### **Electronic Supplementary Information**

# Mechanofluorochromism, polymorphism and thermochromism of novel $D-\pi$ -A piperidin-1-yl-substitued isoquinoline derivatives

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### **Contents:**

### 1. Experimental

### Measurements and materials

<sup>1</sup>H, <sup>13</sup>C, and <sup>19</sup>F NMR spectra were tested on a Bruker DRX 500 NMR or DRX 400 NMR spectrometer. Melting points were tested on WRS-1B digital melting point meter (uncorrected). High-resolution electrospray ionization (HRMS-ESI) mass spectra were performed using a Hitachi Nano Frontier LD spectrometer. Elemental analyses (C, H, and N) were obtained using an Elementar Vario MICRO analyzer. X-Ray diffraction (XRD) patterns were performed using on an Empyrean X-ray diffraction instrument. Fluorescence spectra were obtained using a Cary Eclipse fluorescence spectrophotometer. The absolute fluorescence quantum yields and fluorescence lifetime decays were performed using Jobin Yvon Horiba FluoroMax-4 fluorometer. UV-vis absorption spectra were conducted on a UV-3600 Shimadzu spectrophotometer. Differential scanning calorimetry (DSC) experiments were conducted on a TA-DSC Q2000 at a heating rate of 10 °C/min. Fluorescence microscopic images were obtained using a Leica DMI3000B inverted optical microscope. Single-crystal X-ray data were collected with graphite-monochromated MoK $\alpha$  radiation on a Bruker-Nonius Smart Apex CCD diffractometer. 2,6-Dimethyl-4-pyranone (2), cyanoacetic acid, trifluoromethanesulfonic anhydride (Tf<sub>2</sub>O), 4-bromo-N,N-diphenylaniline (4), 9-(4-bromophenyl)-9H-carbazole (5), trimethylsilyl acetylene, and tetrabutyl ammonium fluoride (TBAF) were commercially available and used directly. 4-Ethynyl- $N_{,N}$ -diphenylaniline (8)<sup>1</sup> and 9-(4-ethynylphenyl)-9Hcarbazole  $(9)^2$  were synthesized using compound 4 and compound 5 as the starting material based on the previous literatures, respectively.

### Synthesis of 2-(2,6-dimethyl-4*H*-pyran-4-ylidene)-3-oxopentanedinitrile (1)

Compound 2 (1.24 g, 10.0 mmol) and cyanoacetic acid (4.25 g, 60.0 mmol) were dissolved in

acetic anhydride (10 mL) with stirring at 140°C for 3 h. After the reaction was completed, the reaction mixture was allowed to cool to room temperature and then poured into water (500 mL). After 30 min of stirring, the mixture was extracted three times with dichloromethane. Anhydrous sodium sulfate was added to dry the organic phase and then the organic solvent was evaporated by decompression. The residue was passed through a silica gel column chromatography (ethyl acetate-petroleum ether = 1:2) to afford a yellow solid (727.8 mg, 34.0% yield). M. p. 175.3-176.1°C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  8.31 (s, 1H), 6.70 (s, 1H), 3.88 (s, 2H), 2.41 (s, 3H), 2.40 (s, 3H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  182.0, 165.74, 165.69, 155.6, 119.5, 114.0, 109.4, 108.2, 85.4, 31.8, 20.3, 20.1 ppm. HRMS (ESI) m/z: [M+Na]<sup>+</sup> calculated for C<sub>12</sub>H<sub>10</sub>N<sub>2</sub>O<sub>2</sub>Na, 237.0640; found, 237.0644.

#### Synthesis of 8-hydroxy-3,6-dimethyl-1-(piperidin-1-yl)isoquinoline-7-carbonitrile (PIQ)

A mixture of compound **1** (642.2 mg, 3.0 mmol) and piperidine (510.5 mg, 6.0 mmol) in acetonitrile (10 mL) was stirred at 90°C for 4 h. After the reaction was completed, the reaction mixture was allowed to cool to room temperature and the organic solvent was evaporated by decompression. The residue was passed through a silica gel column chromatography (ethyl acetate-petroleum ether = 1:5) to afford a white solid (716.9 mg, 85.0% yield). M. p. 169.7-170.1°C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  16.63 (s, 1H), 7.19 (s, 1H), 6.88 (s, 1H), 3.26-3.14 (m, 4H), 2.58 (s, 3H), 2.55 (s, 3H) 1.90-1.82 (m, 5H), 1.54-1.42 (m, 1H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  164.4, 160.3, 154.0, 142.7, 141.9, 117.6, 116.3, 114.9, 109.4, 95.2, 52.8, 25.6, 24.0, 23.1, 21.2 ppm. HRMS (ESI) m/z: [M+H]<sup>+</sup> calculated for C<sub>17</sub>H<sub>20</sub>N<sub>3</sub>O, 282.1601; found, 282.1603.

## Synthesisof7-cyano-3,6-dimethyl-1-(piperidin-1-yl)isoquinolin-8-yltrifluoromethanesulfonate (3)

Pyridine (592.8 mg, 7.5 mmol) was added to the mixture of **PIQ** (702.9 mg, 2.5 mmol) and dichloromethane (10 mL) at 0°C. After being stirring for 5 min at 0°C, trifluoromethane sulfonic anhydride (1.41 g, 5.0 mmol) was added to the reaction mixture drop by drop. The reaction mixture was allowed to warm to the room temperature and further stirred for 3 h. Then 1 mL ether and 2 mL 10% hydrochloric acid solution were added to quench the reaction. The mixture was extracted with dichloromethane, and the organic phase was washed with saturated sodium bicarbonate solution and then dried with anhydrous sodium sulfate. The organic solvent was evaporated by decompression and the residue was passed through a silica gel column chromatography (ethyl acetate-petroleum ether = 1:10) to afford a green solid (831.4 mg, 80.5% yield). M. p. 186.8-187.1 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.47 (s, 1H), 6.88 (s, 1H), 3.45-3.25 (m, 4H), 2.64 (s, 3H), 2.53 (s, 3H) 1.83-1.52 (m, 6H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz): δ 158.5, 154.4, 148.7, 142.8, 140.3, 126.4, 118.7 (q, *J*<sub>C-F</sub> = 319.5 Hz), 113.6, 110.6, 110.0, 103.9, 24.7, 24.5, 24.3, 20.7 ppm. <sup>19</sup>F NMR (CDCl<sub>3</sub>, 470 MHz): δ -73.1 ppm. HRMS (ESI) m/z:

[M+Na]<sup>+</sup> calculated for C<sub>18</sub>H<sub>18</sub>F<sub>3</sub>N<sub>3</sub>O<sub>3</sub>SNa, 436.0919; found, 436.0919.

### Synthesis of compounds 6 and 7

The general procedure is as follows: a mixture of compound 4/5 (10.0 mmol), CuI (95.5 mg, 0.5 mmol), Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (175.3 mg, 0.25 mmol), and trimethylsilyl acetylene (2.95 g, 30.0 mmol) in 20 mL triethylamine and 20 mL THF was stirred for 12 h at 80°C in N<sub>2</sub> atmosphere. The organic solvent was evaporated by decompression and the residue was passed through a silica gel column chromatography (petroleum ether) to afford pure compound 6/7.

*N*,*N*-Diphenyl-4-((trimethylsilyl)ethynyl)aniline (6). White solids (2.69 g), 78.7% yield. <sup>1</sup>H NMR (DMSO- $d_6$ , 500 MHz): 7.36-7.30 (m, 6H), 7.13-7.05 (m, 6H), 6.83 (d, J = 8.5 Hz, 2H), 0.20 (s,9H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  147.7, 146.3, 132.7, 129.6, 124.9, 123.9, 120.9, 114.6, 105.5, 92.7, -0.12 ppm. HRMS (ESI) m/z: [M+Na]<sup>+</sup> calculated for C<sub>23</sub>H<sub>23</sub>NSiNa, 364.1498; found, 364.1493.

**9-(4-((Trimethylsilyl)ethynyl)phenyl)-9***H***-carbazole (7).** White solids (2.79 g), 82.1% yield. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): 8.14 (d, J = 7.5 Hz, 2H), 7.71 (d, J = 8.5 Hz, 2H), 7.53 (d, J = 8.0 Hz, 2H), 7.44-7.40 (m, 4H), 7.33-7.28 (m, 2H), 0.31 (s,9H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  140.6, 137.8, 133.5, 126.7, 126.0, 123.6, 122.2, 120.3, 120.2, 109.7, 104.3, 95.4, -0.04 ppm. HRMS (ESI) m/z: [M +H]<sup>+</sup> calcd for C<sub>23</sub>H<sub>22</sub>NSi, 340.1516, found, 340.1510.

#### Synthesis of compounds 8 and 9

The general procedure is as follows: a mixture of compound 6/7 (5.0 mmol) and tetrabutylammonium fluoride (10.0 mmol) in THF (20 mL) was stirred for 12 h at 0°C in a N<sub>2</sub> atmosphere. The organic solvent was evaporated by decompression and the residue was passed through a silica gel column chromatography (petroleum ether) to afford pure compound 8/9.

**4-Ethynyl-***N***,***N***-diphenylaniline (8).** White solids (1.02 g), 75.8% yield. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 500 MHz): 7.35-7.32 (m, 6H), 7.12-7.05 (m, 6H), 6.86 (d, *J* = 8.5 Hz, 2H), 4.06 (s, 1H) ppm. <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 125 MHz): 147.7, 146.4, 132.8, 129.6, 124.8, 123.9, 121.2, 114.2, 83.6, 79.5 ppm. HRMS (ESI) m/z: [M+Na]<sup>+</sup> calculated for C<sub>20</sub>H<sub>15</sub>NNa, 292.1102; found, 292.1104.

**9-(4-Ethynylphenyl)-9***H***-carbazole (9).** White solids (1.04 g), 78.1% yield. <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>, 500 MHz): 8.25 (d, J = 7.5 Hz, 2H), 7.77 (d, J = 8.5 Hz, 2H), 7.66 (d, J = 8.5 Hz, 2H), 7.46-7.41 (m, 4H), 7.32-7.29 (m, 2H), 4.34 (s, 1H) ppm. <sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>, 125 MHz):  $\delta$  139.7, 137.2, 133.5, 126.7, 126.3, 122.9, 120.6, 120.5, 120.3, 109.6, 82.8, 81.5 ppm. HRMS (ESI) m/z: [M +H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>14</sub>N, 268.1121; found, 268.1124.

### Synthesis of 8-((4-(diphenylamino)phenyl)ethynyl)-3,6-dimethyl-1-(piperidin-1yl)isoquinoline-7-carbonitrile (IQ-TPA)

A mixture of compound **8** (807.4 mg, 3.0 mmol), compound **3** (826.8 mg, 2.0 mmol), CuI (22.9 mg, 0.12 mmol), and Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (84.0 mg, 0.12 mmol) in triethylamine (8 mL) and *N*, *N*-

dimethylformamide (8 mL) was stirred for 12 h at 100°C in a N<sub>2</sub> atmosphere. The mixture was allowed to cool to room temperature and then water (75 mL) was added. The reaction mixture was extracted with dichloromethane and the organic solvent was evaporated by decompression. The residue was passed through a silica gel column chromatography (dichloromethane-petroleum ether = 1:4) to afford a yellow-green solid (505.6 mg, 47.5% yield). M. p. 224.2-225.0 °C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): 7.58 (d, J = 8.5 Hz, 2H), 7.33-7.29 (m, 5H), 7.16 (d, J = 8.0 Hz, 4H), 7.10 (t, J = 14.5 Hz, 2H), 7.05 (d, J = 9.0 Hz, 2H), 6.84 (s, 1H), 3.64 (br, 2H), 3.17 (br, 2H), 2.61 (s, 3H), 2.50 (s, 3H), 1.88-1.49 (m, 6H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  160.6, 152.8, 148.8, 146.9, 141.7, 139.4, 133.0, 129.4, 127.3, 125.8, 125.4, 123.9, 121.5, 117.9, 115.9, 115.2, 115.0, 111.6, 102.4, 87.8, 53.0, 25.4, 24.6, 24.2, 20.9 ppm. HRMS (ESI) m/z: [M+H]<sup>+</sup> calculated for C<sub>37</sub>H<sub>33</sub>N<sub>4</sub>, 533.2700; found, 533.2698. Anal. calcd. for C<sub>37</sub>H<sub>32</sub>N<sub>4</sub>: C, 83.43; H, 6.06; N, 10.52. Found: C, 83.98; H, 6.03; N, 10.47.

## Synthesis of 8-((4-(9*H*-carbazol-9-yl)phenyl)ethynyl)-3,6-dimethyl-1-(piperidin-1-yl)isoquinoline-7-carbonitrile (IQ-PC)

A mixture of compound **9** (801.3 mg, 3.0 mmol), compound **3** (826.8 mg, 2.0 mmol), CuI (22.9 mg, 0.12 mmol), and Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (84.0 mg, 0.12 mmol) in triethylamine (8 mL) and *N*, *N*-dimethylformamide (8 mL) was stirred for 12 h at 100°C in a N<sub>2</sub> atmosphere. After the mixture was cooled to room temperature, a large number of solids were precipitated. The precipitate was filtrated under vacuum and was passed through a silica gel column chromatography (dichloromethane-petroleum ether = 1:3) to afford a yellow-green solid (719.0 mg, 67.8% yield). M. p. 281.0-282.0°C. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): 8.16 (d, *J* = 8.0 Hz, 2H), 7.97 (d, *J* = 8.5 Hz, 2H), 7.67 (d, *J* = 8.5 Hz, 2H), 7.51 (d, *J* = 8.0 Hz, 2H), 7.46-7.42 (m, 3H), 7.34-7.31 (m, 2H), 6.89 (s, 1H), 3.70 (br, 2H), 3.24 (br, 2H), 2.66 (s, 3H), 2.54 (s, 3H), 1.94-1.55 (m, 6H) ppm. <sup>13</sup>C NMR (CDCl<sub>3</sub>, 125 MHz):  $\delta$  160.6, 153.0, 141.8, 140.5, 139.5, 138.6, 133.5, 126.9, 126.63, 126.59, 126.1, 123.7, 120.4, 117.8, 116.2, 115.7, 111.8, 109.8, 100.6, 89.0, 53.2, 25.5, 24.6, 24.2, 21.0 ppm. HRMS (ESI) m/z: [M+H]<sup>+</sup> calculated for C<sub>37</sub>H<sub>31</sub>N<sub>4</sub>, 531.2543; found, 531.2565. Anal. calcd. for C<sub>37</sub>H<sub>30</sub>N<sub>4</sub>: C, 83.74; H, 5.70; N, 10.56. Found: C, 83.27; H, 5.74; N, 10.61.

### Notes and references

- C. Teng, X. Yang, C. Yang, S. Li, M. Cheng, A. Hagfeldt and L. Sun, *J. Phys. Chem. C*, 2010, 114, 9101–9110.
- A. Aguilar-Granda, S. Pérez-Estrada, A. E. Roa, J. Rodríguez-Hernández, S. Hernández-Ortega, M. Rodriguez and B. Rodríguez-Molina, *Cryst. Growth Des.*, 2016, 16, 3435–3442.

### 2. Figures, schemes, and tables



Fig. S1 Normalized fluorescence spectra (a) and XRD curves (b) of the PIQ solid samples before and after grinding. Inset: Emission pictures of the PIQ solid samples before and after grinding under the UV light ( $\lambda_{ex} = 365$  nm).



Scheme S1 Possible mechanism of PIQ prepared from compound 1 and piperidine.



Fig. S2 Normalized emission (a) and absorption (b) spectra of IQ-TPA in different solvents. Concentration:  $1 \times 10^{-5}$  M.



Fig. S3 Normalized emission (a) and absorption (b) spectra of IQ-PC in different solvents. Concentration:  $1 \times 10^{-5}$  M.

 Table S1 Crystal data and details of collection and refinement for the polymorphs of IQ-TPA and IQ-PC.

	IQ-TPA-g	IQ-TPA-y	IQ-PC-g	IQ-PC-y
CCDC (no.)	1940020	1940022	1940021	1940023
Empirical formula	$C_{37}H_{32}N_4$	$C_{37}H_{32}N_4$	$C_{37}H_{30}N_4$	$C_{37}H_{30}N_4$
Formula weight	532.66	532.66	530.65	530.65
Temperature (K)	293(2)	293(2)	293(2)	293(2)
Crystal system	Monoclinic	Monoclinic	Orthorhombic	Orthorhombic
Space group	<i>P</i> 21/c	<i>P</i> 21/c	<i>P</i> n a 21	<i>P</i> n a 21
Ζ	4	4	4	4
D <sub>calcd</sub> [Mg/m <sup>3</sup> ]	1.212	1.214	1.242	1.240
F (000)	1128	1128	1120	1120
$\theta$ range [°]	2.535-24.996	2.629-25.998	2.567-25.495	2.566-25.992
$R_1[I \ge 2\sigma(I)]$	0.0831	0.0447	0.0441	0.0445
$wR_2 [I \ge 2\sigma(I)]$	0.2060	0.1050	0.0845	0.0967
<i>a</i> [Å]	10.951(2)	10.9558(3)	8.1615(2)	8.1632(4)
<i>b</i> [Å]	10.159(2)	10.1374(3)	33.9887(12)	33.999(2)
<i>c</i> [Å]	26.604(5)	26.5962(7)	10.2333(3)	10.2383(5)
$\alpha$ [deg]	90	90	90	90
$\beta$ [deg]	99.365(7)	99.3760(10)	90	90
γ [deg]	90	90	90	90
V[Å <sup>3</sup> ]	2920.3(10)	2914.40(14)	2838.71(15)	2841.5(3)
GOF	1.036	1.034	1.062	1.056
R(int)	0.0832	0.0531	0.0392	0.0563
No. of reflens collected	12707	41504	12972	23645
No. of unique reflens	5102	5723	5168	5532
$R_1$ (all data)	0.1392	0.0685	0.0674	0.0704
$wR_2$ (all data)	0.2602	0.1221	0.0965	0.1109



**Fig. S4** Comparison of experimental XRD curves and the simulated XRD curves obtained from the corresponding single crystals of **IQ-TPA**: **IQ-TPA-***g* (a) and **IQ-TPA-***y* (b).



**Fig. S5** Comparison of experimental XRD curves and the simulated XRD curves obtained from the corresponding single crystals of **IQ-PC**: **IQ-PC**-*g* (a) and **IQ-PC**-*y* (b).



**Fig. S6** Single crystal **IQ-TPA-***y*: (a) Intermolecular interactions; (b) Zigzag stacking arrangement viewed along the *a*-axis; (c) Stacking arrangement along the *b*-axis.



Fig. S7 Selected dihedral angles and torsion angles along with  $\pi$ -plane of IQ-TPA-*g* (a), IQ-TPA-*y* (b), IQ-PC-*g* (c), and IQ-PC-*y* (d).



**Fig. S8** Normalized solid-state absorption spectra of the polymorphs of **IQ-TPA** (a) and **IQ-PC** (b).



**Fig. S9** Single crystal **IQ-PC-***y*: (a) (b) Intermolecular interactions; (c) Zigzag stacking arrangement viewed along the *c*-axis.



Fig. S10 Comparison of the fluorescence spectra of IQ-TPA-*g* and the annealed sample of IQ-TPA-*y*.



Fig. S11 Fluorescence spectra of IQ-PC-g (a) and IQ-PC-y (b) before and after grinding.



Fig. S12 Comparison of the fluorescence spectra of IQ-PC-*y* and the annealed sample of IQ-PC-*g*.



Fig. S13 XRD curves of the polymorphs of IQ-PC before and after grinding.



Fig. S14 Normalized solid-state absorption spectra of IQ-TPA-*y*, the ground sample of IQ-TPA-*y*, and IQ-TPA-*g*.



**Fig. S15** (a) Comparison of the XRD curves of **IQ-TPA-***g* and the annealed sample of **IQ-TPA-***y*. (b) Comparison of the XRD curves of **IQ-PC**-*y* and the annealed sample of **IQ-PC**-*g*.



Fig. S16 DSC curves of IQ-TPA (a) and IQ-PC (b) solid samples under various conditions.



Fig. S18<sup>13</sup>C NMR of compound 1 (CDCl<sub>3</sub>, 125 MHz).





Fig. S22 <sup>13</sup>C NMR of compound 3 (CDCl<sub>3</sub>, 125 MHz).



Fig. S23 <sup>19</sup>F NMR of compound 3 (CDCl<sub>3</sub>, 470 MHz).









Fig. S26 <sup>1</sup>H NMR of compound 7 (CDCl<sub>3</sub>, 500 MHz).



**Fig. S28** <sup>1</sup>H NMR of compound **8** (DMSO-*d*<sub>6</sub>, 500 MHz).



**Fig. S30** <sup>1</sup>H NMR of compound **9** (DMSO-*d*<sub>6</sub>, 500 MHz).







Fig. S34 <sup>1</sup>H NMR of IQ-PC (CDCl<sub>3</sub>, 500 MHz).

