Supporting Information

A panchromatic electrochromic device composed of Ru(II)/Fe(II)-based heterometallosupramolecular polymer

Li-Yin Hsiao,^{a, d} Ting-Hsiang Chang,^a Hsin-Che Lu,^a Yen-Chun Wang,^a Yen-An Lu,^a Kuo-

Chuan Ho,^{a, b, c}* and Masayoshi Higuchi^d*

^a Department of Chemical Engineering, National Taiwan University, No. 1, Sec. 4, Roosevelt Road, Taipei 10617, Taiwan

^b Institute of Polymer Science and Engineering, National Taiwan University, Taipei 10617, Taiwan

^c Advanced Research Center for Green Materials Science and Technology, National Taiwan University, Taipei 10617, Taiwan

^d Electronic Functional Macromolecules Group, Research Center for Functional Materials, National Institute for Materials Science (NIMS), Tsukuba 305-0044, Japan

*Corresponding Author

E-mail: kcho@ntu.edu.tw (Dr. K. C. Ho)

E-mail: HIGUCHI.Masayoshi@nims.go.jp (Dr. M. Higuchi)



Figure S1. The XPS spectra of the PolyRuFe for (a) Fe 2p and (b) Ru 3p.



Figure S2. The EDS spectrum of the PolyRuFe.

Element	Weight (%)	Atomic (%)
СК	67.10	78.93
N K	13.20	13.32
O K	4.55	4.02
Cl K	4.35	1.73
Fe L	4.32	1.09
Ru L	6.48	0.91
Totals	100.00	

 Table S1. The atomic composition of PolyRuFe.



Figure S3. (a) The CV of water-dispersible PB thin film cycled in 0.1 M NaClO₄/acetone at 50 mV/s. (b) The UV-vis absorbance spectra of water-dispersible PB thin film collected in 0.1 M NaClO₄/acetone at different potential biases from -0.4 to 0.3 V (vs. Ag/Ag⁺).



Figure S4. Enlarged view of dynamic transmittance response of the PolyRuFe/PB ECD at 503, 580 and 690 nm.



Figure S5. Colorimetry (x-y diagram) of the PolyRuFe/PB ECD at the bleached state (2.2 V) and colored state (-1.3 V).



Figure S6. SEM images of (a) PolyRuFe and (b) PolyRuFe-MWCNT thin films.



Figure S7. CVs and accumulated mass changes of (a) PolyRuFe and (b) PolyRuFe-MWCNT thin films coated on the gold-disk electrode. The data were obtained in 0.1 M NaClO₄/acetone at a scan rate of 20 mV/s.