## Electronic Supplementary Information

# Asymmetric thermally activated delayed fluorescence (TADF) emitters with 5,9-dioxa-13b-boranaphtho[3,2,1-de]anthracene (OBA) as acceptor and highly efficient blue-emitting OLEDs 

Dongdong Song, ${ }^{\text {a }}$ Yue Yu, ${ }^{\text {b }}$ Ling Yue, ${ }^{\text {a }}$ Daokun Zhong, ${ }^{a}$ Yindi Zhang, ${ }^{a}$ Xiaolong Yang, ${ }^{\text {a }}$ Yuanhui Sun, ${ }^{\text {a }}$ Guijiang Zhou, ${ }^{* a}$ Zhaoxin Wu*b

${ }^{\text {a }}$ MOE Key Laboratory for Nonequilibrium Synthesis and Modulation of Condensed Matter, Department of Chemistry, School of Science, State Key Laboratory for Mechanical Behavior of Materials, Xi'an Jiaotong University, Xi'an 710049, P. R. China.
${ }^{\mathrm{b}}$ Key Laboratory of Photonics Technology for Information, School of Electronic and Information Engineering, Xi'an Jiaotong University, Xi'an 710049, P. R. China.
*Guijiang Zhou: E-mail, zhougj@mail.xjtu.edu.cn
*Zhaoxin Wu: E-mail, zhaoxinwu@mail.xjtu.edu.cn

## Contents

Synthesis of the key intermediate compounds S3
Figure S1. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for OBA-O. S8
Figure S2. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}-$ NMR spectra for OBA-S. S9
Figure S3. ${ }^{1} \mathrm{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for OBA-BrO. $\quad$ S10

Figure S4. ${ }^{\mathbf{1}} \mathrm{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for OBA-BrS. S11
Figure S5. TGA curves of these OBA-based emitters. $\quad$ S12
Figure S6. Cyclic curves of these asymmetric OBA-based emitters S12 measured in $\mathrm{CH}_{3} \mathrm{CN}$ and ferrocene/ferrocenium $\left(\mathrm{Fc} / \mathrm{Fc}^{+}\right)$couple at room temperature. (a) OBA-O, (b) OBA-S, (c) OBA-BrO, (d) OBA-BrS.

Figure S7. Current density-voltage-luminance $(J-V-L)$ curves for the devices S13 except the optimized ones.

Figure S8. Relationship between EL efficiencies and luminance for the S14 devices except the optimized ones. (a) Device A1, (b) Device B1, (c) Device C1 and (d) Device D1.

Figure S9. Relationship between EL efficiencies and luminance for the devices except the optimized ones. (a) Device A3, (b) Device B3, (c) Device C3 and (d) Device D3.

## 1,3-Diphenoxybenzene



Under a nitrogen atmosphere, 1,3-dibromobenzene ( $5.00 \mathrm{~g}, 0.02 \mathrm{~mol}$ ), phenol ( $5.19 \mathrm{~g}, 0.06 \mathrm{~mol}$ ), $\mathrm{Cs}_{2} \mathrm{CO}_{3}(20.70 \mathrm{~g}, 0.06 \mathrm{~mol})$ and $\mathrm{CuBr}(0.91 \mathrm{~g}, 0.006 \mathrm{~mol})$ were heated to $160^{\circ} \mathrm{C}$ in 35 mL degassed 2-pyrrolidinone for 24 h . After cooling to room temperature, the mixture was extracted with petroleum ether. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether as the eluent. White solid of the title compound was obtained ( $4.06 \mathrm{~g}, 72 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm})$ $7.32(\mathrm{t}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.23(\mathrm{t}, J=8 \mathrm{~Hz}, 1 \mathrm{H}), 7.10(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H})$, 6.72-6.69 (m, 3H). ${ }^{13} \mathrm{C}$ NMR (100 MHz, $\mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 160.3,158.3,132.0,131.5,125.3,120.8$, 114.8, 110.9

## 1,4-Benzoxaborino-2,3,4-kl-phenoxaborin



A solution of $n$-butyllithium in hexane $(3.66 \mathrm{~mL}, 2.5 \mathrm{M}, 9.16 \mathrm{mmol})$ was added slowly to a solution of 1,3-diphenoxybenzene ( $2.00 \mathrm{~g}, 7.63 \mathrm{mmol}$ ) in o-xylene $(30.0 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ under a nitrogen atmosphere. After stirring at $70{ }^{\circ} \mathrm{C}$ for 4 h , borontribromide ( $0.87 \mathrm{~mL}, 9.16 \mathrm{mmol}$ ) was added slowly at $-40^{\circ} \mathrm{C}$. After the reaction mixture was allowed to warm to room temperature for $1 \mathrm{~h}, \mathrm{~N}, \mathrm{~N}$ -
diisopropylethylamine ( $2.66 \mathrm{~mL}, 15.26 \mathrm{mmol}$ ) was added at $0^{\circ} \mathrm{C}$. After stirring at $120^{\circ} \mathrm{C}$ for 5 h , $N, N$-diisopropylethylamine ( $1.33 \mathrm{~mL}, 7.64 \mathrm{mmol}$ ) was added at $0^{\circ} \mathrm{C}$, the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether as the eluent. White solid of the title compound was obtained ( $1.27 \mathrm{~g}, 62 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 8.69(\mathrm{dd}, J=1.6,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 7.79(\mathrm{t}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.70(\mathrm{ddd}, J=1.6,7.1,8.5$ $\mathrm{Hz}, 2 \mathrm{H}), 7.54(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.38(\mathrm{~d}, J=7.1,7.8 \mathrm{~Hz}, 2 \mathrm{H}), 7.22(\mathrm{~d}, J=8 \mathrm{~Hz}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 160.5,157.4,134.6,134.5,133.6,122.7,118.5,108.4$.

## 10-(4-bromophenyl)-10H-phenothiazine (SBr)



Under a nitrogen atmosphere, phenothiazine ( $2.00 \mathrm{~g}, 10.90 \mathrm{mmol}$ ), 4-bromoiodobenzene ( 2.80 g , 10.00 mmol ) and Copper powder ( $3.00 \mathrm{~g}, 23.40 \mathrm{mmol}$ ) were heated to $160^{\circ} \mathrm{C}$ in $35 \mathrm{~mL} \mathrm{N,N-}$ dimethylformamide for 48 h . After cooling to room temperature, the mixture was washed with water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}(5: 1, \mathrm{v}: \mathrm{v})$ as the eluent. A white solid of the title compound was obtained ( 2.50 g , $66 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 7.70(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.25(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.04$ (dd, $J=7.3,1.8 \mathrm{~Hz}, 2 \mathrm{H}), 6.86(\mathrm{~m}, 4 \mathrm{H}), 6.24(\mathrm{dd}, J=8.0,1.4 \mathrm{~Hz}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 144.2,143.8,140.9,133.8,131.9,129.7,129.5,122.5,122.2,121.2,117.5,116.2$

## SB



Under a nitrogen atmosphere, bis-(pinacolato)diboro ( $1.10 \mathrm{~g}, 4.33 \mathrm{mmol}$ ), $N$-(4-bromophenyl)phenothiazine ( $1.00 \mathrm{~g}, 2.833 \mathrm{mmol}$ ), $\mathrm{KOAc}(0.931 \mathrm{~g}, 9.482 \mathrm{mmol})$ and $\mathrm{Pd}(\mathrm{dppf}) \mathrm{Cl}_{2}(0.178 \mathrm{~g}, 0.632$ mmol) were heated to $110^{\circ} \mathrm{C}$ in 25 mL degassed 1,4-dioxane for 12 h . After cooling to room temperature, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ / petroleum ether (1:3, v:v) as the eluent. White solid of the title compound was obtained ( $0.62 \mathrm{~g}, 53 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d_{6}$ ): $\delta(\mathrm{ppm}) 7.88$ (d, $\left.J=8.3 \mathrm{~Hz}, 2 \mathrm{H}\right), 7.35(\mathrm{~d}, J$ $=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.18(\mathrm{dd}, J=7.5,1.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{ddd}, J=7.9,7.7,1.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.95(\mathrm{ddd}, J=7.5$, $7.5,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.41(\mathrm{dd}, J=8.1,1.3 \mathrm{~Hz}, 2 \mathrm{H}), 1.33(\mathrm{~s}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ) $\delta$ (ppm) 144.5, 143.4, 137.3, 127.9, 127.6, 123.9, 122.4, 118.6, 84.3, 25.2.

## 10-(4-bromophenyl)-10H-phenoxazine ( OBr )



Under a nitrogen atmosphere, phenoxazine ( $2.00 \mathrm{~g}, 10.00 \mathrm{mmol}$ ), 4-bromoiodobenzene ( 2.80 g , 10.00 mmol ) and Copper powder ( $3.00 \mathrm{~g}, 23.40 \mathrm{mmol}$ ) were heated to $160^{\circ} \mathrm{C}$ in $35 \mathrm{~mL} \mathrm{~N}, \mathrm{~N}$ dimethylformamide for 48 h . After cooling to room temperature, the mixture was washed with water and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using petroleum ether $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}(5: 1, \mathrm{v}: \mathrm{v})$ as the eluent. White solid of the title compound was obtained ( $2.30 \mathrm{~g}, 68 \%$ ). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 5.91(\mathrm{dd}, J=7.8,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.57-6.70(\mathrm{~m}, 6 \mathrm{H}), 7.23(\mathrm{~d}, J$ $=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.72(\mathrm{~d}, J=8.6 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta(\mathrm{ppm}) 113.3,115.7$, 121.7, $122.5,123.4,132.9,134.1,134.6,138.2,144.0$.

OB


Under a nitrogen atmosphere, bis-(pinacolato)diboro ( $1.10 \mathrm{~g}, 4.33 \mathrm{mmol}$ ), N -(4-Bromophenyl)Phenoxazine ( $1.00 \mathrm{~g}, 2.97 \mathrm{mmol}$ ), $\mathrm{KOAc}(0.93 \mathrm{~g}, 9.48 \mathrm{mmol})$ and $\operatorname{Pd}(\mathrm{dppf}) \mathrm{Cl}_{2}(0.18 \mathrm{~g}, 0.63 \mathrm{mmol})$ were heated to $110^{\circ} \mathrm{C}$ in 25 mL degassed 1,4-dioxane for 12 h . After cooling to room temperature, the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated under reduced pressure, then purified by column chromatography over silica gel using $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ / petroleum ether (1:5, v:v) as the eluent. White solid of the title compound was obtained $(0.65 \mathrm{~g}, 55 \%) .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 8.02(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{~d}, J=8.0 \mathrm{~Hz}$,
$2 \mathrm{H}), 6.68(\mathrm{dd}, J=7.8 \mathrm{~Hz}, 1.4 \mathrm{~Hz}, 2 \mathrm{H}), 6.63(\mathrm{t}, J=7.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.56(\mathrm{td}, J=7.6 \mathrm{~Hz}, 1.5 \mathrm{~Hz}, 2 \mathrm{H})$, $5.91(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 2 \mathrm{H}), 1.38(\mathrm{~s}, 12 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta(\mathrm{ppm}) 145.9,139.2,134.5$, 134.3, 125.9, 122.8, 122.7, 139.2, 134.5, 88.1, 24.7.


Figure S1. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for OBA-O.



Figure S2. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for OBA-S.


ボ

$\begin{array}{llllllllllllllllllllll}175 & 170 & 165 & 160 & 155 & 150 & 145 & 140 & 135 & 130 & 125 & 120 & 115 & 110 & 105 & 100 & 95 & 90 & 85 & 80 & & \mathrm{ppm}\end{array}$

Figure S3. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for $\mathbf{O B A}-\mathbf{B r O}$.



$\begin{array}{lllllllllllllllllllllllllllllllllll}180 & 175 & 170 & 165 & 160 & 155 & 150 & 145 & 140 & 135 & 130 & 125 & 120 & 115 & 110 & 105 & 100 & 95 & 90 & 85 & 80 & \mathrm{ppm}\end{array}$

Figure S4. ${ }^{1} \mathbf{H}$ - and ${ }^{13} \mathbf{C}$-NMR spectra for $\mathbf{O B A}-\mathbf{B r S}$.


Figure S5. TGA curves of these OBA-based emitters.


Figure S6. Cyclic curves of these asymmetric OBA-based emitters measured in $\mathrm{CH}_{3} \mathrm{CN}$ and ferrocene/ferrocenium ( $\mathrm{Fc} / \mathrm{Fc}^{+}$) couple at room temperature. (a) OBA-O, (b) OBA-S, (c) OBA-BrO, (d) OBA-BrS.


Figure S7. Current density-voltage-luminance ( $J-V-L$ ) curves for the devices except the optimized ones.


Figure S8. Relationship between EL efficiencies and luminance for the devices except the optimized ones. (a) Device A1, (b) Device B1, (c) Device C1 and (d) Device D1.


Figure S9. Relationship between EL efficiencies and luminance for the devices except the optimized ones. (a) Device A3, (b) Device B3, (c) Device C3 and (d) Device D3.

