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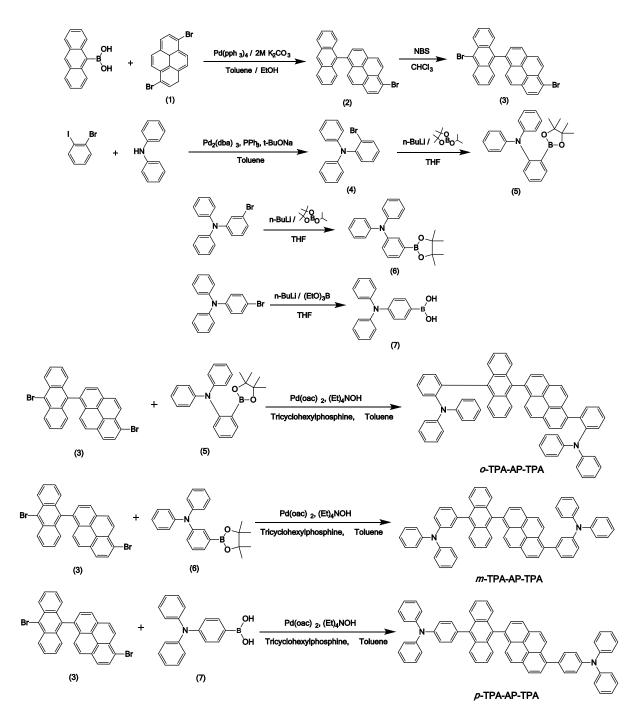
## **Supporting information**

## Highly Efficient Dual-Core Derivatives with EQEs as high as 8.38% at High Brightness for OLED Blue Emitters

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## **General information**

Reagents and solvents were purchased as reagent grade and used without further purification. Analytical TLC was carried out on Merck 60 F254 silica gel plate, and column chromatography was performed on Merck 60 silica gel (230-400 mesh). The <sup>1</sup>H-NMR spectra were recorded on Bruker Avance 300 spectrometers. The FAB+-mass and El+-spectra were recorded on a JMS-600W, JMS-700, 6890 Series and Flash1112, Flash2000. The optical UV-Vis absorption spectra were obtained using a Lambda 1050 UV/Vis/NIR spectrometer (Perkin Elmer). A Perkin-Elmer luminescence spectrometer LS55 (Xenon flash tube) was used to perform PL spectroscopy. The transient PL was measured with a Quantaurus-Tau fluorescence life time measurement system. The glass transition temperatures (T<sub>g</sub>) of the compounds were DSC under a nitrogen atmosphere using a DSC 4000 (Perkin Elmer). T<sub>d</sub> of the compounds were measured with TGA using a TGA4000 (Perkin Elmer). The HOMO energy levels were determined with ultraviolet photoelectron yield spectroscopy (Riken Keiki AC-2). The LUMO energy levels were derived from the HOMO energy levels and the band gaps. For the EL devices, all organic layers were deposited under 10<sup>-6</sup> torr, with a rate of deposition of 1 Å/s to give an emitting area of 4 mm<sup>2</sup>. The LiF and aluminum layers were continuously deposited under the same vacuum conditions. The current-voltage-luminance (I-V-L) characteristics of the fabricated EL devices were obtained with a Keithley 2400 electrometer. Light intensities were obtained with a Minolta CS-1000A. To calibrate the EQE values considering the angular dependence, emission angular distributions were also measured. The operational stabilities of the devices were measured under encapsulation in a glovebox.





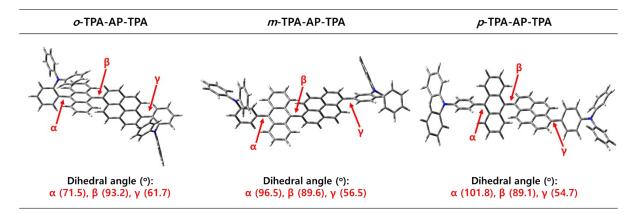
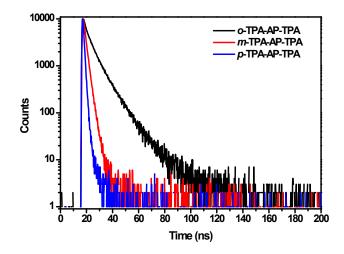


Fig. S2 Optimization structure and dihedral angle of synthesized compounds



**Fig. S3** Time-resolved PL of the synthesized compounds (film states in which the synthesized material in PMMA was dispersed at 1wt%).

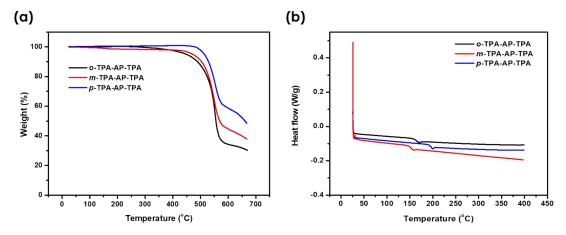


Fig. S4 (a) TGA and (b) DSC curves of synthesized compounds

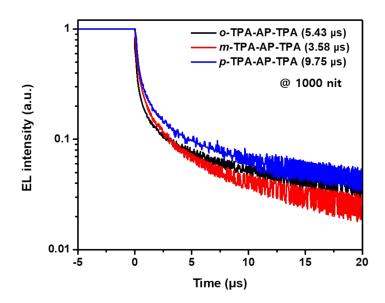


Fig. S5 Transient electroluminescence (EL) responses of an ITO/2-TNATA (60 nm)/NPB (15 nm)/ $\alpha$ , $\beta$ -ADN: synthesized material 4% (30 nm)/TPBi (30 nm)/LiF (1 nm)/AI (200 nm)

 Table S1 Photophysical properties of synthesized compounds.

Compounds	$\Phi_{\text{F}}{}^{\text{a}}$	τ <sub>F</sub> (ns) <sup>b</sup>	k <sub>rad</sub> <sup>c</sup> x 10 <sup>8</sup> /s	k <sub>nr</sub> <sup>c</sup> x 10 <sup>7</sup> /s	k <sub>rad</sub> /k <sub>n</sub> r
o-TPA-AP-TPA	0.45	6.78	0.66	8.14	0.81
<i>m</i> -TPA-AP- TPA	0.74	1.89	3.91	13.8	2.83
<i>p</i> -TPA-AP-TPA	0.85	0.78	10.9	19.3	5.64

<sup>a,b</sup> Film states in which the synthesized material in PMMA was dispersed at 1wt%, <sup>c</sup> Values of  $k_{rad}$  and  $k_{nr}$  were calculated according to the equations  $k_{rad} = \Phi_F/\tau_F$  and  $k_{nr} = (1/\tau_F) - k_{rad}$ .

 Table S2 Absorption frequencies and oscillator strengths for the synthesized compounds calculated using TD-DFT (B3LYP/6-311G(d) method).

Compounds	λ <sub>abs</sub> (nm) (oscillator strength, f)	Characteristic transition	Distribution of character <sup>a</sup>	Contribution (%) <sup>b</sup>
o-TPA-AP-TPA	419 (0.136)	HOMO → LUMO+1	P + TPA → P	88
		$HOMO \rightarrow LUMO$	$P + TPA \rightarrow A$	5
	406 (0.223)	HOMO-2 → LUMO	$A + TPA \rightarrow A$	68
		HOMO−2 $\rightarrow$ LUMO+1	$A + TPA \rightarrow P$	10
		HOMO−1 $\rightarrow$ LUMO	$A + TPA \rightarrow A$	7
<i>m</i> -TPA-AP-TPA	399 (0.370)	HOMO-2 → LUMO	A  A	93

	358 (0.503)	HOMO-2 → LUMO	$A \to A$	4
<i>p</i> -TPA-AP-TPA	424 (0.204)	HOMO-1 → LUMO	$A + TPA \rightarrow A$	93
		HOMO $\rightarrow$ LUMO+1	$P + TPA \rightarrow P$	6
	414 (0.526)	HOMO → LUMO+1	P + TPA → P	83
		HOMO-2 → LUMO	$A + TPA \rightarrow A$	9
		HOMO−1 $\rightarrow$ LUMO	$A + TPA \rightarrow A$	4
	396 (0.117)	HOMO−2 $\rightarrow$ LUMO	$A + TPA \rightarrow A$	86
		HOMO → LUMO+1	P + TPA → P	7

<sup>a</sup> A: anthracene, P: pyrene, TPA: triphenylamine. <sup>b</sup> Where the sums of the contributions are less than 100%, the remaining contributions are from various minor transitions that are not listed.