

Supporting Information

Towards an efficient anode material for Li-ion batteries: understanding the conversion mechanism of nickel hydroxy chloride with Li -ions

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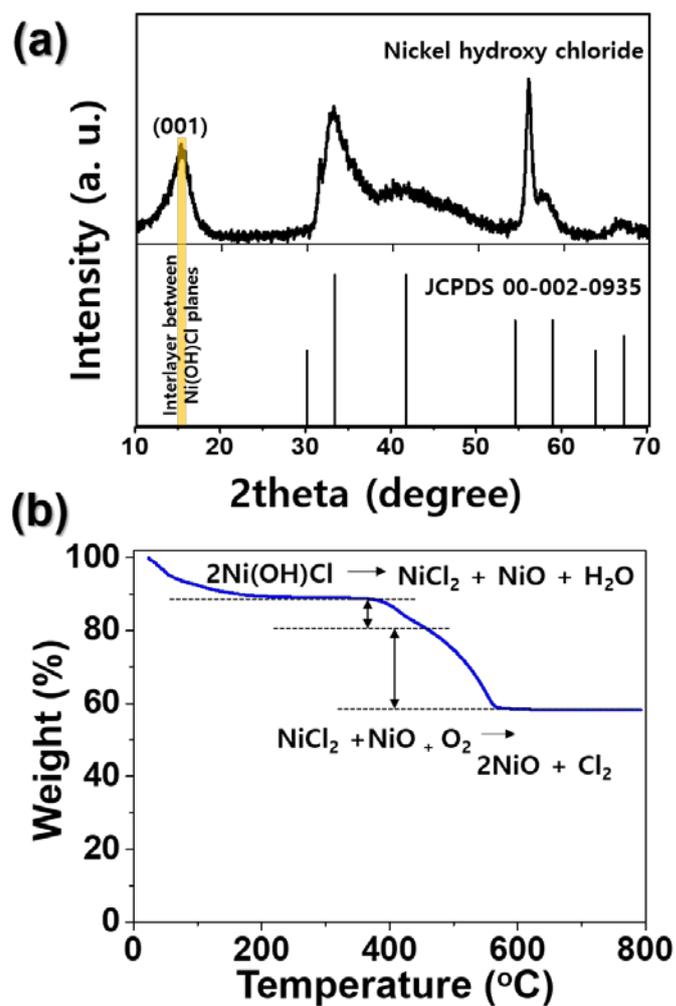


Fig. S1 (a) XRD pattern and (b) TG curve measured under air atmosphere of Ni(OH)Cl.

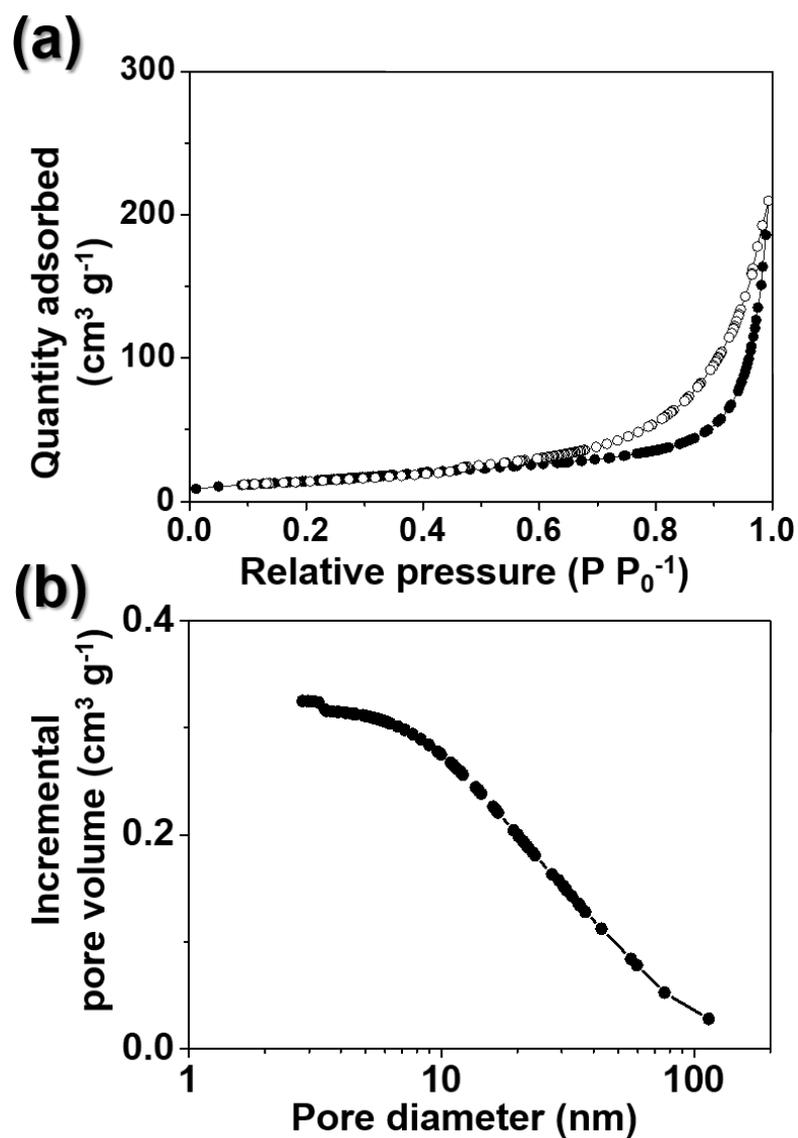


Fig. S2 (a) N₂ gas adsorption and desorption isotherm and (b) pore size distribution of Ni(OH)Cl.

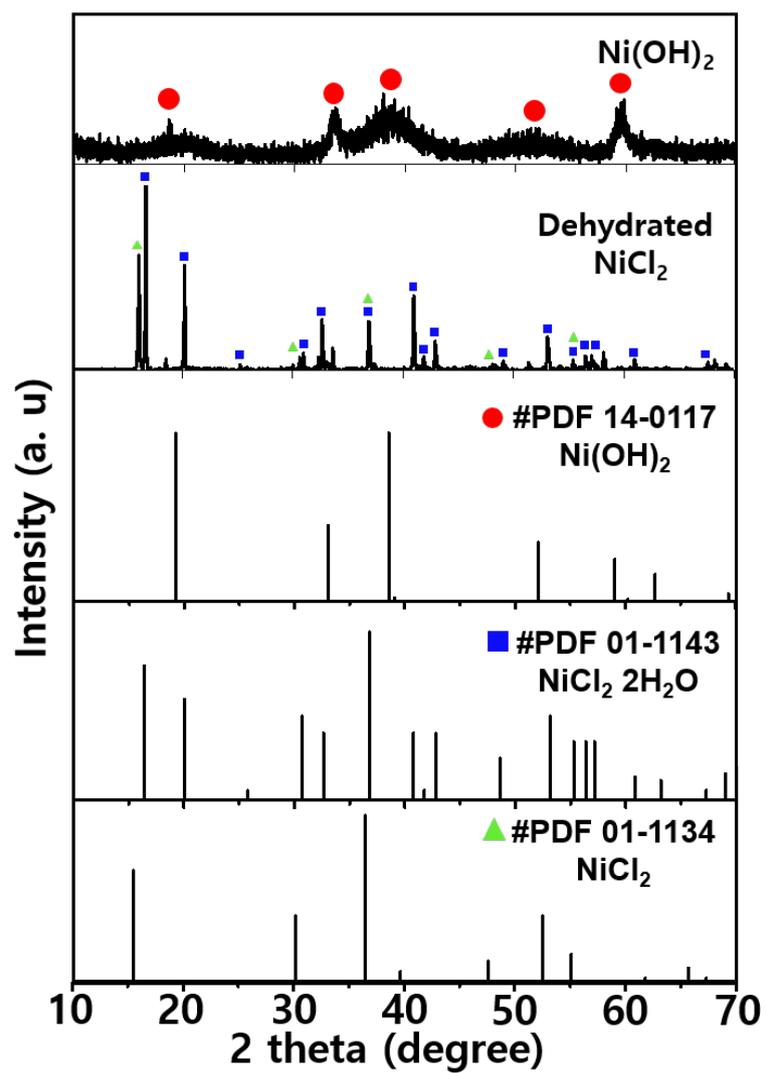


Fig. S3 XRD patterns of Ni(OH)_2 and dehydrated NiCl_2 .

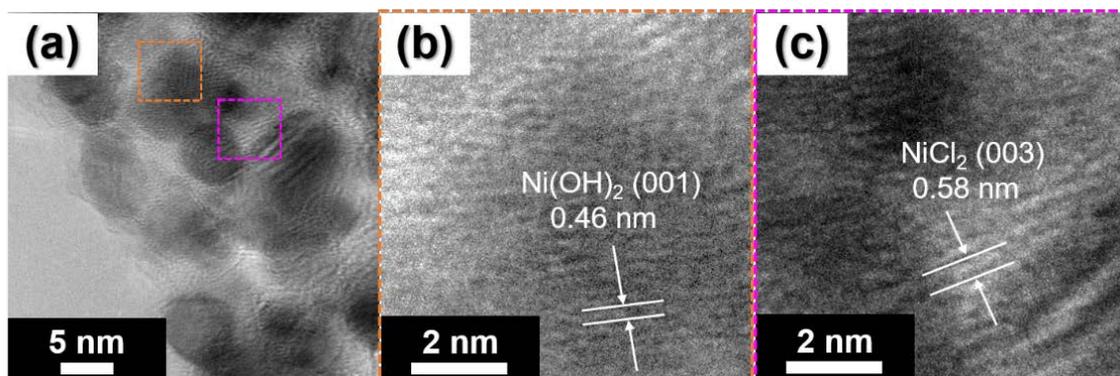
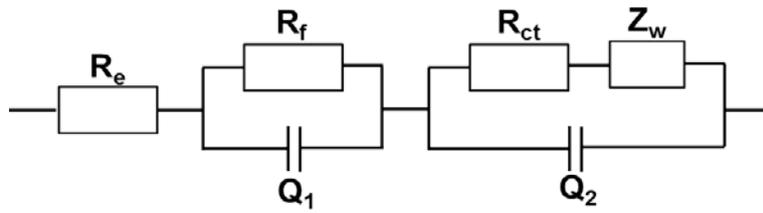


Fig. S4 Ex-situ HR-TEM images of Ni(OH)₂ and NiCl₂ at the fully charged states.



R_e : the electrolyte resistance, corresponding to the intercept of high frequency semicircle at Z_{re} axis

R_f : the SEI layer resistance corresponding to the high-frequency semicircle

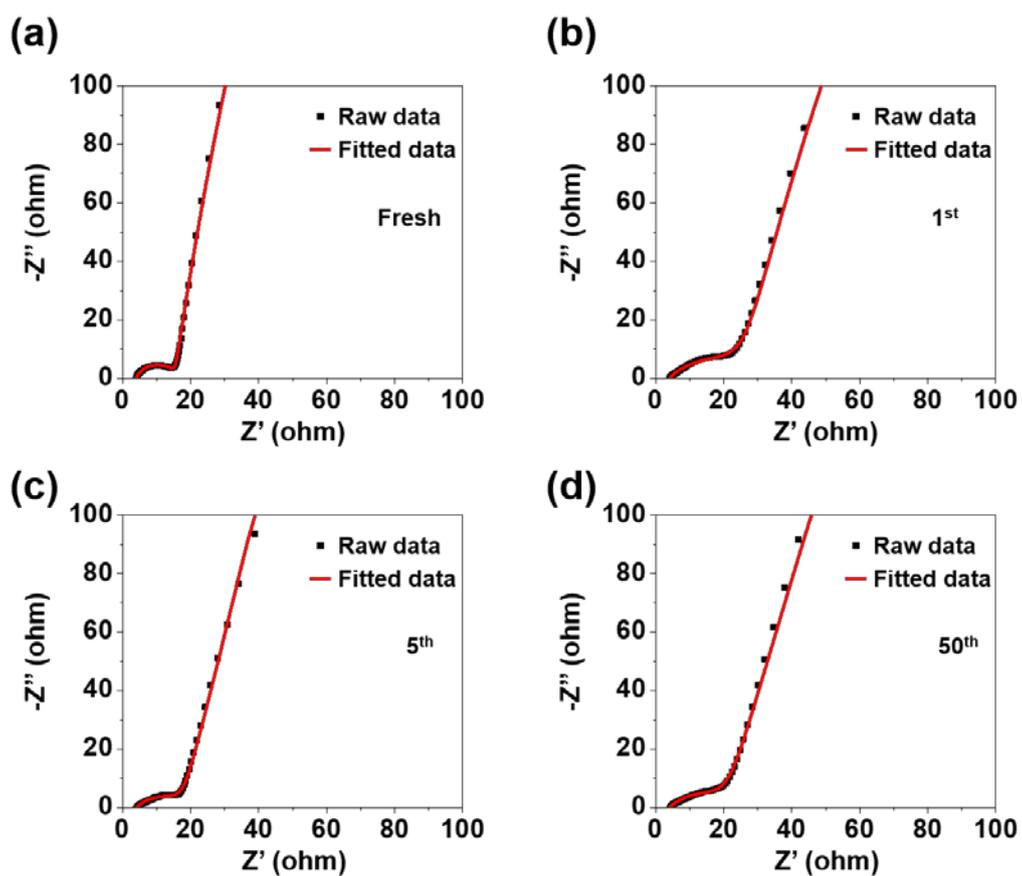
Q_1 : the dielectric relaxation capacitance corresponding to the high-frequency semicircle

R_{ct} : the denote the charger transfer resistance related to the middle-frequency semicircle

Q_2 : the associated double-layer capacitance related to the middle-frequency semicircle

Z_w : the Li-ion diffusion resistance

Fig. S5 Equivalent circuit model used for ac impedance fitting.



| Cycle | R_s | R_f | R_{ct} |
|------------------|-------|-------|----------|
| Fresh | 4.1 | - | 12.5 |
| 1 st | 4.0 | 11.8 | 15.3 |
| 5 th | 4.1 | 10.7 | 7.0 |
| 50 th | 4.0 | 12.0 | 12.9 |

Fig. S6 Nyquist plots and fitted curves and equivalent element parameters obtained from the model of equivalent circuit at fresh state and after 1st, 5th, and 50th cycle.