

## Supporting Information

for

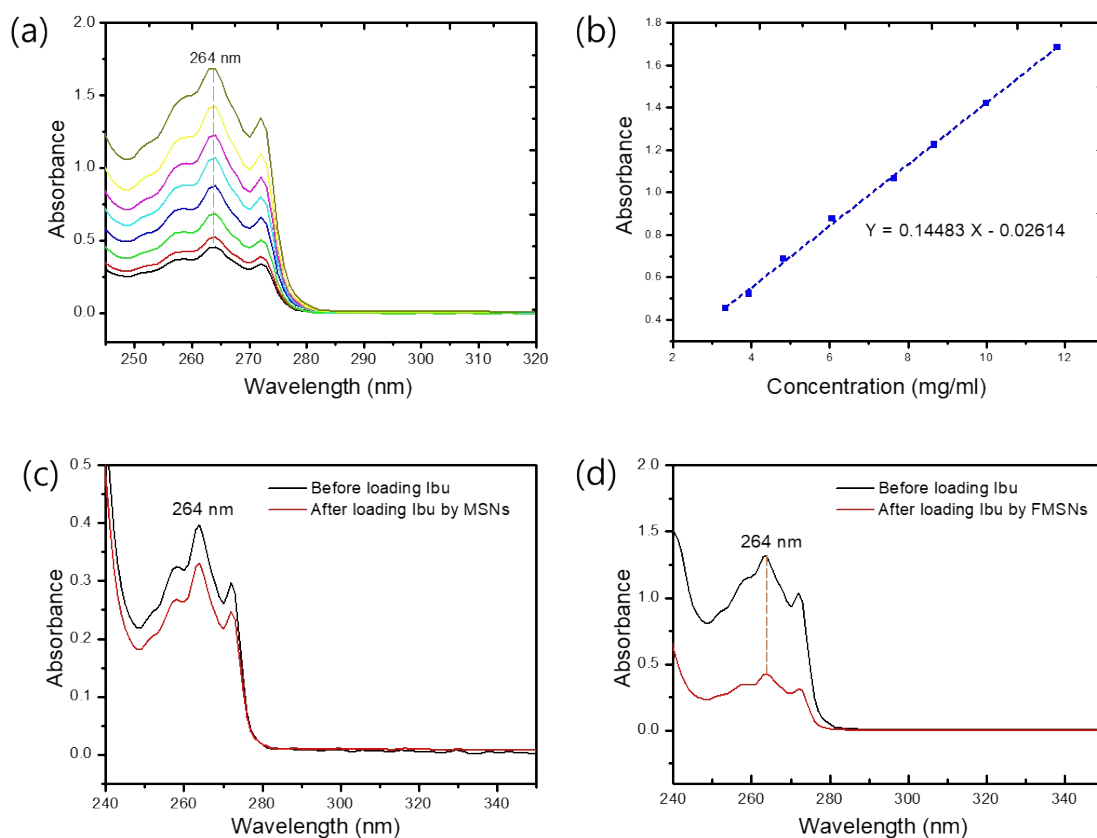
### **Comparative release kinetics of small drugs (ibuprofen and acetaminophen) from multifunctional mesoporous silica nanoparticles**

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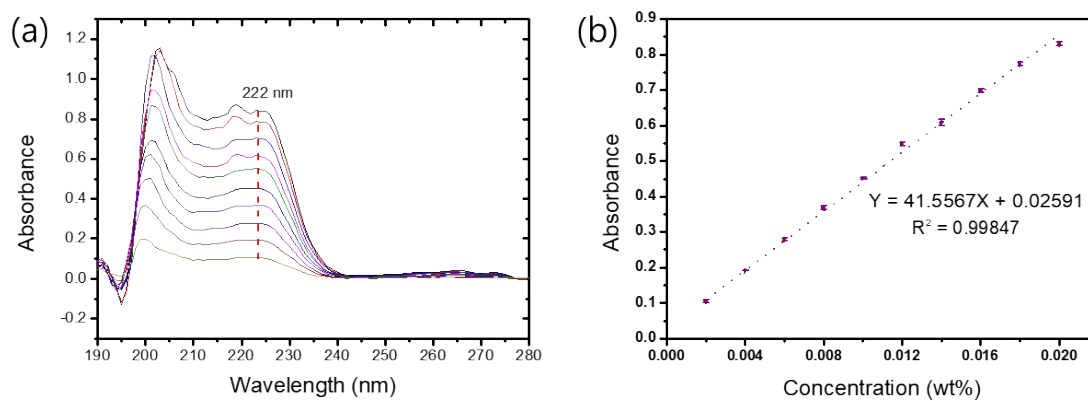
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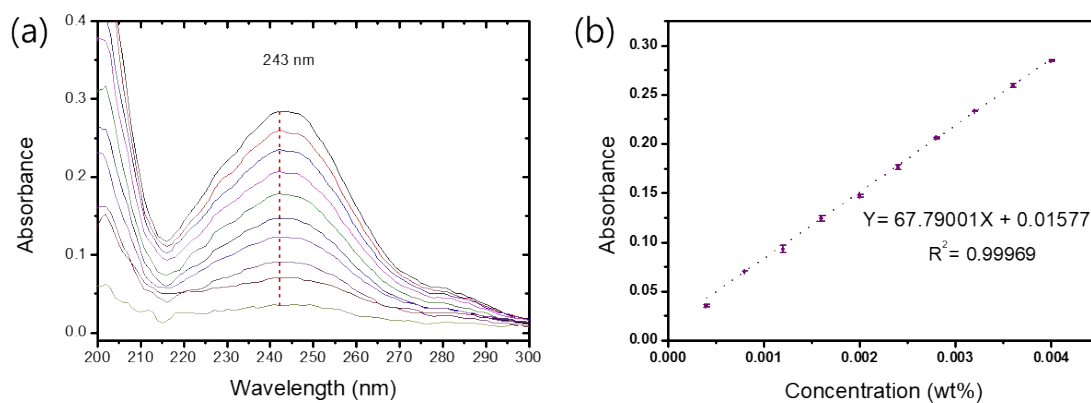
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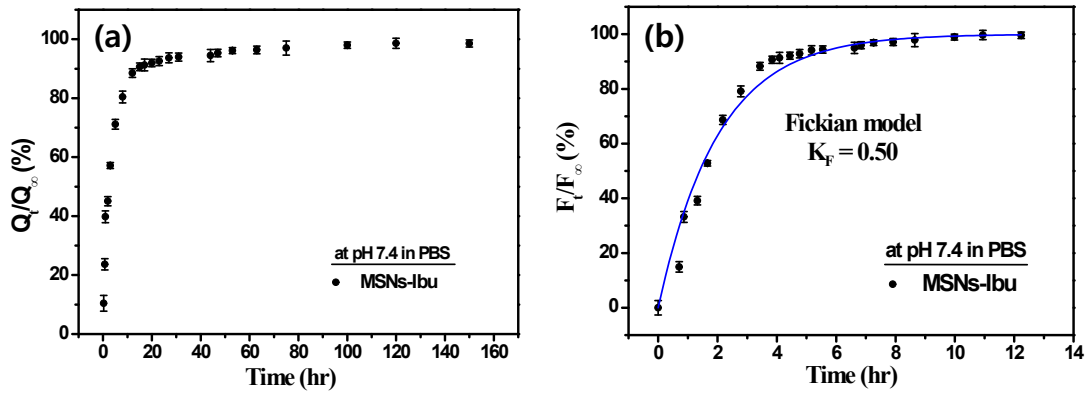
**Fig. S1.** (a) The UV-VIS spectrum of Ibuprofen in ethanol, (b) the linear part of the standard curve of ibuprofen absorption at 264nm in ethanol over concentration ranges (3-12 mg/mL), (c) UV-Vis absorption spectra of the solution before and after loading ibuprofen by MSNs, (d) UV-Vis absorption spectra of the solution before and after loading ibuprofen by FMSNs.



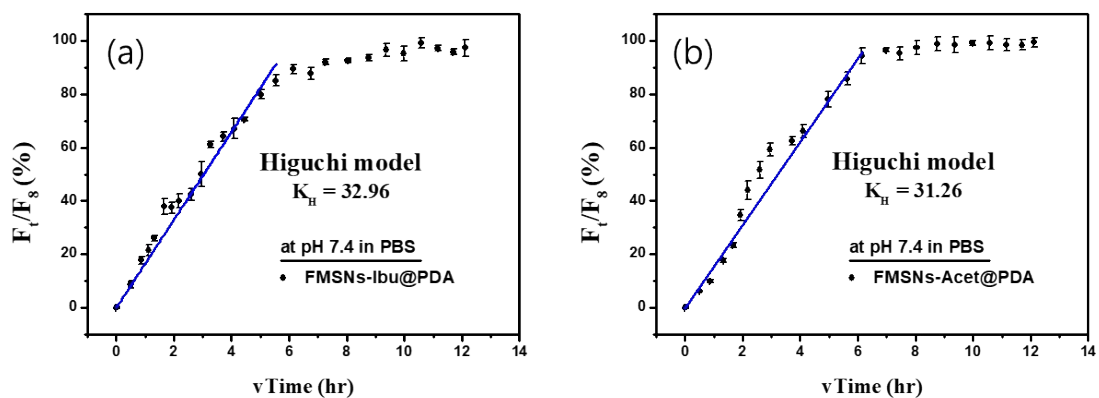
**Fig. S2.** (a) The UV-VIS spectrum of Ibuprofen in PBS, (b) Linear part of the standard curve of ibuprofen absorption at 222 nm in PBS: concentration ranges from 0.002 to 0.02 wt%.



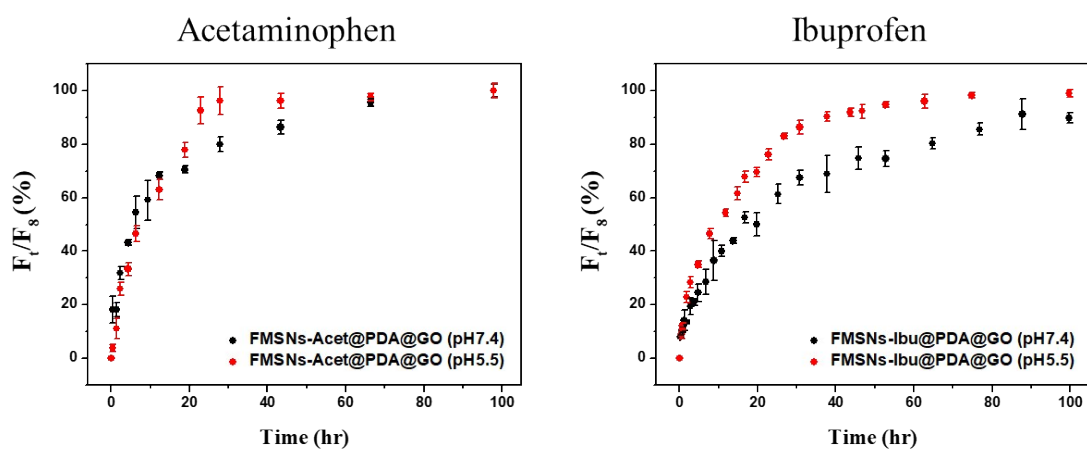
**Fig. S3.** (a) The UV-VIS spectrum of Acetaminophen in PBS, (b) Linear part of the standard curve of acetaminophen absorption at 243 nm in PBS: concentration ranges from 0.0004 to 0.004 wt%.



**Fig. S4.** Absorbance profiles of released Ibuprofen from MSNs in PBS at 37 °C; (a) the correlation between cumulative release fraction and release times; (c) the model fits of release kinetics of MSNs-Ibu by the exponential Fickian model.



**Fig. S5.** The Model fits of (a) Ibuprofen and (b) Acetaminophen release from FMSNs-Drug@PDA in PBS at 37 °C by the Higuchi model versus cumulative time or square root time.



**Fig S6.** The comparative release rate of acetaminophen and ibuprofen of FMSNs-drug@PDA@GO at different pHs.

**Table S1.** Physicochemical properties of as-prepared samples measured by BET-BJT, DLS and SEM instruments.

Particle type	BET- BJH method			SEM Particle size (nm)	DLS Particle size (nm)
	surface area <sup>1)</sup> (m <sup>2</sup> /g)	Pore volume <sup>2)</sup> (cm <sup>3</sup> /g)	Pore size <sup>2)</sup> (nm)		
MSNs	209	0.46	8.7	125 ± 5	124 ± 10
FMSNs	3391	4.00	4.7	123 ± 3	123 ± 2
FMSNs@PDA	-	-	-	127 ± 5	126 ± 14
FMSNs@PDA@GO	-	-	-	129 ± 2	129 ± 9

1) The surface area was estimated according to the BET method.

2) The pore size and pore volume were estimated by the BJH analysis.



**Table S2.** The summary of fitted parameter values of kinetic models applied to the release data of MSNs-Drug@A-F, MSNs-Drug@A-F@PDA, MSNs-Drug@A-F@PDA@GO (A: Fick's law, B: Higuchi model, C: K-P model)

Case	As-prepared MSNs	Diffusion models	Formula	Parameters
A	FMSNs-Ibu	Fick's law	$F_t/F_\infty = 1 - e^{-k_F t}$	$k_F = 0.38$
	FMSNs-Acet	Fick's law	$F_t/F_\infty = 1 - e^{-k_F t}$	$k_F = 0.62$
B	FMSNs-Ibu@PDA@GO (pH7.4)	Higuchi model	$F_t/F_\infty = k_{Ht}^{1/2}$	$k_{H1} = 23.62$ $k_{H2} = 8.87$
	FMSNs-Ibu@PDA@GO (pH5.5)	Higuchi model	$F_t/F_\infty = k_{Ht}^{1/2}$	$k_H = 32.04$
	FMSNs-Acet@PDA@GO	Higuchi model	$F_t/F_\infty = k_{Ht}^{1/2}$	$k_H = 25.40$
	FMSNs-Ibu@PDA	Higuchi model	$F_t/F_\infty = k_{Ht}^{1/2}$	$k_H = 32.96$
	FMSNs-Acet@PDA	Higuchi model	$F_t/F_\infty = k_{Ht}^{1/2}$	$k_H = 31.26$
C	FMSNs-Ibu@PDA	K-P model	$F_t/F_\infty = k_R t^n$	$k_R = 19.61, n = 0.47$
	FMSNs-Acet@PDA	K-P model	$F_t/F_\infty = k_R t^n$	$k_R = 15.00, n = 0.62$

\* K-P model indicate the Korsmeyer-Peppas model.

\*\* Ibu indicates Ibuprofen.

\*\*\* Acet indicates Acetaminophen.