

Electronic Supplementary Information for

Pure *E/Z* isomers of *N*-methylpyrrole-benzohydrazide-based BF_2 complexes: Remarkable aggregation-, crystallization-induced emission switching properties and application in sensing intracellular pH microenvironment

*Yizhe Yu,^a Changjiang Yu,^{*ab} Qinghua Wu,^a Hua Wang,^a Lijuan Jiao,^a Wai-Yeung Wong^b and Erhong Hao^{*a}*

^a The Key Laboratory of Functional Molecular Solids, Ministry of Education; Anhui Laboratory of Molecule-Based Materials (State Key Laboratory Cultivation Base); School of Chemistry and Materials Science, Anhui Normal University, Wuhu, China 241000.

^b Department of Applied Biology and Chemical Technology, The Hong Kong Polytechnic University, 8 Hung Lok Road, Hung Hom, Hong Kong, China.

*To whom correspondence should be addressed.

E-mail: yuchj@ahnu.edu.cn; haoehong@ahnu.edu.cn

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1. Crystal packings and selected parameters

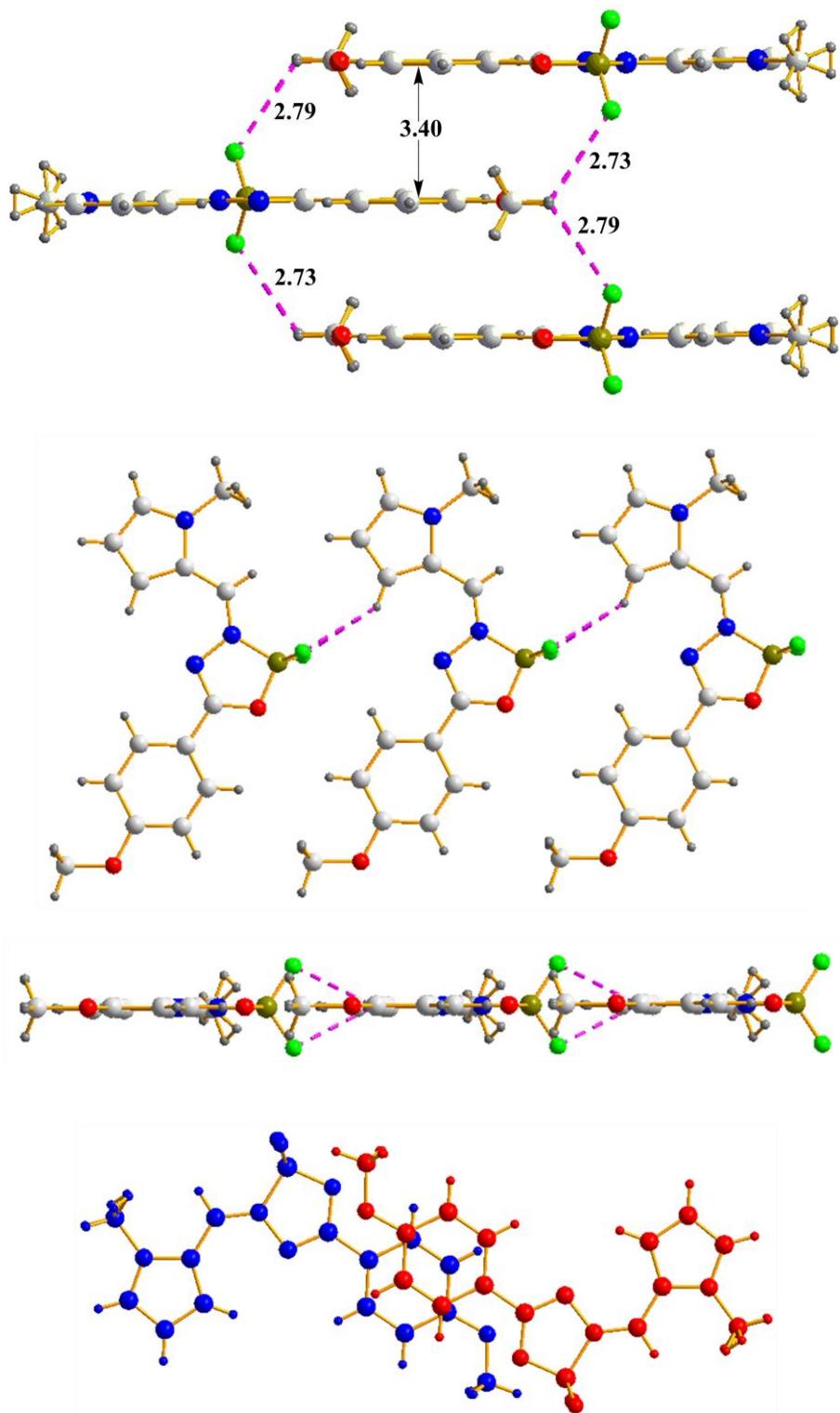


Figure S1. Crystal-packing pattern of **1a** between the adjacent interlayered crystals from side view. Interlayer distance is 3.40 Å. Intermolecular hydrogen bond lengths are 2.79 and 2.73 Å. C, light gray; H, gray; N, blue; O, red; B, dark yellow; F, bright green.

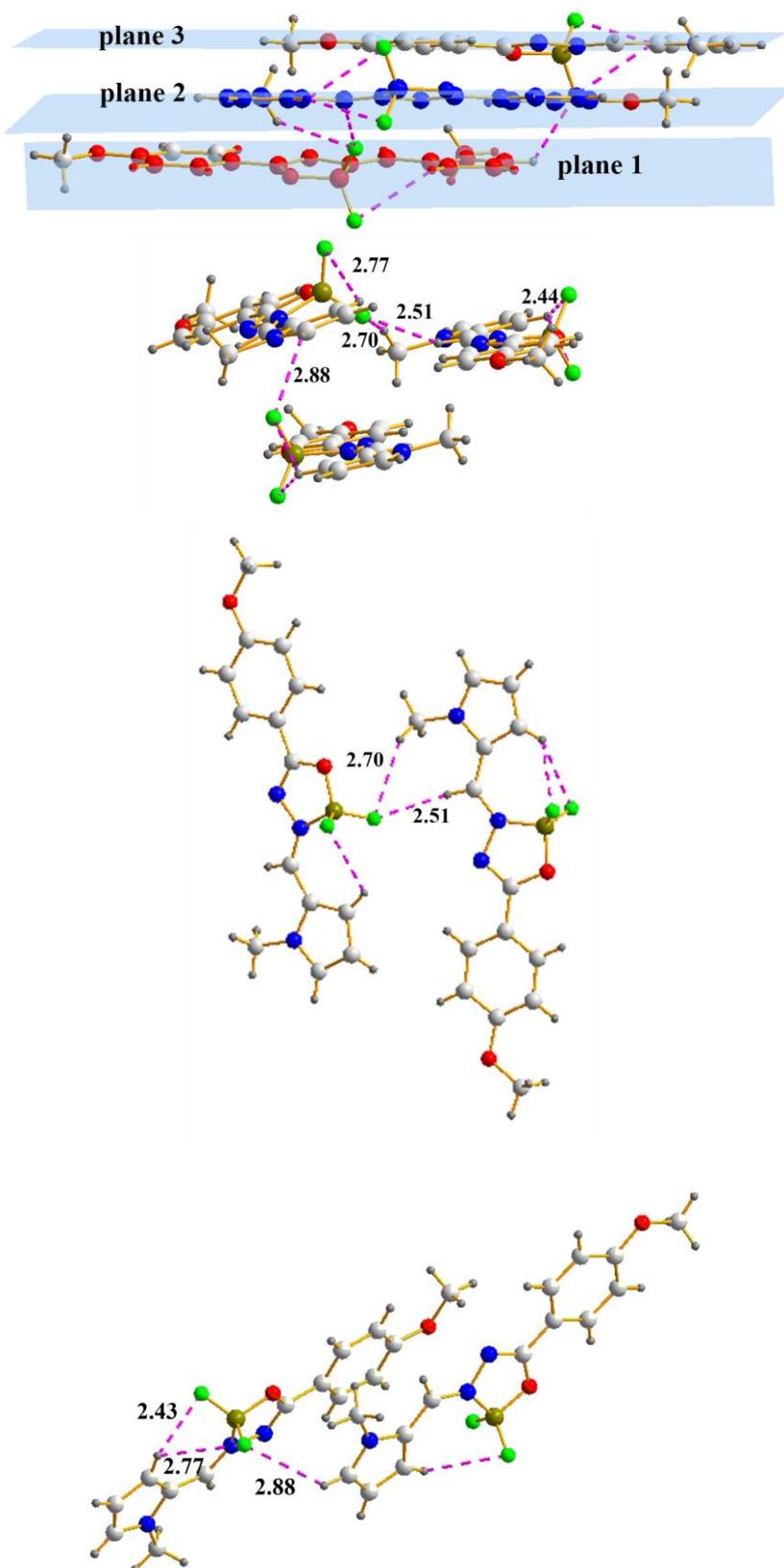


Figure S2. Crystal-packing pattern of **1b** between the adjacent interlayered crystals from side view. Multiple intermolecular hydrogen bond and intramolecular hydrogen bond are given above. C, light gray; H, gray; N, blue; O, red; B, dark yellow; F, bright green.

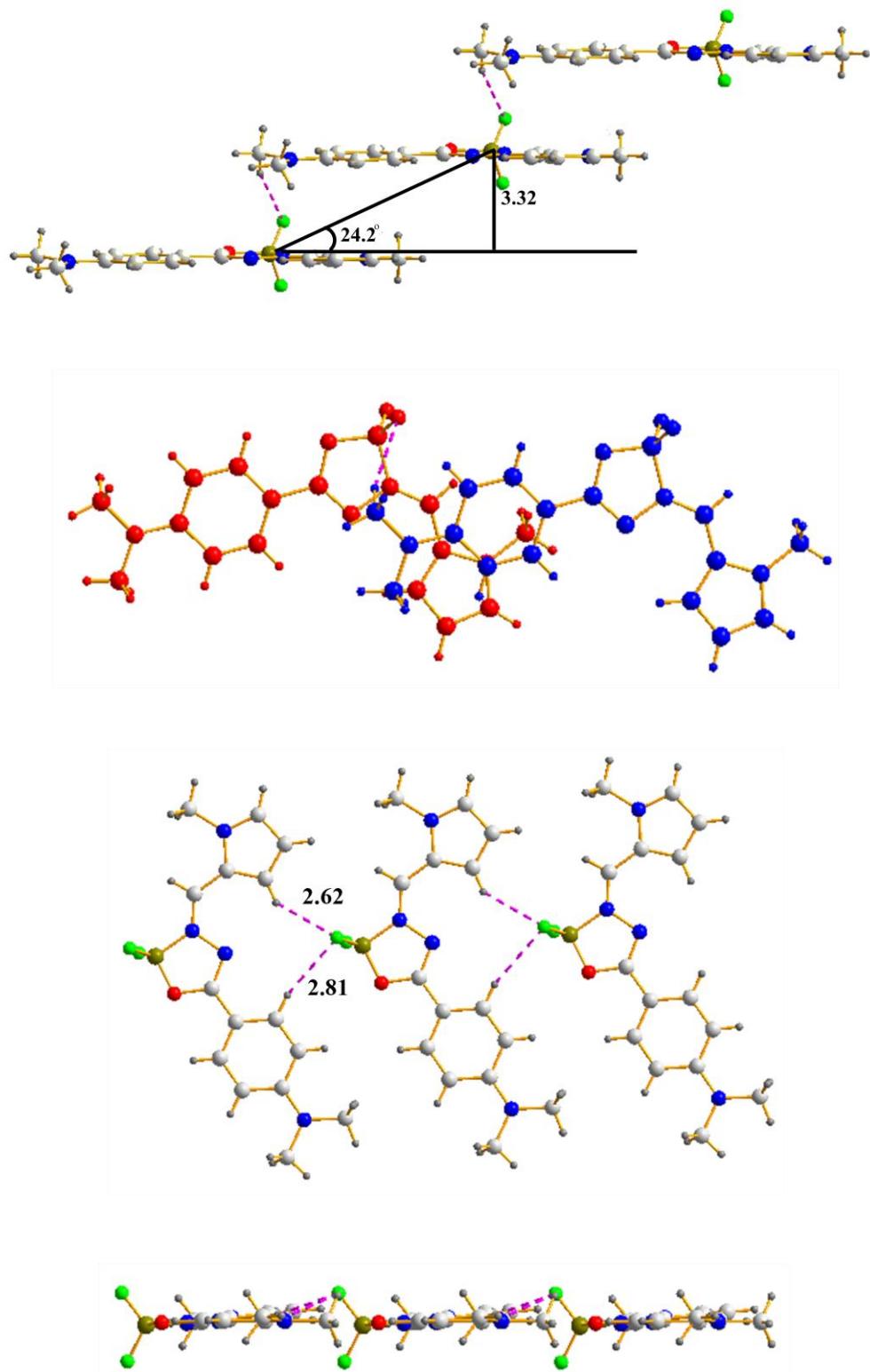


Figure S3. Crystal-packing pattern of **2a** between the adjacent interlayered crystals from side view. Interlayer distance is 3.32 Å. Slip angel of 24.2° for coplanar inclined arrangements of its transition dipole. Intermolecular hydrogen bond lengths are 2.62 and 2.81 Å. C, light gray; H, gray; N, blue; O, red; B, dark yellow; F, bright green.

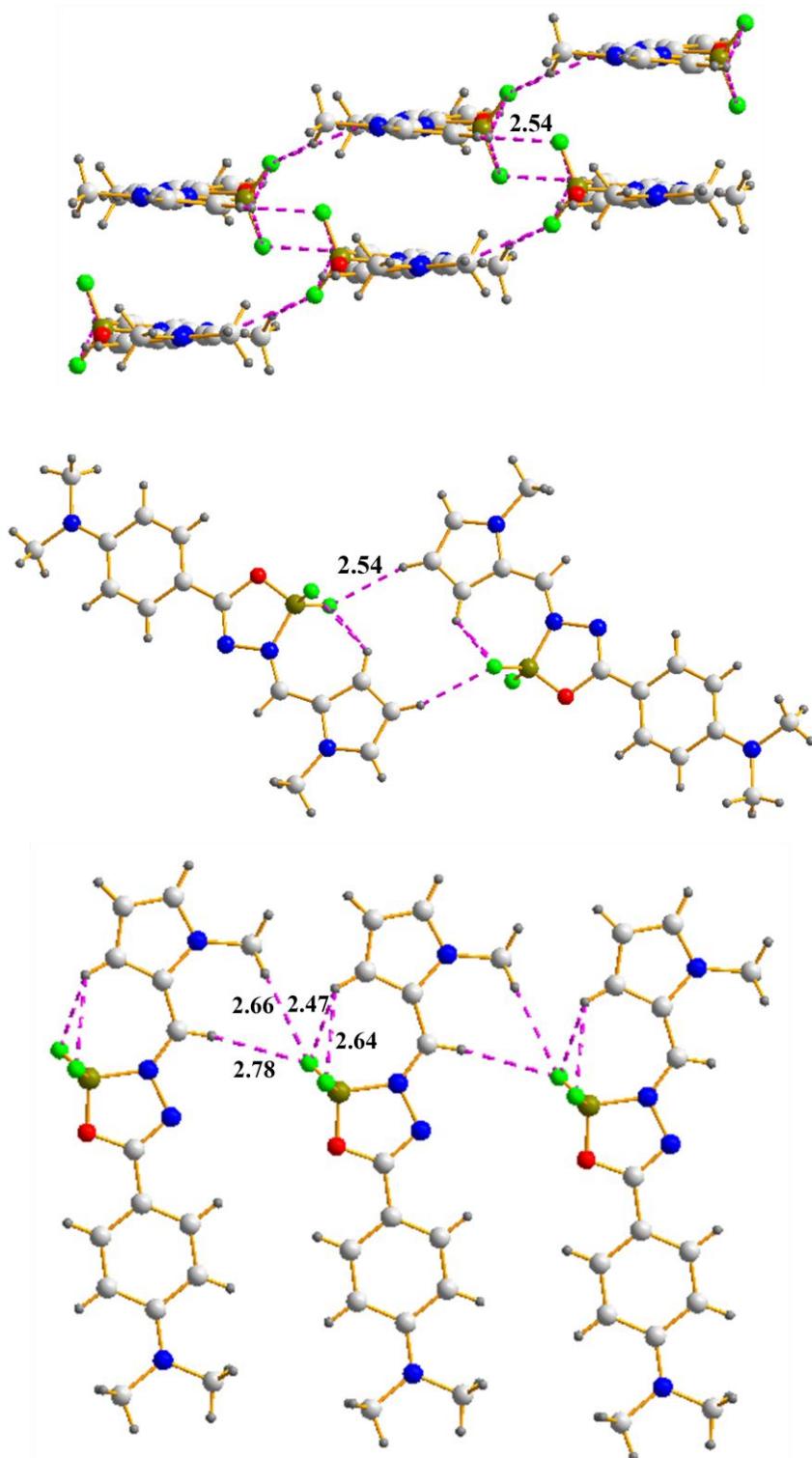


Figure S4. Crystal-packing pattern of **2b** between the adjacent interlayered crystals from side view. Multiple intermolecular hydrogen bond (2.54, 2.66 and 2.78 Å) and intramolecular hydrogen bond (2.47 and 2.64 Å) are given above. C, light gray; H, gray; N, blue; O, red; B, dark yellow; F, bright green.

Table S1. Selected bond lengths [\AA] and dihedral angles [deg] of **1a**, **1b**, **2a** and **2b** obtained from crystallography.

| dyes | B-N bond distances (\AA) | B-O bond distances (\AA) | N-N bond distances (\AA) | dihedral angles between N-methylpyrrolic ring and newly formed ring (deg) | dihedral angles of N-methylpyrrolic ring and the aromatic ring (deg) | dihedral angles between the newly formed ring and the aromatic ring (deg) |
|-----------|--|---|---|---|---|---|
| 1a | 1.579 | 1.472 | 1.397 | 0.000 | 0.000 | 0.000 |
| 1b | 1.587 | 1.467 | 1.407 | 8.63(2) | 3.07(2) | 5.58(2) |
| 2a | 1.571 | 1.469 | 1.404 | 4.23(2) | 5.62(2) | 2.40(2) |
| 2b | 1.584 | 1.468 | 1.408 | 6.05(2) | 10.08(2) | 7.39(2) |

Table S2. Crystal data and structure refinement for **1a** and **1b**

| Identification code | 1a | 1b |
|---|---|---|
| Empirical formula | C ₁₄ H ₁₄ BF ₂ N ₃ O ₂ | C ₁₄ H ₁₅ BF ₂ N ₃ O ₂ |
| Formula weight | 305.09 | 306.10 |
| Temperature/K | 293(2) | 273.15 |
| Crystal system | monoclinic | monoclinic |
| Space group | P2 ₁ /m | P2 ₁ /n |
| a/Å | 7.3108(5) | 10.8986(2) |
| b/Å | 6.7983(5) | 11.8661(2) |
| c/Å | 14.1226(11) | 11.1113(2) |
| α/° | 90 | 90 |
| β/° | 91.598(2) | 91.8440(10) |
| γ/° | 90 | 90 |
| Volume/Å ³ | 701.63(9) | 1436.21(4) |
| Z | 2 | 4 |
| ρ _{calc} g/cm ³ | 1.444 | 1.416 |
| μ/mm ⁻¹ | 0.115 | 0.956 |
| F(000) | 316.0 | 636.0 |
| Crystal size/mm ³ | 0.23 × 0.18 × 0.06 | 0.23 × 0.12 × 0.08 |
| Radiation | MoKα ($\lambda = 0.71073$) | CuKα ($\lambda = 1.54178$) |
| 2Θ range for data collection/° | 6.206 to 55.024 -9 ≤ h ≤ 9, -8 ≤ k ≤ 8, -18 ≤ l ≤ 18 | 10.912 to 133.096 -12 ≤ h ≤ 12, -14 ≤ k ≤ 14, -10 ≤ l ≤ 13 |
| Index ranges | | |
| Reflections collected | 19360 | 10675 |
| Independent reflections | 1744 [R _{int} = 0.0462, R _{sigma} = 0.0232] | 2501 [R _{int} = 0.0252, R _{sigma} = 0.0219] |
| Data/restraints/parameters | 1744/0/132 | 2501/0/201 |
| Goodness-of-fit on F ² | 1.055 | 1.038 |
| Final R indexes [I>=2σ (I)] | R ₁ = 0.0441, wR ₂ = 0.1016 | R ₁ = 0.0395, wR ₂ = 0.1089 |
| Final R indexes [all data] | R ₁ = 0.0663, wR ₂ = 0.1120 | R ₁ = 0.0439, wR ₂ = 0.1133 |
| Largest diff. peak/hole / e Å ⁻³ | 0.19/-0.27 | 0.15/-0.50 |

Table S3. Crystal data and structure refinement for **2a** and **2b**

| Identification code | 2a | 2b |
|---|--|--|
| Empirical formula | C ₁₆ H ₁₇ F ₂ N ₄ O | C ₁₅ H ₁₇ BF ₂ N ₄ O |
| Formula weight | 319.33 | 318.13 |
| Temperature/K | 273.15 | 293(2) |
| Crystal system | triclinic | triclinic |
| Space group | P-1 | P-1 |
| a/Å | 7.2781(12) | 6.7077(2) |
| b/Å | 8.5866(14) | 8.4933(3) |
| c/Å | 13.586(2) | 14.4139(5) |
| α/° | 72.984(6) | 91.8700(10) |
| β/° | 80.919(6) | 102.8680(10) |
| γ/° | 72.590(6) | 107.3170(10) |
| Volume/Å ³ | 772.4(2) | 759.93(4) |
| Z | 2 | 2 |
| ρ _{calc} g/cm ³ | 1.373 | 1.390 |
| μ/mm ⁻¹ | 0.106 | 0.106 |
| F(000) | 334.0 | 332.0 |
| Crystal size/mm ³ | 0.28 × 0.19 × 0.13 | 0.23 × 0.15 × 0.12 |
| Radiation | MoKα ($\lambda = 0.71073$) | MoKα ($\lambda = 0.71073$) |
| 2Θ range for data collection/° | 5.884 to 49.992 -8 ≤ h ≤ 8, -9 ≤ k ≤ 10, -16 ≤ l ≤ 16 | 5.832 to 49.982 -7 ≤ h ≤ 7, -10 ≤ k ≤ 10, -17 ≤ l ≤ 17 |
| Index ranges | | |
| Reflections collected | 11719 | 14692 |
| Independent reflections | 2615 [R _{int} = 0.0409, R _{sigma} = 0.0361] | 2644 [R _{int} = 0.0602, R _{sigma} = 0.0276] |
| Data/restraints/parameters | 2615/0/211 | 2644/0/211 |
| Goodness-of-fit on F ² | 1.075 | 1.158 |
| Final R indexes [I>=2σ (I)] | R ₁ = 0.0933, wR ₂ = 0.2249 | R ₁ = 0.0422, wR ₂ = 0.1204 |
| Final R indexes [all data] | R ₁ = 0.1169, wR ₂ = 0.2369 | R ₁ = 0.0509, wR ₂ = 0.1344 |
| Largest diff. peak/hole / e Å ⁻³ | 0.35/-0.39 | 0.33/-0.32 |

2. Photophysical properties

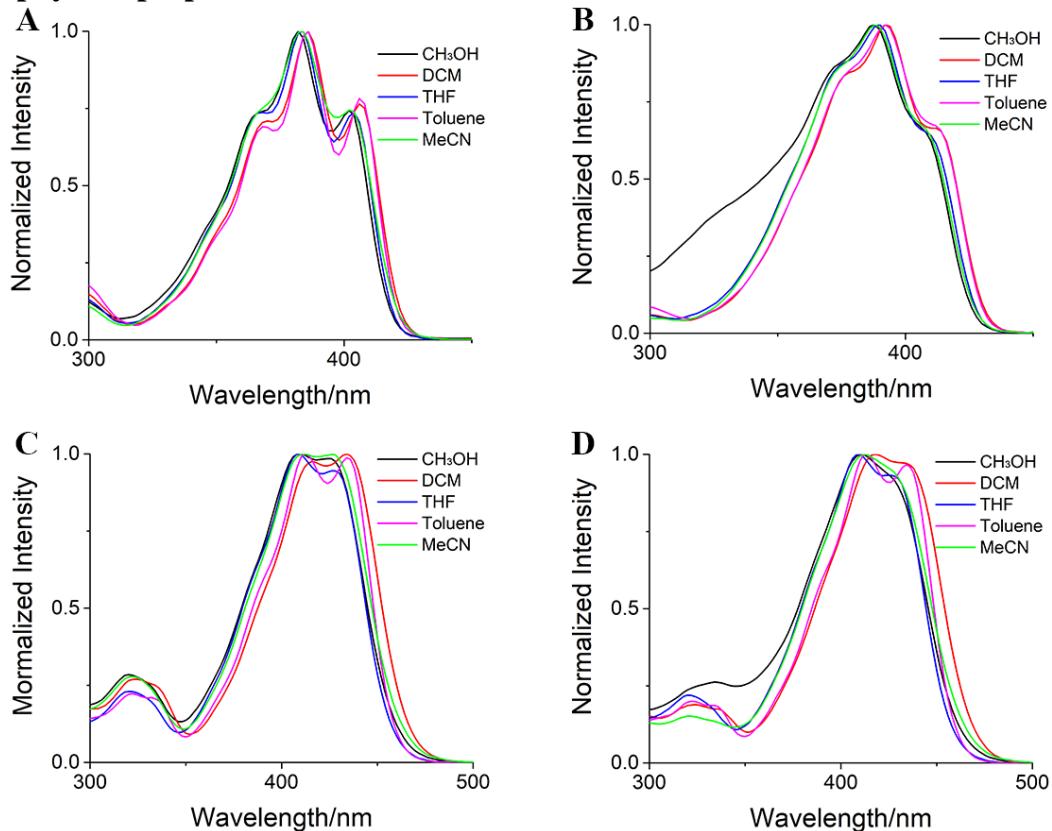


Figure S5. UV-vis spectra of (A) **1a**, (B) **1b**, (C) **2a** and (D) **2b** in different solvents.

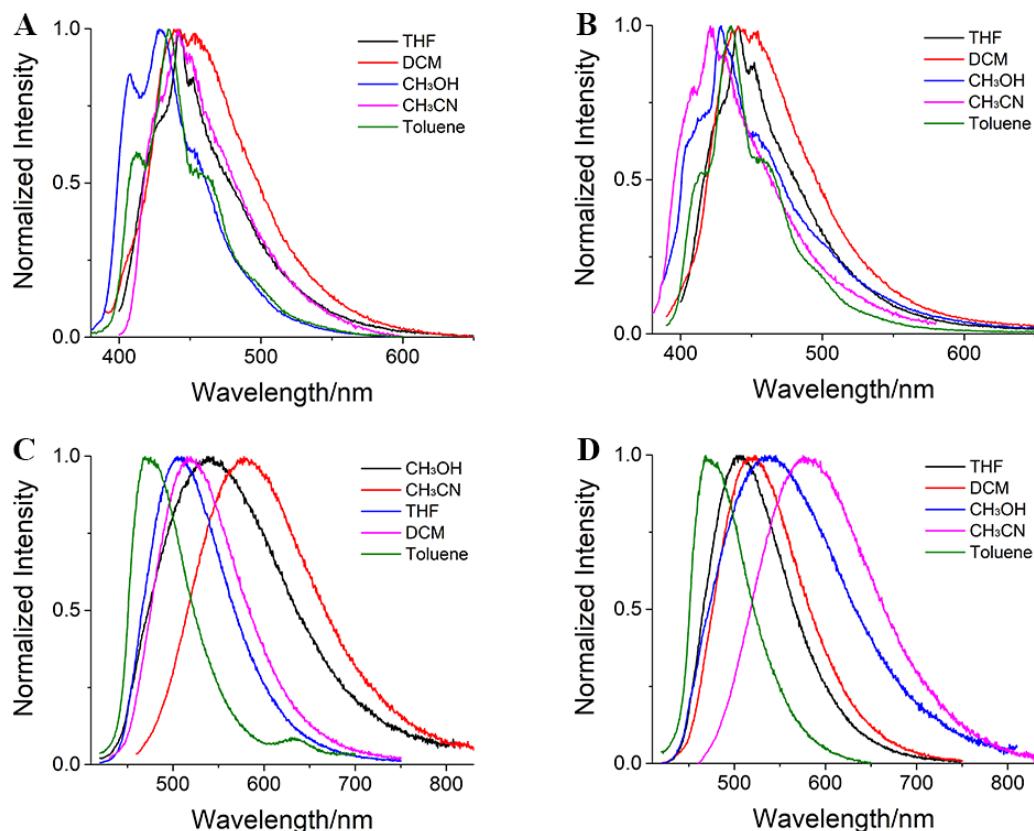


Figure S6. PL spectra of (A) **1a**, (B) **1b**, (C) **2a** and (D) **2b** in different solvents.

Table S4. Photophysical properties in organic solvents and in solid state

| dye s | solvents | $\lambda_{\text{abs}}^{\text{max}}$ (nm) | $\lambda_{\text{Em}}^{\text{max}}$ (nm) | $\log \epsilon^{\text{max}}$ | ϕ |
|-----------|--------------------|---|--|------------------------------|--------|
| 1a | DCM | 386 | 443 | 4.52 | 0.018 |
| | CH ₃ OH | 382 | 428 | 4.71 | 0.050 |
| | Toluene | 386 | 435 | 4.54 | 0.016 |
| | THF | 384 | 442 | 4.54 | 0.017 |
| | MeCN | 384 | 442 | 4.76 | 0.014 |
| | solid state | \ | 509 | \ | 0.137 |
| 1b | DCM | 392 | 442 | 4.65 | 0.019 |
| | CH ₃ OH | 388 | 428 | 4.65 | 0.046 |
| | Toluene | 390 | 435 | 4.59 | 0.012 |
| | THF | 390 | 442 | 4.69 | 0.035 |
| | MeCN | 388 | 441 | 4.79 | 0.014 |
| | solid state | \ | 505 | \ | 0.140 |
| 2a | DCM | 434 | 518 | 4.69 | 0.064 |
| | CH ₃ OH | 410 | 539 | 4.69 | 0.071 |
| | Toluene | 412 | 467 | 4.77 | 0.018 |
| | THF | 408 | 507 | 4.75 | 0.067 |
| | MeCN | 412 | 579 | 4.76 | 0.029 |
| | solid state | \ | 552 | \ | 0.108 |
| 2b | DCM | 418 | 522 | 4.61 | 0.089 |
| | CH ₃ OH | 410 | 538 | 4.77 | 0.193 |
| | Toluene | 414 | 468 | 4.63 | 0.049 |
| | THF | 410 | 506 | 4.64 | 0.061 |
| | MeCN | 412 | 579 | 4.81 | 0.027 |
| | solid state | \ | 571 | \ | 0.269 |

3. DFT calculations

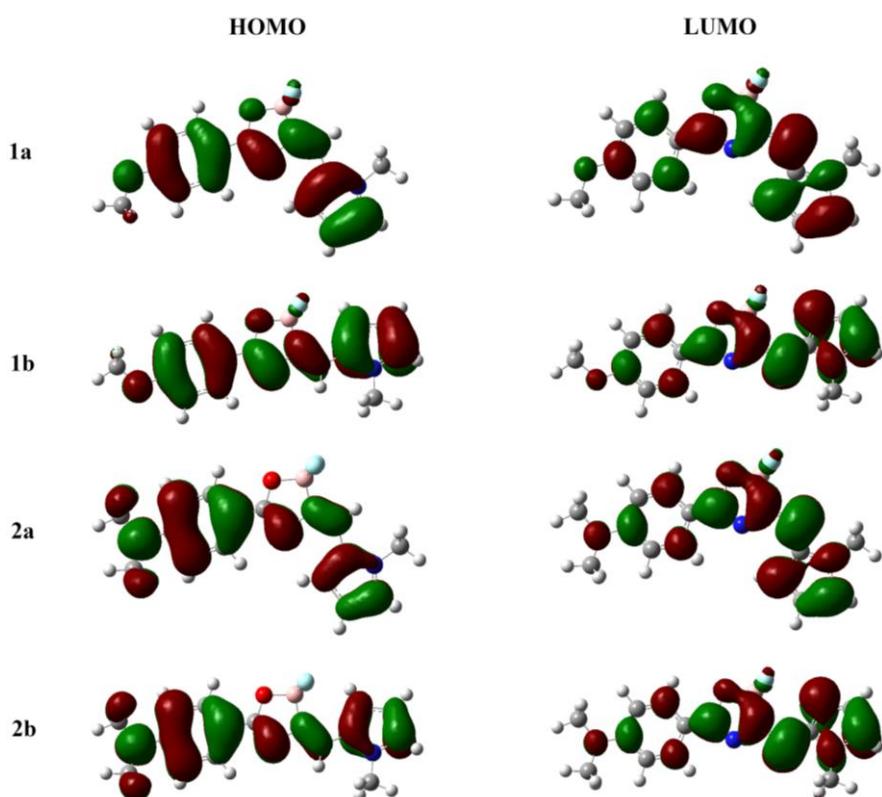


Figure S7. Optimized structures and orbits of **1** and **2** at the B3LYP/6-31G (d, p) level.

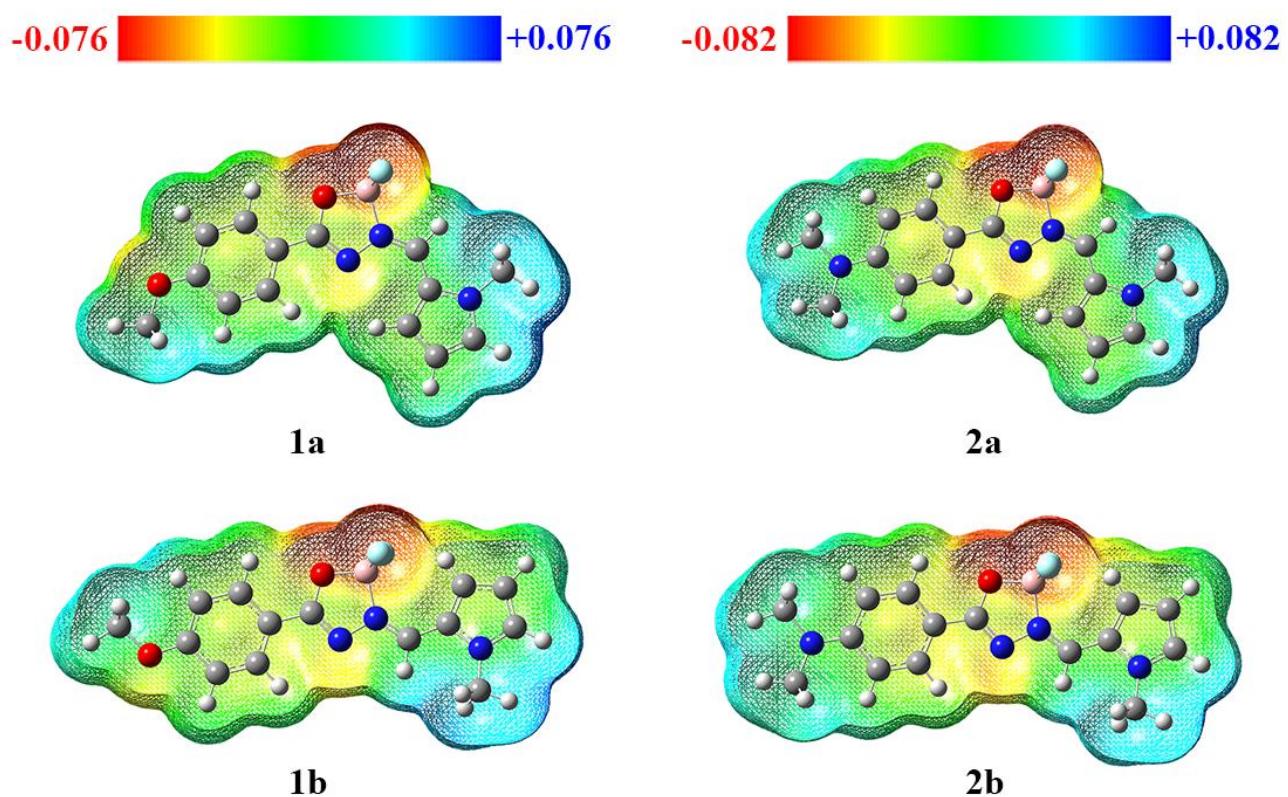


Figure S8. The molecular electrostatic potential in S_0 state.

Table S5. (E)-MPOAB **1a** coordinates.

| | | | | | | | |
|---|----------|----------|----------|---|----------|----------|----------|
| C | -3.59985 | -1.38832 | -2.3E-05 | N | 4.555853 | -1.24793 | 9.33E-06 |
| C | -4.64615 | -0.45095 | -1E-05 | C | 5.667953 | -0.30303 | -0.00016 |
| C | -4.35173 | 0.923944 | 1.95E-05 | O | -5.96422 | -0.77064 | -2.9E-05 |
| C | -3.03565 | 1.353926 | 3.81E-05 | C | -6.33435 | -2.14991 | -6.9E-05 |
| C | -1.97834 | 0.424323 | 3.53E-05 | H | -3.80404 | -2.45167 | -4.8E-05 |
| C | -2.2816 | -0.94585 | -3.8E-06 | H | -5.17393 | 1.631517 | 2.8E-05 |
| C | -0.59511 | 0.889934 | 6.28E-05 | H | -2.80998 | 2.414011 | 6.24E-05 |
| N | 0.43288 | 0.06719 | 0.000051 | H | -1.47351 | -1.66883 | -1.5E-05 |
| N | 1.548307 | 0.888191 | 1.76E-05 | H | 3.525365 | 1.200431 | -4.3E-05 |
| O | -0.36036 | 2.196162 | -3.1E-05 | H | 1.410554 | -2.21868 | 0.000133 |
| C | 2.769024 | 0.421621 | 6.08E-06 | H | 3.224533 | -4.24649 | 5.3E-05 |
| B | 1.111908 | 2.420724 | -7.7E-06 | H | 5.667361 | -3.05275 | 1.73E-05 |
| F | 1.541687 | 3.062478 | -1.14246 | H | 6.600007 | -0.86732 | 0.000252 |
| F | 1.541628 | 3.062491 | 1.14247 | H | 5.641967 | 0.327968 | -0.89213 |
| C | 3.193128 | -0.9299 | 3.89E-05 | H | 5.641628 | 0.328645 | 0.891318 |
| C | 2.485686 | -2.14619 | 8.45E-05 | H | -7.42411 | -2.16494 | -8.5E-05 |
| C | 3.424046 | -3.18476 | 3.94E-05 | H | -5.96302 | -2.66182 | -0.89461 |
| C | 4.687647 | -2.59701 | 2.98E-05 | H | -5.96304 | -2.66187 | 0.894452 |

Table S6. (Z)-MPOAB **1b** coordinates.

| | | | | | | | |
|---|----------|----------|----------|---|----------|----------|----------|
| C | -4.08563 | 1.644598 | 0.142539 | N | 4.791146 | 1.251367 | -0.05135 |
| C | -5.01126 | 0.588357 | 0.042428 | C | 4.681133 | 2.686606 | -0.29148 |
| C | -4.55176 | -0.73172 | -0.0695 | O | -6.31878 | 0.950899 | 0.063266 |
| C | -3.18173 | -0.98409 | -0.07992 | C | -7.31278 | -0.06957 | -0.03373 |
| C | -2.25373 | 0.061034 | 0.019242 | H | -4.46226 | 2.658313 | 0.228454 |
| C | -2.72765 | 1.383495 | 0.131166 | H | -5.24535 | -1.55933 | -0.14903 |
| C | -0.82129 | -0.22187 | 0.004946 | H | -2.82444 | -2.0039 | -0.16709 |
| N | 0.085419 | 0.722116 | 0.075473 | H | -2.01595 | 2.197775 | 0.209087 |
| N | 1.314534 | 0.064562 | 0.043313 | H | 2.206672 | 1.875664 | 0.064237 |
| O | -0.41406 | -1.48492 | -0.08256 | H | 3.739471 | -1.82615 | 0.475656 |
| C | 2.392035 | 0.804463 | 0.065796 | H | 6.432662 | -1.55776 | 0.365143 |
| C | 3.733724 | 0.347723 | 0.096415 | H | 6.904069 | 1.086348 | -0.06995 |
| B | 1.06922 | -1.515 | -0.10584 | H | 5.686225 | 3.102385 | -0.35483 |
| F | 1.562539 | -1.97478 | -1.30736 | H | 4.158102 | 2.885826 | -1.23003 |
| F | 1.582642 | -2.21983 | 0.967346 | H | 4.153097 | 3.177866 | 0.529663 |
| C | 4.301358 | -0.9251 | 0.287429 | H | -8.27243 | 0.445827 | 0.003168 |
| C | 5.692005 | -0.78064 | 0.245013 | H | -7.23016 | -0.61783 | -0.97865 |
| C | 5.96017 | 0.570118 | 0.030046 | H | -7.24503 | -0.77312 | 0.803405 |

Table S7. (E)-MPOAB **2a** coordinates.

| | | | | | | | |
|---|----------|----------|----------|---|----------|----------|----------|
| C | 3.247144 | -1.38395 | -0.00109 | C | 6.720474 | 0.079662 | 0.002251 |
| C | 4.328414 | -0.45964 | -0.00112 | C | 5.937582 | -2.31196 | 0.000581 |
| C | 4.004745 | 0.923814 | -0.0004 | H | 3.441163 | -2.44878 | -0.00111 |
| C | 2.686989 | 1.346621 | -0.00021 | H | 4.790845 | 1.667915 | 3.04E-06 |
| C | 1.62271 | 0.426607 | -0.00051 | H | 2.466369 | 2.408307 | 0.000251 |
| C | 1.935881 | -0.94643 | -0.00088 | H | 1.130789 | -1.67333 | -0.00092 |
| C | 0.247318 | 0.889641 | -0.00025 | H | -3.87565 | 1.205961 | -5.8E-05 |
| N | -0.78614 | 0.067141 | -0.00023 | H | -1.76445 | -2.21465 | 0.000641 |
| N | -1.89869 | 0.887815 | -0.00015 | H | -3.57921 | -4.24391 | 0.000946 |
| O | 0.007685 | 2.197843 | 7.57E-06 | H | -6.02198 | -3.04961 | 0.000161 |
| C | -3.12102 | 0.425723 | -5.3E-05 | H | -5.99373 | 0.332495 | 0.891404 |
| B | -1.46147 | 2.419679 | 0.00015 | H | -5.99353 | 0.332586 | -0.89169 |
| F | -1.89477 | 3.062594 | -1.14196 | H | -6.95254 | -0.86263 | -0.00035 |
| C | -3.54649 | -0.92714 | 5.14E-05 | H | 7.67048 | -0.45321 | 0.000473 |
| C | -2.83961 | -2.14275 | 0.000502 | H | 6.692229 | 0.726193 | -0.88372 |
| C | -3.77904 | -3.18216 | 0.000675 | H | 6.691696 | 0.719918 | 0.892853 |
| C | -5.04176 | -2.59497 | 0.00029 | H | 7.018849 | -2.4433 | -0.00244 |
| N | -4.90874 | -1.24462 | -0.00011 | H | 5.537234 | -2.81254 | 0.891154 |
| C | -6.01985 | -0.29924 | -0.00017 | H | 5.53224 | -2.81602 | -0.88562 |
| N | 5.630659 | -0.88753 | -0.00196 | F | -1.89472 | 3.061958 | 1.142674 |

Table S8. (Z)-MPOAB **2b** coordinates.

| | | | | | | | |
|---|----------|----------|----------|---|----------|----------|----------|
| C | -3.81657 | 1.368086 | 0.107518 | C | -6.98437 | -0.66613 | -0.0772 |
| C | -4.72139 | 0.274132 | 0.02257 | H | -4.19157 | 2.380917 | 0.179649 |
| C | -4.16296 | -1.02818 | -0.07332 | H | -4.80793 | -1.8947 | -0.14179 |
| C | -2.79122 | -1.21717 | -0.08091 | H | -2.39103 | -2.22238 | -0.15509 |
| C | -1.90315 | -0.13055 | 0.005029 | H | -1.78048 | 2.015582 | 0.165704 |
| C | -2.449 | 1.163789 | 0.099091 | H | 2.473648 | 1.884503 | 0.051339 |
| C | -0.466 | -0.34448 | -0.00384 | H | 4.161685 | -1.74259 | 0.515886 |
| N | 0.4022 | 0.640181 | 0.064668 | H | 6.842476 | -1.36184 | 0.406103 |
| N | 1.657519 | 0.038299 | 0.040291 | H | 7.202057 | 1.292158 | -0.07083 |
| O | 0.002553 | -1.58919 | -0.08322 | H | 5.898002 | 3.247754 | -0.40255 |
| C | 2.703321 | 0.821935 | 0.061391 | H | 4.371546 | 2.953977 | -1.25748 |
| C | 4.064853 | 0.421516 | 0.101491 | H | 4.371868 | 3.275751 | 0.496683 |
| B | 1.482674 | -1.55137 | -0.09927 | H | -8.01249 | -0.30684 | -0.05423 |
| F | 2.004727 | -1.99683 | -1.29516 | H | -6.83654 | -1.21438 | -1.01629 |
| F | 2.023787 | -2.22566 | 0.981579 | H | -6.85325 | -1.37149 | 0.752945 |
| C | 4.685383 | -0.8217 | 0.314279 | C | -6.62716 | 1.81654 | 0.123415 |
| C | 6.070013 | -0.61861 | 0.272065 | H | -7.71492 | 1.758983 | 0.130307 |
| C | 6.281102 | 0.737513 | 0.035802 | H | -6.30935 | 2.322057 | 1.043704 |
| N | 5.082732 | 1.36677 | -0.05908 | H | -6.32529 | 2.438606 | -0.72899 |
| C | 4.911668 | 2.791839 | -0.32142 | N | -6.07902 | 0.469837 | 0.033198 |

4. Supporting Figures

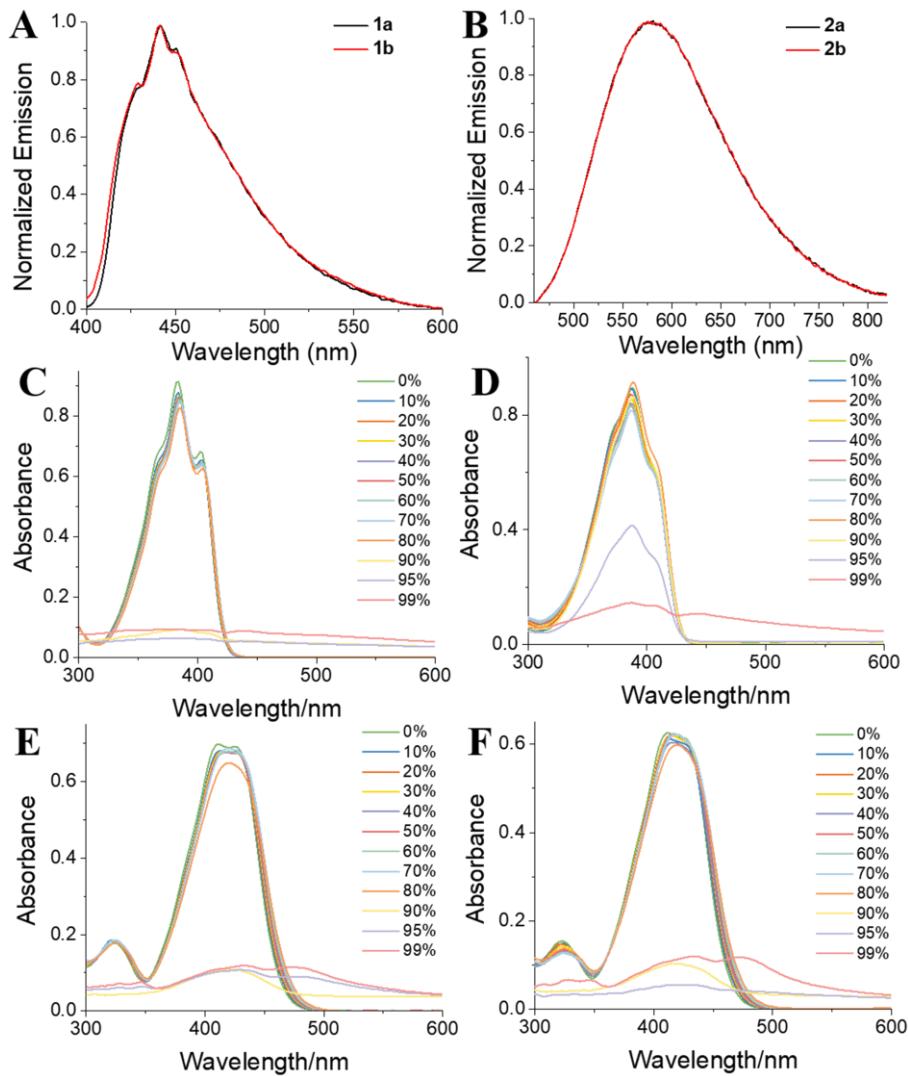


Figure S9. PL spectra of **1a**, **1b** (A) and **2a**, **2b** (B) in acetonitrile; Absorbance spectra of **1a** (1.8×10^{-5} mol/L, C), **1b** (1.7×10^{-5} mol/L, D), **2a** (1.4×10^{-5} mol/L, E) and **2b** (1.4×10^{-5} mol/L, F) in acetonitrile/water with different water fractions (f_w).

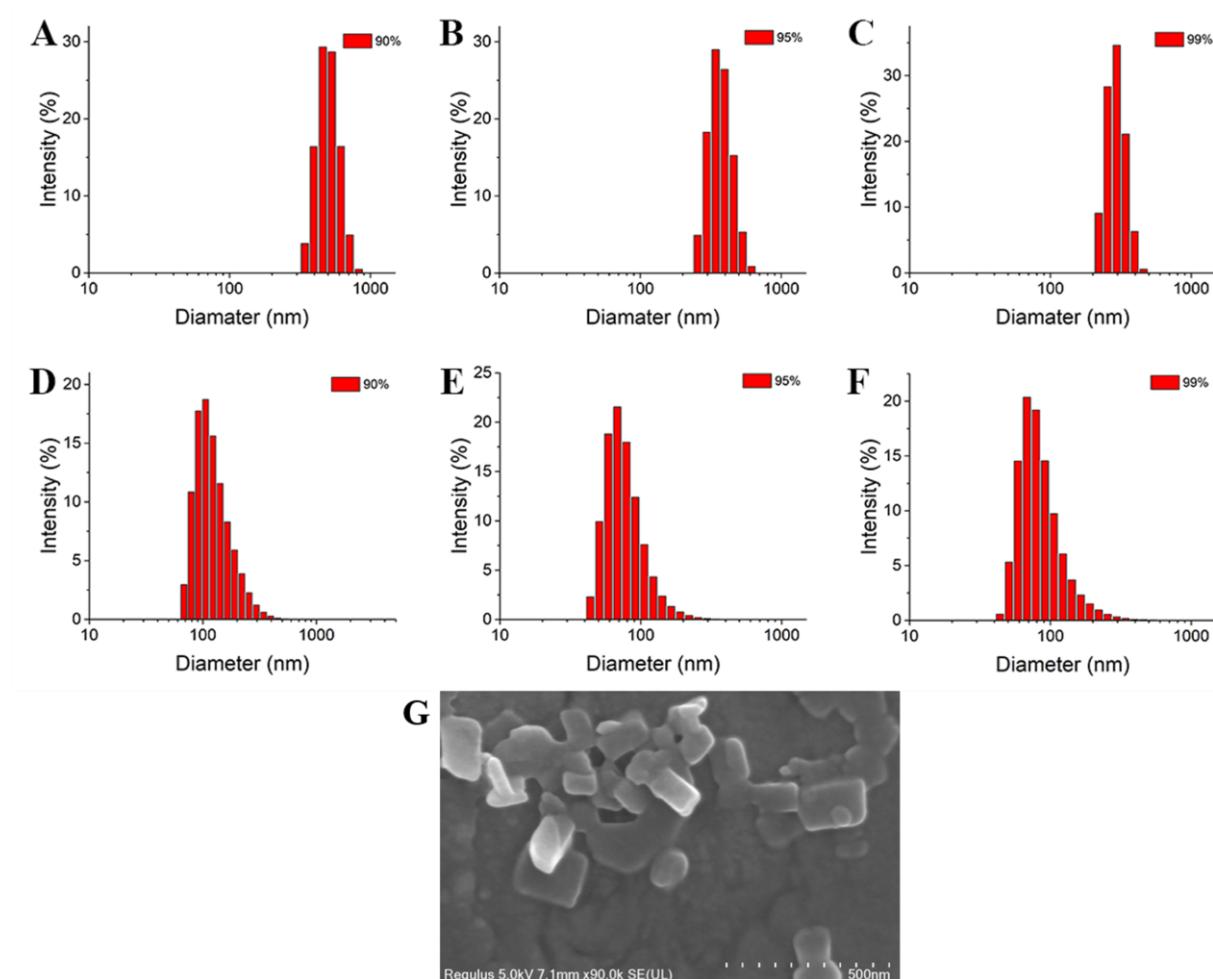
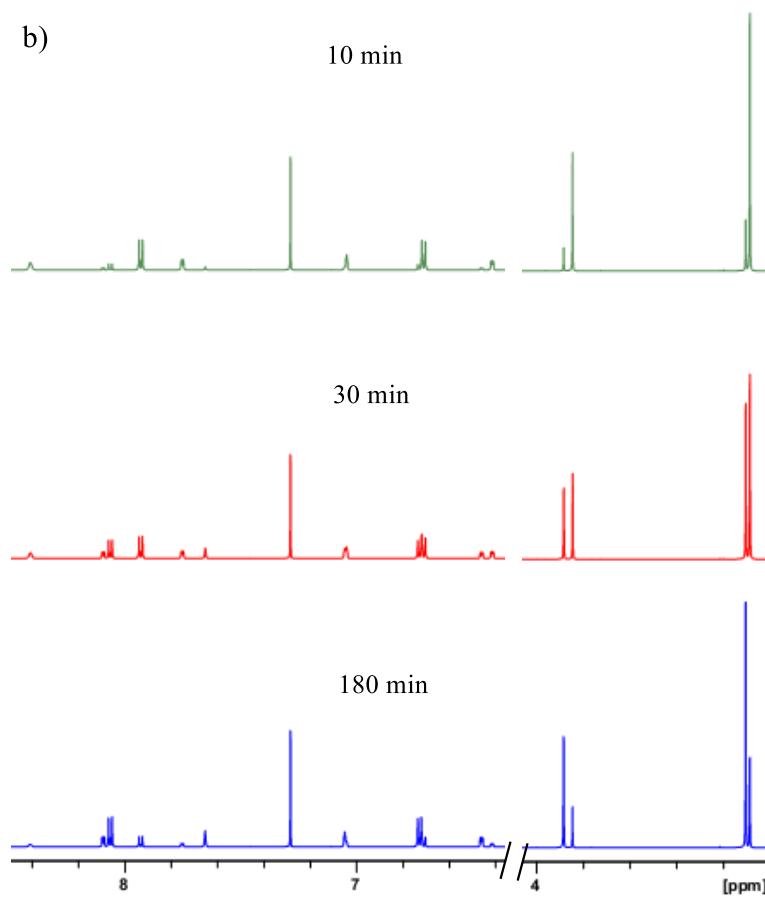
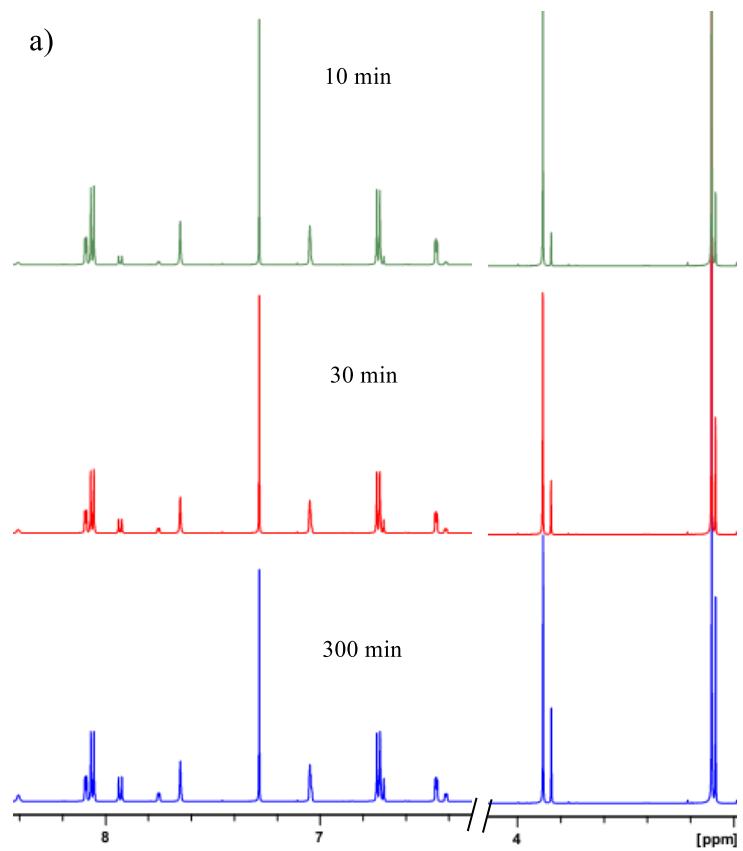


Figure S10. Particle size distributions of **2a** (A, B and C) and **2b** (D, E and F) in acetonitrile/water mixtures with $f_w = 90\%$ (A and D), 95% (B and E) and 99% (C and F). The concentration of samples is $15 \mu\text{M}$. Average diameter (nm) = A) 505, B) 373, C) 294, D) 129, E) 79, F) 89 and G) SEM of **2b** in acetonitrile/water mixtures with $f_w = 99\%$.



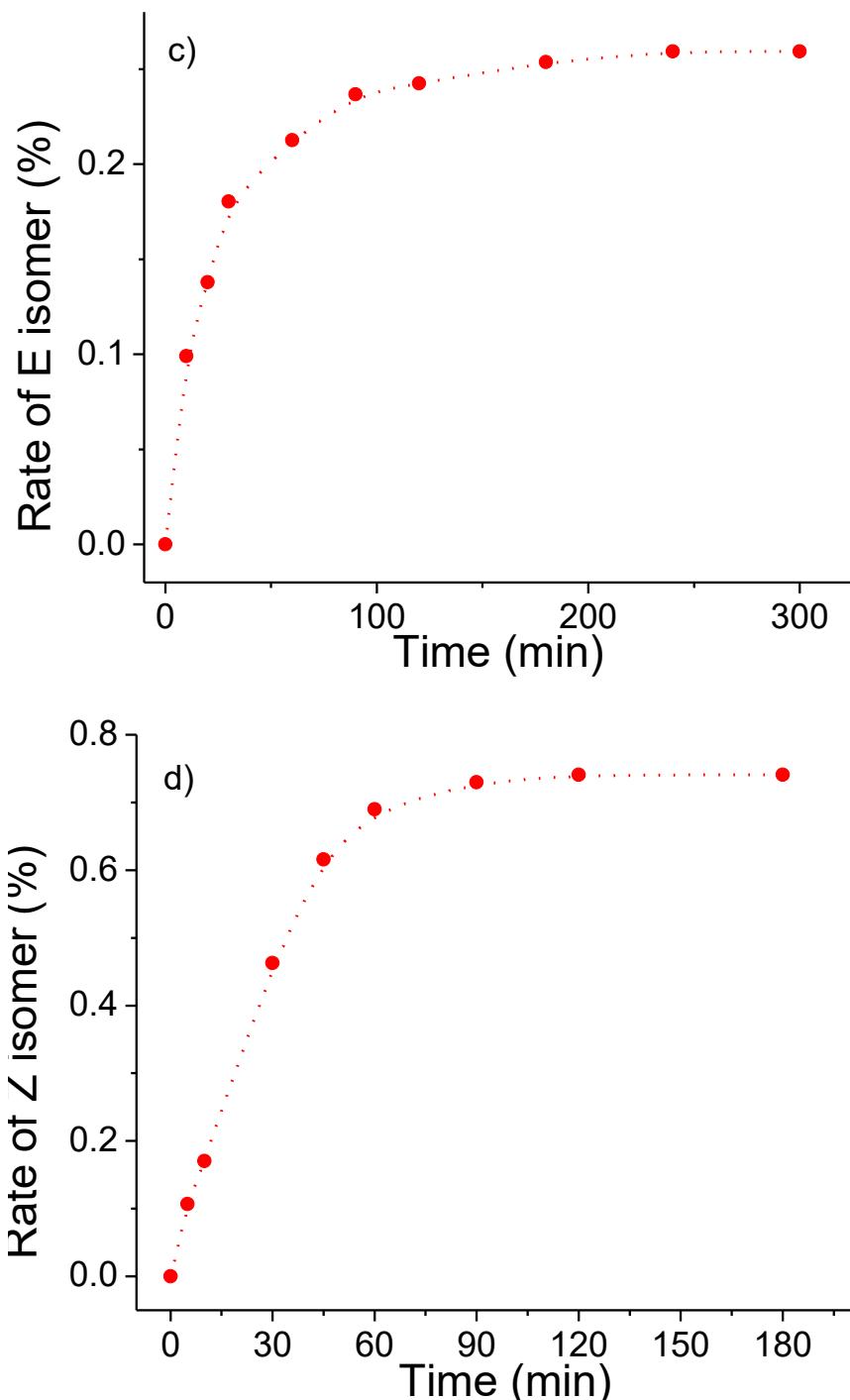


Figure S11. ^1H NMR spectra of **2a** (a) and **2b** (b) irradiated for 10 min, 30 min and 300 min in CDCl_3 . Variations in the amount of E isomer **2b** (c) by irradiating **2a** and Z isomer **2a** (c) by irradiating **2b** with a 18 W white light LED.

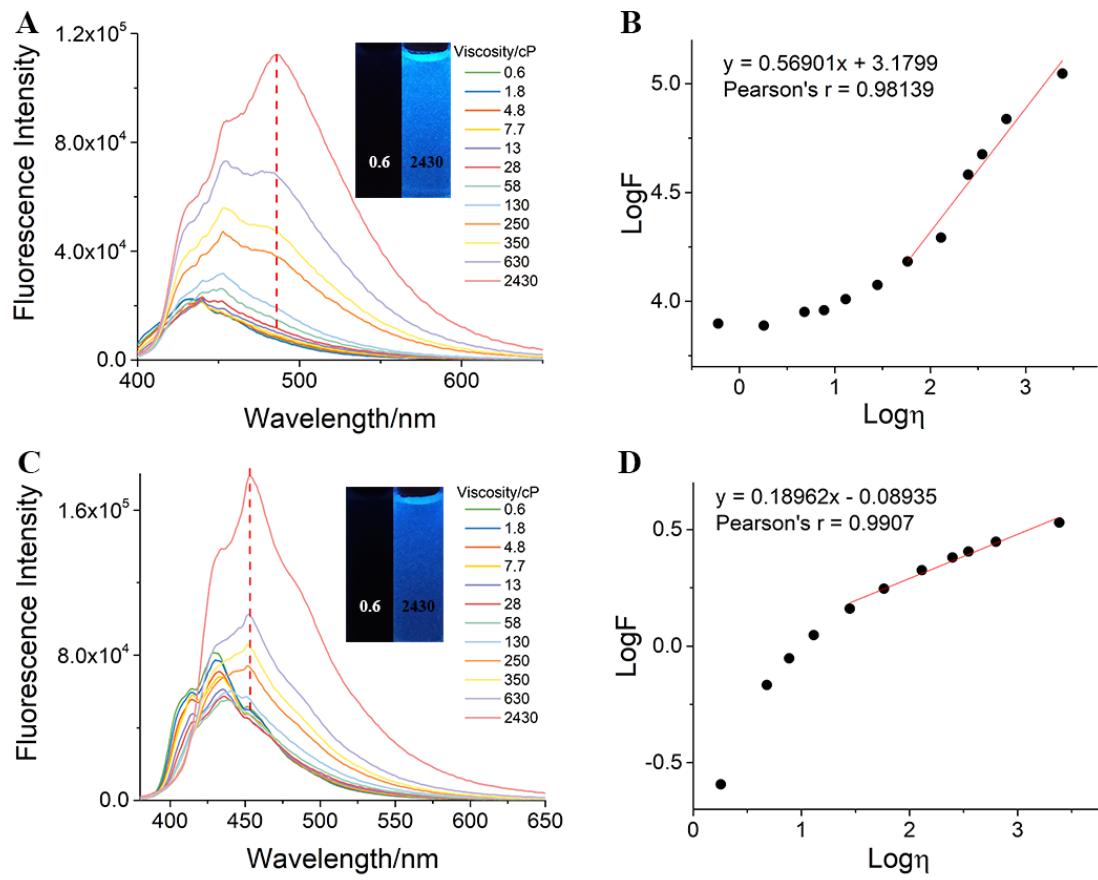


Figure S12. (A) PL spectra of **1a** in $\text{CH}_3\text{OH}/\text{glycerol}$ solutions with different viscosity. The concentration of **1a** is 2.47×10^{-5} mol/L in each solution. $\lambda_{\text{ex}} = 390$ nm. (B) **1a** working curve of $\log_{10}(\eta)$ (the logarithm of viscosity value) vs. $\log_{10}(F)$ (the logarithm of fluorescence intensity). (C) PL spectra of **1b** in $\text{CH}_3\text{OH}/\text{glycerol}$ solutions with different viscosity. The concentration of **1b** is 1.44×10^{-5} mol/L in each solution. $\lambda_{\text{ex}} = 370$ nm. (D) **1b** working curve of $\log_{10}(\eta)$ vs. $\log_{10}(F)$.

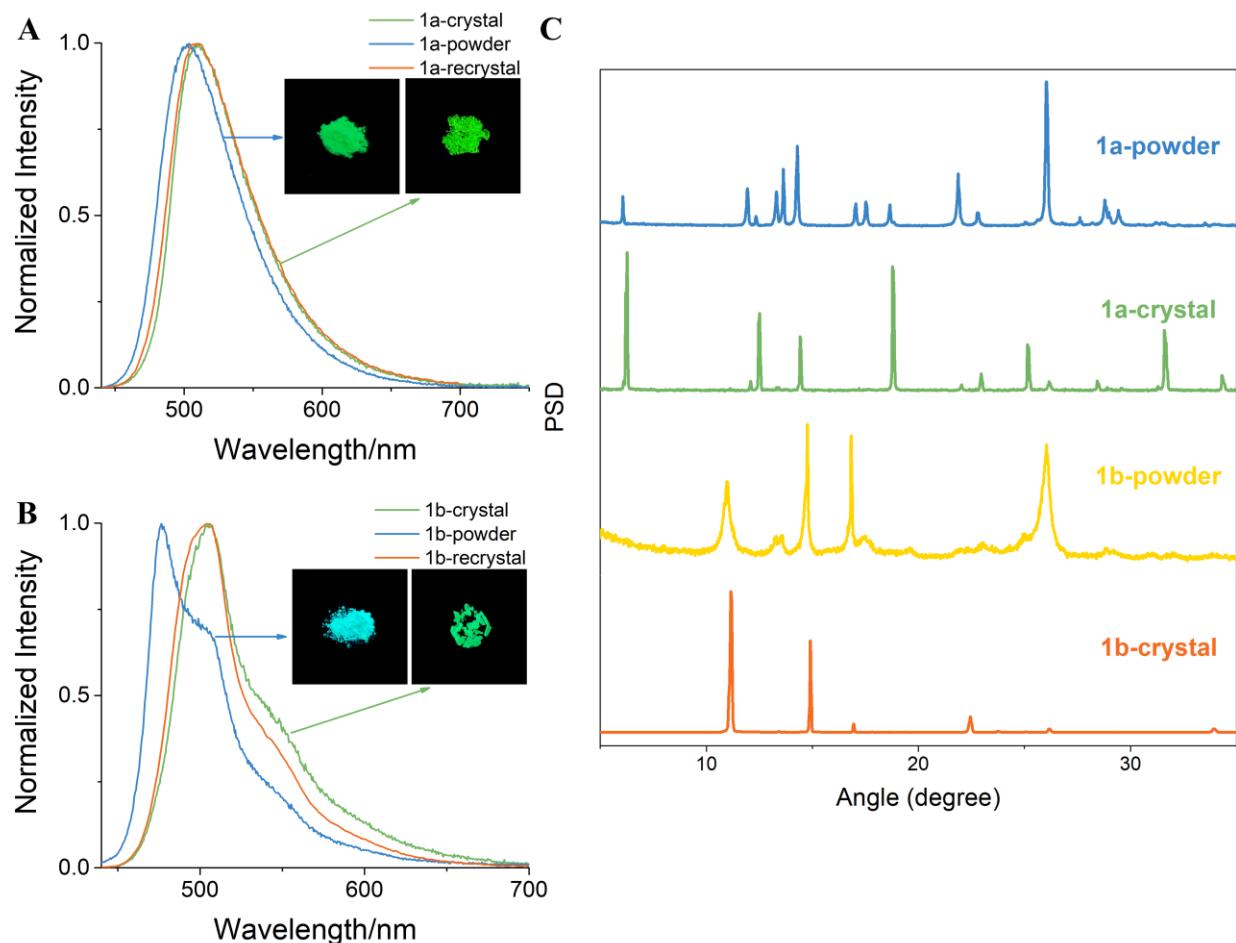


Figure S13. (A) The comparison of the PL spectra between **1a**-powder, **1a**-crystal and **1a**-recrystallized. (B) The comparison of the PL spectra between **1b**-powder, **1b**-crystal and **1b**-recrystallized. (C) XRD patterns of **1a** and **1b** in powder and crystal states. All powder samples were prepared by grinding the corresponding crystal in the agate mortar for 1 min.

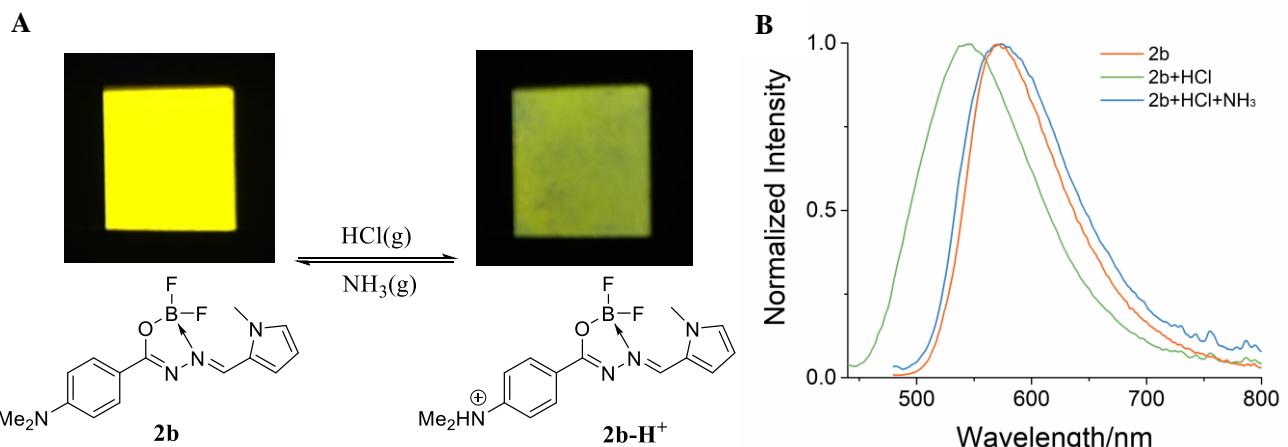


Figure S14. (A) Diagrammatic sketch of acid-base fluorescence response of **2b**. (B) The PL spectra comparison of **2b**, **2b**+HCl and **2b**+HCl+NH₃. The sample solution was dropped upon silex glass and dried to obtain the sample solid film. Every silex glass with sample just be exposed to the HCl or NH₃ vapors about 20 s.

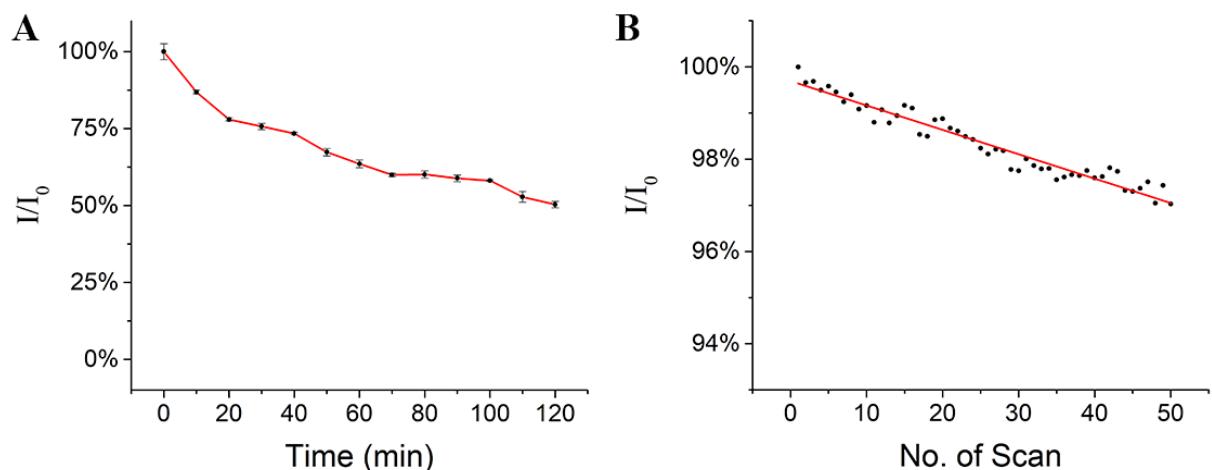
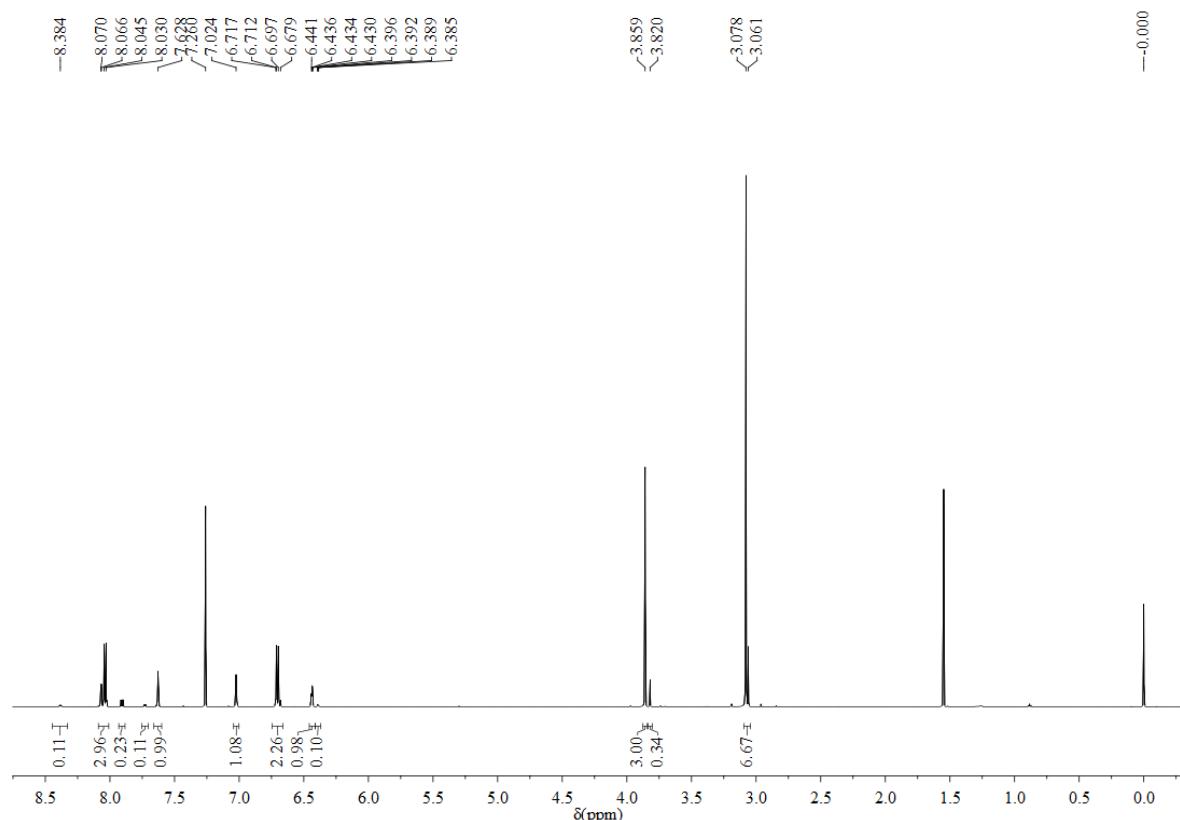
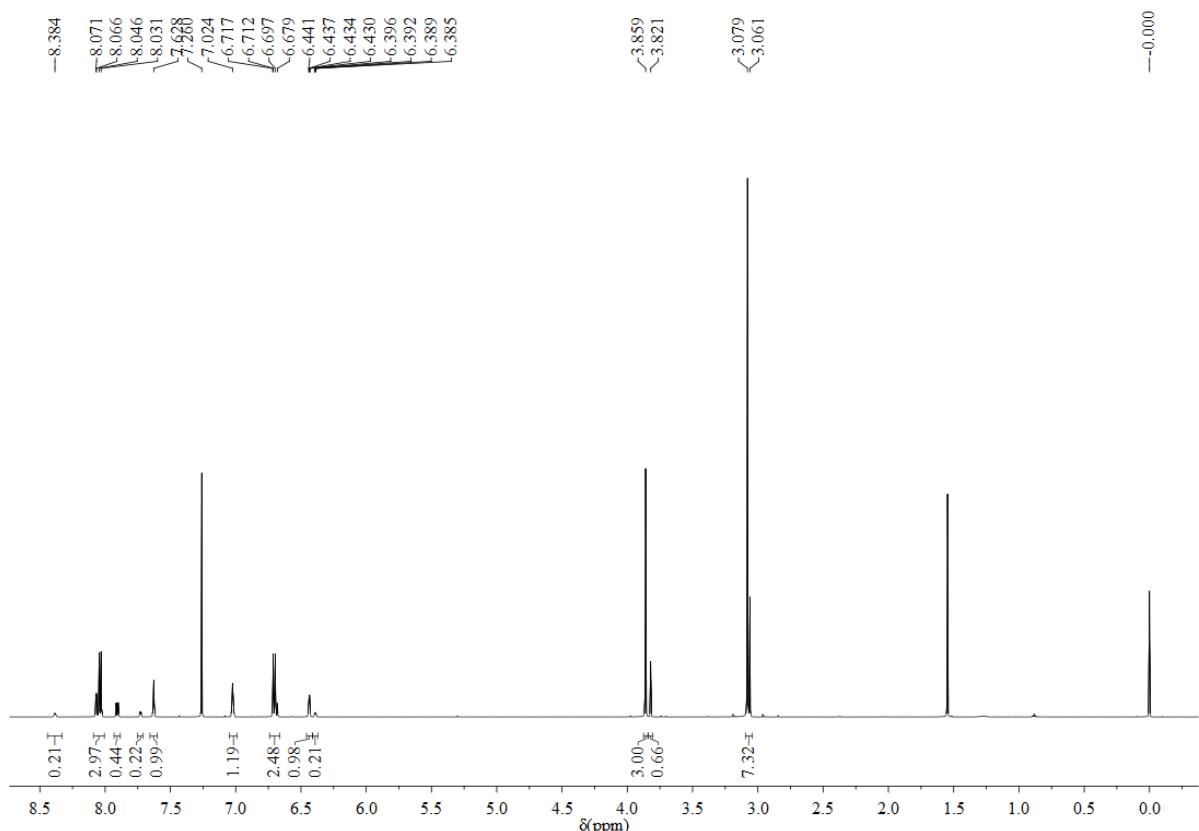


Figure S15. (A) Loss in fluorescence of **2a** NPs in acetonitrile/water mixtures ($f_w = 90\%$) with 2 h Green LED light irradiation. **2a** concentration is 15 μ M. The LED light's rated power is 18 W with 490 nm-530 nm green light. The average power density is about 900 mW/cm². And the data is shown as the mean value plus a standard deviation (\pm SD). (B) Loss in fluorescence of MCF-7 cells stained with **2a** NPs with increasing the number of scans of laser irradiation. Concentration: 15 μ M; λ_{ex} : 488 nm; scanning rate: 10 s frame⁻¹.

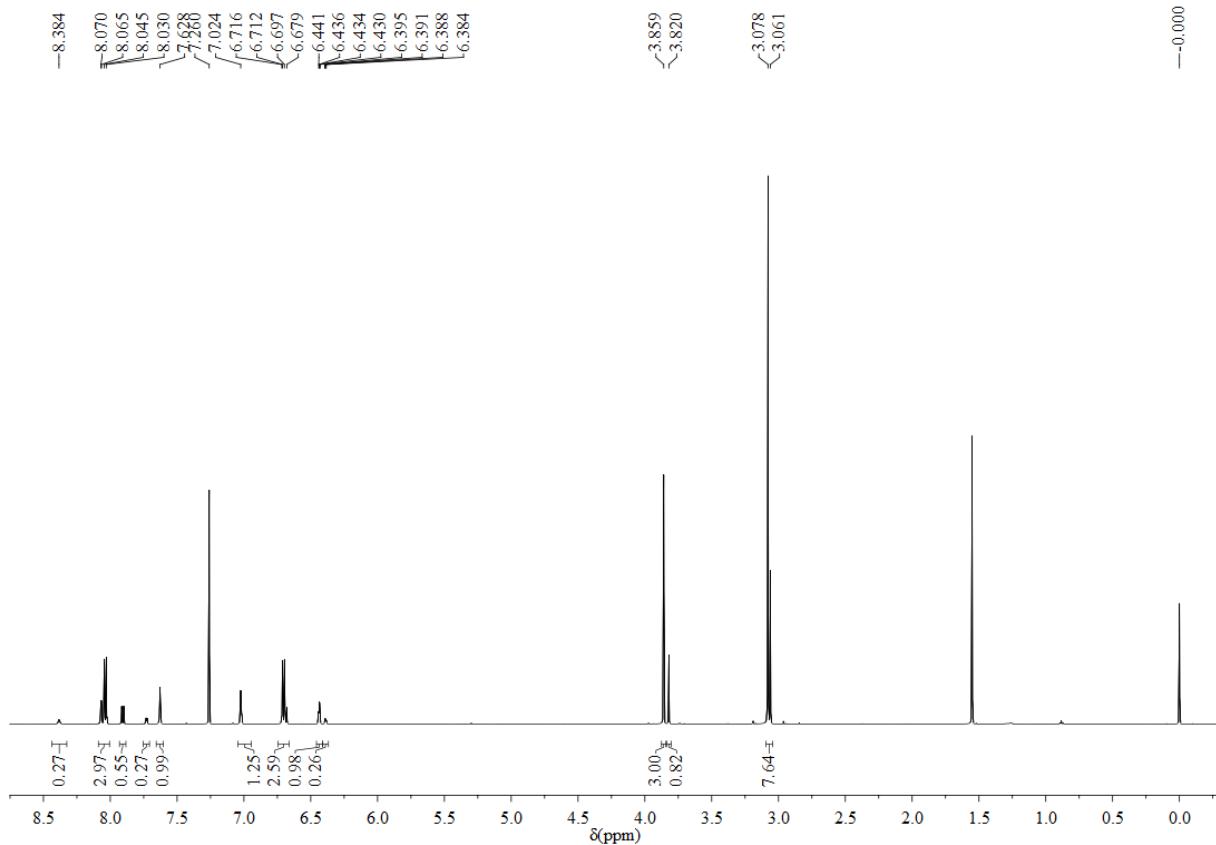
5. Photodynamic isomerization studies by ^1H NMR spectra



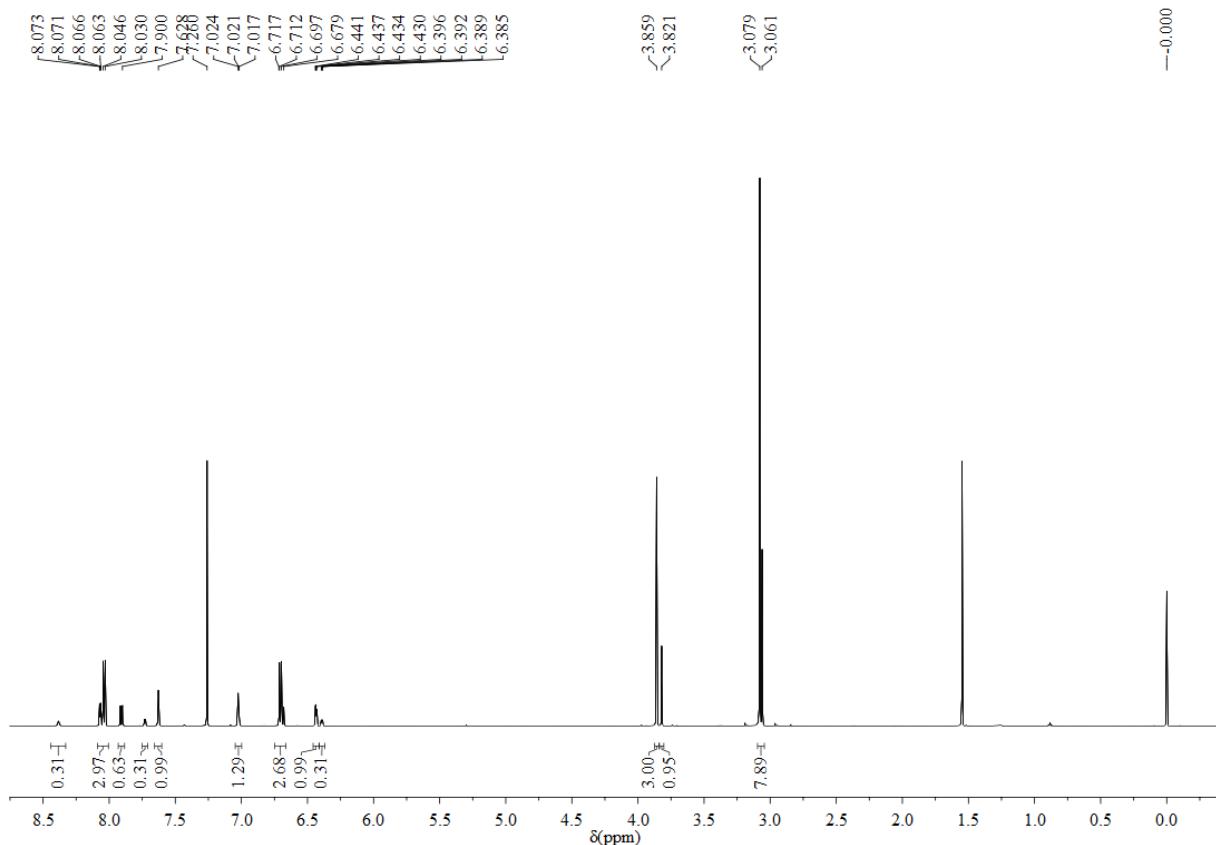
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 10 min in CDCl₃.



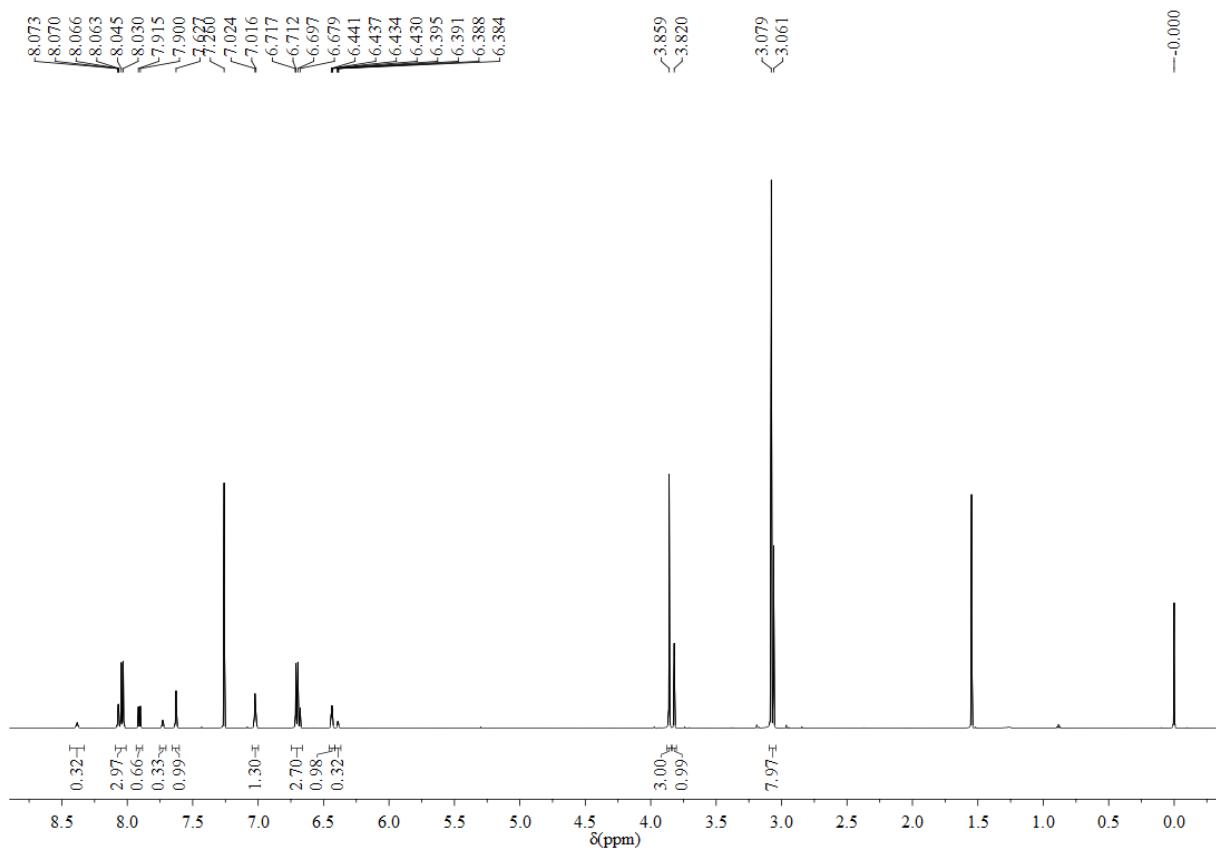
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 30 min in CDCl₃.



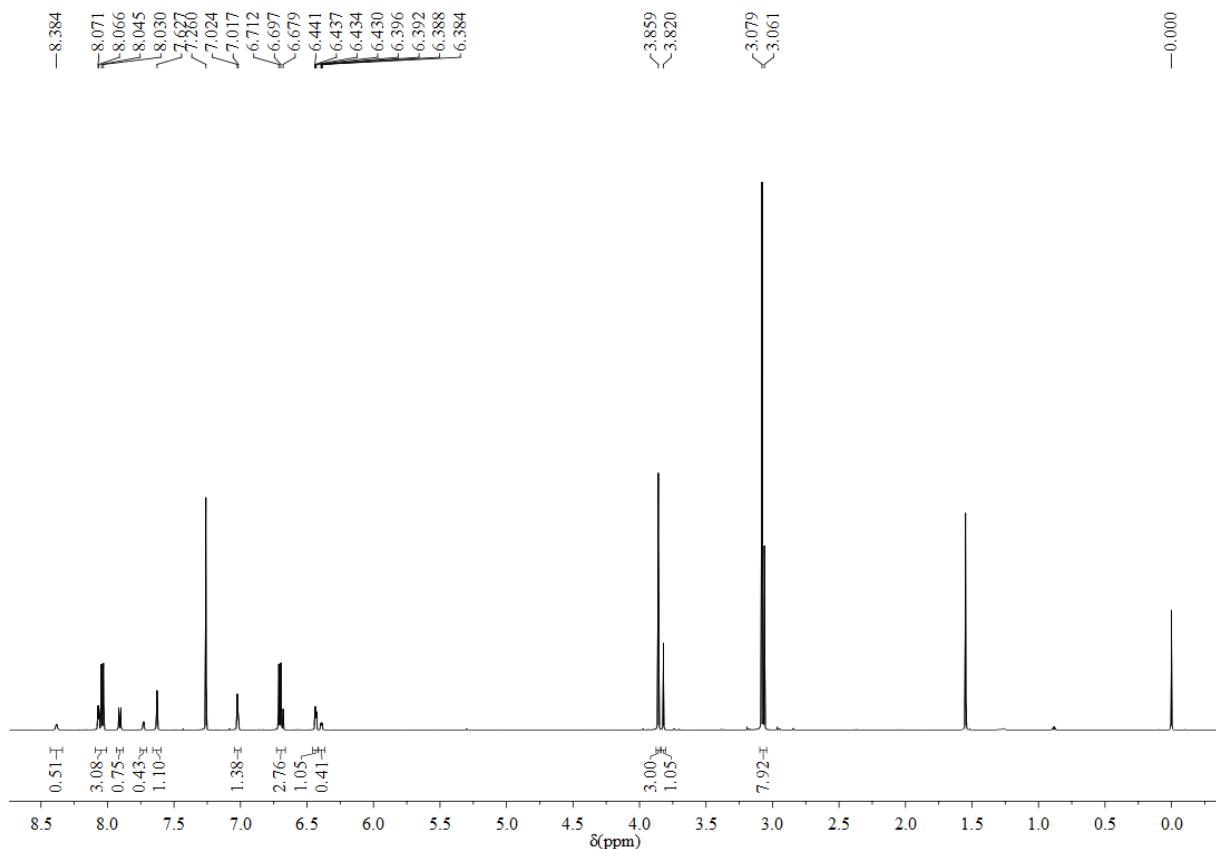
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 60 min in CDCl₃.



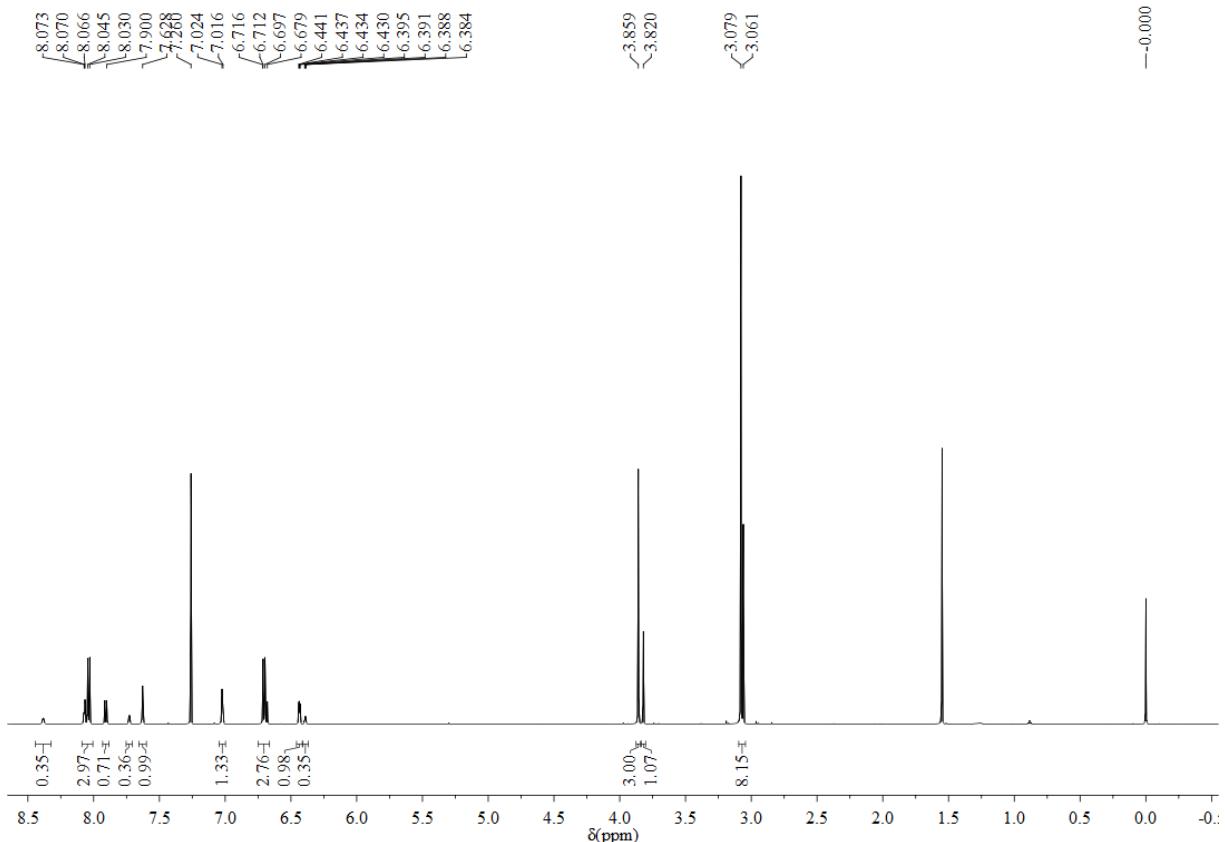
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 90 min in CDCl₃.



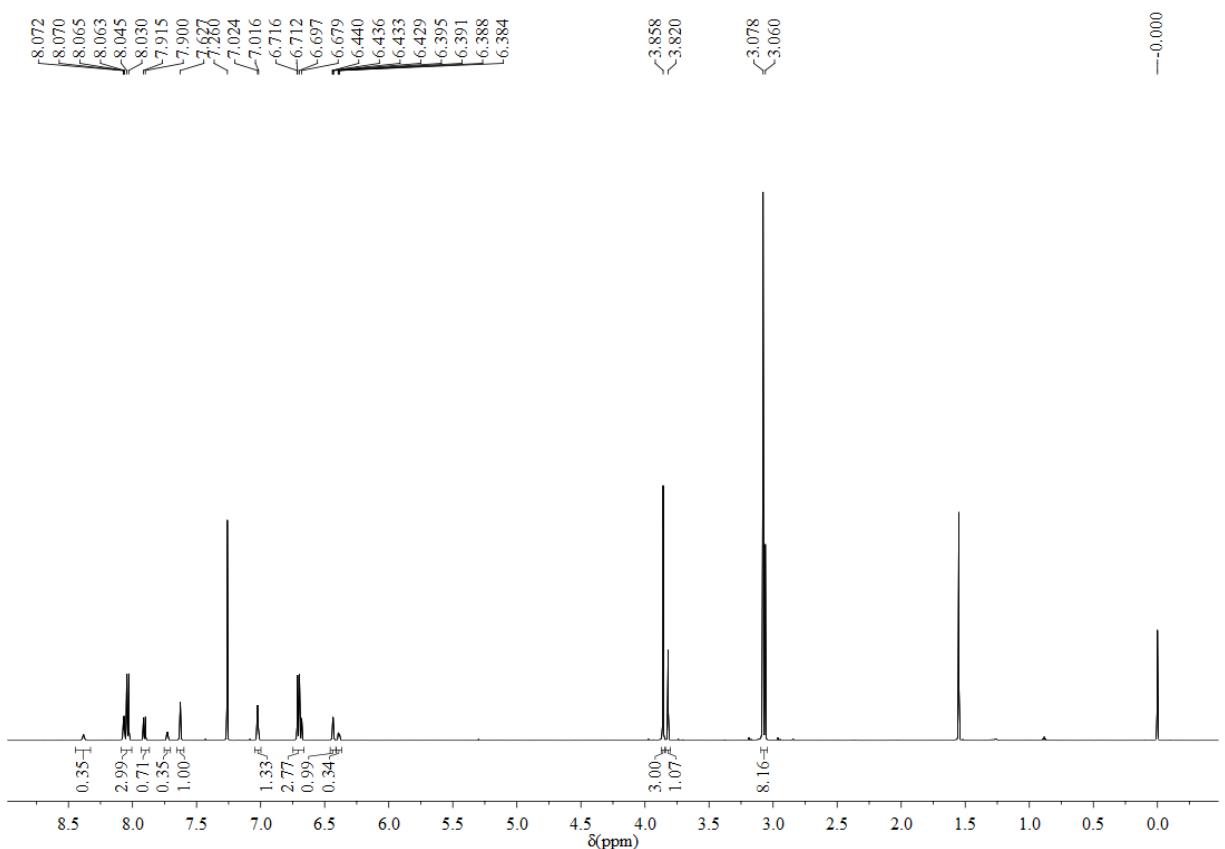
^1H NMR spectrum of **2a** which was irradiated by 18W white light LED for 120 min in CDCl_3 .



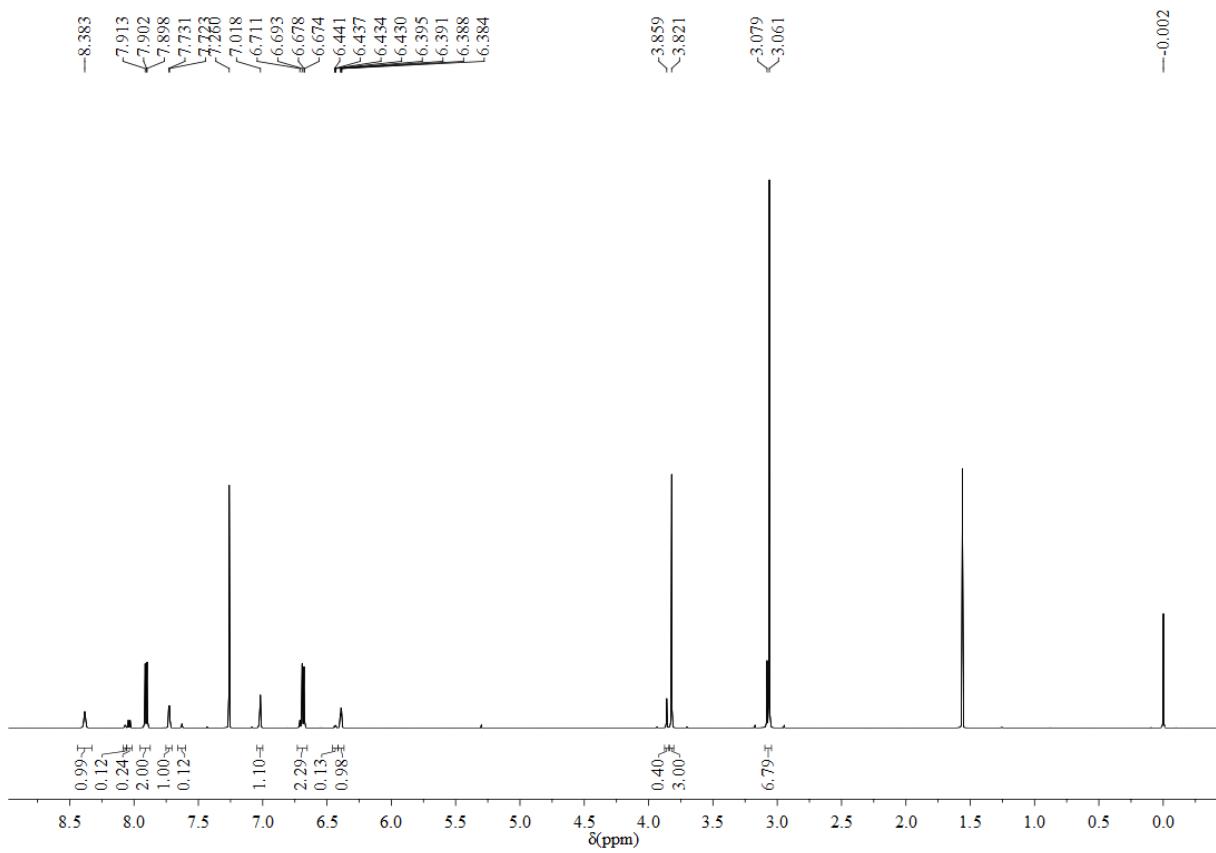
^1H NMR spectrum of **2a** which was irradiated by 18W white light LED for 180 min in CDCl_3 .



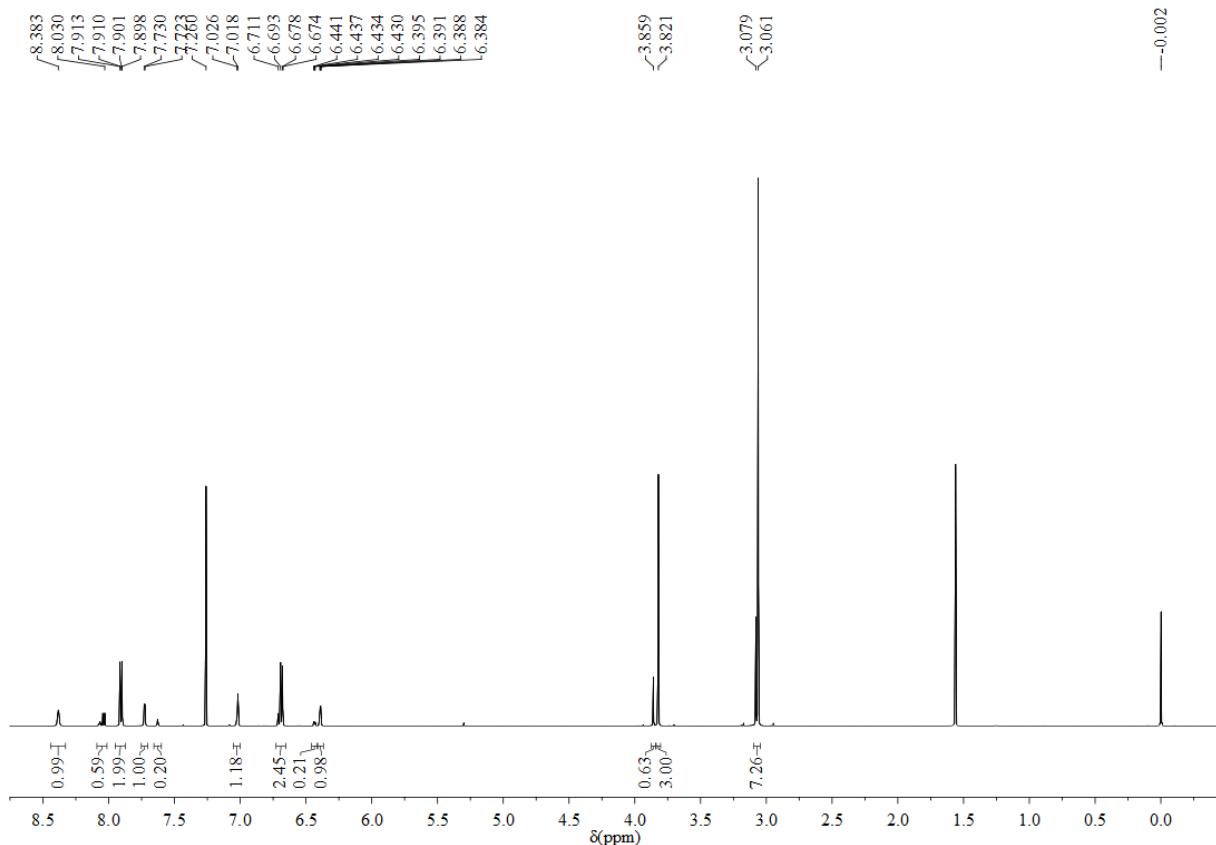
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 240 min in CDCl₃.



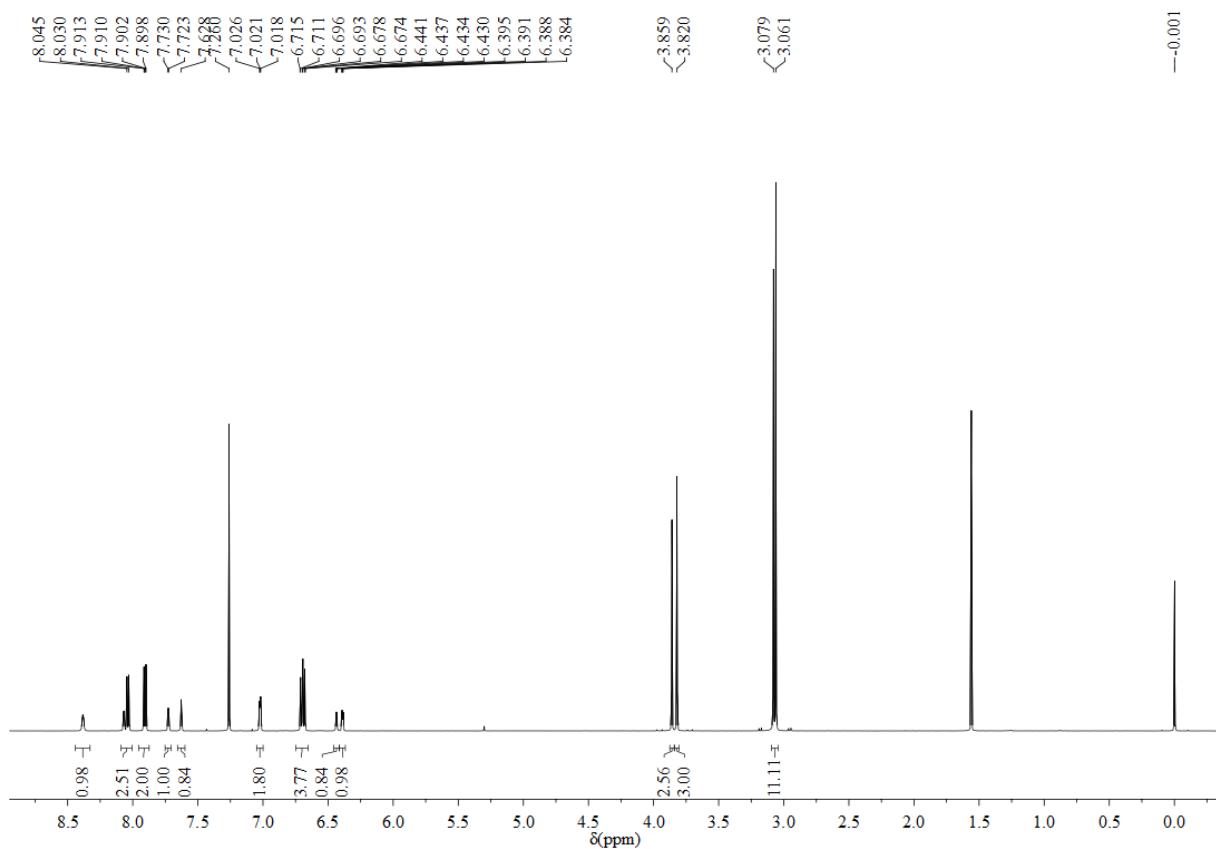
¹H NMR spectrum of **2a** which was irradiated by 18W white light LED for 300 min in CDCl₃.



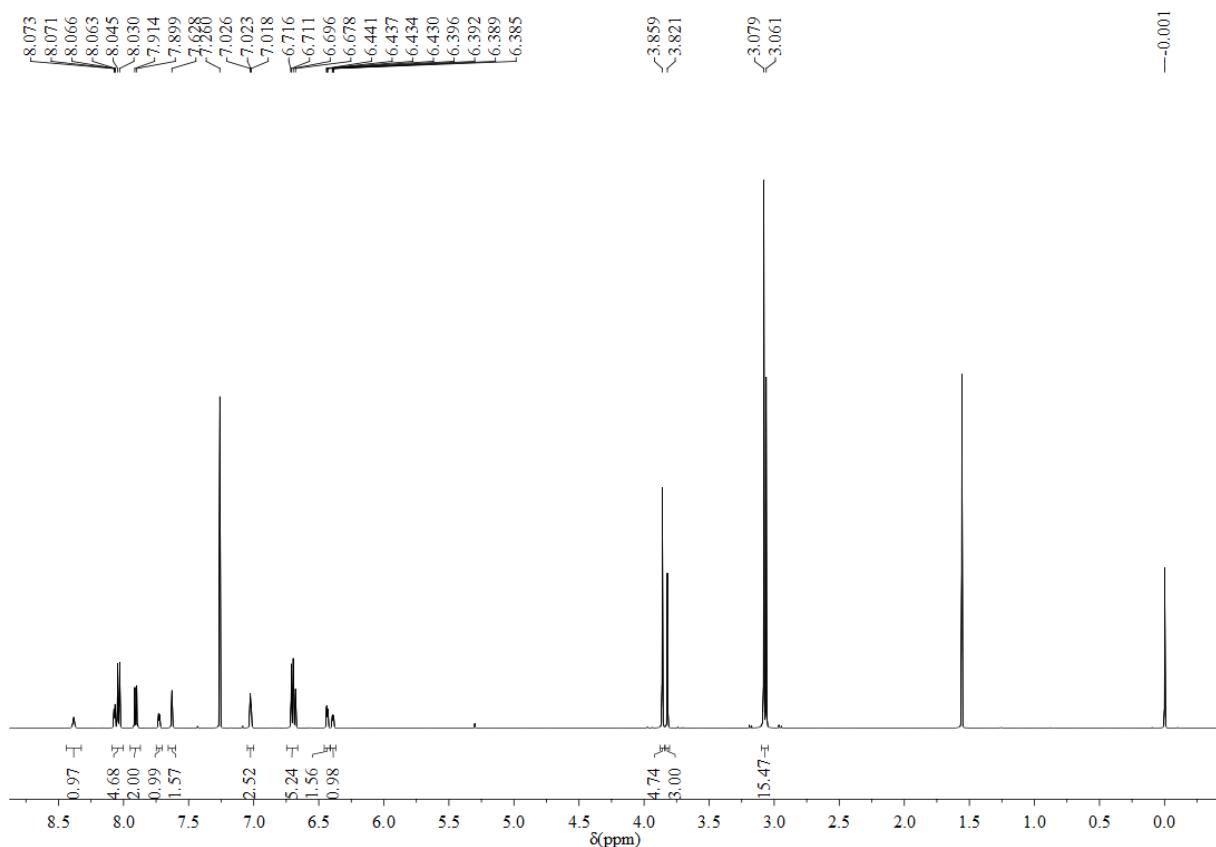
¹H NMR spectrum of **2b** which was irradiated by 18W white light LED for 5 min in CDCl₃.



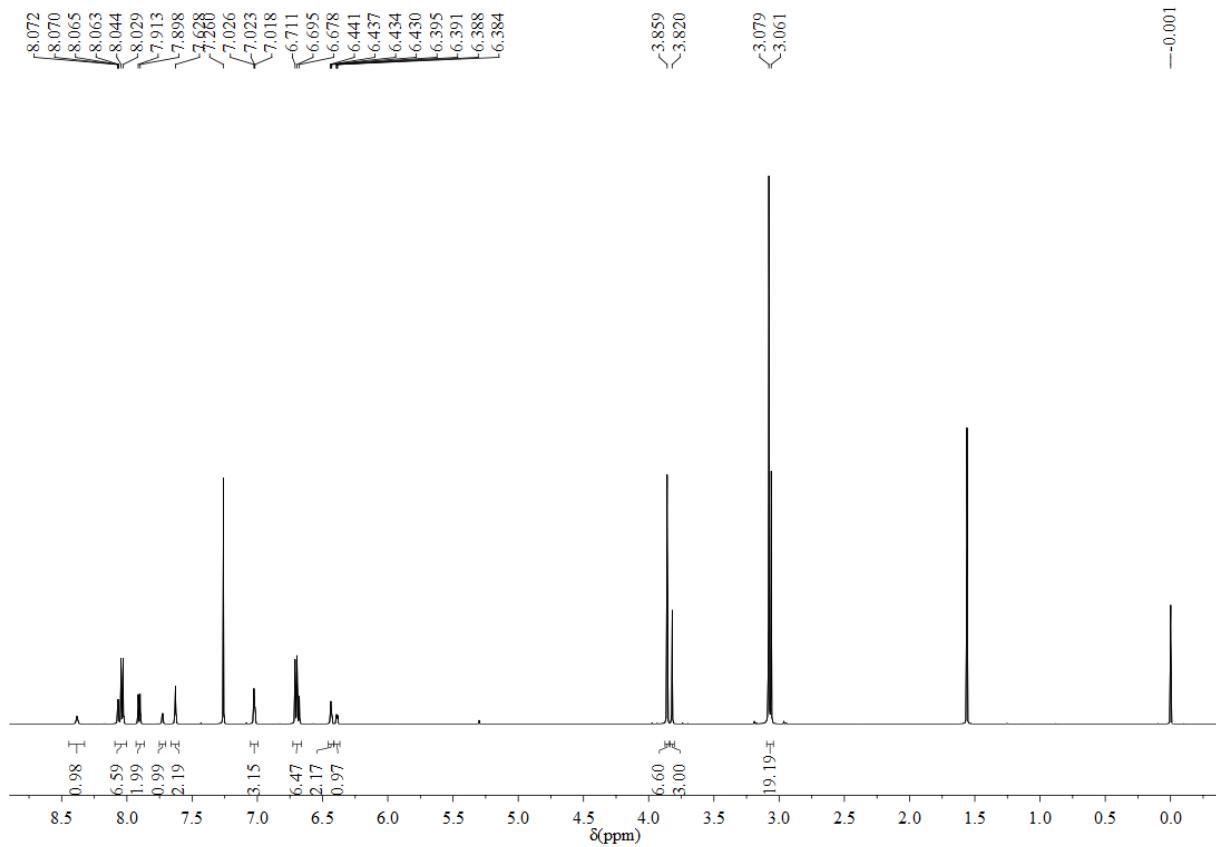
¹H NMR spectrum of **2b** which was irradiated by 18W white light LED for 10 min in CDCl₃.



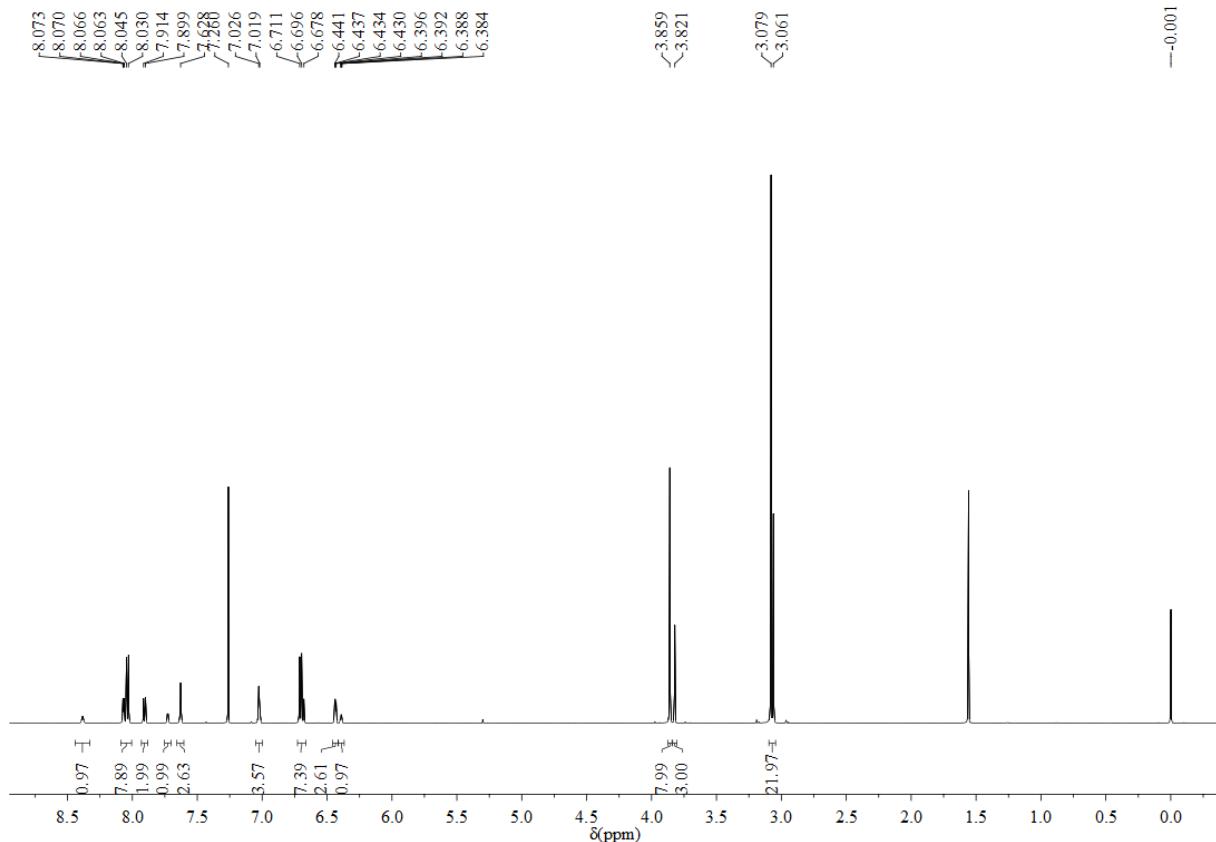
¹H NMR spectrum of **2b** which was irradiated by 18W white light LED for 30 min in CDCl₃.



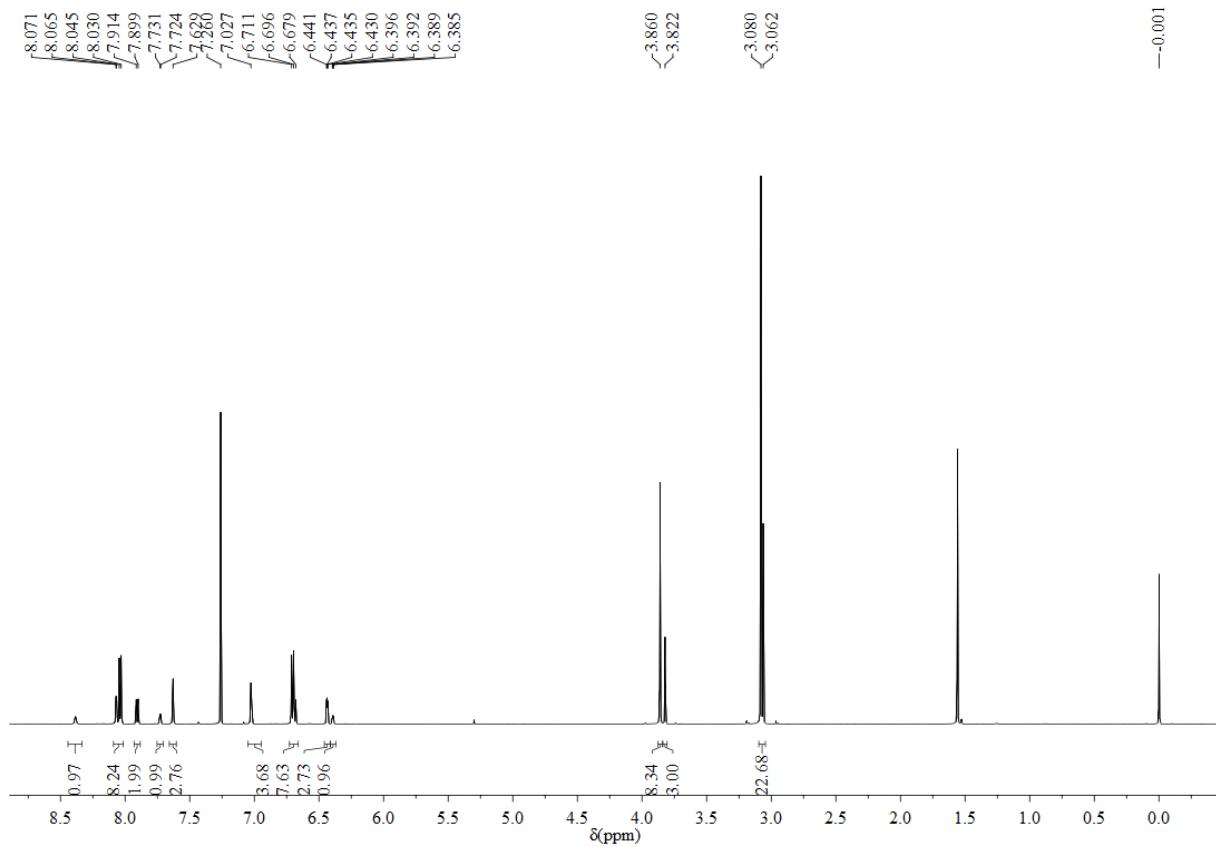
¹H NMR spectrum of **2b** which was irradiated by 18W white light LED for 45 min in CDCl₃.



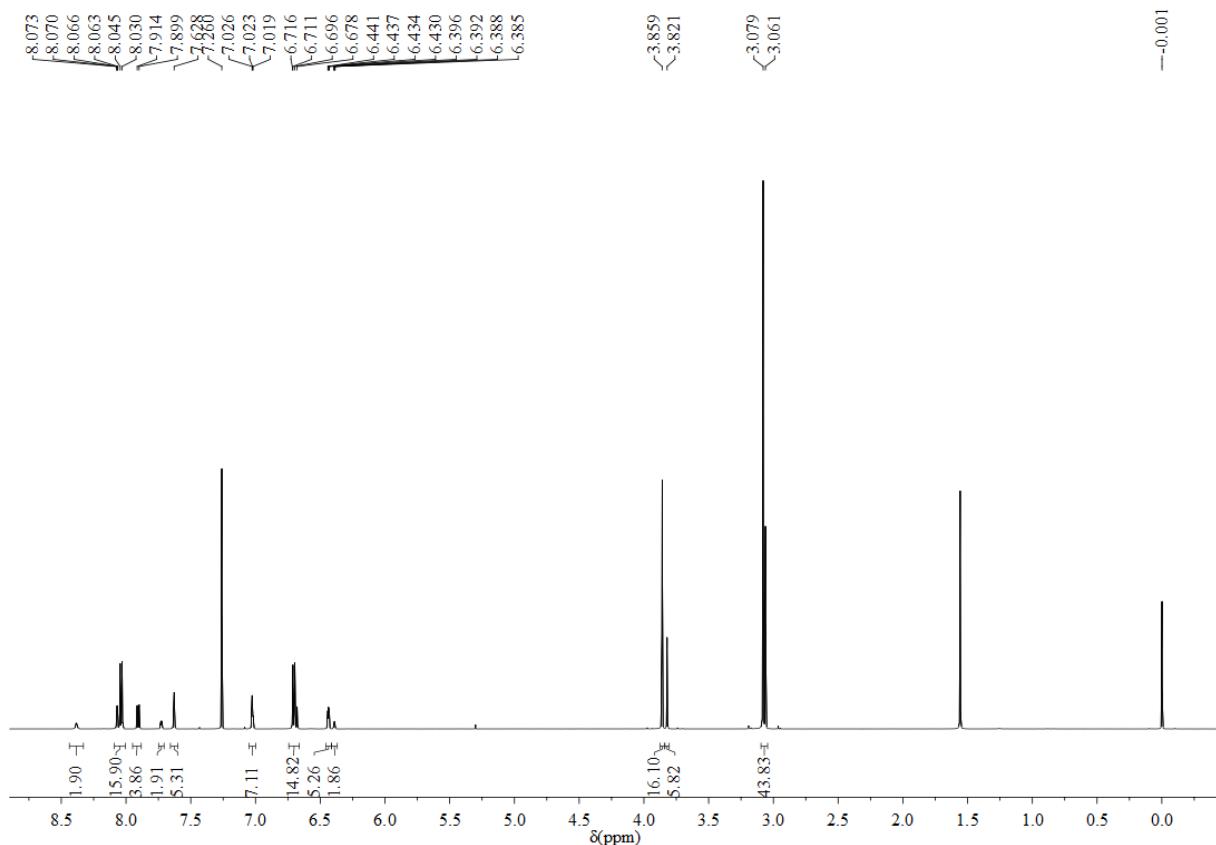
^1H NMR spectrum of **2b** which was irradiated by 18W white light LED for 60 min in CDCl_3 .



^1H NMR spectrum of **2b** which was irradiated by 18W white light LED for 90 min in CDCl_3 .

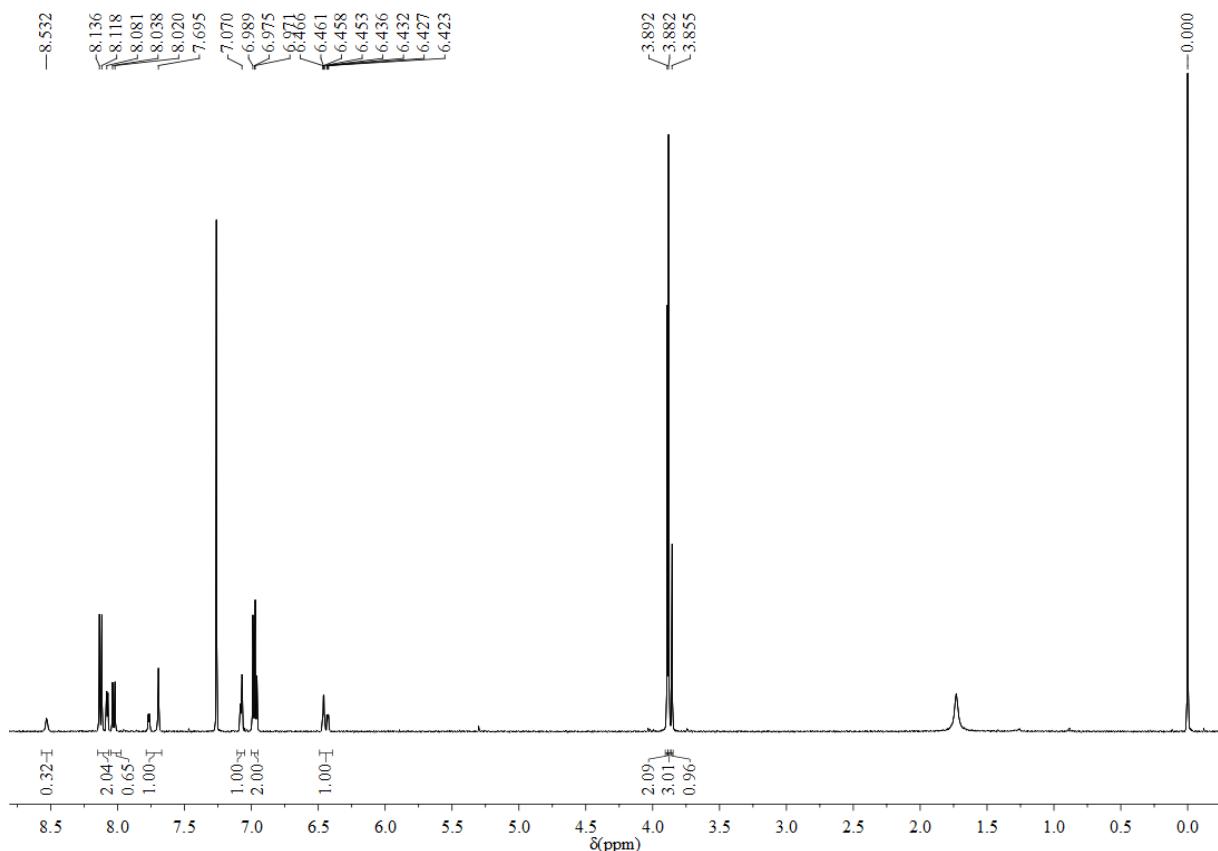


^1H NMR spectrum of **2b** which was irradiated by 18W white light LED for 120 min in CDCl_3 .

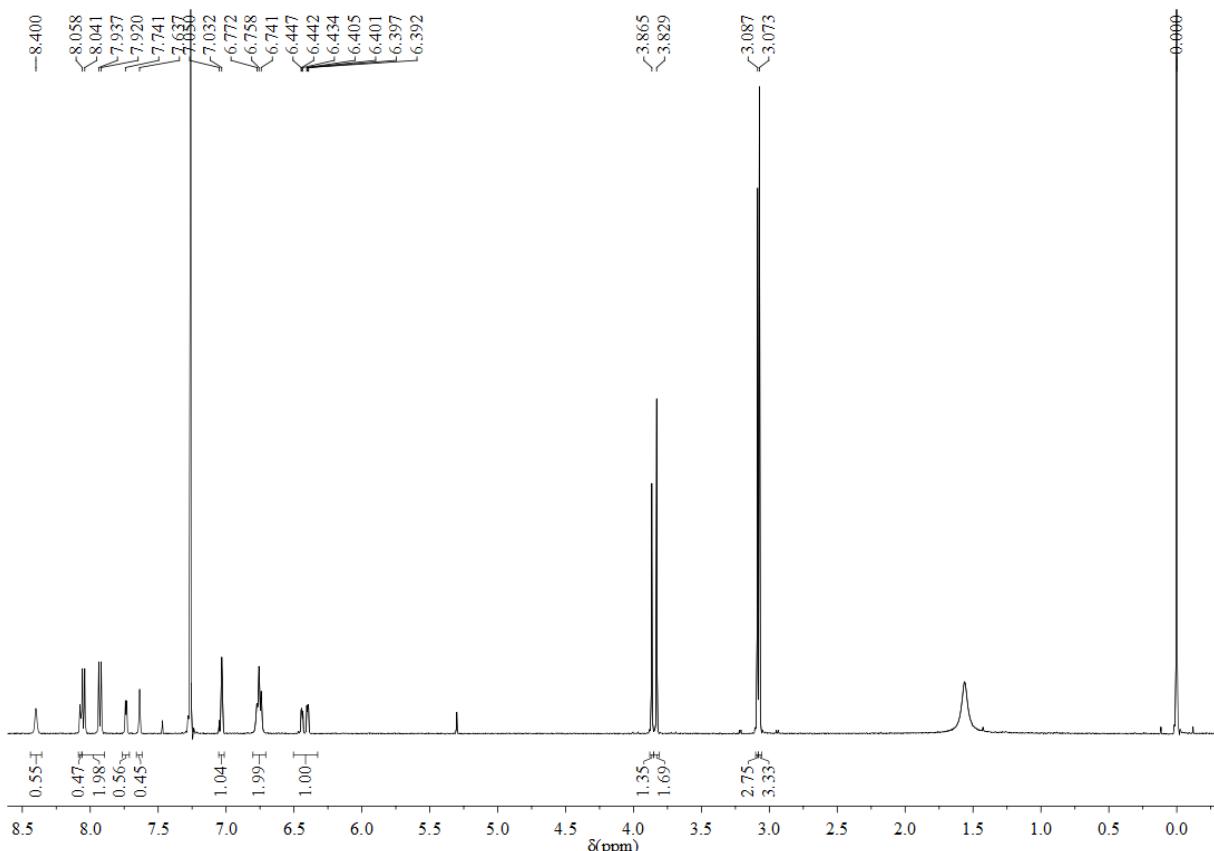


^1H NMR spectrum of **2b** which was irradiated by 18W white light LED for 180 min in CDCl_3 .

6. NMR spectra for prepared isomers mixtures

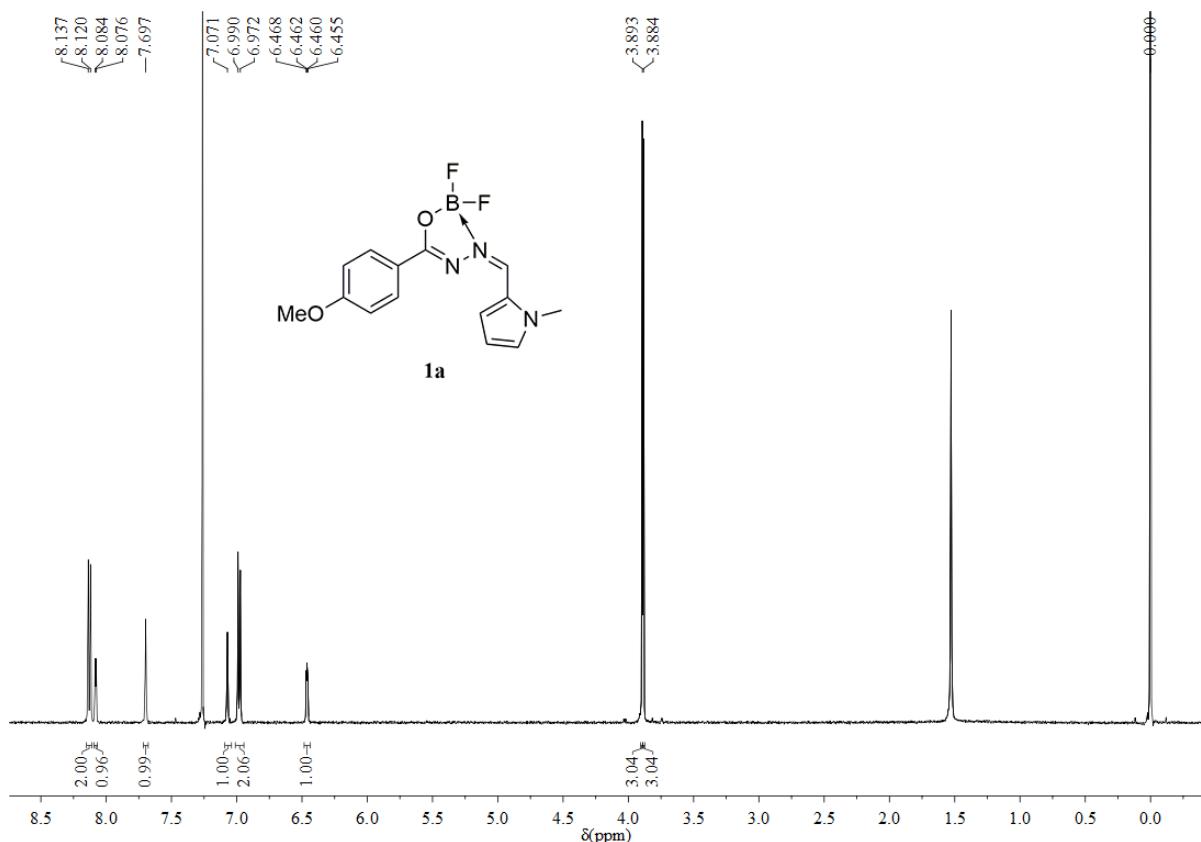


¹H NMR spectrum of the as-prepared crude MPOAB 1 in CDCl₃. The sample was obtained by column chromatography after the end of the reaction. Both isomers **1a** and **1b** were not separated/purified to demonstrate the ratio of **1a** and **1b** obtained under the reaction conditions.

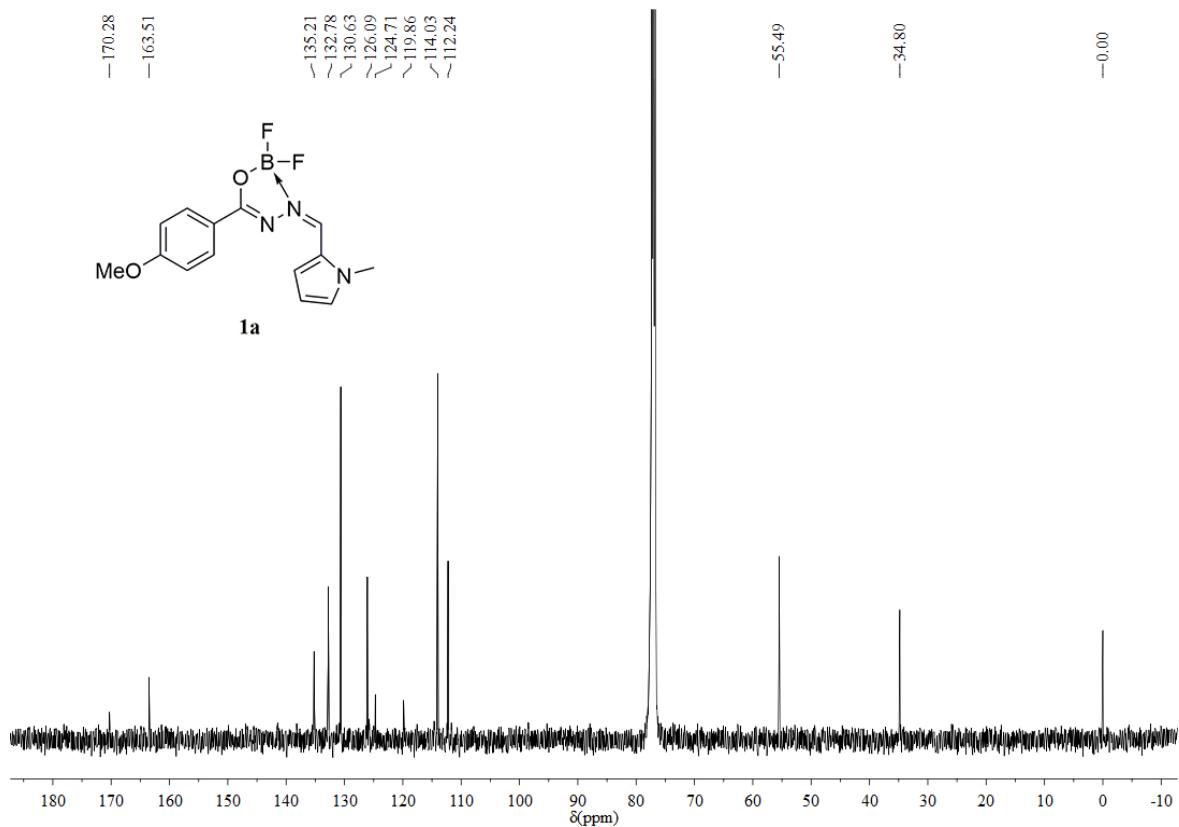


¹H NMR spectrum of the as-prepared crude MPOAB **2** in CDCl₃. The sample was obtained by column chromatography after the end of the reaction. Both isomers **2a** and **2b** were not separated/purified to demonstrate the ratio of **2a** and **2b** obtained under the reaction conditions.

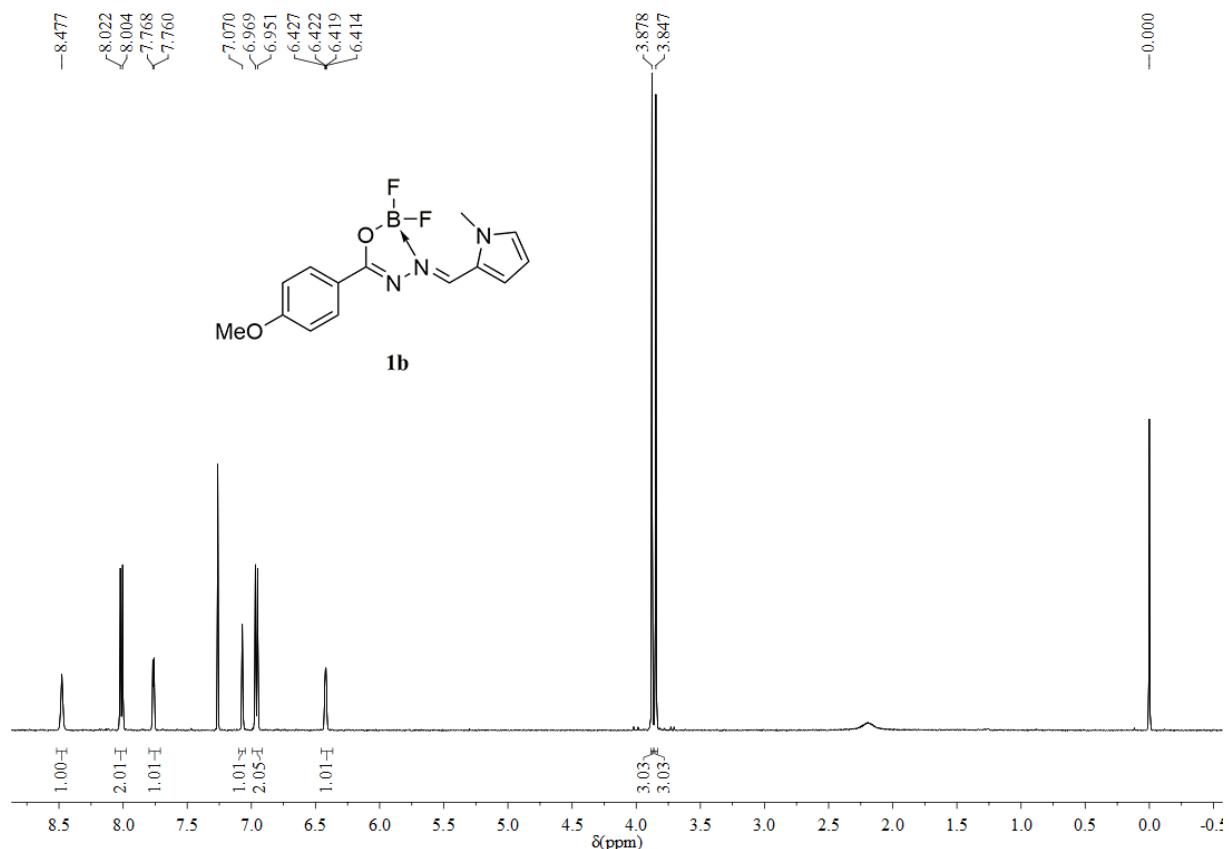
7. NMR spectra of pure isomers



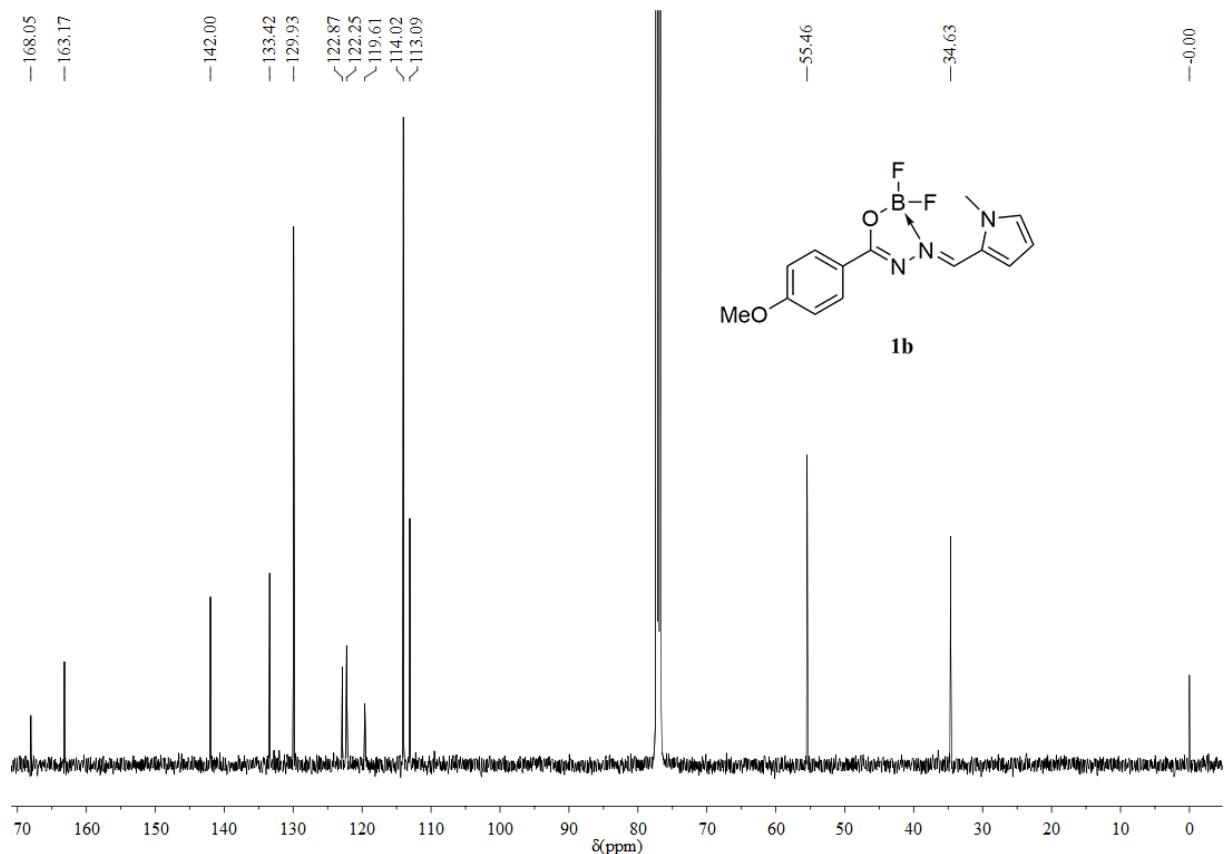
¹H NMR spectrum of (Z)-MPOAB **1a** in CDCl₃.



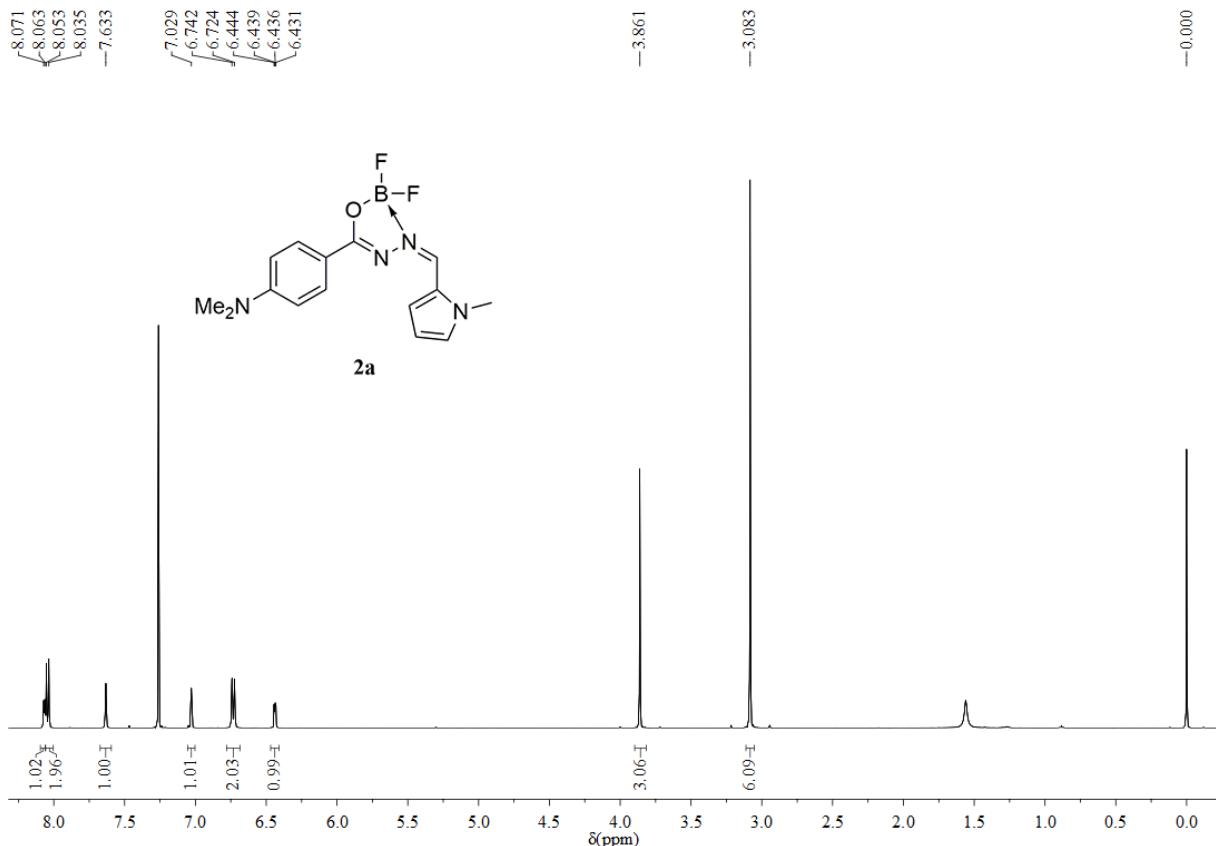
¹³C NMR spectrum of (Z)-MPOAB **1a** in CDCl₃.



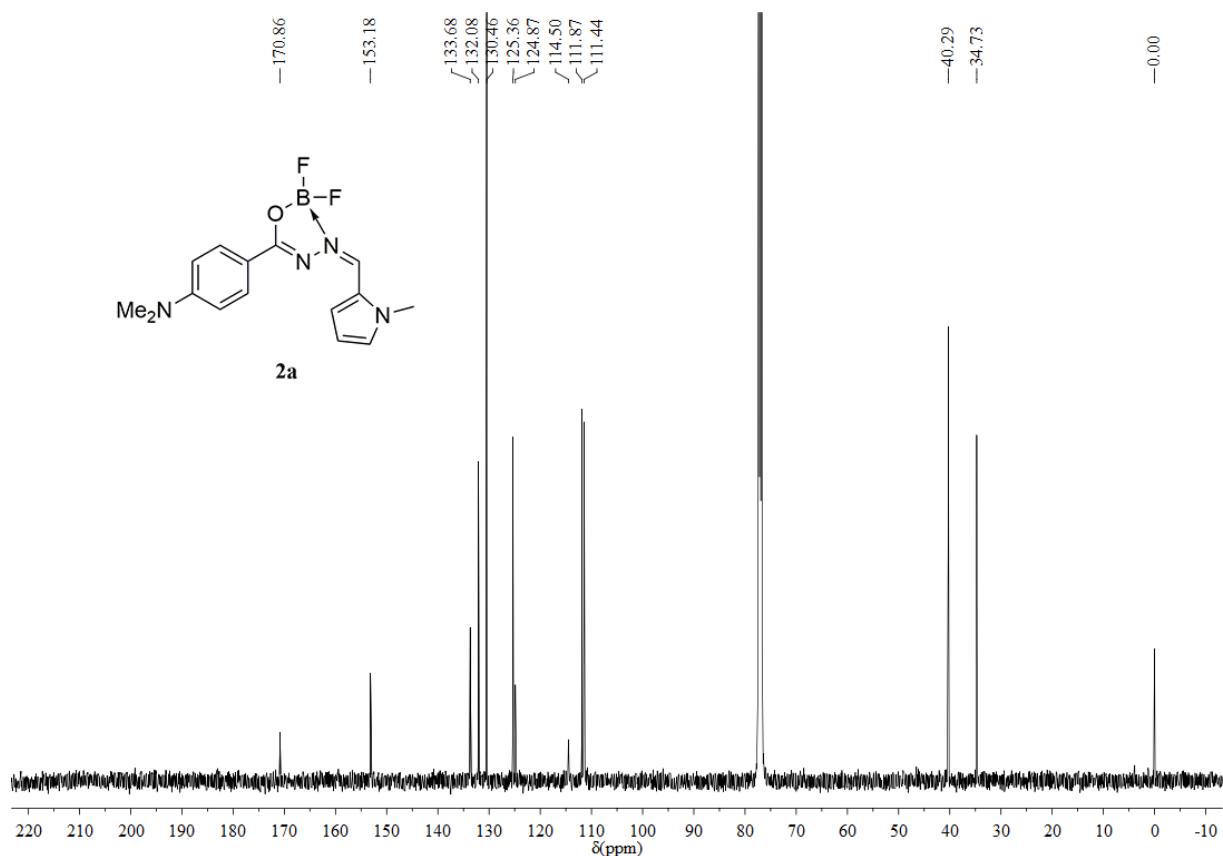
¹H NMR spectrum of (*E*)-MPOAB **1b** in CDCl₃.



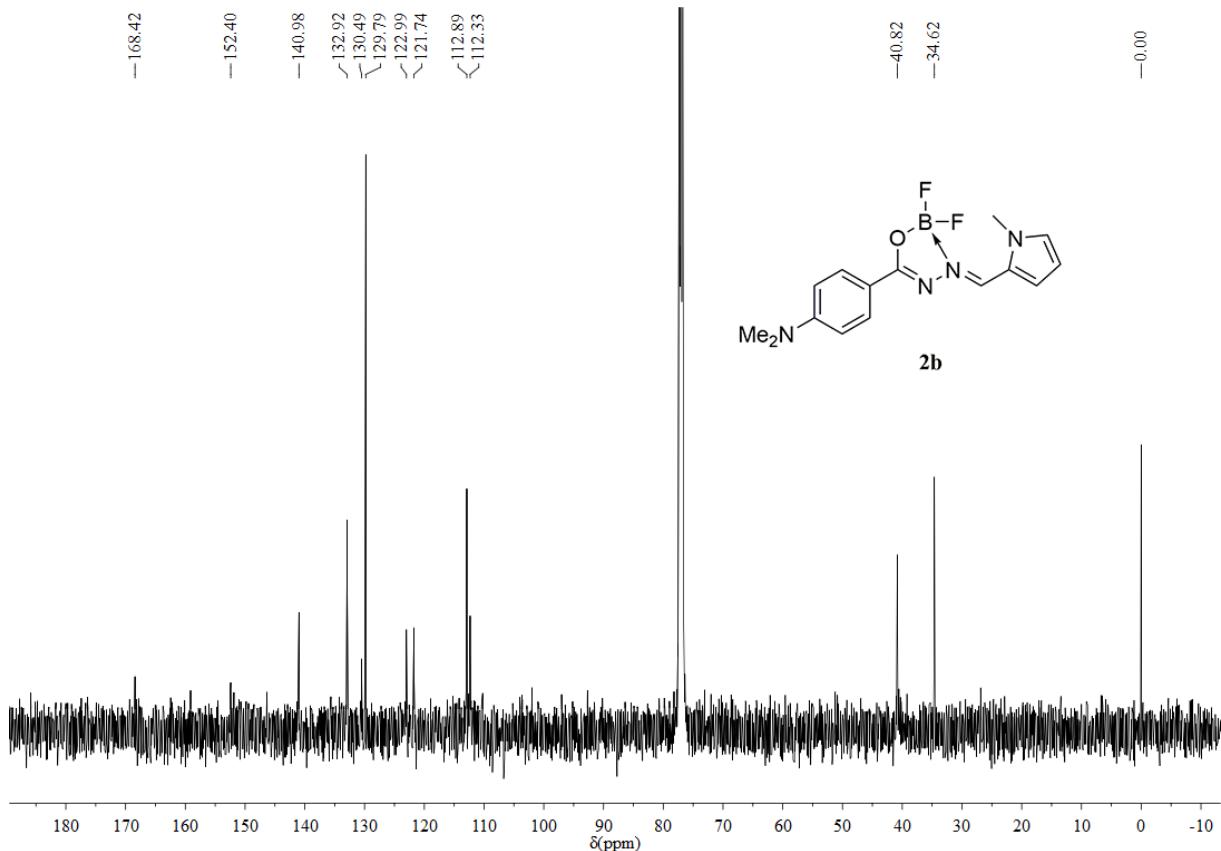
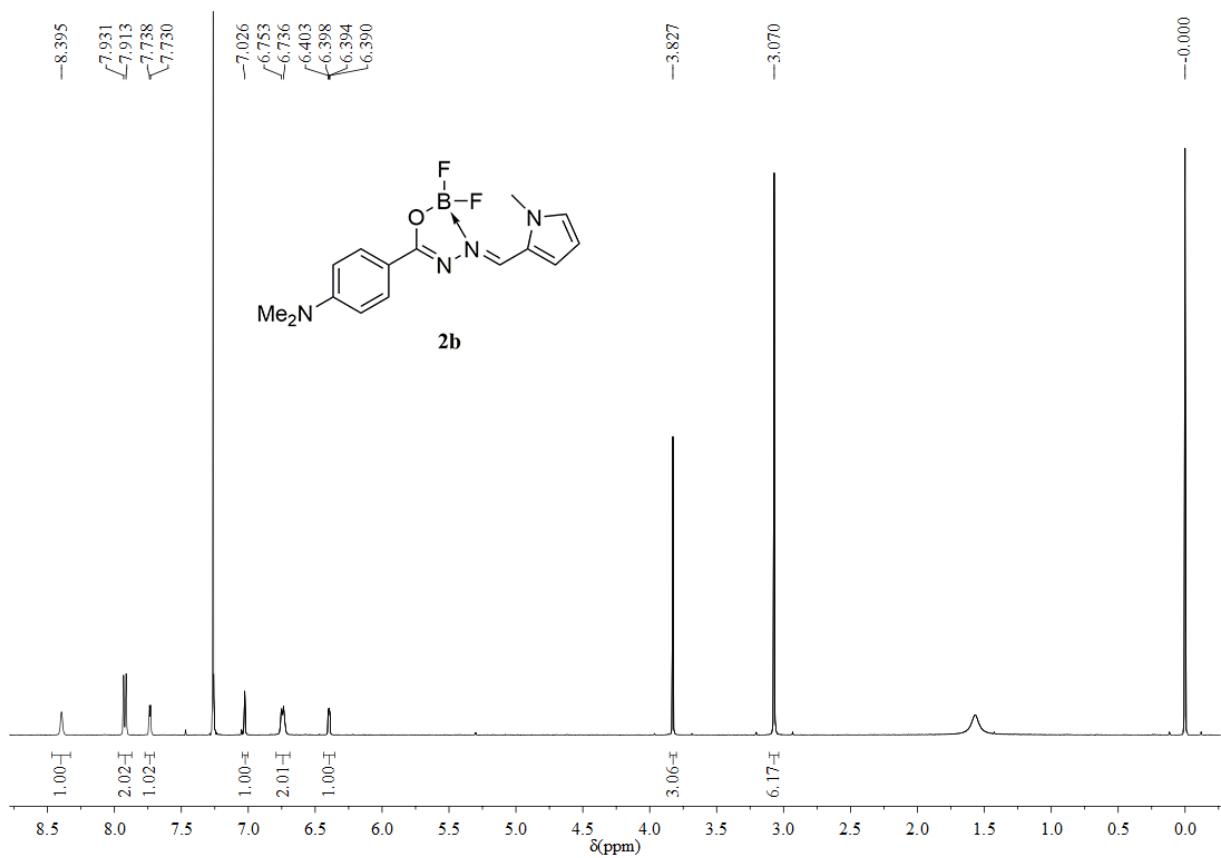
¹³C NMR spectrum of (*E*)-MPOAB **1b** in CDCl₃.



¹H NMR spectrum of (*Z*)-MPOAB **2a** in CDCl₃.



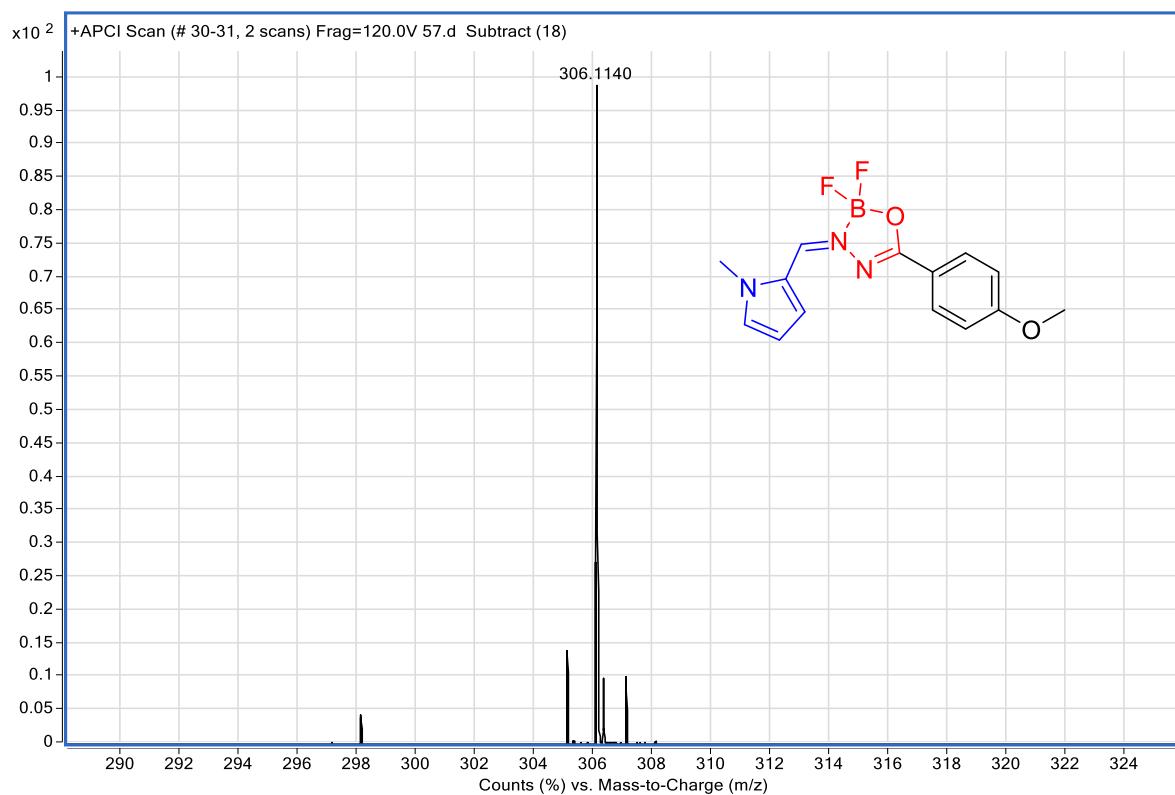
¹³C NMR spectrum of (*Z*)-MPOAB **2a** in CDCl₃.



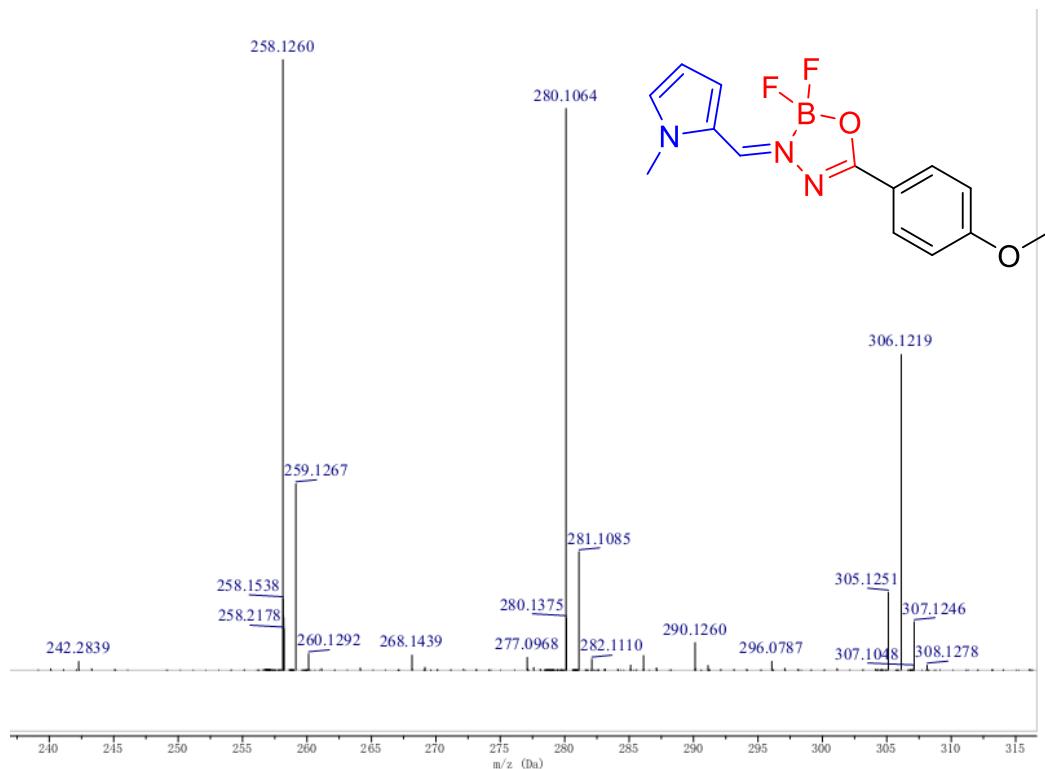
¹³C NMR spectrum of (E)-MPOAB **2b** in CDCl₃.

8. High-resolution mass spectrum spectra

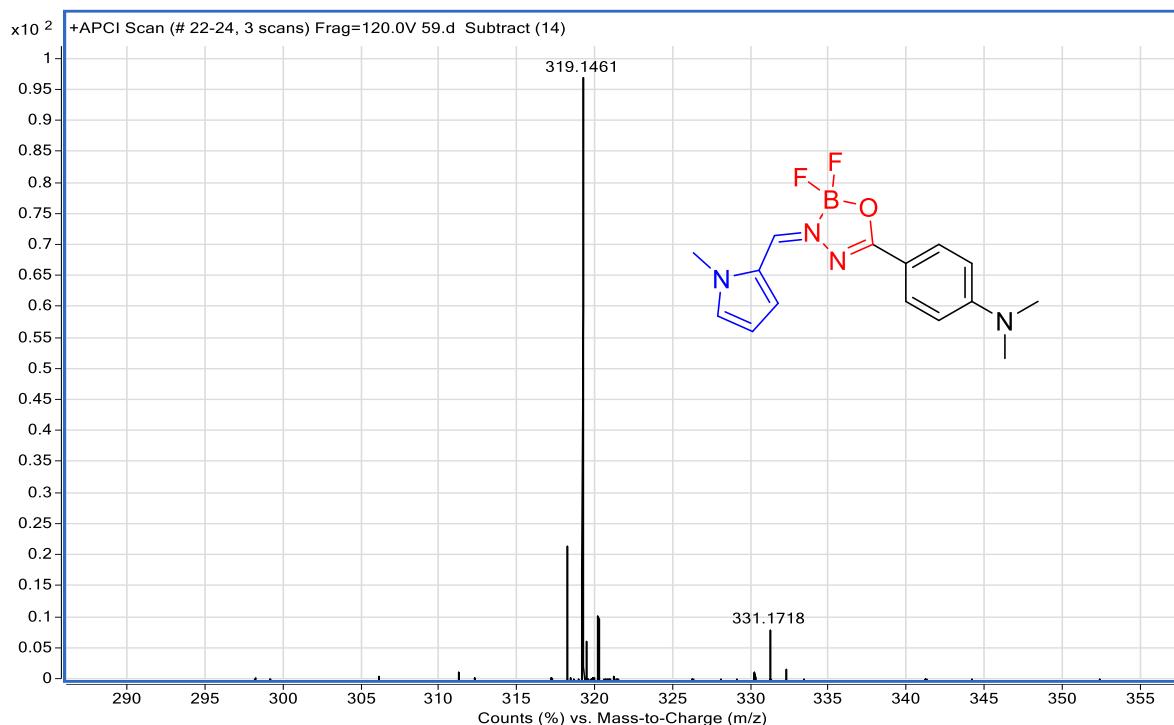
HRMS for 1a



HRMS for 1b



HRMS for 2a



HRMS for 2b

