

Supporting Information

Non-enzymatic electrochemical biosensor based on Au@PBA(Ni-Fe):MoS₂ nanocubes for stable and sensitive detection of hydrogen peroxide released from living cells

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The supporting information contains a supplement of results and discussion.

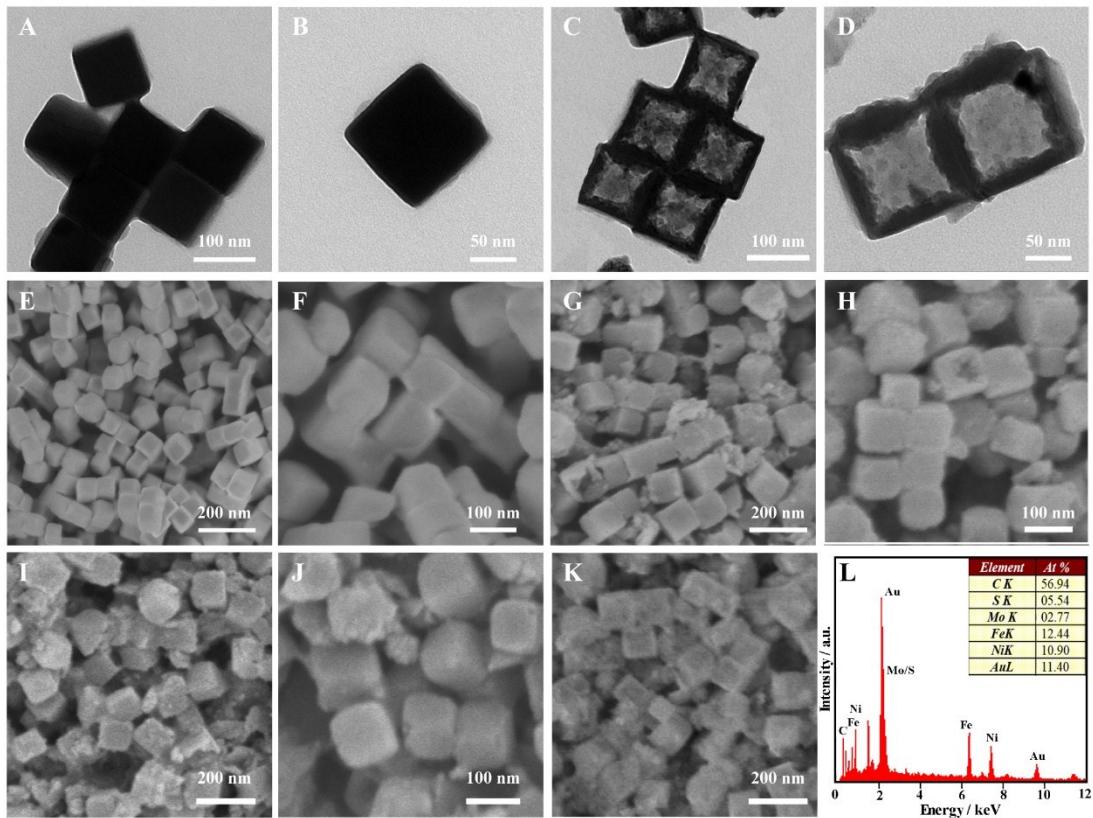


Fig. S1 (A-D) TEM images of PBA(Ni-Fe) and PBA(Ni-Fe):MoS₂, (E-H) SEM images of PBA(Ni-Fe) and PBA(Ni-Fe):MoS₂, (I-J) SEM images of Au@PBA(Ni-Fe):MoS₂, (K) SEM images of Au@PBA(Ni-Fe):MoS₂ after electrochemical test, (L) EDS spectrum of Au@PBA(Ni-Fe):MoS₂.

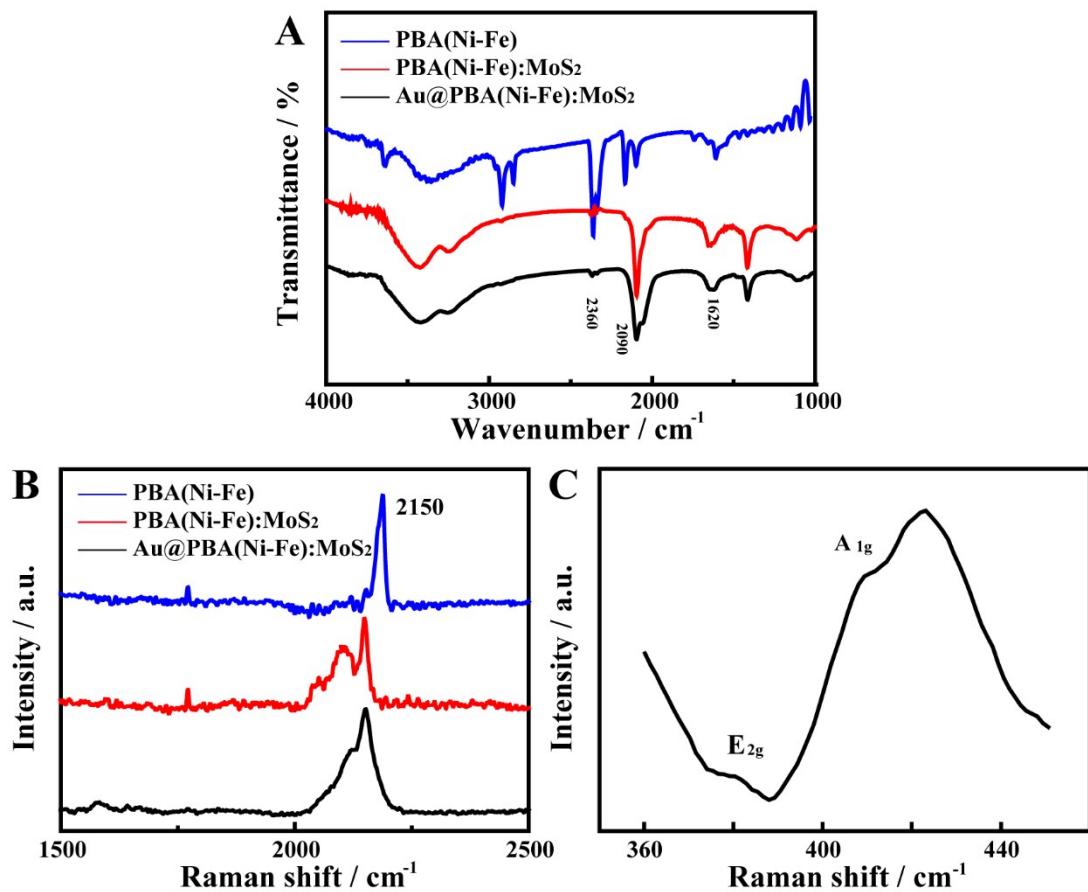


Fig. S2 (A) FT-IR spectra and (B-C) Raman spectra of PBA(Ni-Fe), PBA(Ni-Fe):MoS₂, Au@PBA(Ni-Fe):MoS₂

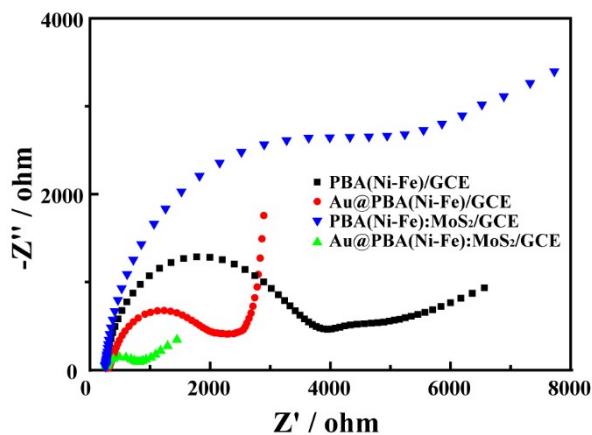


Fig. S3 EIS behaviors of the PBA(Ni-Fe)/GCE, Au@PBA(Ni-Fe)/GCE, PBA(Ni-Fe):MoS₂/GCE and Au@PBA(Ni-Fe):MoS₂/GCE in presence of 5 mM [Fe(CN)₆]^{3-/4-} solution containing 0.1 M KCl at 0.235 V.

Table S1.Comparison of Different Electrochemical Sensors for the Detection of H₂O₂.

Sensor	Linear range (μM)	Detection limit (μM)	Referenc e
PEDOT-PBNPs	5-1000	1.4	[1]
Ag NWs	6-10000	2	[2]
Bimetallic Nanocrystals	-	8.4	[3]
CAT/MoS₂-Au/chitosan/GCE	0.5-200	0.1	[4]
PBNPs/PoPD/GCE	1-58.22	0.8	[5]
Sol-gel-PB glassy carbon electrode	3-210	0.6	[6]
MoS₂/Pt	1-100	0.686	[7]
Au@PBA(Ni-Fe):MoS₂/GCE	0.5-200 and 210-3000	0.23	This work

References

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