

Supporting Information

**Directly Predicting Limiting Potentials from Easily Obtainable Physical Properties of Graphene-Supported Single-Atom Electrocatalysts by Machine Learning**

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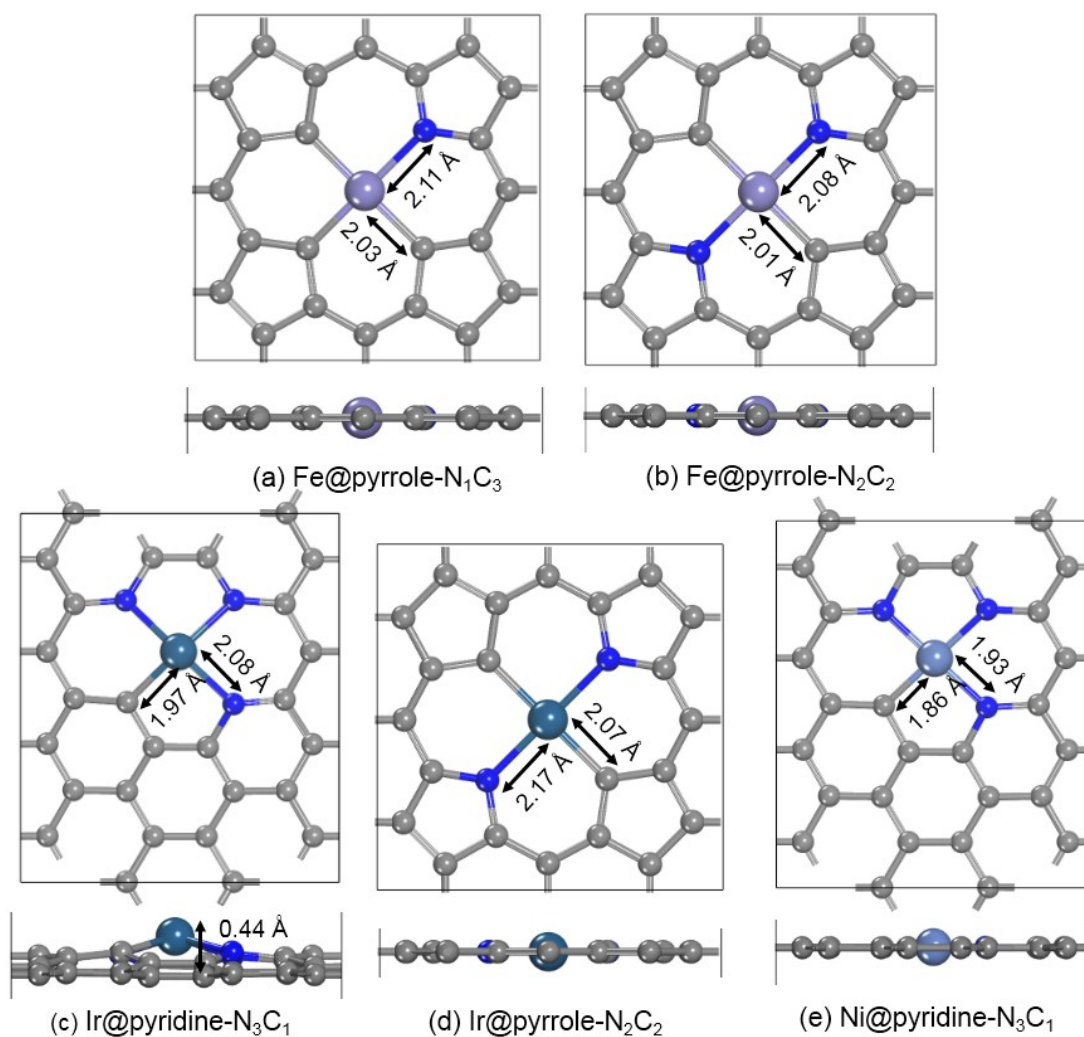
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**Data pretreatment.** For the pretreatment of data, the data were divided into training data and test data randomly with a ratio around 0.85 for training data. Therefore, 104 input data were split into 88 training set and 16 test set for ORR/HER models. For OER model, the small data set for M-N-pyrrole systems affect the training performance, thus, the small data set (with 26 data) was amplified three times with random noises. Totally, 156 input data were produced for OER model, which were divided into 135 training data and 21 validation data.

**DFT computations for catalysis performance.** The changes of Gibbs free energy ( $\Delta G$ ) for each elementary step along the ORR/OER were evaluated using the computational hydrogen electrode (CHE) model developed by Nørskov and co-workers.<sup>1</sup> The computed electronic energy was converted into Gibbs free energy by adding zero-point energy. The energy of the triplet O<sub>2</sub> molecule cannot be accurately computed by the DFT method<sup>2,3</sup> thus, its free energy was computed relative to the free energies of H<sub>2</sub>O(*l*) and H<sub>2</sub>(*g*). The chemical potential of the H<sup>+</sup>/e<sup>-</sup> pair is equal to half of the gas-phase H<sub>2</sub> at standard hydrogen electrode (SHE) conditions.

**Github website link.** The three well-trained models were uploaded to Github: [https://github.com/shirulin1026/ML\\_screening\\_ORROERHER](https://github.com/shirulin1026/ML_screening_ORROERHER), which is public now. The model files can be downloaded and employed for predictions through scikit-learn package. Training data and prediction data were also uploaded in the same folder on Github.



**Figure S1.** The geometries of ML-selected best SACs, (a, b) for ORR, (c, d) for OER, and (e) for HER.

**Table S1.** The structural compositions of 260 other graphene-supported SACs ( $M@N_xC_y$ ) and their limiting potentials towards ORR/OER/HER predicted by ML models.

| Num | Metal | Base                          | ORR   | OER  | HER   |
|-----|-------|-------------------------------|-------|------|-------|
| 1   | Sc    | C <sub>2</sub> N <sub>1</sub> | -2.19 | 3.78 | -2.16 |
| 2   | Ti    | C <sub>2</sub> N <sub>1</sub> | -1.85 | 3.25 | -1.90 |
| 3   | V     | C <sub>2</sub> N <sub>1</sub> | -1.31 | 3.00 | -1.35 |
| 4   | Cr    | C <sub>2</sub> N <sub>1</sub> | -0.98 | 3.05 | -0.83 |
| 5   | Mn    | C <sub>2</sub> N <sub>1</sub> | -0.99 | 2.80 | -0.96 |

| Num | Metal | Base                          | ORR   | OER  | HER   |
|-----|-------|-------------------------------|-------|------|-------|
| 6   | Fe    | C <sub>2</sub> N <sub>1</sub> | -0.53 | 3.14 | -0.55 |
| 7   | Co    | C <sub>2</sub> N <sub>1</sub> | -0.24 | 3.05 | -0.35 |
| 8   | Ni    | C <sub>2</sub> N <sub>1</sub> | -0.14 | 3.01 | -0.17 |
| 9   | Cu    | C <sub>2</sub> N <sub>1</sub> | 0.19  | 2.55 | 0.08  |
| 10  | Zn    | C <sub>2</sub> N <sub>1</sub> | 0.10  | 2.33 | 0.01  |
| 11  | Y     | C <sub>2</sub> N <sub>1</sub> | -2.18 | 3.61 | -2.03 |
| 12  | Zr    | C <sub>2</sub> N <sub>1</sub> | -1.77 | 3.17 | -1.72 |
| 13  | Nb    | C <sub>2</sub> N <sub>1</sub> | -1.36 | 3.26 | -1.42 |
| 14  | Mo    | C <sub>2</sub> N <sub>1</sub> | -0.76 | 3.48 | -0.75 |
| 15  | Tc    | C <sub>2</sub> N <sub>1</sub> | -0.52 | 3.35 | -0.41 |
| 16  | Ru    | C <sub>2</sub> N <sub>1</sub> | -0.17 | 2.83 | -0.09 |
| 17  | Rh    | C <sub>2</sub> N <sub>1</sub> | -0.01 | 2.62 | -0.04 |
| 18  | Pd    | C <sub>2</sub> N <sub>1</sub> | 0.28  | 2.58 | 0.34  |
| 19  | Cd    | C <sub>2</sub> N <sub>1</sub> | 0.19  | 2.27 | 0.20  |
| 20  | Hf    | C <sub>2</sub> N <sub>1</sub> | -1.86 | 3.39 | -1.83 |
| 21  | Ta    | C <sub>2</sub> N <sub>1</sub> | -1.65 | 3.24 | -1.66 |
| 22  | W     | C <sub>2</sub> N <sub>1</sub> | -0.93 | 3.16 | -1.16 |
| 23  | Re    | C <sub>2</sub> N <sub>1</sub> | -0.99 | 3.07 | -0.71 |
| 24  | Os    | C <sub>2</sub> N <sub>1</sub> | -0.32 | 3.42 | -0.39 |
| 25  | Ir    | C <sub>2</sub> N <sub>1</sub> | -0.17 | 3.05 | -0.11 |
| 26  | Pt    | C <sub>2</sub> N <sub>1</sub> | 0.17  | 2.68 | 0.07  |
| 27  | Sc    | C <sub>1</sub> N <sub>2</sub> | -2.22 | 3.76 | -2.15 |
| 28  | Ti    | C <sub>1</sub> N <sub>2</sub> | -1.80 | 3.20 | -1.88 |
| 29  | V     | C <sub>1</sub> N <sub>2</sub> | -1.20 | 2.92 | -1.27 |
| 30  | Cr    | C <sub>1</sub> N <sub>2</sub> | -0.65 | 2.82 | -0.57 |
| 31  | Mn    | C <sub>1</sub> N <sub>2</sub> | -0.67 | 2.65 | -0.77 |
| 32  | Fe    | C <sub>1</sub> N <sub>2</sub> | -0.34 | 2.76 | -0.44 |
| 33  | Co    | C <sub>1</sub> N <sub>2</sub> | -0.13 | 2.70 | -0.26 |
| 34  | Ni    | C <sub>1</sub> N <sub>2</sub> | -0.18 | 2.76 | -0.18 |
| 35  | Cu    | C <sub>1</sub> N <sub>2</sub> | -0.07 | 2.53 | -0.06 |
| 36  | Zn    | C <sub>1</sub> N <sub>2</sub> | -0.25 | 2.47 | -0.22 |
| 37  | Y     | C <sub>1</sub> N <sub>2</sub> | -2.22 | 3.79 | -2.02 |
| 38  | Zr    | C <sub>1</sub> N <sub>2</sub> | -1.75 | 3.18 | -1.71 |
| 39  | Nb    | C <sub>1</sub> N <sub>2</sub> | -1.12 | 3.35 | -1.20 |
| 40  | Mo    | C <sub>1</sub> N <sub>2</sub> | -0.45 | 3.02 | -0.49 |
| 41  | Tc    | C <sub>1</sub> N <sub>2</sub> | -0.31 | 3.00 | -0.27 |
| 42  | Ru    | C <sub>1</sub> N <sub>2</sub> | -0.08 | 2.53 | -0.07 |
| 43  | Rh    | C <sub>1</sub> N <sub>2</sub> | -0.12 | 2.49 | -0.09 |
| 44  | Pd    | C <sub>1</sub> N <sub>2</sub> | -0.05 | 2.72 | -0.01 |
| 45  | Cd    | C <sub>1</sub> N <sub>2</sub> | -0.26 | 2.48 | -0.10 |
| 46  | Hf    | C <sub>1</sub> N <sub>2</sub> | -1.85 | 3.39 | -1.82 |
| 47  | Ta    | C <sub>1</sub> N <sub>2</sub> | -1.57 | 3.06 | -1.60 |
| 48  | W     | C <sub>1</sub> N <sub>2</sub> | -0.68 | 3.29 | -0.87 |

| Num | Metal | Base          | ORR   | OER  | HER   |
|-----|-------|---------------|-------|------|-------|
| 49  | Re    | C1N2          | -0.59 | 2.72 | -0.44 |
| 50  | Os    | C1N2          | -0.18 | 2.88 | -0.28 |
| 51  | Ir    | C1N2          | -0.08 | 2.62 | -0.06 |
| 52  | Pt    | C1N2          | -0.04 | 2.59 | 0.02  |
| 53  | Sc    | N3            | -2.15 | 3.72 | -2.17 |
| 54  | Ti    | N3            | -1.80 | 3.26 | -1.86 |
| 55  | V     | N3            | -1.22 | 3.03 | -1.29 |
| 56  | Cr    | N3            | -0.62 | 3.32 | -0.48 |
| 57  | Mn    | N3            | -0.64 | 2.85 | -0.62 |
| 58  | Fe    | N3            | -0.24 | 2.39 | -0.34 |
| 59  | Co    | N3            | -0.05 | 1.93 | -0.18 |
| 60  | Ni    | N3            | 0.00  | 1.86 | -0.08 |
| 61  | Cu    | N3            | 0.19  | 1.89 | 0.11  |
| 62  | Zn    | N3            | 0.03  | 2.03 | -0.02 |
| 63  | Y     | N3            | -2.13 | 3.67 | -2.02 |
| 64  | Zr    | N3            | -1.71 | 3.23 | -1.67 |
| 65  | Nb    | N3            | -1.17 | 3.25 | -1.24 |
| 66  | Mo    | N3            | -0.51 | 3.25 | -0.51 |
| 67  | Tc    | N3            | -0.22 | 2.83 | -0.26 |
| 68  | Ru    | N3            | -0.06 | 2.09 | -0.04 |
| 69  | Rh    | N3            | -0.03 | 1.99 | 0.02  |
| 70  | Pd    | N3            | 0.10  | 2.34 | 0.16  |
| 71  | Cd    | N3            | 0.10  | 2.03 | 0.11  |
| 72  | Hf    | N3            | -1.83 | 3.25 | -1.78 |
| 73  | Ta    | N3            | -1.57 | 3.13 | -1.59 |
| 74  | W     | N3            | -0.75 | 3.36 | -0.93 |
| 75  | Re    | N3            | -0.67 | 3.45 | -0.45 |
| 76  | Os    | N3            | -0.16 | 2.56 | -0.27 |
| 77  | Ir    | N3            | -0.07 | 2.16 | -0.05 |
| 78  | Pt    | N3            | 0.05  | 2.25 | 0.06  |
| 79  | Sc    | pyridine-C3N1 | -2.16 | 3.73 | -2.11 |
| 80  | Ti    | pyridine-C3N1 | -1.57 | 3.17 | -1.56 |
| 81  | V     | pyridine-C3N1 | -1.12 | 3.02 | -1.19 |
| 82  | Cr    | pyridine-C3N1 | -0.60 | 3.13 | -0.58 |
| 83  | Mn    | pyridine-C3N1 | -0.48 | 2.73 | -0.47 |
| 84  | Fe    | pyridine-C3N1 | 0.12  | 2.50 | 0.29  |
| 85  | Co    | pyridine-C3N1 | 0.31  | 2.04 | 0.36  |
| 86  | Ni    | pyridine-C3N1 | 0.41  | 2.06 | 0.49  |
| 87  | Cu    | pyridine-C3N1 | 0.50  | 2.13 | 0.56  |
| 88  | Zn    | pyridine-C3N1 | 0.34  | 2.16 | 0.39  |
| 89  | Y     | pyridine-C3N1 | -2.17 | 3.81 | -1.97 |
| 90  | Zr    | pyridine-C3N1 | -1.57 | 3.23 | -1.60 |
| 91  | Nb    | pyridine-C3N1 | -1.03 | 3.25 | -1.05 |

| Num | Metal | Base                                   | ORR   | OER  | HER   |
|-----|-------|--|-------|------|-------|
| 92  | Mo    | pyridine-C <sub>3</sub> N <sub>1</sub> | -0.43 | 3.06 | -0.50 |
| 93  | Tc    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.10  | 2.75 | 0.05  |
| 94  | Ru    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.27  | 2.19 | 0.36  |
| 95  | Rh    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.38  | 2.16 | 0.54  |
| 96  | Pd    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.45  | 2.49 | 0.30  |
| 97  | Cd    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.28  | 2.34 | 0.33  |
| 98  | Hf    | pyridine-C <sub>3</sub> N <sub>1</sub> | -1.56 | 3.29 | -1.57 |
| 99  | Ta    | pyridine-C <sub>3</sub> N <sub>1</sub> | -1.32 | 3.09 | -1.31 |
| 100 | W     | pyridine-C <sub>3</sub> N <sub>1</sub> | -0.82 | 3.32 | -0.87 |
| 101 | Re    | pyridine-C <sub>3</sub> N <sub>1</sub> | -0.64 | 2.97 | -0.48 |
| 102 | Os    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.18  | 2.54 | 0.19  |
| 103 | Ir    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.27  | 2.24 | 0.30  |
| 104 | Pt    | pyridine-C <sub>3</sub> N <sub>1</sub> | 0.41  | 2.22 | 0.61  |
| 105 | Sc    | pyridine-C <sub>2</sub> N <sub>2</sub> | -2.20 | 3.73 | -2.11 |
| 106 | Ti    | pyridine-C <sub>2</sub> N <sub>2</sub> | -1.56 | 3.17 | -1.54 |
| 107 | V     | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.98 | 3.02 | -1.12 |
| 108 | Cr    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.15 | 3.13 | -0.14 |
| 109 | Mn    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.11 | 2.73 | -0.28 |
| 110 | Fe    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.31  | 2.50 | 0.47  |
| 111 | Co    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.38  | 2.04 | 0.44  |
| 112 | Ni    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.23  | 2.06 | 0.20  |
| 113 | Cu    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.04  | 2.13 | 0.05  |
| 114 | Zn    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.21 | 2.16 | -0.29 |
| 115 | Y     | pyridine-C <sub>2</sub> N <sub>2</sub> | -2.20 | 3.81 | -1.97 |
| 116 | Zr    | pyridine-C <sub>2</sub> N <sub>2</sub> | -1.59 | 3.23 | -1.57 |
| 117 | Nb    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.83 | 3.25 | -0.82 |
| 118 | Mo    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.02  | 3.06 | -0.01 |
| 119 | Tc    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.30  | 2.75 | 0.24  |
| 120 | Ru    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.26  | 2.19 | 0.38  |
| 121 | Rh    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.07  | 2.16 | 0.18  |
| 122 | Pd    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.11 | 2.49 | -0.37 |
| 123 | Cd    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.26 | 2.34 | -0.36 |
| 124 | Hf    | pyridine-C <sub>2</sub> N <sub>2</sub> | -1.58 | 3.29 | -1.59 |
| 125 | Ta    | pyridine-C <sub>2</sub> N <sub>2</sub> | -1.26 | 3.09 | -1.24 |
| 126 | W     | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.55 | 3.32 | -0.61 |
| 127 | Re    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.16 | 2.97 | 0.01  |
| 128 | Os    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.24  | 2.54 | 0.39  |
| 129 | Ir    | pyridine-C <sub>2</sub> N <sub>2</sub> | 0.26  | 2.24 | 0.39  |
| 130 | Pt    | pyridine-C <sub>2</sub> N <sub>2</sub> | -0.04 | 2.22 | 0.25  |
| 131 | Sc    | pyridine-C <sub>1</sub> N <sub>3</sub> | -2.26 | 3.73 | -2.11 |
| 132 | Ti    | pyridine-C <sub>1</sub> N <sub>3</sub> | -1.54 | 3.13 | -1.52 |
| 133 | V     | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.86 | 2.83 | -1.03 |
| 134 | Cr    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.28  | 2.60 | 0.25  |

| Num | Metal | Base                                   | ORR   | OER  | HER   |
|-----|-------|--|-------|------|-------|
| 135 | Mn    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.30  | 2.46 | -0.05 |
| 136 | Fe    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.55  | 2.04 | 0.64  |
| 137 | Co    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.47  | 1.66 | 0.54  |
| 138 | Ni    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.03  | 1.94 | -0.03 |
| 139 | Cu    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.44 | 2.32 | -0.36 |
| 140 | Zn    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.81 | 2.65 | -0.83 |
| 141 | Y     | pyridine-C <sub>1</sub> N <sub>3</sub> | -2.25 | 3.98 | -1.97 |
| 142 | Zr    | pyridine-C <sub>1</sub> N <sub>3</sub> | -1.62 | 3.19 | -1.54 |
| 143 | Nb    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.58 | 3.43 | -0.58 |
| 144 | Mo    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.44  | 2.51 | 0.41  |
| 145 | Tc    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.55  | 2.41 | 0.46  |
| 146 | Ru    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.25  | 1.69 | 0.38  |
| 147 | Rh    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.28 | 2.04 | -0.12 |
| 148 | Pd    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.67 | 2.99 | -0.97 |
| 149 | Cd    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.92 | 2.88 | -0.93 |
| 150 | Hf    | pyridine-C <sub>1</sub> N <sub>3</sub> | -1.60 | 3.35 | -1.59 |
| 151 | Ta    | pyridine-C <sub>1</sub> N <sub>3</sub> | -1.17 | 2.86 | -1.18 |
| 152 | W     | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.29 | 3.45 | -0.32 |
| 153 | Re    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.30  | 2.37 | 0.46  |
| 154 | Os    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.30  | 1.96 | 0.60  |
| 155 | Ir    | pyridine-C <sub>1</sub> N <sub>3</sub> | 0.25  | 1.64 | 0.47  |
| 156 | Pt    | pyridine-C <sub>1</sub> N <sub>3</sub> | -0.50 | 2.28 | -0.07 |
| 157 | Sc    | pyrrole-C <sub>4</sub>                 | -2.22 | 3.72 | -2.11 |
| 158 | Ti    | pyrrole-C <sub>4</sub>                 | -1.55 | 3.17 | -1.54 |
| 159 | V     | pyrrole-C <sub>4</sub>                 | -1.00 | 3.00 | -1.11 |
| 160 | Cr    | pyrrole-C <sub>4</sub>                 | -0.16 | 3.08 | -0.18 |
| 161 | Mn    | pyrrole-C <sub>4</sub>                 | -0.07 | 2.68 | -0.24 |
| 162 | Fe    | pyrrole-C <sub>4</sub>                 | 0.36  | 2.40 | 0.46  |
| 163 | Co    | pyrrole-C <sub>4</sub>                 | 0.40  | 1.97 | 0.47  |
| 164 | Ni    | pyrrole-C <sub>4</sub>                 | 0.21  | 2.02 | 0.26  |
| 165 | Cu    | pyrrole-C <sub>4</sub>                 | 0.02  | 2.15 | 0.14  |
| 166 | Zn    | pyrrole-C <sub>4</sub>                 | -0.26 | 2.27 | -0.16 |
| 167 | Y     | pyrrole-C <sub>4</sub>                 | -2.22 | 3.83 | -1.97 |
| 168 | Zr    | pyrrole-C <sub>4</sub>                 | -1.59 | 3.24 | -1.57 |
| 169 | Nb    | pyrrole-C <sub>4</sub>                 | -0.78 | 3.26 | -0.81 |
| 170 | Mo    | pyrrole-C <sub>4</sub>                 | -0.01 | 2.99 | -0.08 |
| 171 | Tc    | pyrrole-C <sub>4</sub>                 | 0.34  | 2.71 | 0.27  |
| 172 | Ru    | pyrrole-C <sub>4</sub>                 | 0.26  | 2.14 | 0.36  |
| 173 | Rh    | pyrrole-C <sub>4</sub>                 | 0.03  | 2.14 | 0.25  |
| 174 | Pd    | pyrrole-C <sub>4</sub>                 | -0.11 | 2.58 | -0.30 |
| 175 | Cd    | pyrrole-C <sub>4</sub>                 | -0.38 | 2.45 | -0.23 |
| 176 | Hf    | pyrrole-C <sub>4</sub>                 | -1.58 | 3.28 | -1.58 |
| 177 | Ta    | pyrrole-C <sub>4</sub>                 | -1.24 | 3.03 | -1.25 |

| Num | Metal | Base         | ORR   | OER  | HER   |
|-----|-------|--------------|-------|------|-------|
| 178 | W     | pyrrole-C4   | -0.55 | 3.34 | -0.59 |
| 179 | Re    | pyrrole-C4   | -0.18 | 2.92 | -0.03 |
| 180 | Os    | pyrrole-C4   | 0.25  | 2.42 | 0.40  |
| 181 | Ir    | pyrrole-C4   | 0.27  | 2.13 | 0.38  |
| 182 | Pt    | pyrrole-C4   | -0.05 | 2.23 | 0.30  |
| 183 | Sc    | pyrrole-C3N1 | -2.22 | 3.73 | -2.11 |
| 184 | Ti    | pyrrole-C3N1 | -1.55 | 3.13 | -1.54 |
| 185 | V     | pyrrole-C3N1 | -1.00 | 2.83 | -1.11 |
| 186 | Cr    | pyrrole-C3N1 | -0.16 | 2.60 | -0.18 |
| 187 | Mn    | pyrrole-C3N1 | -0.07 | 2.46 | -0.24 |
| 188 | Fe    | pyrrole-C3N1 | 0.36  | 2.04 | 0.46  |
| 189 | Co    | pyrrole-C3N1 | 0.40  | 1.66 | 0.47  |
| 190 | Ni    | pyrrole-C3N1 | 0.21  | 1.94 | 0.26  |
| 191 | Cu    | pyrrole-C3N1 | 0.02  | 2.32 | 0.14  |
| 192 | Zn    | pyrrole-C3N1 | -0.26 | 2.65 | -0.16 |
| 193 | Y     | pyrrole-C3N1 | -2.22 | 3.98 | -1.97 |
| 194 | Zr    | pyrrole-C3N1 | -1.59 | 3.19 | -1.57 |
| 195 | Nb    | pyrrole-C3N1 | -0.78 | 3.43 | -0.81 |
| 196 | Mo    | pyrrole-C3N1 | -0.01 | 2.51 | -0.08 |
| 197 | Tc    | pyrrole-C3N1 | 0.34  | 2.41 | 0.27  |
| 198 | Ru    | pyrrole-C3N1 | 0.26  | 1.69 | 0.36  |
| 199 | Rh    | pyrrole-C3N1 | 0.03  | 2.04 | 0.25  |
| 200 | Pd    | pyrrole-C3N1 | -0.11 | 2.99 | -0.30 |
| 201 | Cd    | pyrrole-C3N1 | -0.38 | 2.88 | -0.23 |
| 202 | Hf    | pyrrole-C3N1 | -1.58 | 3.35 | -1.58 |
| 203 | Ta    | pyrrole-C3N1 | -1.24 | 2.86 | -1.25 |
| 204 | W     | pyrrole-C3N1 | -0.55 | 3.45 | -0.59 |
| 205 | Re    | pyrrole-C3N1 | -0.18 | 2.37 | -0.03 |
| 206 | Os    | pyrrole-C3N1 | 0.25  | 1.96 | 0.40  |
| 207 | Ir    | pyrrole-C3N1 | 0.27  | 1.64 | 0.38  |
| 208 | Pt    | pyrrole-C3N1 | -0.05 | 2.28 | 0.30  |
| 209 | Sc    | pyrrole-C2N2 | -2.26 | 3.73 | -2.11 |
| 210 | Ti    | pyrrole-C2N2 | -1.54 | 3.13 | -1.52 |
| 211 | V     | pyrrole-C2N2 | -0.86 | 2.83 | -1.03 |
| 212 | Cr    | pyrrole-C2N2 | 0.28  | 2.60 | 0.25  |
| 213 | Mn    | pyrrole-C2N2 | 0.30  | 2.46 | -0.05 |
| 214 | Fe    | pyrrole-C2N2 | 0.55  | 2.04 | 0.64  |
| 215 | Co    | pyrrole-C2N2 | 0.47  | 1.66 | 0.54  |
| 216 | Ni    | pyrrole-C2N2 | 0.03  | 1.94 | -0.03 |
| 217 | Cu    | pyrrole-C2N2 | -0.44 | 2.32 | -0.36 |
| 218 | Zn    | pyrrole-C2N2 | -0.81 | 2.65 | -0.83 |
| 219 | Y     | pyrrole-C2N2 | -2.25 | 3.98 | -1.97 |
| 220 | Zr    | pyrrole-C2N2 | -1.62 | 3.19 | -1.54 |



| Num | Metal | Base                                  | ORR   | OER  | HER   |
|-----|-------|---------------------------------------|-------|------|-------|
| 221 | Nb    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.58 | 3.43 | -0.58 |
| 222 | Mo    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.44  | 2.51 | 0.41  |
| 223 | Tc    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.55  | 2.41 | 0.46  |
| 224 | Ru    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.25  | 1.69 | 0.38  |
| 225 | Rh    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.28 | 2.04 | -0.12 |
| 226 | Pd    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.67 | 2.99 | -0.97 |
| 227 | Cd    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.92 | 2.88 | -0.93 |
| 228 | Hf    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -1.60 | 3.35 | -1.59 |
| 229 | Ta    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -1.17 | 2.86 | -1.18 |
| 230 | W     | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.29 | 3.45 | -0.32 |
| 231 | Re    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.30  | 2.37 | 0.46  |
| 232 | Os    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.30  | 1.96 | 0.60  |
| 233 | Ir    | pyrrole-C <sub>2</sub> N <sub>2</sub> | 0.25  | 1.64 | 0.47  |
| 234 | Pt    | pyrrole-C <sub>2</sub> N <sub>2</sub> | -0.50 | 2.28 | -0.07 |
| 235 | Sc    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -2.26 | 3.73 | -2.11 |
| 236 | Ti    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -1.54 | 3.13 | -1.52 |
| 237 | V     | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.86 | 2.83 | -1.03 |
| 238 | Cr    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.28  | 2.60 | 0.25  |
| 239 | Mn    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.30  | 2.46 | -0.05 |
| 240 | Fe    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.55  | 2.04 | 0.64  |
| 241 | Co    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.47  | 1.66 | 0.54  |
| 242 | Ni    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.03  | 1.94 | -0.03 |
| 243 | Cu    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.44 | 2.32 | -0.36 |
| 244 | Zn    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.81 | 2.65 | -0.83 |
| 245 | Y     | pyrrole-C <sub>1</sub> N <sub>3</sub> | -2.25 | 3.98 | -1.97 |
| 246 | Zr    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -1.62 | 3.19 | -1.54 |
| 247 | Nb    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.58 | 3.43 | -0.58 |
| 248 | Mo    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.44  | 2.51 | 0.41  |
| 249 | Tc    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.55  | 2.41 | 0.46  |
| 250 | Ru    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.25  | 1.69 | 0.38  |
| 251 | Rh    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.28 | 2.04 | -0.12 |
| 252 | Pd    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.67 | 2.99 | -0.97 |
| 253 | Cd    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.92 | 2.88 | -0.93 |
| 254 | Hf    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -1.60 | 3.35 | -1.59 |
| 255 | Ta    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -1.17 | 2.86 | -1.18 |
| 256 | W     | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.29 | 3.45 | -0.32 |
| 257 | Re    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.30  | 2.37 | 0.46  |
| 258 | Os    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.30  | 1.96 | 0.60  |
| 259 | Ir    | pyrrole-C <sub>1</sub> N <sub>3</sub> | 0.25  | 1.64 | 0.47  |
| 260 | Pt    | pyrrole-C <sub>1</sub> N <sub>3</sub> | -0.50 | 2.28 | -0.07 |

## References

- (1) Nørskov, J. K.; Rossmeisl, J.; Logadottir, A.; Lindqvist, L.; Kitchin, J. R.; Bligaard, T.; Jonsson, H., *J. Phys. Chem. B* **2004**, *108*, 17886-17892.
- (2) Jones, R. O.; Gunnarsson, O., *Rev. Modern Physics* **1989**, *61*, 689.
- (3) Kurth, S.; Perdew, J. P.; Blaha, P., *Int. J. Quantum Chem.* **1999**, *75*, 889-909.