

## Supporting information

### Investigation on voltage loss in organic triplet photovoltaic devices based on Ir complexes

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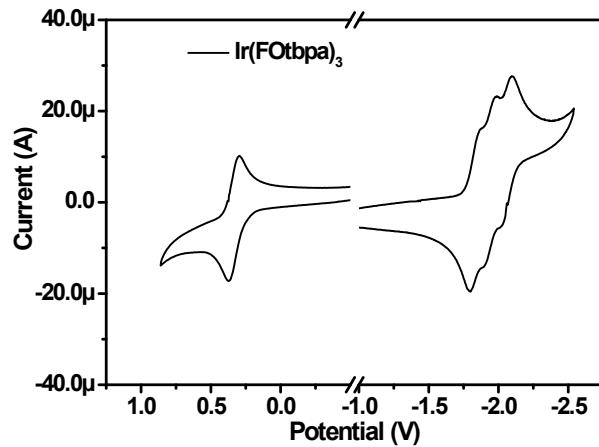
**Table S1** Photophysical Characteristics of Ir(Ftbpa)<sub>3</sub> and Ir(FOtbp)<sub>3</sub>.

	Degassed CH <sub>2</sub> Cl <sub>2</sub> solution <sup>a</sup>					Neat film			
	$\lambda_{\text{abs}}$ [nm] [ $\varepsilon$ ( $\times 10^4$ L mol <sup>-1</sup> cm <sup>-1</sup> )]	$\lambda_{\text{em}}$ [nm]	$\Phi_{\text{PL}}^{\text{b}}$ [%]	$\tau_p$ [ns]	$K_r/K_{\text{nr}}$ [ $\times 10^5$ s <sup>-1</sup> ]	$\lambda_{\text{em}}$ [nm]	$\Phi_{\text{PL}}^{\text{b}}$ [%]	$\tau_p$ [ns]	$K_r/K_{\text{nr}}$ [ $\times 10^5$ s <sup>-1</sup> ]
Ir(Ftbpa) <sub>3</sub>	263 (12.3), 335 (3.5), 355 (3.6), 374 (3.3), 490 (2.3), 550 (1.8)	765, 820(s)	14.7	730	2.0/11.7	784	2.6	19	13.7/512.6
Ir(FOtbp) <sub>3</sub>	260 (10.8), 326 (3.0), 349 (2.9), 349 (2.8), 383 (2.1), 472 (1.9), 537 (1.1)	767, 827(s)	10.8	489	2.2/18.2	780	2.4	50	4.9/199.2

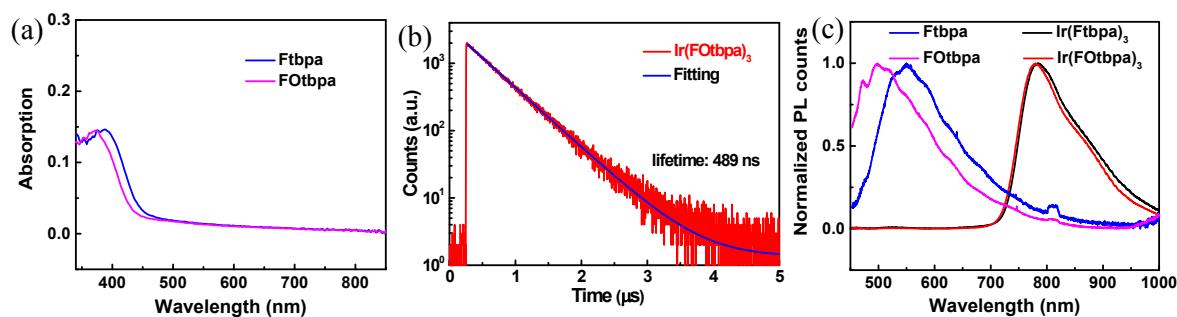
<sup>a</sup> The concentration of the solution is  $2 \times 10^{-5}$  M.  $\varepsilon$  denotes the molar extinction coefficients. <sup>b</sup>  $\Phi_{\text{PL}}$  denotes the PLQY.

**Table S2** Summary of photovoltaic parameters of T-OPVs based on Ftbpa and FOtbp ligands.

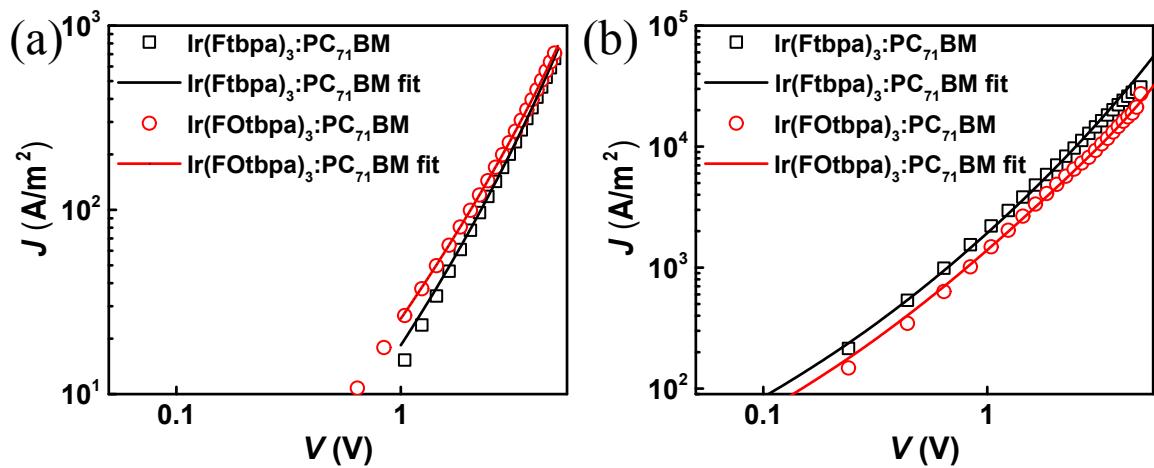
Active layer	$V_{\text{oc}}$ (V)	$J_{\text{sc}}$ (mA/cm <sup>2</sup> )	FF	PCE (%)
Ftbpa:PC <sub>71</sub> BM	0.27	0.02	0.24	0.001
FOtbp:PC <sub>71</sub> BM	0.45	0.04	0.37	0.007



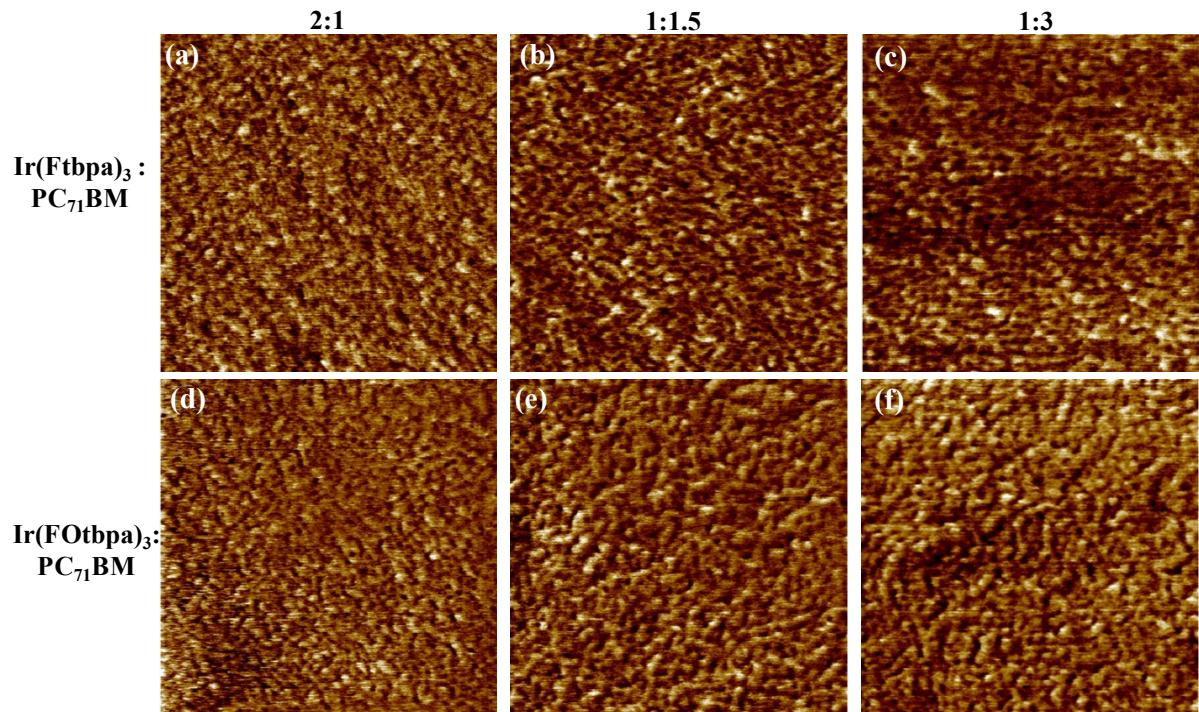
**Fig. S1** CV of and  $\text{Ir}(\text{FOtbpda})_3$  recorded versus  $\text{Fc}^+/\text{Fc}$  in anhydrous DMF solutions at 298 K under a  $\text{N}_2$  atmosphere;



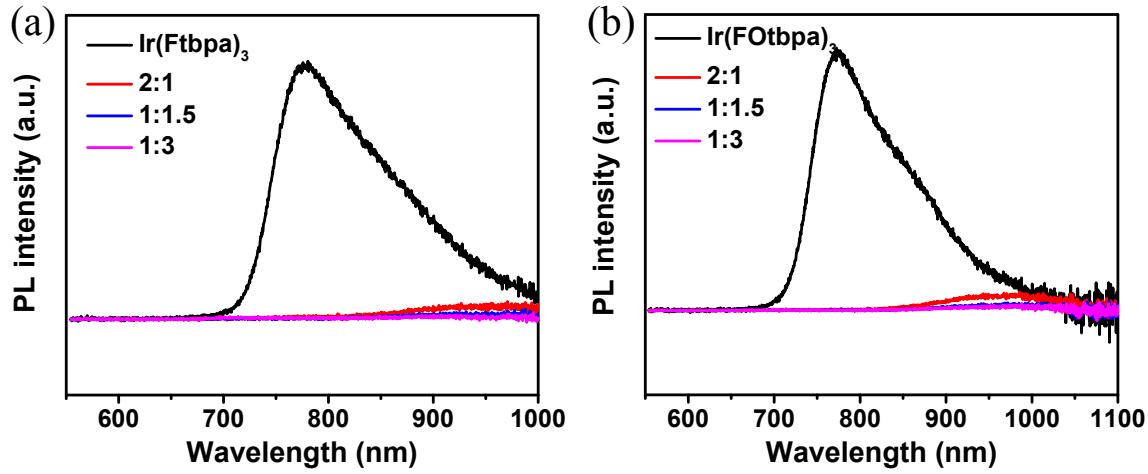
**Fig. S2** (a) Absorption spectra of Ftbpa and FOtbpda films; (b) Transient PL decay curves of  $\text{Ir}(\text{FOtbpda})_3$  in degassed  $\text{CH}_2\text{Cl}_2$  solution; (c) Normalized PL spectra of  $\text{Ir}(\text{Ftbpa})_3$ ,  $\text{Ir}(\text{FOtbpda})_3$ , Ftbpa, and FOtbpda films; The films were excited by a 405 nm laser.



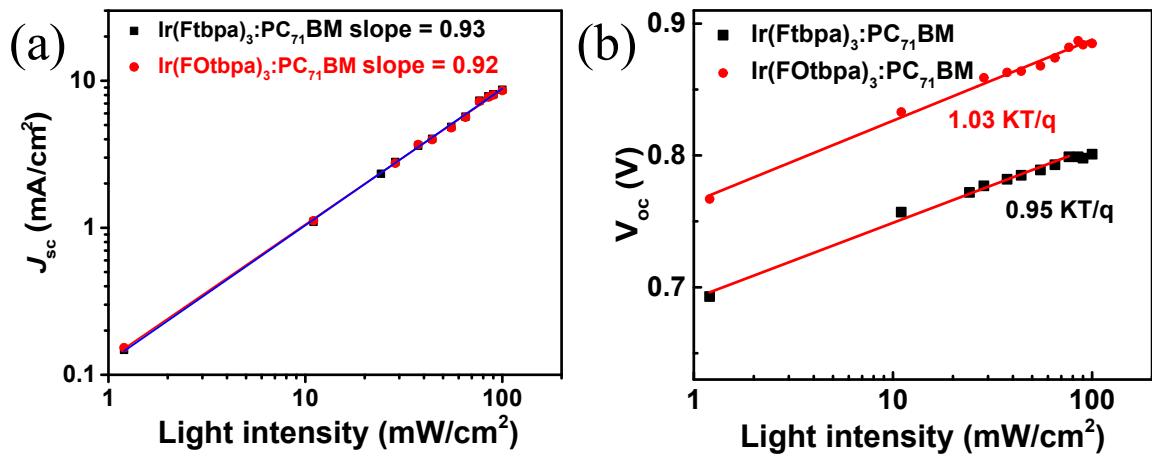
**Fig. S3**  $J$ - $V$  curves for hole (a) and electron (b) only devices based on Ir(Ftbpa)<sub>3</sub>:PC<sub>71</sub>BM and Ir(FOtbp)<sub>3</sub>:PC<sub>71</sub>BM blends with a weight ratio of 1:1.5.



**Fig. S4** AFM ( $2 \mu\text{m} \times 2 \mu\text{m}$ ) phase images of Ir(Ftbpa)<sub>3</sub>:PC<sub>71</sub>BM blends with weight ratio of 2:1 (a), 1:1.5 (b), 1:3 (c) and Ir(FOtbp)<sub>3</sub>:PC<sub>71</sub>BM blends with weight ratio of 2:1 (d), 1:1.5 (e), 1:3 (f).



**Fig. S5** (a) PL spectra of the pristine Ir(Ftbpa)<sub>3</sub> and blended films with different weight ratios. (b) PL spectra of the pristine Ir(FOtbp)<sub>3</sub> and blended films with different weight ratios. The PL intensities are corrected by their absorptions at the excitation wavelength (532 nm).



**Fig. S6** (a)  $J_{sc}$  and (b)  $V_{oc}$  versus light intensity for T-OPVs based on Ir(Ftbpa)<sub>3</sub>:PC<sub>71</sub>BM blends (1:1.5) and Ir(FOtbp)<sub>3</sub>:PC<sub>71</sub>BM blends (1:1.5).