## **Electronic Supplementary Information**

Conjugation of RTHLVFFARK to Human Lysozyme Creates a Potent Multifunctional Modulator for Cu<sup>2+</sup>-Mediated Amyloid β-Protein Aggregation and Cytotoxicity

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Agent	Molecular weight	Size	$\zeta$ potential	
	(Da)	(nm)	(mV)	
hLys	14689.7	$4.5 \pm 0.4$	5.8 ± 0.4	
PDEA-hLys	15092.1	N.D.	N.D.	
R-hLys	R-hLys 17471.3		$13.2 \pm 0.7$	

Table S1. Physicochemical properties of hLys, PDEA-hLys and R-hLys at pH 7.4

N. D. not determined.

Table S2. Quantitative analysis of CD spectra of hLys and R-hLys<sup>*a*</sup>

	Helix1	Helix2	Strand1	Strand2	Turns	Unordered	Total
hLys (0 h)	0.573	0.223	0.000	0.006	0.076	0.161	1.039
R-hLys (0 h)	0.573	0.221	0.001	0.005	0.070	0.159	1.028
hLys (48 h)	0.571	0.222	0.000	0.006	0.077	0.162	1.038
R-hLys (48 h)	0.570	0.221	0.002	0.008	0.081	0.168	1.049

<sup>*a*</sup> The result was analyzed by CD Dicroweb with the self-consistent method (SELCON3) (http://dichroweb.cryst.bbk.ac.uk/html/home.shtml).<sup>1,2</sup>



**Fig. S1** Analysis of the purity of R-hLys by size exclusion chromatography (SEC). The retention volumes of hLys and R-hLys were 20.04 and 19.94 mL, respectively.



**Fig. S2** Mass spectra of (A) native hLys, (B) PDEA-hLys (intermediate product, see Scheme 1) and (C) R-hLys.



Fig. S3 Zeta potentials of hLys and R-hLys as a function of pH. Native hLys and R-hLys were both 25  $\mu$ M in deionized water. Each point was the mean of three different measurements and error bars represent the standard deviations.



Fig. S4 Comparison of the far-UV circular dichroism spectra of native hLys and R-hLys after incubation for (A) 0 h and (B) 48 h and their fluorescence spectra after incubation for (C) 0 h and (D) 48 h. Native hLys or R-hLys was dissolved in buffer A at 25  $\mu$ M and kept static at 37 °C.



**Fig. S5** Aggregation kinetics of  $A\beta_{40}$  (25 µM) incubated in the absence and presence of (A) hLys, (B) RC11 and (C) R-hLys. RC11 concentrations for lines P1 to P3 in (B) were approximately equal to those of RC11 coupled to R-hLys for lines L1 to L3 in (A), respectively. All measurements were conducted in buffer A at 37 °C.



**Fig. S6** Cell viability assay at different (yellow) hLys, (blue) RC11 or (red) R-hLys concentrations using MTT assay. Lanes denote the following conditions: lane 1, buffer A; lane 2, 0.5  $\mu$ M hLys/R-hLys, 1  $\mu$ M RC11; lane 3, 1.25  $\mu$ M hLys/R-hLys, 2.5  $\mu$ M RC11; lane 4, 2.5  $\mu$ M hLys/R-hLys, 5  $\mu$ M RC11; lane 5, 5  $\mu$ M hLys/R-hLys, 10  $\mu$ M RC11. All concentration data represent those in the cell culture medium.



**Fig. S7** ITC analysis of the interactions between  $Cu^{2+}$  and (A) hLys, (B) R-hLys. The concentrations of  $Cu^{2+}$ , glycine and chelator were 0.5 mM, 2 mM and 0.1 mM in buffer A, respectively. The top panel shows the heat rate for each injection and the positive peaks indicate an exothermic process; the bottom panel shows the integrated heat for each injection.



**Fig. S8** Aggregation kinetics of Cu<sup>2+</sup>-mediated A $\beta_{40}$  aggregation incubated in the absence and presence of (A) hLys, (B) RC11 and (C) R-hLys. The concentration of A $\beta_{40}$  was 25 µM. Lines denote the following conditions: (A) line 1, A $\beta_{40}$  alone; line 2, Cu<sup>2+</sup>:A $\beta_{40}$  = 0.4:1; line L1, hLys:Cu<sup>2+</sup>:A $\beta_{42}$  = 0.1:0.4:1; line L2, hLys:Cu<sup>2+</sup>:A $\beta_{42}$  = 0.25:0.4:1; line L3, hLys:Cu<sup>2+</sup>:A $\beta_{42}$  = 0.5:0.4:1. (B) line 1, A $\beta_{40}$  alone; line 2, Cu<sup>2+</sup>:A $\beta_{40}$  = 0.4:1; line P1, RC11:Cu<sup>2+</sup>:A $\beta_{42}$  = 0.2:0.4:1; line P2, RC11:Cu<sup>2+</sup>:A $\beta_{42}$  = 0.5:0.4:1; line P3, RC11:Cu<sup>2+</sup>:A $\beta_{42}$  = 1:0.4:1. (C) line 1, A $\beta_{40}$  alone; line 2, Cu<sup>2+</sup>:A $\beta_{40}$  = 0.4:1; line R1, R-hLys:Cu<sup>2+</sup>:A $\beta_{40}$  = 0.1:0.4:1, line R2, R-hLys:Cu<sup>2+</sup>:A $\beta_{40}$  = 0.25:0.4:1; line R3, R-hLys:Cu<sup>2+</sup>:A $\beta_{40}$  = 0.5:0.4:1. All measurements were conducted

in buffer A at 37 °C.



Fig. S9 Analysis of the stability of R-hLys in the presence of ascorbic acid by SEC. The concentrations of R-hLys and ascorbic acid were 25  $\mu$ M and 500  $\mu$ M, respectively.



**Fig. S10** Time-dependent ROS production measured by CCA fluorescence assays under a reducing environment.  $Cu^{2+}$  incubated with hLys/RC11/R-hLys in the absence of A $\beta_{42}$ . Lines denote the following conditions: line 0,  $Cu^{2+}$ ; line L1,  $Cu^{2+}$  + 2.5 µM hLys; line L2,  $Cu^{2+}$  + 6.25 µM hLys; line L3,  $Cu^{2+}$  + 12.5 µM hLys; line P1,  $Cu^{2+}$  + 5 µM RC11; line P2,  $Cu^{2+}$  + 12.5 µM RC11; line P3,  $Cu^{2+}$  + 25 µM RC11; line R1,  $Cu^{2+}$  + 2.5 µM R-hLys; line R2,  $Cu^{2+}$  + 6.25 µM R-hLys; line R3,  $Cu^{2+}$  + 12.5 µM R-hLys.



Fig. S11 Normalized ThT fluorescence of  $A\beta_{42}$  fibrils incubated in the absence and presence of various concentrations of RC11 for 48 h. The final concentration of  $A\beta_{42}$  was 25  $\mu$ M.



**Fig. S12** ThT fluorescence kinetics for the remodeling of  $A\beta_{42}$  fibrils with or without (A) hLys or (B) R-hLys at different concentrations. The final concentration of  $A\beta_{42}$  fibrils was 25  $\mu$ M.  $A\beta_{42}$  fibrils were obtained by incubation in buffer A at 37 °C for 24 h.

## References

- 1 L. Whitmore and B. A. Wallace, *Biopolymers*, 2008, 89, 392-400.
- 2 N. Sreerama and R. W. Woody, Anal. Biochem. 2000, 287, 252-260.