Lattice Restraint Induced Ultra-large Bandgap Widening of

ZnO Nanoparticles

An Xie,^a Dandan Yang,^b Xiaoming Li,^{*b} Haibo Zeng^b

^{a.} School of Materials Science and Engineering, Key Laboratory of Functional Materials and Applications of Fujian Province, Xiamen University of Technology, Xiamen 361024, China, E-mail: anxie@xmut.edu.cn

^{b.} MIIT Key Laboratory of Advanced Display Materials and Devices, Institute of Optoelectronics & Nanomaterials, College of Materials Science and Engineering, Nanjing University of Science and Technology, Nanjing 210094, China. E-mail: lixiaoming@njust.edu.cn



Fig. S1. XRD patterns of ZnMgO nanoparticle alloys prepared with different precursors and nominal Mg concentrations. (a) $Zn(St)_2+Mg(St)_2$, (b) $Zn(St)_2+Mg(acac)_2$, (c) $Zn(acac)_2+Mg(acac)_2$, (d) $Zn(acac)_2+Mg(St)_2$.



Fig. S2. Absorption spectra of ZnMgO nanoparticle alloys prepared with different precursors and nominal Mg concentrations. (a) $Zn(St)_2+Mg(St)_2$, (b) $Zn(St)_2+Mg(acac)_2$, (c) $Zn(acac)_2+Mg(acac)_2$, (d) $Zn(acac)_2+Mg(St)_2$.



Fig. S3. XRD pattern of Be-0.2Mg-MB70 sample.



Fig. S4. Normalized absorption curves of samples with different Be concentration at a 60% nominal Mg concentration.



Fig. S5. PL spectra of samples with different Mg nominal concentrations.



Fig. S6. Calculated Eg values and emission peak energies as a function of Mg nominal concentration. The Be to Mg ratio is 0.4.