

Supporting Information

High-performance 3 V “water in salt” aqueous asymmetric supercapacitors based on VN nanowires electrode

Mingyu Ma¹, Zude Shi¹, Yan Li¹, Yifan Yang¹, Yaxiong Zhang¹, Yin Wu¹, Hao Zhao², and Erqing Xie^{1*}

Affiliations

1. School of Physical Science & Technology, Lanzhou University, Lanzhou, 730000, China.
2. School of Science and Technology for Opto-Electronic Information, Yantai University, Yantai, Shandong 264005, China.

* Corresponding author: xieeq@lzu.edu.cn.

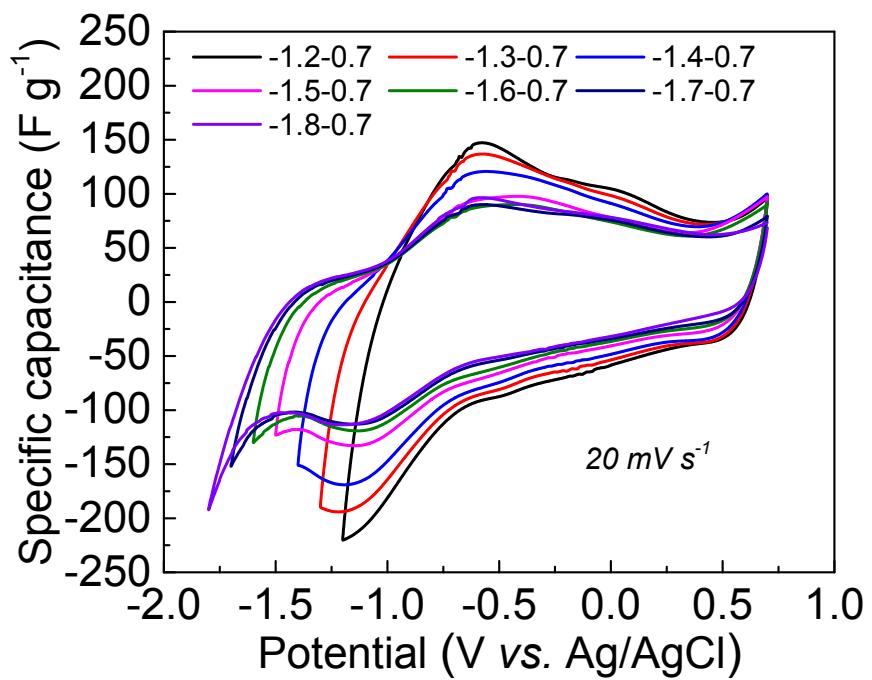


Figure S1. the CV curves of VN-NWs@CC in 21 m LiTFSI aqueous electrolyte at different potential ranges

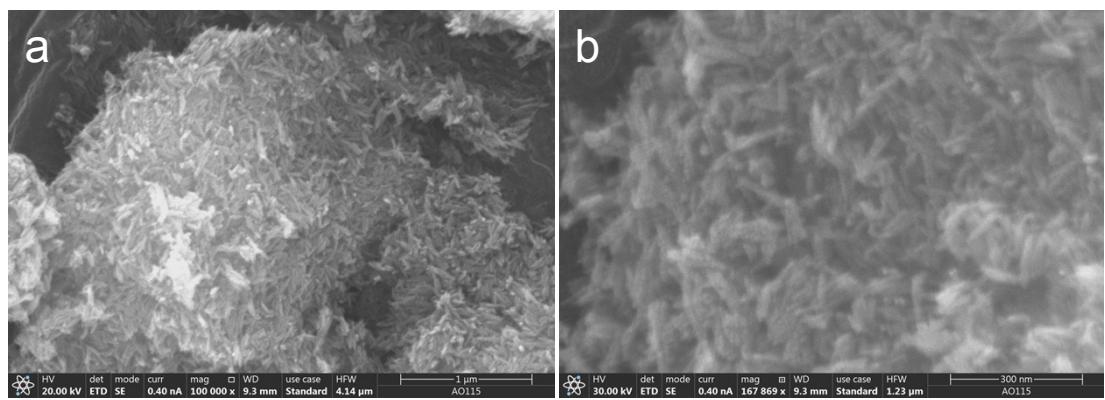


Figure S2. The SEM images of magnification of (a) 100000 \times and (b) 167869 \times for the α -Fe₂O₃ materials.

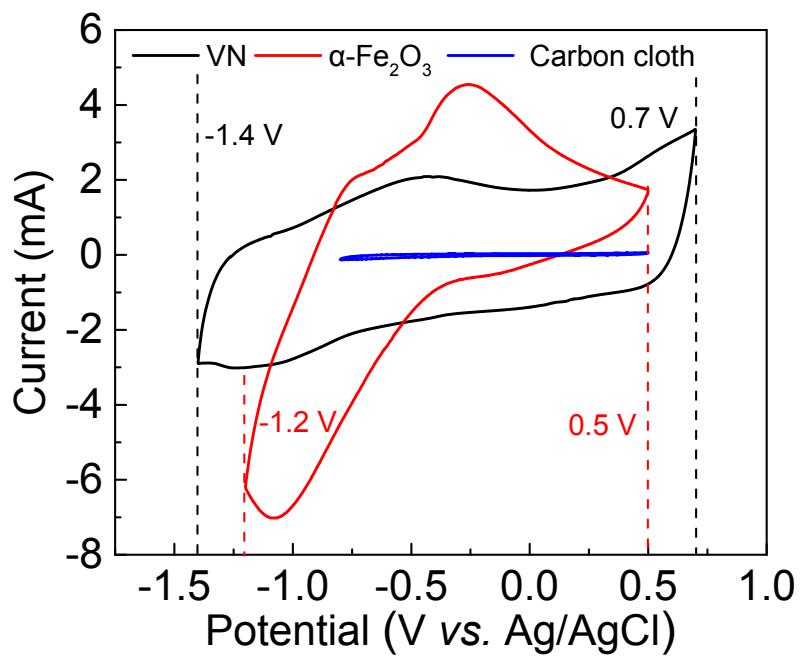


Figure S3. The CV curves of VN, $\alpha\text{-Fe}_2\text{O}_3$, and Carbon cloth electrode in 21m LITFSI electrolyte at a scan rate of 20 mV s^{-1} .

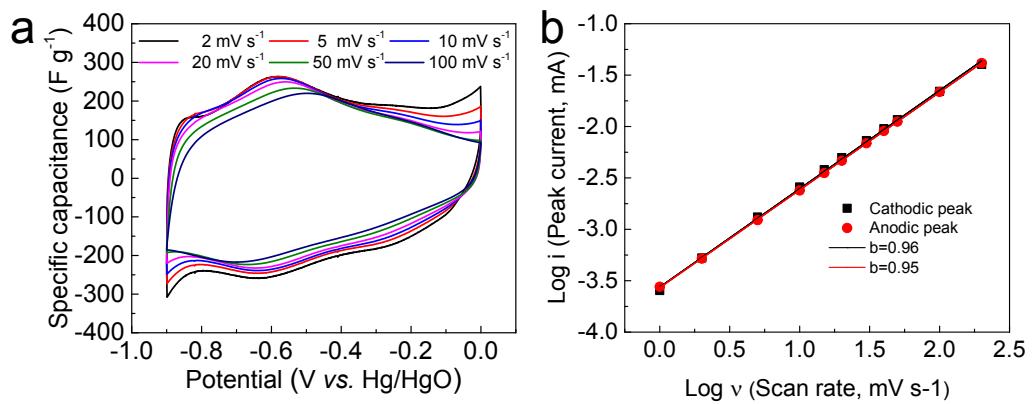


Figure S4. (a) CV curves of VN-NWs@CC at various scan rates in aqueous electrolyte; (b) b-value analysis using the relationship between the peak currents and the scan rates.

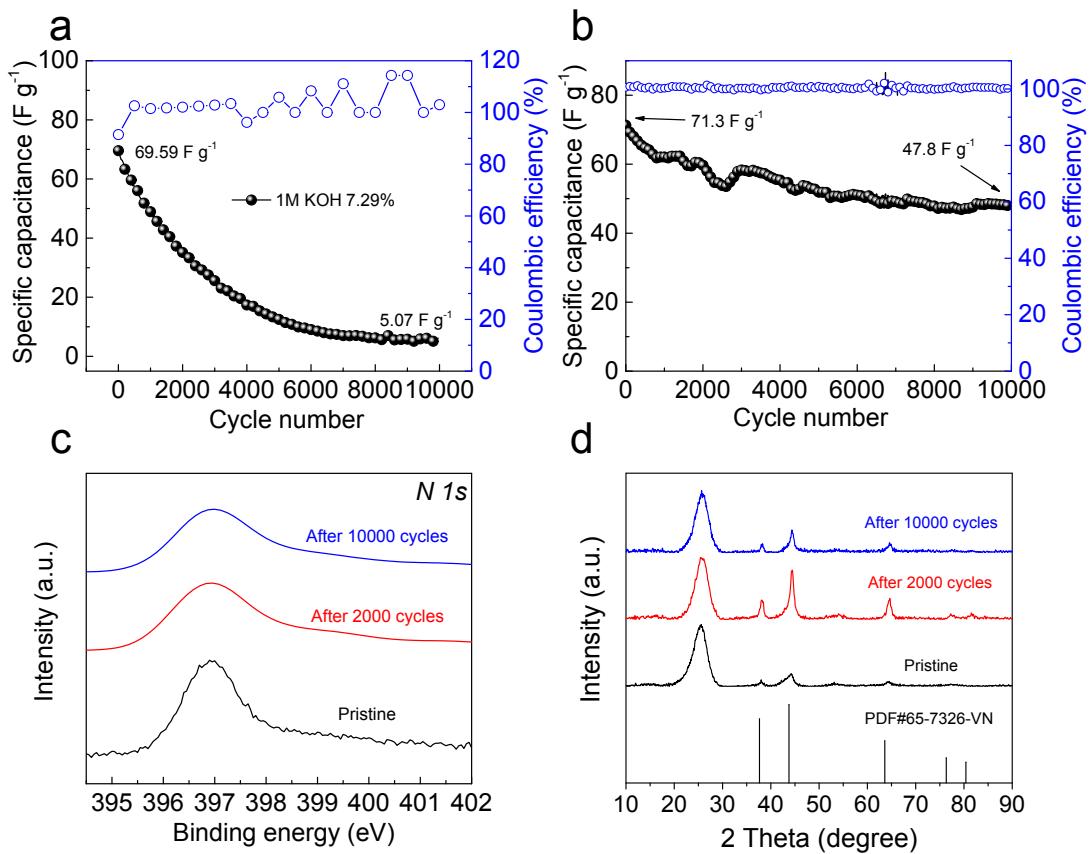


Figure S5. The cycling performance of VN-NWs@CC electrode for 10,000 charge/discharge tests (a) in 1M KOH electrolyte at a current density of 20 A g^{-1} , (b) in 21 m LITFSI electrolyte at a current density of 5 A g^{-1} , (c) the high-resolution XPS spectra of N 1s and (d) XRD patterns for pristine VN, after 2000 cycles, and after 10000 cycles in 21 m LITFSI electrolyte.

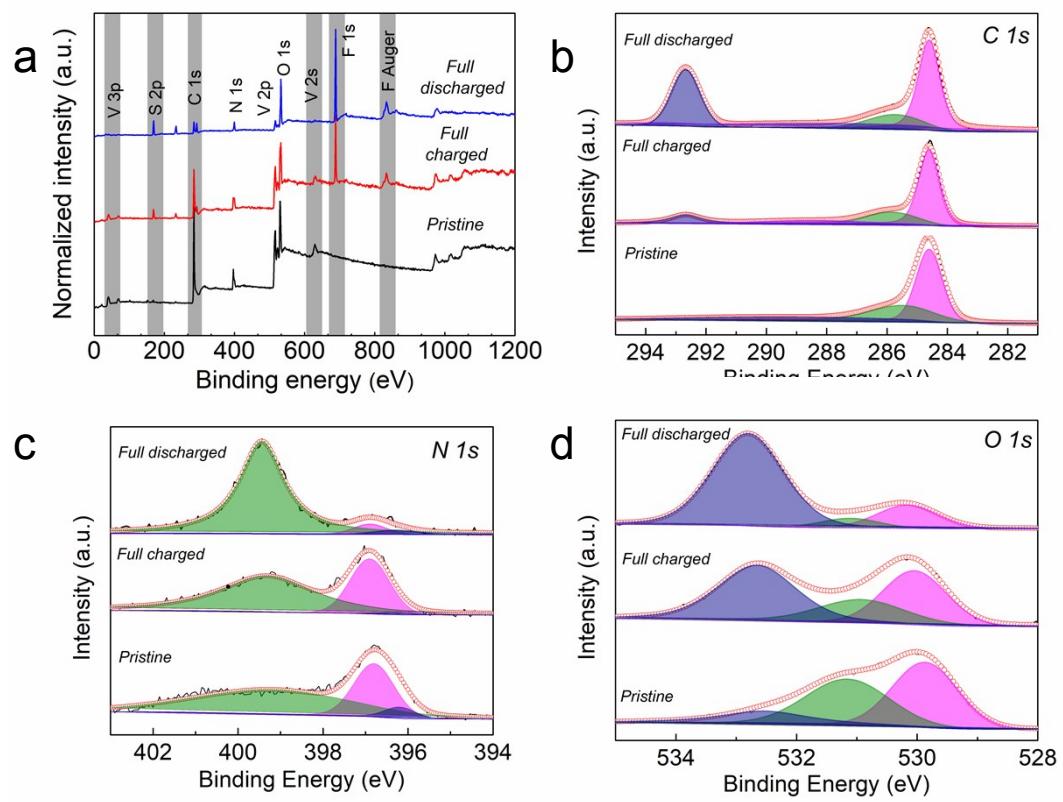


Figure S6. (a) The XPS spectra of full spectrum, high-resolution (b) C 1s, (c) N 1s, and (d) O 1s at pristine, full charge, and full discharge state.

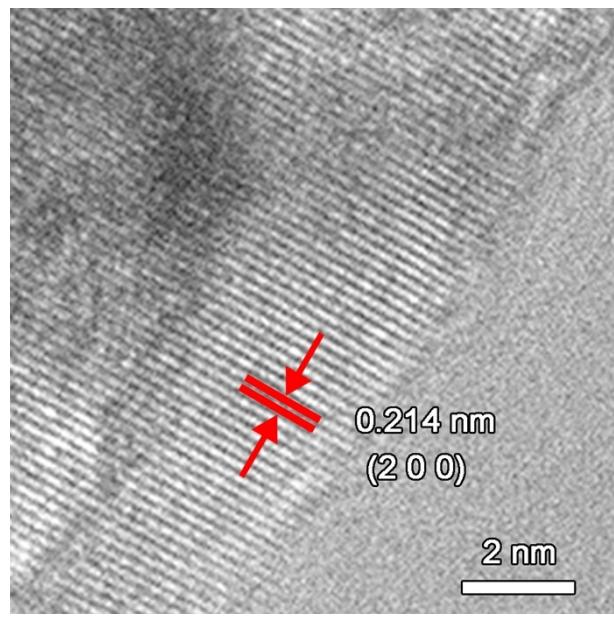


Figure S7. The high-resolution TEM image of VN-NWs@CC of full-discharged state.

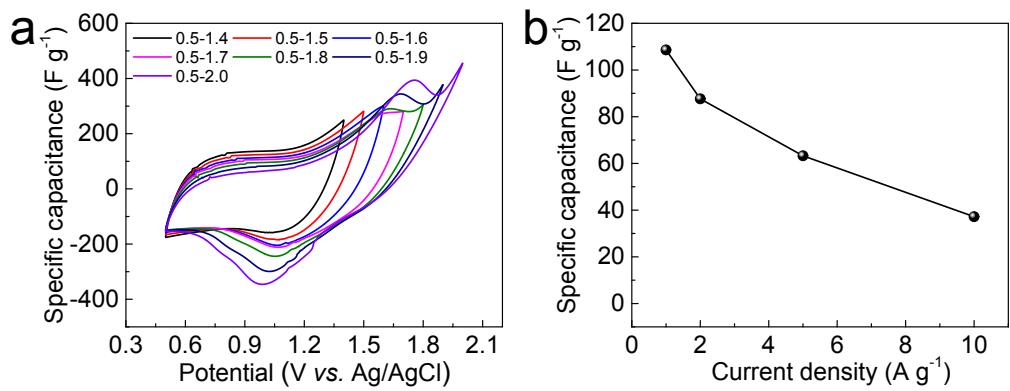


Figure S8. (a) The CV curves of VN-NWs@CC in 21 m LITFSI aqueous electrolyte at different potential ranges, (b) the specific capacitance as a function of current densities for the MnO_2 electrode at 21m LITFSI electrolyte.

Table S1. The comparison of fitting result derived from the impedance spectra with EIS analysis.

Samples	VN	$\alpha\text{-Fe}_2\text{O}_3$	Carbon cloth
R_s (ohm)	5.975	6.142	5.523
R_{ct} (ohm)	10.09	9.95	10.02
CPE_{EDL} (S sec ⁿ)	2.89×10^{-7}	6.91×10^{-7}	5.11×10^{-7}
n_{EDL}	0.9682	0.9279	0.9172
CPE_p (S sec ⁿ)	0.1067	0.0025	0.0069
n_p	0.9878	0.8683	0.7879
Z_w (ohm sec ^{-1/2})	6.96	11.50	17.45

The R_s refers to resistance of electrode and electrolyte, R_{ct} refers to the charge-transfer resistance, Z_w is Warburg impedance corresponding to semi-finite diffusion, and CPE is the constant phase element.

Table S2. The comparison of Warburg coefficient (σ), and chemical diffusion coefficient (D) for VN, $\alpha\text{-Fe}_2\text{O}_3$, and Carbon cloth samples, respectively.

Samples	σ (ohm s ^{-1/2})	D (cm ² s ⁻¹)
VN	15.17	5.20×10^{-14}
$\alpha\text{-Fe}_2\text{O}_3$	32.20	3.17×10^{-14}
Carbon cloth	36.40	3.07×10^{-15}