

SUPPLEMENTARY INFORMATION

Lamellar columnar liquid-crystalline mesophases as a 2D platform for anhydrous proton conduction

Cristián Cuerva,^{*a,b} José A. Campo,^a Mercedes Cano^{*a} and Rainer Schmidt^c

^a Department of Inorganic Chemistry, Faculty of Chemical Sciences, Complutense University of Madrid, 28040, Madrid, Spain; Tel: 34 91394 4340

^b BIOSCOPE Research Group, LAQV@REQUIMTE, Chemistry Department, Faculty of Science and Technology, NOVA University Lisbon, 2829-516 Caparica, Portugal.

^c GFMC, Department of Materials Physics, Faculty of Physics, Complutense University of Madrid, Ciudad Universitaria, 28040 Madrid, Spain.

* Correspondence: Mercedes Cano (mmcana@quim.ucm.es)
Cristián Cuerva (c.alaiz@fct.unl.pt)

Structural characterisation of Pd(II) compounds

Compounds [PdBr₂(Hpz^{R(n,n)py})] 9-13

[PdBr₂(Hpz^{R(6,6)py})] (**9**): orange solid (44%). Found: C, 44.8; H, 5.0; N, 6.2. PdC₂₆H₃₅N₃O₂Br₂·0.1CH₂Cl₂ requires C, 45.0; H, 5.1; N, 6.0%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3263m $\nu(\text{N}-\text{H})$, 2926 – 2857s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1597s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 778m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.92 (6H, t, ³J 6.9, CH₃), 1.35 (12H, m, CH₂), 1.80 (4H, qt, ³J 6.6, CH₂), 3.98 (4H, t, ³J 6.5, OCH₂), 5.30 (s, CH₂Cl₂), 6.53 (1H, t, ⁴J 2.1, Hp), 6.63 (2H, d, ⁴J 2.1, Ho), 7.06 (1H, d, ⁴J 2.0, H4'), 7.39 (1H, ddd, ³J 7.3, 5.8, ⁴J 1.3, H5), 7.92 (1H, d, ³J 7.3, H3), 8.09 (1H, ddd, ³J 7.7, 7.7, ⁴J 1.4, H4), 9.20 (1H, d, ³J 5.7, H6), 11.00 (1H, br, NH).

[PdBr₂(Hpz^{R(12,12)py})] (**10**): orange solid (58%). Found: C, 54.0; H, 6.8; N, 5.1. PdC₃₈H₅₉N₃O₂Br₂·0.3CH₃CN requires C, 53.4; H, 7.0; N, 5.3%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3260m $\nu(\text{N}-\text{H})$, 2921 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1597s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 779m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, ³J 6.9, CH₃), 1.27 (36H, m, CH₂), 1.80 (4H, qt, ³J 6.7, CH₂), 2.10 (s, CH₃CN), 3.98 (4H, t, ³J 6.5, OCH₂), 6.53 (1H, t, ⁴J 2.0, Hp), 6.63 (2H, d, ⁴J 2.1, Ho), 7.04 (1H, s, H4'), 7.41 (1H, ddd, ³J 7.3, 5.7, ⁴J 1.3, H5), 7.90 (1H, d, ³J 7.4, H3), 8.08 (1H, ddd, ³J 7.7, 7.7, ⁴J 1.4, H4), 9.25 (1H, d, ³J 5.7, H6), 11.00 (1H, s, NH). δ_{C} (75.48 MHz; CDCl₃; TMS): 14.1 (CH₃), 22.7 – 31.9 (CH₂), 68.5 (OCH₂), 102.4 (C4'), 103.4 (Cp), 104.0 (Co), 123.1 (C3), 125.2 (C5), 127.2 (Ci), 140.6 (C4), 145.5 (C3'), 150.1 (C2), 151.7 (C6), 152.1 (C5'), 161.0 (Cm).

[PdBr₂(Hpz^{R(14,14)py})] (**11**): orange solid (61%). Found: C, 55.2; H, 7.2; N, 4.8. PdC₄₂H₆₇N₃O₂Br₂ requires C, 55.3; H, 7.4; N, 4.6%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3290m $\nu(\text{N}-\text{H})$, 2920 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1598s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 779m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, ³J 6.9, CH₃), 1.26 (44H, m, CH₂), 1.80 (4H, qt, ³J 6.6, CH₂), 3.98 (4H, t, ³J 6.5, OCH₂), 6.53 (1H, t, ⁴J 2.0, Hp), 6.64 (2H, d, ⁴J 2.0, Ho), 7.04 (1H, s, H4'), 7.41 (1H, ddd, ³J 7.4, 5.7, ⁴J 1.1, H5), 7.89 (1H, d, ³J 7.5, H3), 8.09 (1H, ddd, ³J 7.7, 7.7, ⁴J 1.2, H4), 9.25 (1H, d, ³J 5.6, H6), 11.05 (1H, s, NH).

[PdBr₂(Hpz^{R(16,16)py})] (**12**): orange solid (64%). Found: C, 56.6; H, 7.5; N, 4.4. PdC₄₆H₇₅N₃O₂Br₂ requires C, 57.0; H, 7.8; N, 4.3%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3278w $\nu(\text{N}-\text{H})$, 2919 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1600m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 779w $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, ³J 6.9, CH₃), 1.26 (52H, m, CH₂), 1.80 (4H, qt, ³J 6.5, CH₂), 3.98 (4H, t, ³J 6.5, OCH₂), 6.54 (1H, t, ⁴J 2.0, Hp), 6.64 (2H, d, ⁴J 2.1, Ho), 7.02 (1H, d, ⁴J

1.8, H4’), 7.43 (1H, ddd, 3J 7.4, 5.7, 4J 1.1, H5), 7.87 (1H, d, 3J 7.5, H3), 8.08 (1H, ddd, 3J 7.7, 7.7, 4J 1.3, H4), 9.29 (1H, d, 3J 5.8, H6), 11.08 (1H, br, NH).

[PdBr₂(Hpz^{R(18,18)py})] (**13**): orange solid (64%). Found: C, 59.0; H, 7.8; N, 4.0. PdC₅₀H₈₃N₃O₂Br₂ requires C, 58.6; H, 8.2; N, 4.1%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3267w $\nu(\text{N}-\text{H})$, 2919 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1597m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 779w $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, 3J 6.9, CH₃), 1.25 (60H, m, CH₂), 1.80 (4H, qt, 3J 6.8, CH₂), 3.98 (4H, t, 3J 6.5, OCH₂), 6.54 (1H, t, 4J 2.0, Hp), 6.65 (2H, d, 4J 2.1, Ho), 7.01 (1H, d, 4J 1.7, H4’), 7.44 (1H, ddd, 3J 7.4, 5.7, 4J 1.3, H5), 7.86 (1H, d, 3J 7.6, H3), 8.06 (1H, ddd, 3J 7.7, 7.7, 4J 1.4, H4), 9.30 (1H, d, 3J 5.7, H6), 11.07 (1H, d, 4J 1.5, NH).

Compounds [PdI₂(Hpz^{R(n,n)py})] 14-18

[PdI₂(Hpz^{R(6,6)py})] (**14**): red solid (53%). Found: C, 39.7; H, 4.4; N, 5.4. PdC₂₆H₃₅N₃O₂I₂ requires C, 39.9; H, 4.5; N, 5.4%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3311m $\nu(\text{N}-\text{H})$, 2926 – 2855m $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1595s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 778m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.93 (6H, t, 3J 6.9, CH₃), 1.36 (12H, m, CH₂), 1.81 (4H, qt, 3J 6.8, CH₂), 3.99 (4H, t, 3J 6.5, OCH₂), 6.53 (1H, t, 4J 2.1, Hp), 6.64 (2H, d, 4J 2.1, Ho), 7.01 (1H, d, 4J 1.7, H4’), 7.38 (1H, ddd, 3J 7.4, 5.7, 4J 1.3, H5), 7.91 (1H, d, 3J 7.4, H3), 8.09 (1H, ddd, 3J 7.7, 7.7, 4J 1.4, H4), 9.45 (1H, d, 3J 5.6, H6), 11.00 (1H, d, 4J 1.4, NH).

[PdI₂(Hpz^{R(12,12)py})] (**15**): red solid (67%). Found: C, 48.0; H, 6.1; N, 4.5. PdC₃₈H₅₉N₃O₂I₂ requires C, 48.0; H, 6.3; N, 4.4%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3308m $\nu(\text{N}-\text{H})$, 2920 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1595m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 778m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, 3J 6.9, CH₃), 1.27 (36H, m, CH₂), 1.81 (4H, qt, 3J 6.6, CH₂), 3.99 (4H, t, 3J 6.5, OCH₂), 6.53 (1H, t, 4J 2.0, Hp), 6.64 (2H, d, 4J 1.9, Ho), 6.98 (1H, d, 4J 1.9, H4’), 7.41 (1H, ddd, 3J 7.5, 5.8, 4J 1.2, H5), 7.87 (1H, d, 3J 7.5, H3), 8.08 (1H, ddd, 3J 7.7, 7.7, 4J 1.4, H4), 9.51 (1H, d, 3J 5.8, H6), 11.03 (1H, d, 4J 1.4, NH). δ_{C} (75.48 MHz; CDCl₃; TMS): 14.1 (CH₃), 22.7 – 31.9 (CH₂), 68.5 (OCH₂), 102.6 (C4’), 103.1 (Cp), 104.0 (Co), 123.2 (C3), 125.5 (C5), 127.2 (Ci), 140.3 (C4), 145.2 (C3’), 150.1 (C2), 152.5 (C5’), 153.1 (C6), 161.0 (Cm).

[PdI₂(Hpz^{R(14,14)py})] (**16**): red solid (61%). Found: C, 50.2; H, 6.6; N, 4.3. PdC₄₂H₆₇N₃O₂I₂ requires C, 50.1; H, 6.7; N, 4.2%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3308m $\nu(\text{N}-\text{H})$, 2917 – 2849s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1595m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 778m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃;

TMS): 0.88 (6H, t, 3J 6.8, CH₃), 1.26 (44H, m, CH₂), 1.81 (4H, qt, 3J 6.6, CH₂), 3.99 (4H, t, 3J 6.4, OCH₂), 6.54 (1H, t, 4J 2.0, Hp), 6.65 (2H, d, 4J 1.9, Ho), 6.98 (1H, d, 4J 1.7, H4'), 7.41 (1H, ddd, 3J 7.7, 5.5, 4J 1.1, H5), 7.87 (1H, d, 3J 7.6, H3), 8.08 (1H, ddd, 3J 7.7, 7.7, 4J 1.2, H4), 9.52 (1H, d, 3J 5.4, H6), 11.04 (1H, br, NH).

[PdI₂(Hpz^{R(16,16)py]) (17)}:

red solid (72%). Found: C, 52.5; H, 7.0; N, 4.0. PdC₄₆H₇₅N₃O₂I₂ requires C, 52.0; H, 7.1; N, 4.0%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3309m $\nu(\text{N}-\text{H})$, 2919 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1597m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 779w $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, 3J 6.9, CH₃), 1.26 (52H, m, CH₂), 1.81 (4H, qt, 3J 6.8, CH₂), 3.99 (4H, t, 3J 6.5, OCH₂), 6.55 (1H, t, 4J 2.0, Hp), 6.66 (2H, d, 4J 2.1, Ho), 6.96 (1H, d, 4J 2.0, H4'), 7.43 (1H, ddd, 3J 7.4, 5.7, 4J 1.2, H5), 7.85 (1H, d, 3J 7.5, H3), 8.07 (1H, ddd, 3J 7.7, 7.7, 4J 1.3, H4), 9.57 (1H, d, 3J 5.8, H6), 11.10 (1H, br, NH).

[PdI₂(Hpz^{R(18,18)py]) (18)}:

red solid (72%). Found: C, 53.7; H, 7.3; N, 3.9. PdC₅₀H₈₃N₃O₂I₂ requires C, 53.7; H, 7.5; N, 3.8%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3306m $\nu(\text{N}-\text{H})$, 2917 – 2849s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1595m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 778m $\gamma(\text{C}-\text{H})_{\text{py}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, 3J 6.9, CH₃), 1.26 (60H, m, CH₂), 1.81 (4H, qt, 3J 6.8, CH₂), 3.99 (4H, t, 3J 6.5, OCH₂), 6.55 (1H, t, 4J 2.2, Hp), 6.66 (2H, d, 4J 2.0, Ho), 6.96 (1H, d, 4J 2.0, H4'), 7.43 (1H, ddd, 3J 7.4, 5.7, 4J 1.4, H5), 7.84 (1H, d, 3J 7.8, H3), 8.08 (1H, dd, 3J 7.7, 7.7, 4J 1.4, H4), 9.58 (1H, d, 3J 5.9, H6), 11.09 (1H, br, NH).

Compounds [PdCl₂(Hpz^{R(n,n)iq})] 19-26

[PdCl₂(Hpz^{R(4,4)iq]) (19)}:

pale orange solid (60%). Found: C, 52.0; H, 4.8; N, 7.0. PdC₂₆H₂₉N₃O₂Cl₂·0.1CH₂Cl₂ requires C, 52.1; H, 4.9; N, 7.0%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3296w $\nu(\text{N}-\text{H})$, 2919 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1636 – 1595s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 1.02 (6H, t, 3J 7.3, CH₃), 1.53 (4H, m, CH₂), 1.80 (4H, qt, 3J 6.7, CH₂), 4.00 (4H, t, 3J 6.5, OCH₂), 5.30 (s, CH₂Cl₂), 6.53 (1H, t, 4J 2.0, Hp), 6.60 (2H, d, 4J 1.9, Ho), 7.24 (1H, s, H4'), 7.69 (2H, m, H7, H8), 7.95 (1H, pt, 3J 8.1, H6), 8.07 (1H, d, 3J 8.3, H5), 8.35 (1H, s, H4), 9.33 (1H, s, H1), 10.73 (1H, s, NH).

[PdCl₂(Hpz^{R(6,6)iq]) (20)}:

pale orange solid (70%). Found: C, 56.1; H, 5.7; N, 6.7. PdC₃₀H₃₇N₃O₂Cl₂ requires C, 55.5; H, 5.8; N, 6.5%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3274w $\nu(\text{N}-\text{H})$, 2915 – 2849s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1636 – 1596s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 716m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.94 (6H, t, 3J 7.0, CH₃), 1.37 (12H, m, CH₂), 1.81 (4H, qt, 3J 6.7,

CH_2), 3.98 (4H, t, 3J 6.5, OCH₂), 6.50 (1H, t, 4J 2.0, Hp), 6.58 (2H, d, 4J 2.1, Ho), 7.26 (1H, m, H4'), 7.64 (2H, m, H7, H8), 7.93 (1H, ddd, 3J 8.1, 6.3, 4J 1.7, H6), 8.07 (1H, d, 3J 8.2, H5), 8.37 (1H, s, H4), 9.25 (1H, s, H1), 10.74 (1H, s, NH).

[PdCl₂(Hpz^{R(8,8)iq})] (**21**): pale orange solid (60%). Found: C, 57.8; H, 6.3; N, 6.1. PdC₃₄H₄₅N₃O₂Cl₂ requires C, 57.9; H, 6.4; N, 6.0%. $\nu_{\max}/\text{cm}^{-1}$: 3270w $\nu(\text{N}-\text{H})$, 2919 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1637 – 1596s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.90 (6H, t, 3J 6.9, CH₃), 1.31 (20H, m, CH₂), 1.81 (4H, qt, 3J 6.7, CH₂), 3.98 (4H, t, 3J 6.3, OCH₂), 6.51 (1H, br, Hp), 6.57 (2H, br, Ho), 7.29 (1H, br, H4'), 7.65 (2H, m, H7, H8), 7.93 (1H, pt, 3J 7.2, H6), 8.09 (1H, d, 3J 7.8, H5), 8.40 (1H, s, H4), 9.20 (1H, s, H1), 10.59 (1H, br, NH).

[PdCl₂(Hpz^{R(10,10)iq})] (**22**): pale orange solid (63%). Found: C, 60.1; H, 6.9; N, 5.6. PdC₃₈H₅₃N₃O₂Cl₂ requires C, 60.0; H, 7.0; N, 5.5%. $\nu_{\max}/\text{cm}^{-1}$: 3272w $\nu(\text{N}-\text{H})$, 2921 – 2852s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1636 – 1596s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.89 (6H, t, 3J 6.6, CH₃), 1.29 (28H, m, CH₂), 1.81 (4H, qt, 3J 6.8, CH₂), 3.98 (4H, t, 3J 6.5, OCH₂), 6.51 (1H, t, 4J 1.9, Hp), 6.57 (2H, d, 4J 1.9, Ho), 7.29 (1H, d, 4J 1.5, H4'), 7.64 (2H, m, H7, H8), 7.94 (1H, ddd, 3J 7.9, 6.4, 4J 1.0, H6), 8.09 (1H, d, 3J 8.2, H5), 8.40 (1H, s, H4), 9.22 (1H, s, H1), 10.55 (1H, br, NH).

[PdCl₂(Hpz^{R(12,12)iq})] (**23**): pale orange solid (67%). Found: C, 62.1; H, 7.3; N, 5.1. PdC₄₂H₆₁N₃O₂Cl₂ requires C, 61.7; H, 7.5; N, 5.1%. $\nu_{\max}/\text{cm}^{-1}$: 3272w $\nu(\text{N}-\text{H})$, 2921 – 2852s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1637 – 1595s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, 3J 6.9, CH₃), 1.27 (36H, m, CH₂), 1.81 (4H, qt, 3J 6.8, CH₂), 3.98 (4H, t, 3J 6.3, OCH₂), 6.50 (1H, t, 4J 1.9, Hp), 6.56 (2H, d, 4J 2.0, Ho), 7.29 (1H, d, 4J 1.6, H4'), 7.65 (2H, m, H7, H8), 7.94 (1H, pt, 3J 7.1, H6), 8.09 (1H, d, 3J 7.9, H5), 8.40 (1H, s, H4), 9.19 (1H, s, H1), 10.58 (1H, br, NH). δ_{C} (75.48 MHz; CDCl₃; TMS): 14.1 (CH₃), 22.6 – 31.9 (CH₂), 68.5 (OCH₂), 101.4 (C4'), 103.2 (Co), 103.9 (Cp), 120.8 (C4), 126.6 (Ci), 126.9 (C10), 128.0 (C5), 128.7 (C8), 129.7 (C7), 134.0 (C6), 135.2 (C9), 141.9 (C3), 145.0 (C3'), 151.7 (C5'), 154.1 (C1), 160.9 (Cm).

[PdCl₂(Hpz^{R(14,14)iq})] (**24**): pale orange solid (45%). Found: C, 63.8; H, 7.8; N, 4.6. PdC₄₆H₆₉N₃O₂Cl₂ requires C, 63.3; H, 8.0; N, 4.8%. $\nu_{\max}/\text{cm}^{-1}$: 3271w $\nu(\text{N}-\text{H})$, 2921 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1637 – 1596s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 718m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, 3J 6.8, CH₃), 1.26 (44H, m, CH₂), 1.81 (4H, qt, 3J 6.9, CH₂), 3.99 (4H, t, 3J 6.4, OCH₂), 6.53 (1H, t, 4J 1.7, Hp), 6.63 (2H, d, 4J 1.6, Ho), 7.20

(1H, d, 4J 1.3, H4'), 7.71 (1H, pt, 3J 7.2, H7), 7.79 (1H, d, 3J 7.5, H8), 7.95 (1H, pt, 3J 7.4, H6), 8.06 (1H, d, 3J 8.1, H5), 8.31 (1H, s, H4), 9.43 (1H, s, H1), 10.88 (1H, br, NH).

[PdCl₂(Hpz^{R(16,16)iq})] (**25**): pale orange solid (53%). Found: C, 65.1; H, 8.1; N, 4.4. PdC₅₀H₇₇N₃O₂Cl₂ requires C, 64.6; H, 8.3; N, 4.5%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3262w $\nu(\text{N}-\text{H})$, 2920 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1637 – 1596s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, 3J 6.9, CH₃), 1.26 (52H, m, CH₂), 1.81 (4H, qt, 3J 6.7, CH₂), 3.98 (4H, t, 3J 6.5, OCH₂), 6.53 (1H, t, 4J 1.9, Hp), 6.63 (2H, d, 4J 2.0, Ho), 7.18 (1H, d, 4J 1.3, H4'), 7.71 (1H, pt, 3J 7.1, H7), 7.81 (1H, d, 3J 8.3, H8), 7.95 (1H, pt, 3J 7.5, H6), 8.05 (1H, d, 3J 8.3, H5), 8.29 (1H, s, H4), 9.45 (1H, s, H1), 10.95 (1H, br, NH).

[PdCl₂(Hpz^{R(18,18)iq})] (**26**): pale orange solid (44%). Found: C, 65.4; H, 8.5; N, 3.9. PdC₅₄H₈₅N₃O₂Cl₂·0.1CH₂Cl₂ requires C, 65.4; H, 8.6; N, 4.2%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3247w $\nu(\text{N}-\text{H})$, 2917 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1637 – 1597m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 752 – 719m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, 3J 6.9, CH₃), 1.25 (60H, m, CH₂), 1.81 (4H, qt, 3J 6.6, CH₂), 3.98 (4H, t, 3J 6.4, OCH₂), 5.30 (s, CH₂Cl₂), 6.51 (1H, t, 4J 2.2, Hp), 6.59 (2H, d, 4J 2.2, Ho), 7.26 (1H, m, H4'), 7.65 (2H, m, H7, H8), 7.94 (1H, ddd, 3J 7.9, 5.9, 4J 2.2, H6), 8.08 (1H, d, 3J 8.2, H5), 8.37 (1H, s, H4), 9.27 (1H, s, H1), 10.73 (1H, s, NH).

Compounds [PdBr₂(Hpz^{R(n,n)iq})] 27-30

[PdBr₂(Hpz^{R(12,12)iq})] (**27**): orange solid (62%). Found: C, 55.7; H, 6.6; N, 4.7. PdC₄₂H₆₁N₃O₂Br₂ requires C, 55.7; H, 6.8; N, 4.6%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3325w $\nu(\text{N}-\text{H})$, 2917 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1635 – 1595m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 754 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, 3J 7.0, CH₃), 1.27 (36H, m, CH₂), 1.82 (4H, qt, 3J 6.6, CH₂), 3.99 (4H, t, 3J 6.5, OCH₂), 6.52 (1H, t, 4J 2.0, Hp), 6.57 (2H, d, 4J 2.0, Ho), 7.23 (1H, d, 4J 1.9, H4'), 7.65 (2H, m, H7, H8), 7.94 (1H, ddd, 3J 8.1, 5.7, 4J 1.4, H6), 8.09 (1H, d, 3J 8.2, H5), 8.38 (1H, s, H4), 9.46 (1H, s, H1), 10.64 (1H, d, 4J 1.6, NH). δ_{C} (75.48 MHz; CDCl₃; TMS): 14.1 (CH₃), 22.7 – 31.9 (CH₂), 68.5 (OCH₂), 101.4 (C4'), 103.3 (Co), 103.7 (Cp), 120.8 (C4), 126.9 (Ci), 127.0 (C10), 128.0 (C5), 128.7 (C8), 129.6 (C7), 133.9 (C6), 135.1 (C9), 141.9 (C3), 144.8 (C3'), 151.9 (C5'), 155.2 (C1), 160.8 (Cm).

[PdBr₂(Hpz^{R(14,14)iq})] (**28**): orange solid (58%). Found: C, 57.7; H, 7.1; N, 4.5. PdC₄₆H₆₉N₃O₂Br₂ requires C, 57.4; H, 7.2; N, 4.4%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3291w $\nu(\text{N}-\text{H})$, 2919 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1635 – 1596m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 755 – 716m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, ³J 6.6, CH₃), 1.27 (44H, m, CH₂), 1.82 (4H, qt, ³J 6.7, CH₂), 4.00 (4H, t, ³J 6.5, OCH₂), 6.53 (1H, t, ⁴J 1.9, Hp), 6.59 (2H, d, ⁴J 2.0, Ho), 7.20 (1H, d, ⁴J 1.8, H4'), 7.71 (2H, m, H7, H8), 7.95 (1H, ddd, ³J 8.0, 5.8, ⁴J 1.6, H6), 8.08 (1H, d, ³J 8.1, H5), 8.35 (1H, s, H4), 9.54 (1H, s, H1), 10.72 (1H, br, NH).

[PdBr₂(Hpz^{R(16,16)iq})] (**29**): orange solid (68%). Found: C, 59.5; H, 7.5; N, 4.1. PdC₅₀H₇₇N₃O₂Br₂ requires C, 59.0; H, 7.6; N, 4.1%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3291w $\nu(\text{N}-\text{H})$, 2919 – 2851s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1635 – 1596m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 755 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, ³J 6.8, CH₃), 1.26 (52H, m, CH₂), 1.82 (4H, qt, ³J 6.9, CH₂), 4.00 (4H, t, ³J 6.5, OCH₂), 6.52 (1H, t, ⁴J 1.9, Hp), 6.59 (2H, d, ⁴J 1.9, Ho), 7.21 (1H, d, ⁴J 1.8, H4'), 7.69 (2H, m, H7, H8), 7.95 (1H, ddd, ³J 8.1, 6.0, ⁴J 2.0, H6), 8.09 (1H, d, ³J 8.2, H5), 8.36 (1H, s, H4), 9.52 (1H, s, H1), 10.70 (1H, br, NH).

[PdBr₂(Hpz^{R(18,18)iq})] (**30**): orange solid (61%). Found: C, 60.0; H, 7.6; N, 3.9. PdC₅₄H₈₅N₃O₂Br₂ requires C, 60.4; H, 7.0; N, 3.9%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3276w $\nu(\text{N}-\text{H})$, 2918 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1635 – 1596m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 755 – 716m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, ³J 6.9, CH₃), 1.25 (60H, m, CH₂), 1.81 (4H, qt, ³J 6.8, CH₂), 3.99 (4H, t, ³J 6.6, OCH₂), 6.52 (1H, t, ⁴J 1.9, Hp), 6.59 (2H, d, ⁴J 1.9, Ho), 7.20 (1H, d, ⁴J 1.8, H4'), 7.72 (2H, m, H7, H8), 7.95 (1H, ddd, ³J 7.9, 6.6, ⁴J 1.5, H6), 8.08 (1H, d, ³J 8.2, H5), 8.34 (1H, s, H4), 9.55 (1H, s, H1), 10.77 (1H, br, NH).

Compounds [PdI₂(Hpz^{R(n,n)iq})] 31-34

[PdI₂(Hpz^{R(12,12)iq})] (**31**): red solid (55%). Found: C, 50.7; H, 6.1; N, 4.3. PdC₄₂H₆₁N₃O₂I₂ requires C, 50.4; H, 6.2; N, 4.2%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3381w $\nu(\text{N}-\text{H})$, 2924 – 2855m $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1597s $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 770 – 725m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, ³J 6.9, CH₃), 1.27 (36H, m, CH₂), 1.82 (4H, qt, ³J 6.8, CH₂), 4.00 (4H, t, ³J 6.5, OCH₂), 6.52 (1H, t, ⁴J 1.9, Hp), 6.58 (2H, d, ⁴J 2.0, Ho), 7.11 (1H, d, ⁴J 1.8, H4'), 7.66 (1H, pt, ³J 7.3, H7), 7.76 (1H, d, ³J 8.0, H8), 7.93 (1H, ddd, ³J 7.8, 6.7, ⁴J 1.0, H6), 8.08 (1H, d, ³J 8.1, H5), 8.28 (1H, s, H4), 9.81 (1H, s, H1), 10.73 (1H, br, NH). δ_{C} (75.48 MHz; CDCl₃; TMS): 14.1 (CH₃), 22.7 – 31.9 (CH₂), 68.5 (OCH₂), 101.6

(C4'), 103.3 (Cp), 103.6 (Co), 120.6 (C4), 127.0 (Ci), 127.7 (C10), 128.2 (C5), 128.7 (C8), 129.5 (C7), 133.7 (C6), 135.0 (C9), 142.2 (C3), 144.5 (C3'), 152.4 (C5'), 156.8 (C1), 160.9 (Cm).

[PdI₂(Hpz^{R(14,14)iq})] (32): red solid (63%). Found: C, 52.8; H, 6.5; N, 4.1. PdC₄₆H₆₉N₃O₂I₂ requires C, 52.3; H, 6.6; N, 4.0%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3330w $\nu(\text{N}-\text{H})$, 2918 – 2849s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1632 – 1596m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 794 – 715m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.88 (6H, t, ³J 6.9, CH₃), 1.27 (44H, m, CH₂), 1.82 (4H, qt, ³J 6.7, CH₂), 4.00 (4H, t, ³J 6.5, OCH₂), 6.52 (1H, t, ⁴J 2.0, Hp), 6.58 (2H, d, ⁴J 2.0, Ho), 7.11 (1H, d, ⁴J 1.9, H4'), 7.65 (1H, pt, ³J 7.2, H7), 7.75 (1H, d, ³J 8.0, H8), 7.93 (1H, ddd, ³J 8.0, 6.7, ⁴J 1.0, H6), 8.08 (1H, d, ³J 8.1, H5), 8.28 (1H, s, H4), 9.81 (1H, s, H1), 10.72 (1H, d, ⁴J 1.6, NH).

[PdI₂(Hpz^{R(16,16)iq})] (33): red solid (72%). Found: C, 54.4; H, 7.0; N, 3.8. PdC₅₀H₇₇N₃O₂I₂ requires C, 54.0; H, 7.0; N, 3.8%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3325w $\nu(\text{N}-\text{H})$, 2916 – 2850s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1634 – 1596m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 794 – 715m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, ³J 6.9, CH₃), 1.26 (52H, m, CH₂), 1.82 (4H, qt, ³J 6.7, CH₂), 4.00 (4H, t, ³J 6.5, OCH₂), 6.53 (1H, t, ⁴J 1.9, Hp), 6.61 (2H, d, ⁴J 1.9, Ho), 7.09 (1H, d, ⁴J 1.7, H4'), 7.68 (1H, pt, ³J 7.4, H7), 7.83 (1H, d, ³J 8.2, H8), 7.93 (1H, pt, ³J 7.2, H6), 8.07 (1H, d, ³J 8.2, H5), 8.25 (1H, s, H4), 9.89 (1H, s, H1), 10.81 (1H, br, NH).

[PdI₂(Hpz^{R(18,18)iq})] (34): red solid (66%). Found: C, 54.9; H, 7.1; N, 4.2. PdC₅₄H₈₅N₃O₂I₂·0.4CH₃CN requires C, 55.5; H, 7.3; N, 4.0%. $\nu_{\text{max}}/\text{cm}^{-1}$: 3325w $\nu(\text{N}-\text{H})$, 2916 – 2849s $\nu(\text{C}-\text{H})_{\text{aliph}}$, 1636 – 1595m $\nu(\text{C}=\text{C} + \text{C}=\text{N})$, 796 – 717m $\gamma(\text{C}-\text{H})_{\text{iq}}$. δ_{H} (300.16 MHz; CDCl₃; TMS): 0.87 (6H, t, ³J 6.9, CH₃), 1.25 (60H, m, CH₂), 1.82 (4H, qt, ³J 6.6, CH₂), 2.10 (s, CH₃CN), 4.00 (4H, t, ³J 6.5, OCH₂), 6.52 (1H, t, ⁴J 2.2, Hp), 6.57 (2H, d, ⁴J 1.9, Ho), 7.12 (1H, d, ⁴J 1.7, H4'), 7.64 (1H, pt, ³J 7.0, H7), 7.72 (1H, d, ³J 8.0, H8), 7.93 (1H, pt, ³J 7.4, H6), 8.09 (1H, d, ³J 8.3, H5), 8.29 (1H, s, H4), 9.77 (1H, s, H1), 10.69 (1H, br, NH).

Table S1 Selected bond distances and angles for $[\text{PdBr}_2(\text{Hpz}^{\text{R}(6,6)\text{py}})] \cdot \text{CH}_3\text{CN}$, **9**· CH_3CN .

Bond distances /Å		Bond angles /°	
Pd–N1	1.998(5)	N1–Pd–N3	79.3(3)
Pd–N3	2.082(5)	N1–Pd–Br2	174.6(2)
Pd–Br1	2.408(1)	N1–Pd–Br1	92.5(2)
Pd–Br2	2.394(1)	N3–Pd–Br2	95.6(2)
		N3–Pd–Br1	171.8(2)
		Br2–Pd–Br1	92.5(1)

Table S2 Selected bond distances and angles for $[\text{PdCl}_2(\text{Hpz}^{\text{R}(4,4)\text{iq}})]$ **19**

Bond distances /Å		Bond angles /°	
Pd–N1	1.999(6)	N1–Pd–N3	79.7(2)
Pd–N3	2.041(5)	N1–Pd–Cl2	171.9(3)
Pd–Cl1	2.268(2)	N1–Pd–Cl1	93.2(3)
Pd–Cl2	2.294(3)	N3–Pd–Cl2	95.0(3)
		N3–Pd–Cl1	172.7(3)
		Cl1–Pd–Cl2	92.3(2)

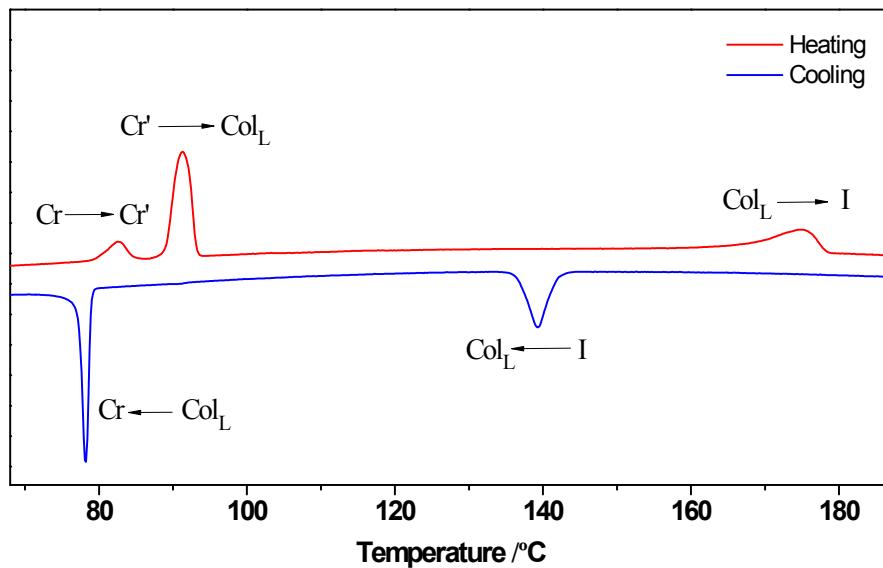


Figure S1 DSC thermogram for $[\text{PdBr}_2(\text{Hpz}^{\text{R}(12,12)\text{py}})]$ **10** after the first heating-cooling cycle.

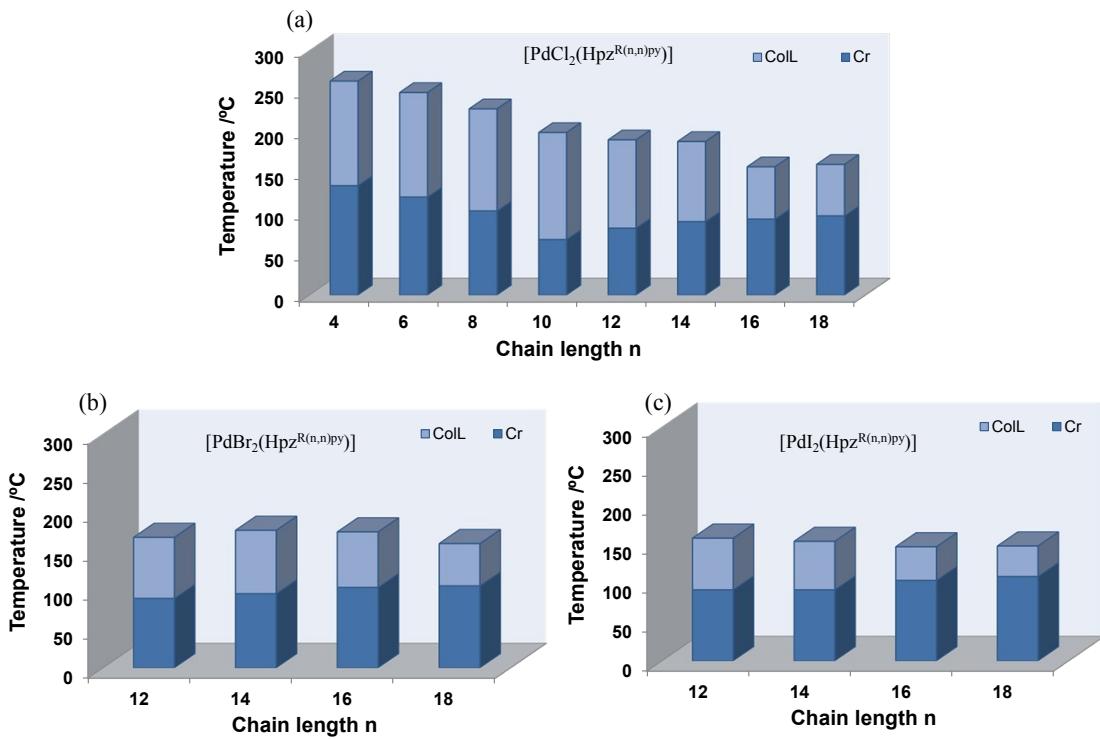


Figure S2 Bar diagram showing the range of existence of solid and liquid crystal phases for the dihalide pyridylpyrazole Pd(II) compounds. Data for the chloride compounds are taken from ref. 1.

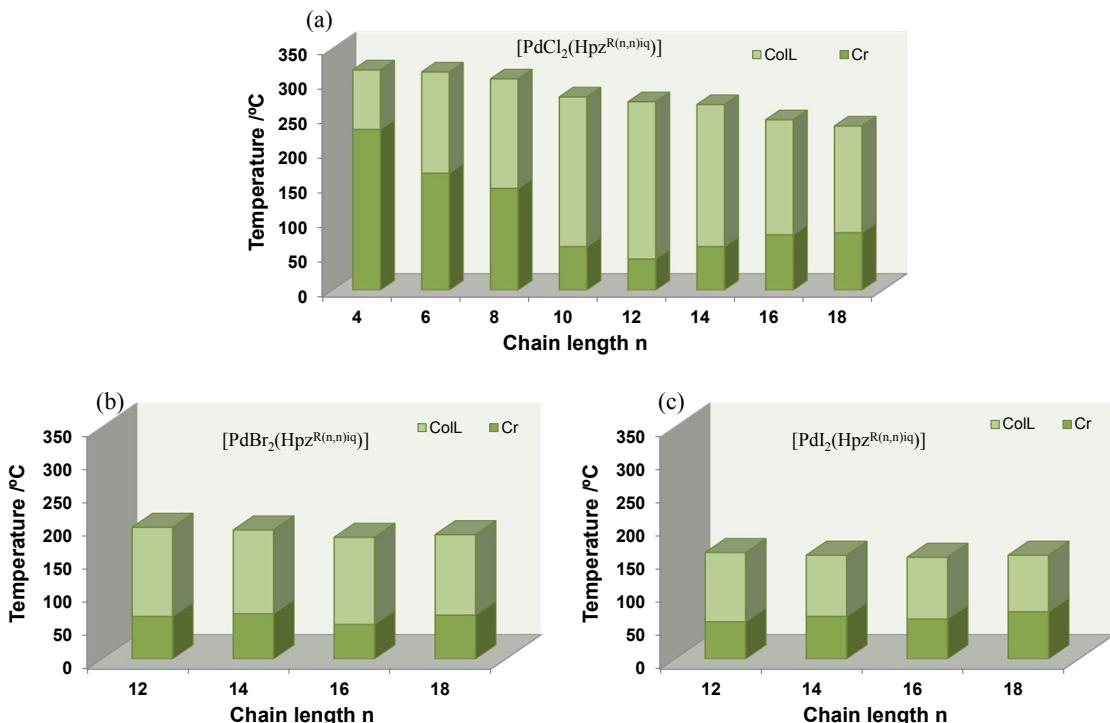


Figure S3 Bar diagram showing the range of existence of solid and liquid crystal phases for the dihalide isoquinolinylpyrazole Pd(II) compounds.

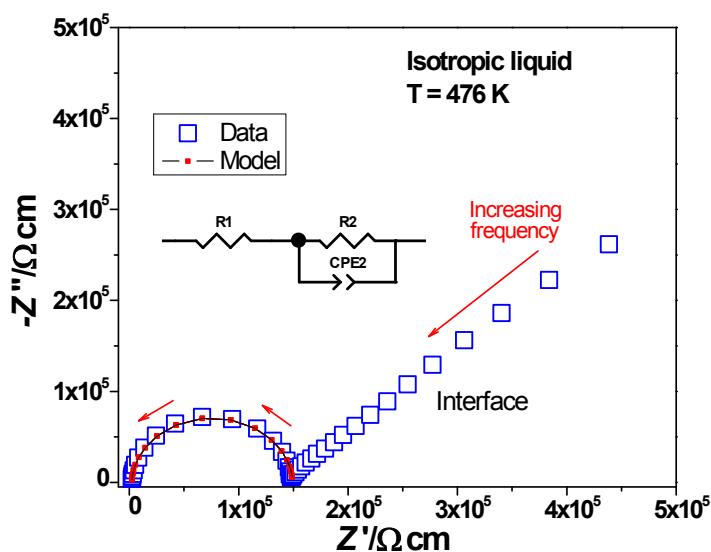


Figure S4 Z'' vs. Z' plots for the dibromide Pd(II) compound $[\text{PdBr}_2(\text{Hpz}^{\text{R}(12,12)\text{iq}})]$ **27** in the isotropic liquid phase at 476 K.

References

- [1] C. Cuerva, J. A. Campo, P. Ovejero, M. R. Torres and M. Cano, *Dalton Trans.*, 2014, **43**, 8849.