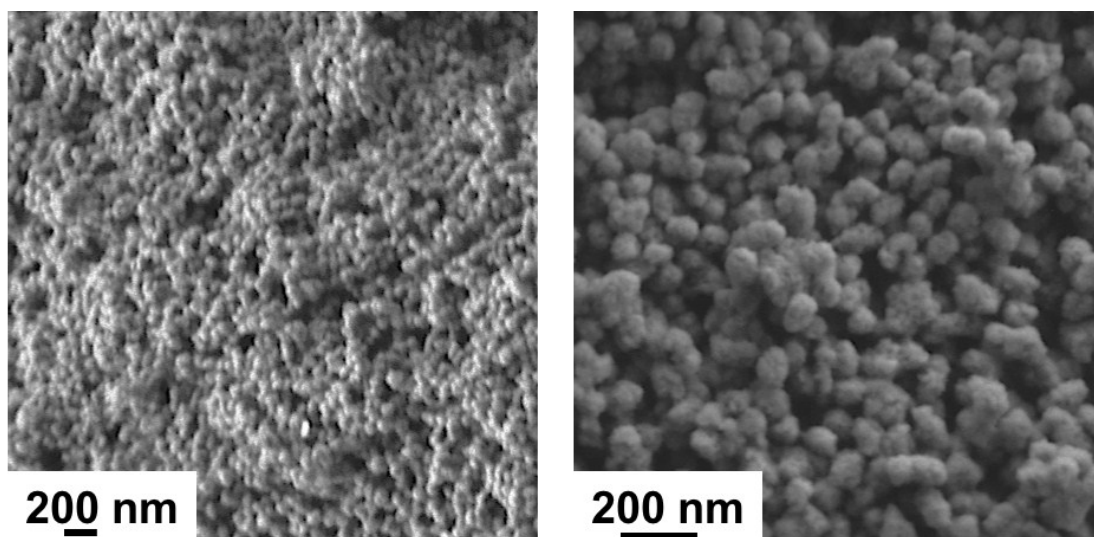


Figure S1. SEM images of PPy NSs.

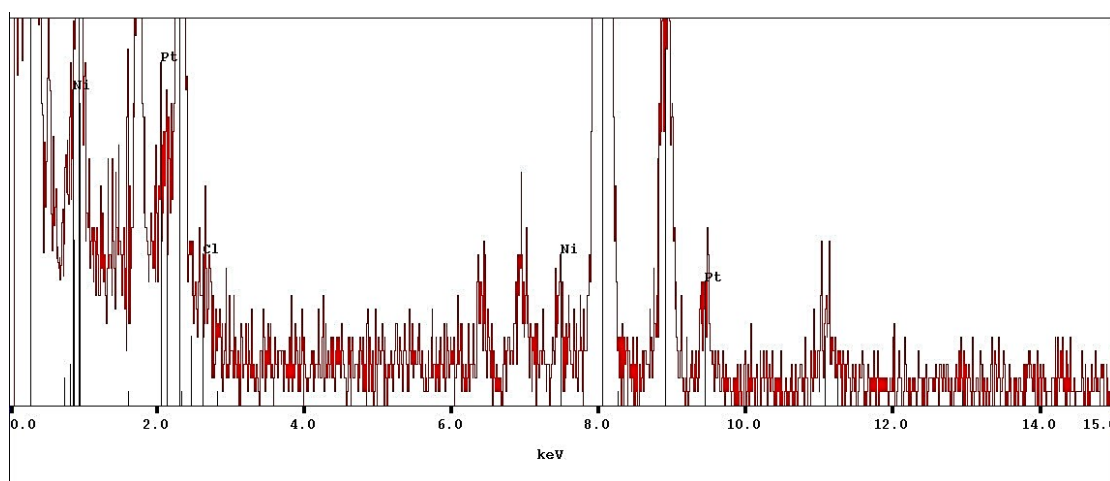


Figure S2. EDS image of 0.05%Ni0.05%Pt-PPy

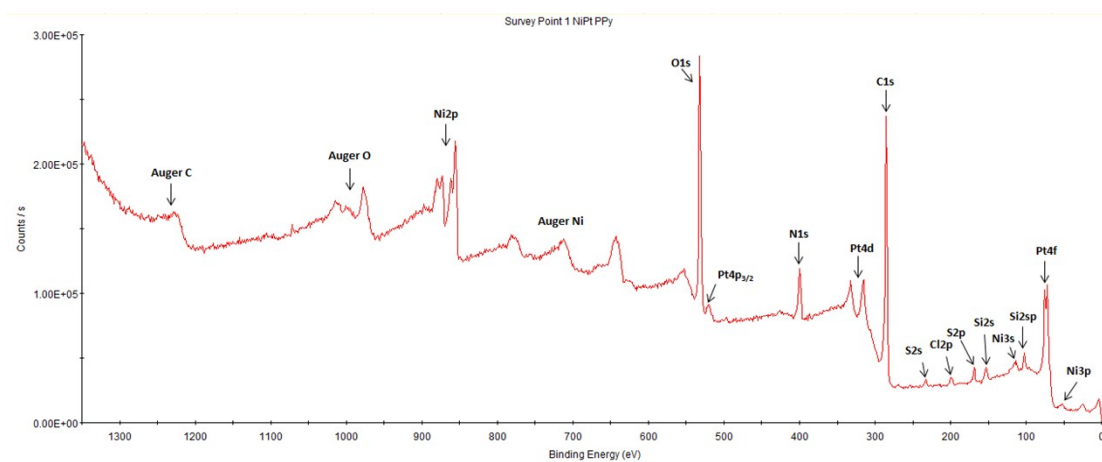


Figure S3. XPS spectra of wide region spectroscopy of PtNi-PPy-NSs.

Table S1. Hydrogen production of 0.2%Pt-PPy and 5%Ni-PPy with different metallic precursors.

Samples	0.2%Pt-PPy	Hydrogen ($\mu\text{mol/h/g}$)	5%Ni-PPy	Hydrogen ($\mu\text{mol/h/g}$)
	Platinum(II) acetylacetonate	1400	Nickel (II) acetylacetonate	289
	Potassium tetrachloroplatinate	398	Nickel formate	200

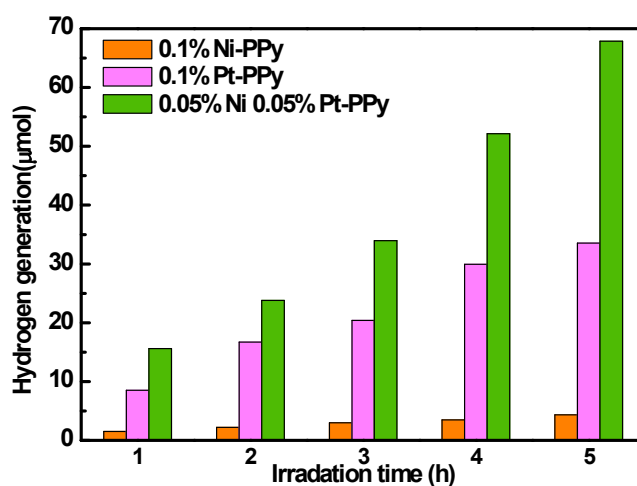


Figure S4. Hydrogen evolution of PtNi-PPy samples with same loading rate.

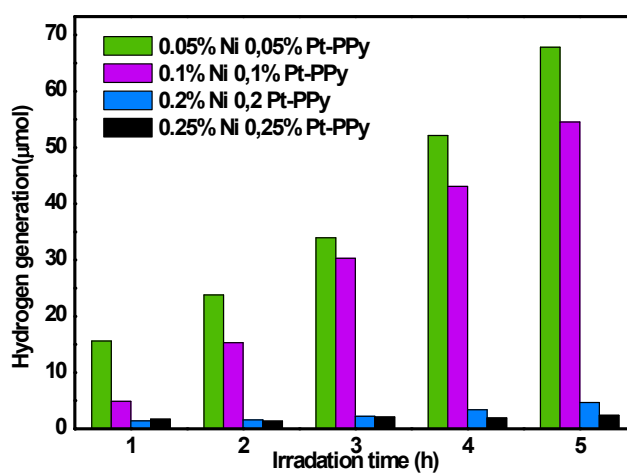


Figure S5. Hydrogen evolution of PtNi-PPy samples with different loading rates.

Table S2. Table 1. Comparison of this work with previously reported materials for H₂ generation

Photocatalyst	Scavenger	Cocatalyst	Light source	hydrogen	Reference
TiO ₂ /ALD-Pt	Ethanol	Pt	300 W Xe lamp	23 μL/h (AM 1.5)	J. Yoo, et al [1]
TiO ₂ /NiS nanofiber	Methanol	NiS	350 W Xe lamp	655 μmol/h/g	F. Xu, et al [2]
2%Au/TiO ₂	Methanol	Au	150 W CERAMICMetal- Halide Lamp	500 μmol/h/g	F. Xu, et al [3]
Ag-TiO ₂ - Graphene	Methanol	Ag	300 W Xe lamp	129.5 μmol/h/g	F. Sheu, et al [4]
10%MoS ₂ /CdS	Acetic acid	MoS ₂	300 W Xe arc lamp	45 mmol/h/g	X. Yin, et al [5]
3%P3HT-g-C ₃ N ₄	S ²⁻ +SO ₂ - 3		300 W Xe lamp	5700 μmol/h/g	X. Zhang, et al [6]
1%Pt/PTh-20	ascorbic acid		300 W Xe arc lamp	2190 μmol/h/g	X. Zong, et al [7]
C ₃ N ₄ - 2%PEDOT-1%Pt	TEA	Pt	300 W Xe lamp	320 μmol/h/g	Z. Xing, et al [8]
Pt/5 wt % g- PAN/g-C ₃ N ₄	TEOA	Pt	300 W Xe lamp	370 μmol/h/g	F. He, et al [9]
0.1%PtNi-PPy	CH ₃ OH	Pt, Ni	300 W Xe lamp	664 μmol/h/g	X. Yuan, et al This work

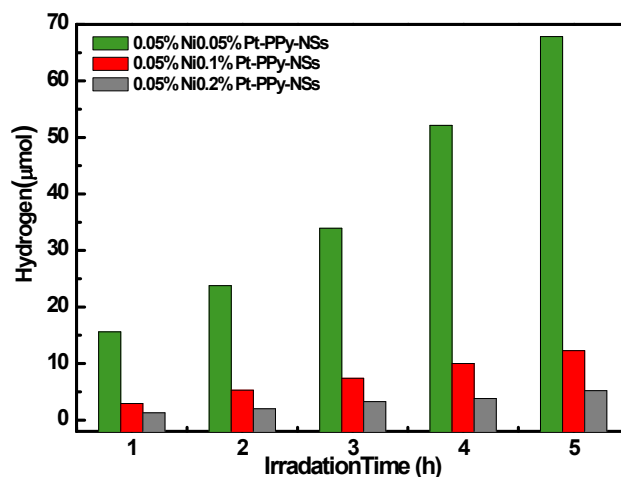


Figure S6. Hydrogen evolution of PtNi-PPy samples with different mass ratio of Pt:Ni.

Reference:

- [1] J. Yoo, R. Zazpe, G. Cha, J. Prikryl, I. Hwang, J.M. Macak, P. Schmuki, Uniform ALD deposition of Pt nanoparticles within 1D anodic TiO₂ nanotubes for photocatalytic H₂ generation, *Electrochemistry Communications*, 86 (2018) 6-11.
- [2] F. Xu, L. Zhang, B. Cheng, J. Yu, Direct Z-scheme TiO₂/NiS core-shell hybrid nanofibers with enhanced photocatalytic H₂-production activity, *ACS Sustainable Chemistry & Engineering*, 6 (2018) 12291-12298.
- [3] C. Marchal, A. Piquet, M. Behr, T. Cottineau, V. Papaefthimiou, V. Keller, V. Caps, Activation of solid grinding-derived Au/TiO₂ photocatalysts for solar H₂ production from water-methanol mixtures with low alcohol content, *Journal of catalysis*, 352 (2017) 22-34.
- [4] F.-J. Sheu, C.-P. Cho, Investigation of the appropriate content of graphene in AgTiO₂-graphene ternary nanocomposites applied as photocatalysts, *International Journal of Hydrogen Energy*, 42 (2017) 17020-17029.
- [5] X.-L. Yin, L.-L. Li, W.-J. Jiang, Y. Zhang, X. Zhang, L.-J. Wan, J.-S. Hu, MoS₂/CdS nanosheets-on-nanorod heterostructure for highly efficient photocatalytic H₂ generation under visible light irradiation, *ACS applied materials & interfaces*, 8 (2016) 15258-15266.
- [6] X. Zhang, B. Peng, S. Zhang, T. Peng, Robust wide visible-light-responsive photoactivity for H₂ production over a polymer/polymer heterojunction photocatalyst:

the significance of sacrificial reagent, ACS Sustainable Chemistry & Engineering, 3 (2015) 1501-1509.

[7] X. Zong, X. Miao, S. Hua, L. An, X. Gao, W. Jiang, D. Qu, Z. Zhou, X. Liu, Z. Sun, Structure defects assisted photocatalytic H₂ production for polythiophene nanofibers, Applied Catalysis B: Environmental, 211 (2017) 98-105.

[8] Z. Xing, Z. Chen, X. Zong, L. Wang, A new type of carbon nitride-based polymer composite for enhanced photocatalytic hydrogen production, Chemical Communications, 50 (2014) 6762-6764.

[9] F. He, G. Chen, Y. Yu, S. Hao, Y. Zhou, Y. Zheng, Facile approach to synthesize g-PAN/g-C₃N₄ composites with enhanced photocatalytic H₂ evolution activity, ACS applied materials & interfaces, 6 (2014) 7171-7179.