**Electronic Supplementary Information** 

# Asymmetric thermally activated delayed fluorescence (TADF) emitters with 5,9-dioxa-13*b*-boranaphtho[3,2,1-*de*]anthracene (OBA) as acceptor and highly efficient blue-emitting OLEDs

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Figure S9. Relationship between EL efficiencies and luminance for the S14 devices except the optimized ones. (a) Device A3, (b) Device B3, (c) Device C3 and (d) Device D3.

### 1,3-Diphenoxybenzene



Under a nitrogen atmosphere, 1,3-dibromobenzene (5.00 g, 0.02 mol), phenol (5.19 g, 0.06 mol), Cs<sub>2</sub>CO<sub>3</sub> (20.70 g, 0.06 mol) and CuBr (0.91 g, 0.006 mol) were heated to 160 °C in 35 mL degassed 2-pyrrolidinone for 24 h. After cooling to room temperature, the mixture was extracted with petroleum ether. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether as the eluent. White solid of the title compound was obtained (4.06 g, 72%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 7.32 (t, *J* = 8.0 Hz, 4H), 7.23 (t, *J* = 8 Hz, 1H), 7.10 (t, *J* = 7.6 Hz, 2H), 7.02 (d, *J* = 8.0 Hz, 4H), 6.72–6.69 (m, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 160.3, 158.3, 132.0, 131.5, 125.3, 120.8, 114.8, 110.9

#### 1,4-Benzoxaborino-2,3,4-kl-phenoxaborin



A solution of *n*-butyllithium in hexane (3.66 mL, 2.5 M, 9.16 mmol) was added slowly to a solution of 1,3-diphenoxybenzene (2.00 g, 7.63 mmol) in o-xylene (30.0 mL) at 0 °C under a nitrogen atmosphere. After stirring at 70 °C for 4 h, borontribromide (0.87 mL, 9.16 mmol) was added slowly at -40 °C. After the reaction mixture was allowed to warm to room temperature for 1 h, *N*,*N*-

diisopropylethylamine (2.66 mL, 15.26 mmol) was added at 0 °C. After stirring at 120 °C for 5 h, *N*,*N*-diisopropylethylamine (1.33 mL, 7.64 mmol) was added at 0 °C, the reaction mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether as the eluent. White solid of the title compound was obtained (1.27 g, 62%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 8.69 (dd, *J* = 1.6, 7.8 Hz, 1H), 7.79 (t, *J* = 8.0Hz, 1H), 7.70 (ddd, *J* = 1.6, 7.1, 8.5 Hz, 2H), 7.54 (d, *J* = 8.5 Hz, 2H), 7.38 (d, *J* = 7.1, 7.8 Hz, 2H), 7.22 (d, *J* = 8 Hz, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 160.5, 157.4, 134.6, 134.5, 133.6, 122.7, 118.5, 108.4.

## 10-(4-bromophenyl)-10H-phenothiazine (SBr)



Under a nitrogen atmosphere, phenothiazine (2.00 g, 10.90 mmol), 4-bromoiodobenzene (2.80 g, 10.00 mmol) and Copper powder (3.00 g, 23.40 mmol) were heated to 160 °C in 35 mL *N*,*N*-dimethylformamide for 48 h. After cooling to room temperature, the mixture was washed with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether / CH<sub>2</sub>Cl<sub>2</sub> (5:1, v:v) as the eluent. A white solid of the title compound was obtained (2.50 g, 66%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 7.70 (d, *J* = 8.6 Hz, 2H), 7.25 (d, *J* = 8.6 Hz, 2H), 7.04 (dd, *J* = 7.3, 1.8 Hz, 2H), 6.86 (m, 4H), 6.24 (dd, *J* = 8.0, 1.4 Hz, 2H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 144.2, 143.8, 140.9, 133.8, 131.9, 129.7, 129.5, 122.5, 122.2, 121.2, 117.5, 116.2



Under a nitrogen atmosphere, bis-(pinacolato)diboro (1.10 g, 4.33 mmol), *N*-(4-bromophenyl)phenothiazine (1.00 g, 2.833 mmol), KOAc (0.931 g, 9.482 mmol) and Pd(dppf)Cl<sub>2</sub> (0.178 g, 0.632 mmol) were heated to 110 °C in 25 mL degassed 1,4-dioxane for 12 h. After cooling to room temperature, the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using CH<sub>2</sub>Cl<sub>2</sub> / petroleum ether (1:3, v:v) as the eluent. White solid of the title compound was obtained (0.62 g, 53%). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  (ppm) 7.88 (d, *J* = 8.3 Hz, 2H), 7.35(d, *J* = 8.3 Hz, 2H), 7.18 (dd, *J* = 7.5, 1.6 Hz, 2H), 7.02 (ddd, *J* = 7.9, 7.7, 1.7 Hz, 2H), 6.95 (ddd, *J* = 7.5, 7.5, 1.3 Hz, 2H), 6.41 (dd, *J* = 8.1, 1.3 Hz, 2H), 1.33 (s, 12H). <sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>):  $\delta$ (ppm) 144.5, 143.4, 137.3, 127.9, 127.6, 123.9, 122.4, 118.6, 84.3, 25.2.

#### 10-(4-bromophenyl)-10H-phenoxazine (OBr)



S5

SB

Under a nitrogen atmosphere, phenoxazine (2.00 g, 10.00 mmol), 4-bromoiodobenzene (2.80 g, 10.00 mmol) and Copper powder (3.00 g, 23.40 mmol) were heated to 160 °C in 35 mL *N*,*N*-dimethylformamide for 48 h. After cooling to room temperature, the mixture was washed with water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether / CH<sub>2</sub>Cl<sub>2</sub> (5:1, v:v) as the eluent. White solid of the title compound was obtained (2.30 g, 68%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 5.91 (dd, *J* = 7.8, 1.4 Hz, 2H), 6.57 - 6.70 (m, 6H), 7.23 (d, *J* = 8.6 Hz, 2H), 7.72 (d, *J* = 8.6Hz, 2H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 113.3, 115.7, 121.7, 122.5,123.4, 132.9, 134.1, 134.6, 138.2, 144.0.

OB



Under a nitrogen atmosphere, bis-(pinacolato)diboro (1.10 g, 4.33 mmol), N-(4-Bromophenyl)-Phenoxazine (1.00 g, 2.97 mmol), KOAc (0.93 g, 9.48 mmol) and Pd(dppf)Cl<sub>2</sub> (0.18 g, 0.63 mmol) were heated to 110 °C in 25 mL degassed 1,4-dioxane for 12 h. After cooling to room temperature, the mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was dried over anhydrous Na<sub>2</sub>SO<sub>4</sub> and evaporated under reduced pressure, then purified by column chromatography over silica gel using CH<sub>2</sub>Cl<sub>2</sub> / petroleum ether (1:5, v:v) as the eluent. White solid of the title compound was obtained (0.65 g, 55%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm) 8.02 (d, *J* = 8.0 Hz, 2H), 7.35 (d, *J* = 8.0 Hz, 2H), 6.68 (dd, *J* = 7.8 Hz, 1.4 Hz, 2H), 6.63 (t, *J* = 7.6 Hz, 2H), 6.56 (td, *J* = 7.6 Hz, 1.5 Hz, 2H), 5.91 (d, *J* = 7.8 Hz, 2H), 1.38 (s, 12H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ (ppm) 145.9, 139.2, 134.5, 134.3, 125.9, 122.8, 122.7, 139.2, 134.5, 88.1, 24.7.



Figure S1. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra for OBA-O.



Figure S2. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra for OBA-S.



Figure S3. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra for OBA-BrO.



Figure S4. <sup>1</sup>H- and <sup>13</sup>C-NMR spectra for OBA-BrS.



Figure S5. TGA curves of these OBA-based emitters.



**Figure S6.** Cyclic curves of these asymmetric OBA-based emitters measured in CH<sub>3</sub>CN and ferrocene/ferrocenium (Fc/Fc<sup>+</sup>) couple at room temperature. (a) **OBA-O**, (b) **OBA-S**, (c) **OBA-BrO**, (d) **OBA-BrS**.



**Figure S7.** Current density–voltage–luminance (*J*–*V*–*L*) curves for the devices except the optimized ones.



Figure S8. Relationship between EL efficiencies and luminance for the devices except the optimized ones. (a) Device A1, (b) Device B1, (c) Device C1 and (d) Device D1.



Figure S9. Relationship between EL efficiencies and luminance for the devices except the optimized ones. (a) Device A3, (b) Device B3, (c) Device C3 and (d) Device D3.