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Supplementary Information

Supplementary Figure 1



Fig. S1 Characterization of the couple of the DAN-BSA.



Fig. S2 Average hydrodynamic diameter of the prepared AgNPRs. (A) The prepared AgNPRs; (B-C) The AgNPRs etched induced by different concentrations of GOx: (B) 15 and (C) 30 ng/mL.



Supplementary Figure 3

Fig. S3 (A) Calibration curve of TMB under H₂O₂ concentration ranging from 1.95 μM to 125 μM;
(B) photographs of TMB under different H₂O₂ concentrations (0, 1.95,3.91, 7,81,15.63, 31.25,

62.5, 125 μM).

Supplementary Figure 4



Fig. S4 Characterization of the mAb by SDS-PAGE. (a) 2.5 μg mAb, (b) 5 μg mAb, (c) 10 μg mAb.



Fig. S5 Characterization of Biotin-mAb and Biotin-GOx. (A) The coupling result characterization of Biotin-mAb. i) ii) were added with 0.01 M PBS as the controls, iii) iv) were added with coupled Biotin-mAb; (B) The coupling result characterization of Biotin-GOx. i) ii) were added with 0.01 M PBS as the controls, iii) iv) were added with coupled Biotin-GOx.

Supplementary Figure 6



Fig. S6 The concentration optimization of Biotin-mAb in ELISA based on TMB as signal transducer. The error bars represent the standard deviation of three measurements.



Fig. S7 The concentration optimization of SA in ELISA based on TMB as signal transducer. The

error bars represent the standard deviation of three measurements.



Supplementary Figure 8

Fig. S8 The concentration optimization of GOx-Biotin in ELISA based on TMB as signal transducer. The error bars represent the standard deviation of three measurements.



Fig. S9 The concentration optimization of glucose in ELISA based on TMB as signal transducer. The error bars represent the standard deviation of three measurements.

LC-MS/MS analysis

The reliability of the proposed AgNPR etching pELISA method was confirmed using LC-MS/MS. The detailed operating procedures were performed according to GB/T 22985-2008. 2 g milk sample was weighed to the nearest 0.01 g and placed in a 50 mL plastic centrifuge tube. The standard of DAN was added to the 50 mL plastic centrifuge tube. Ten milliliters of acetonitrile were added, and the mixture was shaken for 1 min in a vortex shaker. After centrifugation at 5000 rpm for 5 min, the supernatant was filtered into a chicken heart bottle. Five milliliters of a phosphate buffer solution and 10 mL of acetonitrile were added to the residue, the steps described above were repeated, and the supernatant was combined and rotaryevaporated at 50 °C until all the acetonitrile was distilled off. Then, 5 mL of a phosphate buffer solution was added and mixed with the precipitate. The treated solution was transferred to a reservoir, and the chicken heart bottle was washed with 5 mL of phosphate buffer. The wash solution was added to the reservoir and passed through the HLB solid phase extraction cartridge at a flow rate of approximately 1 mL/min. After the sample solution was completely discharged, it was sequentially rinsed with 4 mL of water and 4 mL of 25% an aqueous methanol solution, drained, and eluted with 4 mL of 5% ammonia in methanol in a 10 mL graduated centrifuge tube. The eluate was incubated at 50 °C. When the eluent was reduced to approximately 0.2 mL under a nitrogen flow, the concentration process was terminated. The volume of eluent was adjusted to 1 mL with a methanol-formic acid solution. After centrifugation at 5000 r/min for 5 minutes, the supernatant was passed through a 0.2 µm filter for the LC-MS/MS analysis. Chromatographic separation was achieved with an Agilent Zorbax Eclipse XDBC18 column using solvent A (ammonium acetate, 10 mM) and solvent B (methanol) as the mobile phase. Initial gradient conditions were set to 80% solvent A, reduced with a linear gradient to 40% from 0 min to 5 min, then further decreased to 5% from 5 min to 9 min. At 9 min, the gradient was programmed to the initial conditions to re-equilibrate the column for 2 min. The flow rate was 0.2 mL/min, and the injection volume was 15 µL in the full injection mode. Ionization was achieved using electrospray ionization in positive ion mode. High-purity argon gas was used as the collision gas. The monitored ion pairs for DAN were m/z 358.14/340.00 (quantitative ions) and 358.14/313.80 (qualitative ions).

Table S1

Comparison of recovery and precision between the proposed pELISA and LC–MS/MS for quantitative DAN determination.

DAN added (ng/mL)	Proposed pELISA (n = 3)			LC-MS/MS $(n = 3)$		
	DAN recovered (ng/mL)	Recovery	CV	DAN recovered (ng/mL)	Recovery	CV
		(%)	(%)		(%)	(%)
30	36.38	121.26	0.6	26.22	87.4	5.96
60	62.07	103.45	0.99	43.32	72.2	8.34
90	102.6	114	1.21	73.44	81.6	8.8