

Supporting Information

Tuning the Surface Charge Density of Exfoliated Thin Molybdenum Disulfide Sheets via Non-Covalent Functionalization for Promoting Hydrogen Evolution Reaction

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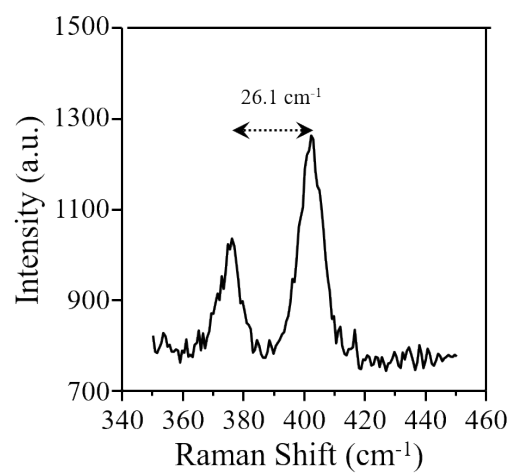


Figure S1. Characteristic peaks of Raman spectra for bulk MoS₂.

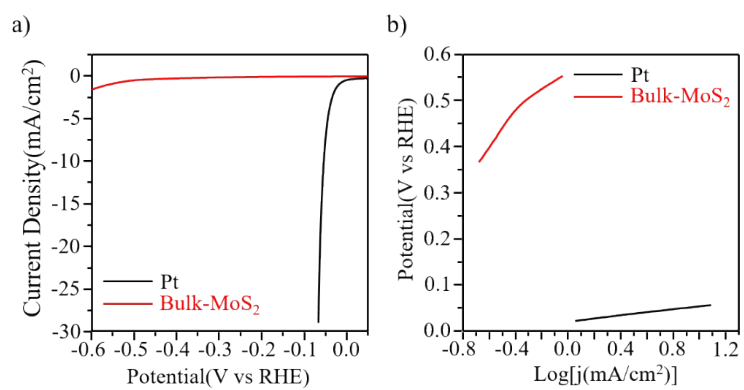


Figure S2. a) LSV and b) Tafel plots of Pt and bulk-MoS₂.

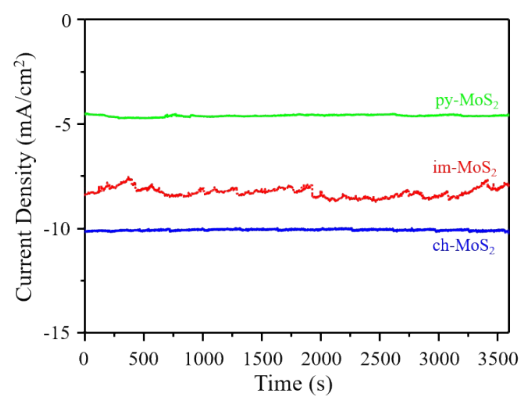


Figure S3. Chronoamperometric response recorded from the py-MoS₂, im-MoS₂, and ch-MoS₂ electrode at a constant potential of -400 mV, -300 mV, and -200 mV, respectively.

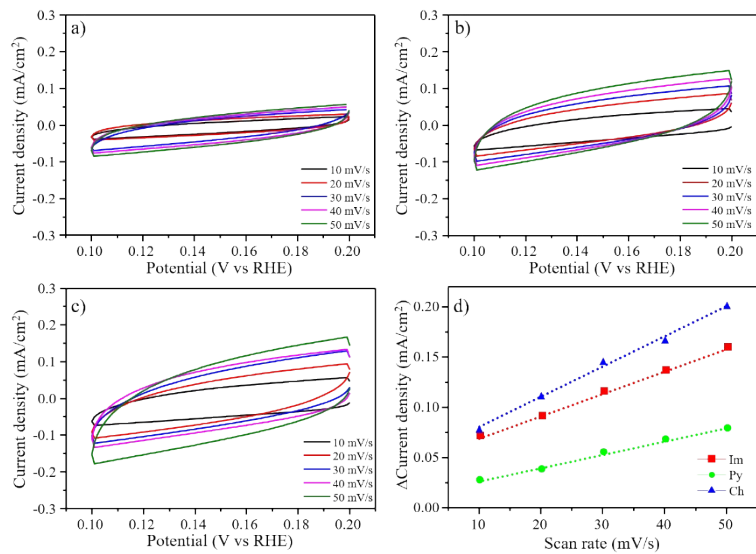


Figure S4. Cyclic voltammetry curves in the region of 0.1 ~ 0.2 V vs. RHE for a) py-MoS₂, b) im-MoS₂, c) ch-MoS₂. d) Capacitive currents at 150 mV as a function of scan rate for py-MoS₂, im-MoS₂, and ch-MoS₂.

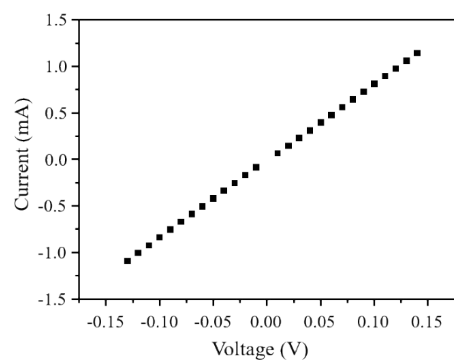


Figure S5. Electrical conductivity of 2 cm (width) x 2cm (length) ch-MoS₂/CNTs paper (thickness: 20 μ m).

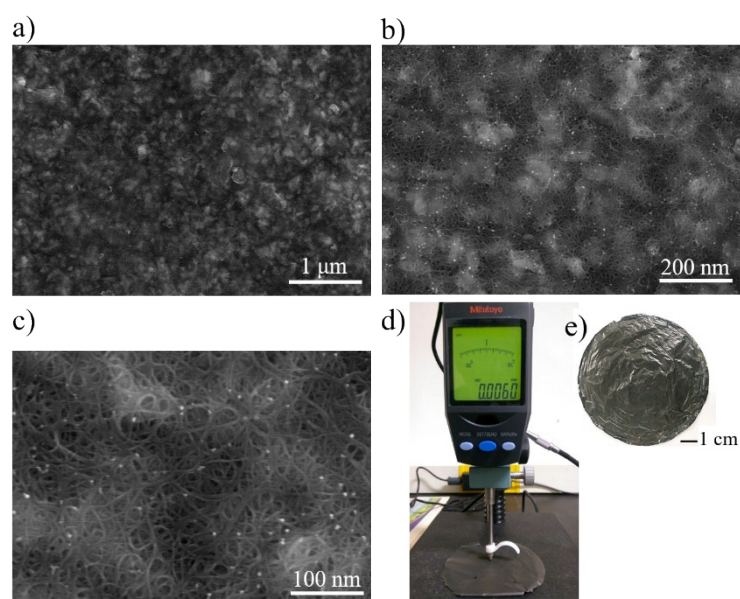


Figure S6. SEM images of the ch-MoS₂/PtNPs/CNTs. a) Low and b)-c) High magnification. d) Thickness measurement of the ch-MoS₂/PtNPs/CNTs paper electrode (unit: mm). e) Photograph of the ch-MoS₂/PtNPs/CNTs.

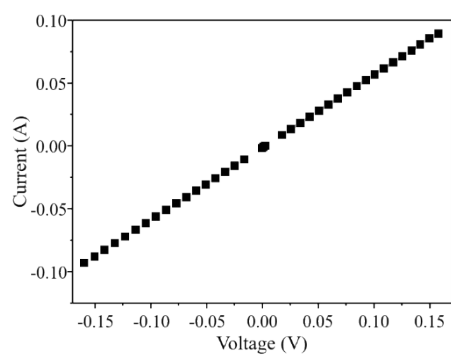


Figure S7. Electrical conductivity of ch-MoS₂/PtNPs/CNTs paper.

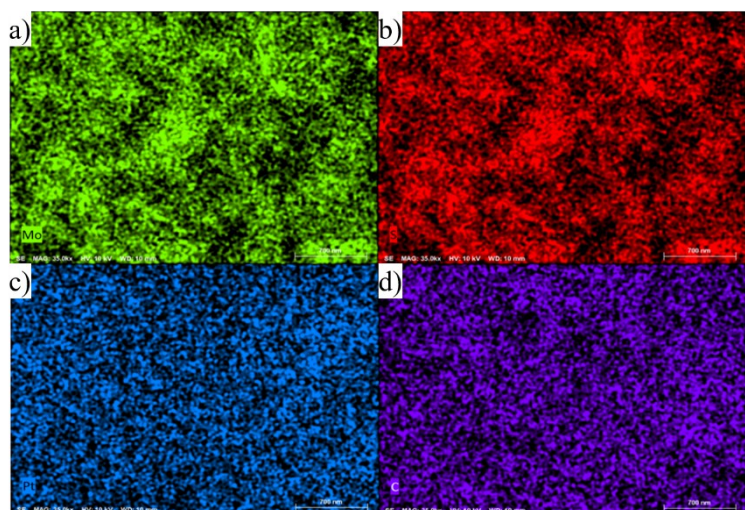


Figure S8. EDS mapping images of a) Mo, b) S, c) Pt, and d) C of the ch-MoS₂/PtNPs/CNTs paper.

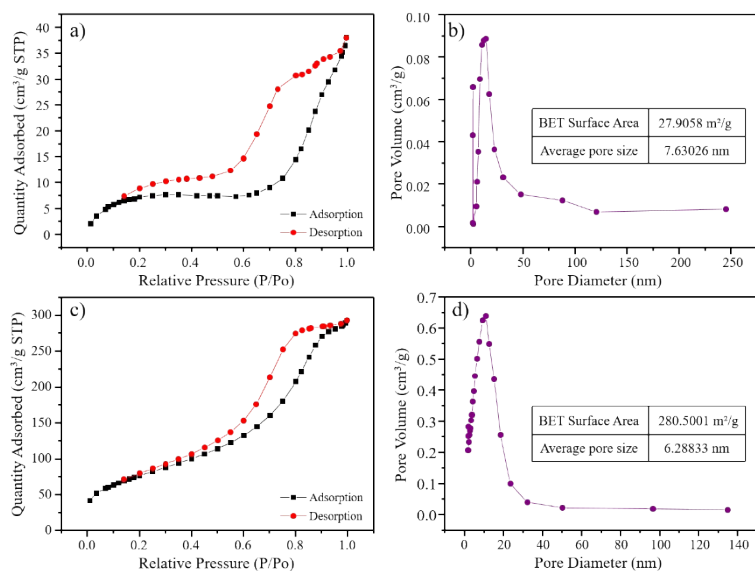


Figure S9. a) N₂ adsorption/desorption isotherms and b) pore size distribution of ch-MoS₂/CNTs. c) N₂ adsorption/desorption isotherms and d) pore size distribution of ch-MoS₂/PtNPs/CNTs.

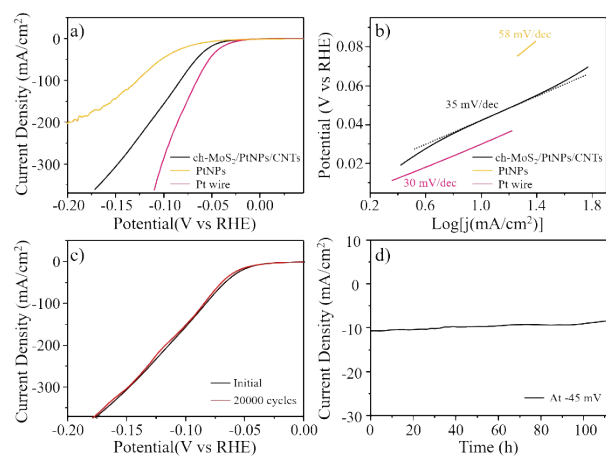


Figure S10. a) LSV, b) Tafel slope, c)-d) HER stability measurements for the flexible ch-MoS₂/PtNPs/CNTs electrode.

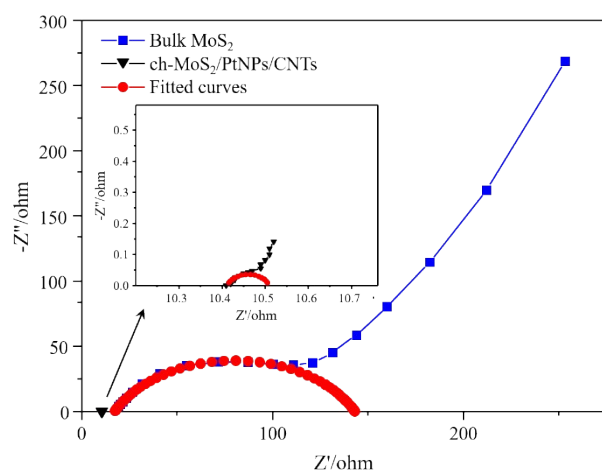


Figure S11. Electrochemical impedance spectroscopy (EIS) curves of ch-MoS₂/PtNPs/CNTs and bulk MoS₂.

Table S1. EDS analysis of the ch-MoS₂/PtNPs/SWCNT paper.

Element	Atom (%)
Platinum	0.52
Molybdenum	3.54
Sulfur	6.86
Carbon	86.12
Oxygen	2.93